Response to Reviewer 1 (Prof. Dr. Werner Eugster)

Dear Prof. Dr. Eugster,

Thank you for your detailed remarks and constructive comments which are all very helpful for improving the manuscript. We are glad that you find our work interesting and innovative. We are particularly grateful for the substantial work to understand all details, which will help us clarify the text and improve the presentation. As noted, our aim is to provide a detailed fully open-source description of the system that can contribute to more extensive data collection and to inspire technical improvements by the broader community. Our response to your specific review comments are given below.

2.1 Major issues

- 2.1 Referee comment (RC)1: The authors use three low-cost sensors, all from Figaro Inc.: the TGS2611-E00, the NGM2611-E13 (which uses the same TGS2611-E00), and the Panterra from Neodym (I assume), which in the version I used had a percursor version of the TGS2611 or so built in. I think the authors should more clearly specify (a) what sensor the Panterra uses (and provide the company names of all sensors), (b) clarify that these are (most likely) all the same sensors in different configurations (as it reads now the reader could be getting the impression that three different sensor types were tested, which is not the case)
- 2.1 Author response (AR)1: It is correct that all three sensors investigated are rather similar sensors from Figaro. The TGS2611-E00 and NGM2611-E13 differ in that the latter is attached to a small board with a potentiometer used for a crude factory calibration at 5000 ppm and with a 5-pin connector for easy plug-in to a system with a corresponding connector (cost efficient when handling many systems). The Panterra sensor is built around the Figaro TGS2610 sensor (order code PN-SM-GMT-A040A-W20A-05-R0- S0-E1-X0-I2-P0-L2-J1-Z0) from Panterra Neodym Technologies, Canada. This Panterra sensor, which was recommended after discussion with a Neodym technician, was the same as used in our study in Duc et al., 2013.

We selected sensors based on specifications and discussions with company representatives regarding several criteria, including methane specificity, a sensitivity that was potentially high enough for our applications, price and power consumption (more on this below), and believed that configuration and signal processing also can make a difference so testing different configurations was of interest to us. The sensor details were too spread out in our manuscript which we now realized was unclear.

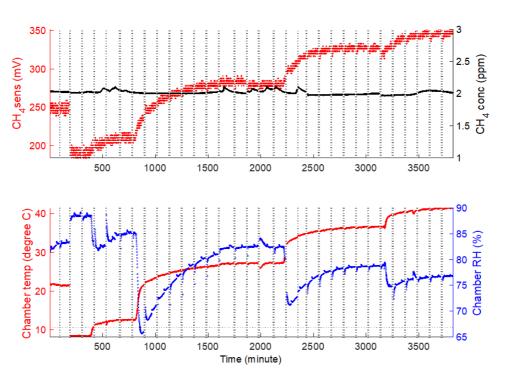
- 2.1 Author changes in manuscript (ACM)1: The sensor details and motives behind the sensor selection will be provided in the same paragraph for all sensors together and early in the text.
- $2.1\ RC2$: I am concerned about the low voltages that the TGS2611-E00 gives, ca. 18–35 mV according to Fig. 3. The TGS2600 that I use delivers 200-400 mV for ambient conditions, and when I look at the specifications it appears that the manufacturer considers the TGS2611-E00 to be useful in the range of 300–10,000 ppm CH₄ which is way above ambient range (the TGS2600 is shown with a sensitivity to CH₄ in the range 1–100 ppm CH₄ that's the reason we selected this one for near ambient measurements in Eugster & Kling, 2012). Now there are some other publications that show that the TGS2611-E00 is actually sensitive also to near ambient conditions, but I am not yet convinced that this is the best choice for your application given the low CH₄ concentrations well below the range

indicated on the technical specification sheet of the manufacturer. Some more critical discussion on the sensor selection would be required in my view.

2.1 AR2: This question highlights and revealed an error in the script for plotting the scale of Fig. 3A. The corrected Figure 3 is pasted below. The background level of the NGM2611-E13 sensor should be in the range of few hundred millivolts, in line with your expectations.

We first intended to try the TGS2600 sensor from your study with proven high sensitivity. When contacting Figaro, they recommended we try the TGS2611-E00 and NGM2611-E13 because we wanted a high specificity for CH₄ and because it have higher sensitivity than specified (the Product Information notes also indicate that the sensors are far from the detection limit at the low end of the tested range; 300 ppm). We decided to follow the Figaro technician suggestions as a start and simply kept working with them because they gave an adequate response for our applications (otherwise we would have tried the TGS2600). It should be noted that previous attempts to measure absolute ambient air levels is much more demanding in terms of sensitivity than our application, which is focusing on relative change, often with a doubling or more in levels over 1-2 hours. When assessing relative changes over time in a closed system it is also important to minimize interferences of other gases that may also change over time – further explaining our sensor choice as a trade-off between sensitivity and specificity.

2.1 ACM2: The error in the scale of Figure 3 have been corrected and clarifications on sensor details and motives behind the sensor selection will be made (see ACM1 above).



2.1 RC3: The authors only sample data every minute, which I find utterly coarse. They may have a reason for this, but in my own tests with the TGS2600 a one-minute measurement interval in combination with a 5-minute data rejection after chamber deployment (page 6, line 26; this information should actually have been given in the Methods section already, because this is an essential flaw in the system in my view) I would have lost all the information relevant to chamber

fluxes (see graph below and description of unpublished internal example graph from my experiments at Toolik Lake, Alaska, USA). Thus, the authors should more precisely describe their method and critically discuss such shortcomings to help others to do better.

2.1 AR3: One goal of our project was to develop an active wireless sensor network in which data from many flux chambers (called clients) are sent by Xbee radio transmitter modules and recorded on a small low-cost Raspberry Pi computer on the lakeshore. The sampling data rate, so far, is constrained to maintain the digimesh network working with minimum labor effort. In bad weather conditions, radio communication is easily broken, therefore it can take some minutes to re-establish the communication depending on the distance of the clients. During this offline period, the limited memory buffer of these data loggers does not allow us to sample very often. A second reason for the one-minute measurement internal is driven by coordinated logging and transmitting data from a CO₂/RH/Temp sensor having a 25 second measurement cycle – which also presently restricts the measurement frequency.

The rejection of data from the initial 5-minutes of measurements, is because the CH_4 sensor signal can be affected by temperature and relative humidity, and when focusing on relative change it is again important to minimize the influence of other confounding factors that may change over time. This rejection period is mainly to wait for temperature and relative humidity in the chamber to stabilize after chamber closure on the water surface to ensure that the CH_4 sensor response reflects CH_4 and not changing humidity and or temperature.

From your attached graph, it appears that the temperature and relative humidity in your chamber reach stable equilibrium quite quickly. In our case, we use the temperature and relative humidity measured from an integrated sensor onboard the CO₂ sensor which is measured every minute along with CO₂ concentration. Of course, an obvious data interpretation improvement would be to modify the length of the initial period during which data is not used to the actual time it takes to reach stable enough relative humidity and temperature, instead of having the static 5-minute period used here for simplicity.

In spite of our low measurement frequency, our results show that the system is still able to capture relative changes adequately. Of course, the situation is very different in applications aiming for accurate absolute levels in ambient air, and for such applications a high measurement frequency is of course more important to cancel out sensor noise in data processing than in our flux chamber application.

2.1 ACM3: The explanations and motivations provided above in AR3 will be clarified in the manuscript in a discussion paragraph devoted to measurement frequency.

RC4: The ebullition (bubble) counter is quite interesting, but with a bubble volume of 3-4 mL required to actually leave a defensible signal, this does not yet seem to be an optimum choice. Here, a reference to and comparison with the (commercial) system of Andreas Mäck (doi:10.5194/bg-11-2925-2014) would be helpful. Since the Varadharajan et al. (2010) reference (pages 2,3,8, lines 30,22,21) is not listed in the References, I could not convince myself that this bubble counter system is really thoroughly tested and reliable. In the discussion you only say "For a long term solution, the recent study using optical sensors in an open path funnel (Delwiche and Hemond, 2017) 30 suggests an alternative and interesting design for ebullition studies, which could be combined with the present sensor approach to also quantify CH₄ content in the bubbles." – thus does this mean that you are satisfied with the performance for short-term investigations? I am not really convinced and would appreciate a somewhat clearer statement what your recommendation is for studies that are shorter than a "long-term solution".

2.1 AR4: We apologize for the missing Varadharajan et al. (2010) reference and will correct this. We were not aware of the Meack et al 2014 paper and will integrate this to the manuscript.

Most previous bubble counter systems are based on bubble volume quantification by differential pressure sensors (e.g. Varadharajan et al. 2010; Maeck et al 2014) or optical sensors (Delwiche and Hemond, 2017). The detection limit of the differential pressure measurement, in our case corresponding to 3-4 ml gas, depends on the shape of the cylinder where the bubbles accumulate. Therefore, the longer and narrower a cylinder, the lower the detection limit. In turn, this leads to a trade-off, where the more sensitive systems become too tall for deployment in shallow waters, which often have proportionally higher ebullition rates. Our detection limit was chosen to allow deployment in shallow water. The shorter funnel of Delwiche and Hemond, 2017 (based on the optical sensor) cobined with our system could solve the challenge we face deploying in shallow waters. We apologize that this statement was not clear in the manuscript.

- 2.1 ACM4: The response above including explanation of trade-offs and choices will be clarified in the revised manuscript. We will also more clearly relate to similar studies and have added proper references to Varadharajan et al 2010 and Maeck et al 2014.
- 2.1 RC5: Your regressions (I assume you use ordinary least-squares regressions) are not correct from a statistical viewpoint (Figs 4 and 5): you must reverse the dependent and the independent variable: you want to find out how to use the signal to compute the true concentration using your regressions, not the other way around (i.e. to predict the signal based on knowledge of the concentration that's what your regressions show).
- 2.1 AR5: Figure 4 is to be seen as a calibration curve where the CH₄ concentrations are measured by reference instruments (GC-FID or an LGR greenhouse gas analyzer) and thereby represents the independent (x-axis) data. The CH₄ sensor response in the calibration case becomes the dependent signal. We agree that in a case when wanting to calculate concentrations from sensor signals, reversing the dependent and independent variables, would make more sense.

Figure 5 is just showing the sensor signal over time and simply provide temporal information of multiple variables (time on x-axis) and no regressions are made directly in this graph.

- 2.1 AMC5: A clarification that Figure 4 represents a calibration curve will be added.
- 2.1 RC6: I always use a fan in chambers, you don't. I understand that this corresponds to some static chambers that people use with syringe sampling, but in your case I am concerned that without a fan to mix the volume of the chamber the CO₂ (which is heavier than air) starts to accumulate above the water surface, and then a steep gradient creeps upwards where I expect your sensors; hence this linear increase in Fig. 5. Contrastingly, CH₄ which is lighter than air, quickly would accumulate under the top of the chamber, and hence probably the curvature although I would have expected that the CO₂ saturation should occur earlier than the CH₄ saturation in such a chamber. Please comment on this and justify why not to mix the air inside the chamber to ascertain representative concentration measurements inside the flux chamber.

2.1 AR6: Our floating chamber is light-weight and freely moves up and down with the water. There is also some wind induced drifting around the separate anchored float. The sensors in the chamber are about 10 cm above water surface. We believe that the natural movement of chamber caused by winds and waves mixes the volume in the chamber.

We are aware of the vital need of air mixing in chambers holding vegetation, but for open water cases we see a risk that adding a fan will create an unknown bias from the added fan-induced turbulence and weight.

Some chambers designed differently have reported biased fluxes, while the chamber design used here without fans have repeatedly shown negligible bias compared to non-invasive techniques under variable conditions ranging from coastal water, small lakes and streams (Cole at al 2010; Gålfalk et al 2013; Lorke et al. 2015). Hence, we preferred to keep this tested chamber design.

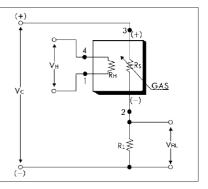
- 2.1 ACM6: The choice of the chamber design and evidence in support of it will be clarified.
- 2.1 RC7: According to the manufacturers information the sensor resistance decreases as the CH₄ concentration increases, thus in principle the voltage you measure should decrease not increase with increasing CH₄ but your measurements in Fig. 4 show the exact opposite of what one would expect from the manufacturer information. Do you have an explanation for this? I must admit that on short timescales I see the same (see Figure inserted below), but on longer timescales I see what I would expect from the manufacturer's data sheet. Our procedure suggested by Eugster & Kling (2012) solve this issue with the calibration after linearisation the use of a high and low calibration point simply reverses the sign if the response is of the kind that you show in Fig. 4. If you have an explanation why the TGS2611-E00 has increasing voltage with increasing CH₄ concentration then this would be a helpful insight for the reader. If you don't have an explanation, maybe you have an opinion?
- 2.1 AR7: From our understanding, the NGM2611-E00 and the TGS 2611 sensor in a circuit essentially operates as a voltage divider. A figure and associated text from the Product Information note of TGS 2611

(https://www.figaro.co.jp/en/product/docs/tgs2611_product_information_rev02.pdf) is provided below. The resistance of the CH₄ sensor is called Rs which has a resistance value that decreases as CH₄ concentration increases, load resistor R_L has constant value (about 5 k Ω). This circuit is fed by a constant circuit voltage V_C (5V), and the current equals to ratio of 5/(Rs+R_L). Hence, if Rs decreases, the current will increase. As a result, output voltage (V_{RL}) which equals to R_L*5/(Rs+R_L) will increase. We have not studied the circuit details for other sensors so perhaps interpretations of the output voltage differ among sensors which may be the reason for this comment. We will double check our understanding and if it seems OK we will try to clarify this in the supplementary material.

Basic Measuring Circuit:

The sensor requires two voltage inputs: heater voltage (VH) and circuit voltage (VC). The heater voltage (VH) is applied to the integrated heater in order to maintain the sensing element at a specific temperature which is optimal for sensing. Circuit voltage (VC) is applied to allow measurement of voltage (VRL) across a load resistor (RL) which is connected in series with the sensor.

A common power supply circuit can be used for both Vc and VH to fulfill the sensor's electrical requirements. The value of the load resistor (RL) should be chosen to optimize the alarm threshold value, keeping power dissipation (Ps) of the semiconductor below a limit of 15mW. Power dissipation (Ps) will be highest when the value of Rs is equal to RL on exposure to gas.



2.1 ACM7: The above explanation will be clarified in the supplementary material.

2.2 Minor Issues

2.2 RC1: p2/33: "The eddy covariance (EC) technique is increasingly used for long-term monitoring, but it is expensive in terms of equipment and is still being evaluated for aquatic environments." – what do you mean with this statement? I don't consider this to be correct, the method is in use beyond evaluation. Please provide some references and reword the second part. For example, we have authored a couple papers and also written a chapter in the book Eddy Covariance: A Practical Guide to Measurement and Data Analysis (chapter on lakes: doi:10.1007/978-94-007-2351-1_15). Thus, the method is established (at least better than your chambers, to be more direct) – but I agree that it is costly and I agree that such low-cost sensors are important.

2.2 AR1: We did not intend to unfairly describe EC measurements. Given the discussions on issues such as lateral fluxes (land/sea breeze effects), wind shadow zones around forested lake shores, other irregularities in wind patterns over lakes, challenges interpreting footprint locations and shape for small lakes, and other discussions on suitable equipment (e.g. open or closed path gas analyzers), we simply had the impression that method evaluation and development was still ongoing. Because the referee comment clearly signals we were wrong we will remove this statement.

2.2 ACM1: We will reword the sentence to: "The eddy covariance (EC) technique is increasingly used for long-term monitoring of terrestrial and lake-dominated landscapes, but it is expensive in terms of equipment."

2.2 RC2: p2/45: "The CH4 sensor tested here is a Taguchi Gas Sensor (TGS) (Figaro Engineering Inc., Osaka, Japan). It is a high sensitivity CH₄ gas sensor. . . ": I completely disagree, it is a low-sensitivity sensor which (in the version you use) only has a manufacturer specified lower measurement range of 300 ppm CH₄! I already realize that our more cautious wordings about the TGS2600 (which has a higher sensitivity than the TGS2611-E00) is ignored by some others, which can lead to frustration. Be clear that this is experimental work trying to squeeze the tiny bit of information out of a sensor that is not made for ambient concentrations – but I agree that it has some value for such measurements.

2.2 AR1: We meant that the sensor is more sensitive than several other CH₄ sensors in the same prize class, but we agree with the reviewer and now see that the statement can be interpreted in misleading ways.

2.2 ACM2: We will remove this statement (See also AR2.1, AR2)

- 2.2 RC3. You never specified which pressure sensor you used, thus it is unclear to me why you did not use an I2C sensor, there plenty of those on the market. What is the special advantage of your pressure sensor that requires an AD620 amplifier to be useful? This remains obscure to the reader.
- 2.2 AR3 This manuscript describes work that is a follow up from a previous study on an automatic system to measure greenhouse gases from aquatic environments (Duc et al., 2013). In our previous electronic circuit, we used an AD620 amplifier for the pressure sensor (26PCDFA6G, Honeywell, Sensing and Control, Canada) to measure atmospheric pressure. After reading the work of Varadharajan et al. (2010), we adapted our system by simply changing one external resistor to get the proper gain factor to use with our pressure sensor (26PCAFA6D).
- 2.2 ACM3: We will add the sensor information and clarify our reasons for choosing this sensor to the manuscript.
- 2.2 RC4: Eq. 1 should use SI units or at least the same units of the same physical quantity and not include obscure conversion factors. Thus, you must decide whether your time variables should be in hours or in seconds (the primary SI unit) or minutes, please no mixtures.
- 2.2 AR4: The flux time unit was in hours representing a relevant time unit given the accumulation time of the chamber. As this study focuses on evaluating the sensor response to the change in mixing ratio of CH_4 and CO_2 gases in the chamber in which the sensor signal is recorded every minute, we have decided to present $\Delta C/\Delta t$ (ppmv min⁻¹) to avoid applying a conversion factor.
- 2.2 ACM4: The above explanation will be provided in the manuscript.
- 2.2 RC5, p5/25: check your instrument information, most likely this was an LGR FGGA (not a DLT-100, which as I remember is a CH₄-only instrument) that measures CH₄ and CO₂.
- 2.2 AR5: The reviewer is correct instrument we used is an early benchtop version of the FGGA analyzer that have the capacity to measure CH₄, CO₂ and H₂O. It has DLT-100 printed on the cover leading to this confusion.
- 2.2 ACM5: We will edit the manuscript to include "a FGGA with capacity to measure CH₄, CO₂ and H₂O."
- 2.2 RC6, p6/8: you did not specify what your "baseline noise" actually is. Is it the square root of the variance or the noise baseline derived from an Allan variance plot, or anything else? Some more details in the Methods section would be really helpful.
- 2.2 AR6: Our baseline noise is square root of the variance.
- 2.2 ACM6: We will add this clarification to the manuscript.
- 2.2 RC7 p6/13: "The pressure in the trap was affected by air temperature, especially the diel temperature cycle." this sounds like an error, pressure is a physical entity that is independent of temperature, thus this must be wrong. What I can imagine is that you mean that your pressure sensor signal (but not the pressure itself) depended on air temperature. Please correct.

- 2.2 AR7: We did mean to refer to the pressure sensor signal.
- 2.2 ACM7: We will correct this in our manuscript to read "The pressure sensor signal measured in the trap was affected...."
- 2.2 RC8 p8/39: "However, since the sensor is not calibrated for very high concentrations, we could not determine the flux rate observed during these events." I completely disagree, at least for CH₄ (you do not really reveal any necessary details on the CO₂ measurements . . .): the TGS2611-E00 has a specified measurement range from 300 to 10,000 ppm according to the manufacturer. Although the sensors come uncalibrated (at any concentration, not only at high ones), this wording is not correct. Maybe you wanted to say that you did not calibrate the sensor at higher concentrations, but the sensor per se is always uncalibrated from this manufacturer.
- 2.2AR8: Correct and thanks. We did not calibrate the sensor at higher concentrations.
- 2.2 ACM8: We will edit this statement in the manuscript to read "However, since we did not calibrate the sensor for high concentrations, we could not determine the flux rate...".
- 2.2~RC9~p9/14: "The Panterra CH₄ sensor signal has been compensated for the temperature effect, but is probably not applicable for temperatures lower than 15° C." please give the details of the sensor used in the Panterra (it is a TGS if you use the same model that I used years ago and threw away because it was unreliable); as it is, this statement is pure speculation and should either be removed or substantiated with some arguments.
- 2.2AR9: The Panterra sensor product features identifies that it has active temperature compensation (http://neodymsystems.com/download/Panterra-MOS-ALL_Brief_101.pdf). We did try calibrating at temperatures lower than 15°C but still the sensor deviated in its response from other TGS sensors. We did not do further testing to investigate this response because it seemed more efficient to focus on the other sensors.
- 2.2 ACM9: We will remove this sentence from the manuscript as we did not investigate this further.
- 2.3 Feedback on Supplementary Information
- 2.3 RC1: in PowerControlBoard.zip remove the deleted file
- ~\$Copy of Bomexample(1).xlsx
- 2.3 ACM1: We will remove this unused file.
- 2.3 RC2: in BOM_PWCv8c_digikey.xlsx remove unused "Sheet1"
- 2.3 ACM2: We will remove "Sheet1".
- 2.4 Technical issues
- homogenise your variable names in text and figures (d0CH₄sens, d0_CH₄sens, d0_CH₄sensor; d0CH₄conc, d0_CH₄conc, d0_CH₄concentration, d0CH₄concentration)
- decide whether you want to use upper case or lower case letters in figure panels
- use a space between axis title and parentheses around units
- use a degree sign where a degree sign is required (not 0)
- define all your variables that appear in text and figures
- spell out abbreviations upon first occurrence (e.g. AFC on p2/39)
- use some more adequate natural intervals in time axes (e.g. Fig. 5: start at midnight 00:00 and then use 3-hour intervals not some 2:24 hour intervals)

- make figure captions standalone so that the figure is understandable without reading the entire text; also define all symbols and line types (maybe a legend could help)
- do not use the term "plot" for lines or symbols in a panel of Fig. 1
- remove the erroneous and confusing superfluous ticks at the right border of panels (b) and (c) in Fig. 1
- Fig. 2: conventionally panels are labeled from left to right, then top to bottom (which would group a,b and c,d in your caption); explain what each line type indicates and what the symbols with vertical stems indicate.
- Fig. 3: add the 0 label at the beginning of the time axis; I would call this "Elapsed time" and put minute into plural
- add subscript in CH₄ in Fig. 3 caption and everywhere else where this was forgotten.

2.4 ACM1: We will update the manuscript to include all the above changes.

References:

Duc, N. T., Silverstein, S., Lundmark, L., Reyier, H., Crill, P., and Bastviken, D.: Automated Flux Chamber for Investigating Gas Flux at Water–Air Interfaces, Environ. Sci. Technol., 47, 968-975, 10.1021/es303848x, 2013.

Cole, J.J., Bade, D.L., Bastviken, D., Pace, M.L. and Van de Bogert, M. (2010) Multiple approaches to estimating air-water gas exchange in small lakes. Limnology and Oceanography-Methods 8, 285-293.

Gålfalk, M., Bastviken, D., Fredriksson, S. and Arneborg, L. (2013) Determination of the piston velocity for water-air interfaces using flux chambers, acoustic Doppler velocimetry, and IR imaging of the water surface. Journal of Geophysical Research: Biogeosciences 118, 770-782.

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Maeck, A., Hofmann, H., and Lorke, A.: Pumping methane out of aquatic sediments – ebullition forcing mechanisms in an impounded river, Biogeosciences, 11, 2925-2938, https://doi.org/10.5194/bg-11-2925-2014, 2014.

Varadharajan, C., Hermosillo, R., and Hemond, H. F.: A low-cost automated trap to measure bubbling gas fluxes, Limnology and Oceanography: Methods, 8, 363-375, 2010.