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2 **Evaluation of Lacustrine Groundwater Discharge, Hydrologic**  
3 **Partitioning, and Nutrient Budgets in a Proglacial Lake in**  
4 **Qinghai-Tibet Plateau: Using  $^{222}\text{Rn}$  and Stable Isotopes**

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25



26 **Abstract**

27 Proglacial lakes are good natural laboratories to investigate groundwater and  
28 glacier dynamics under current climate condition and to explore primary productivity  
29 under pristine lake status. This study conducted a series of investigations of  $^{222}\text{Rn}$ ,  
30 stable isotopes, nutrients and other hydrogeochemical parameters in Ximen Co Lake,  
31 a remote proglacial lake in the east of Qinghai-Tibet Plateau (QTP). A radon mass  
32 balance model was used to quantify the lacustrine groundwater discharge (LGD) of  
33 the lake, leading to an LGD estimate of  $10.3 \pm 8.2 \text{ mm d}^{-1}$ . Based on the three end  
34 member models of stable  $^{18}\text{O}$  and  $\text{Cl}^-$ , the hydrologic partitioning of the lake is  
35 obtained, which shows that groundwater discharge only accounts for 7.0 % of the  
36 total water input. The groundwater derived DIN and DIP loadings constitute 42.9 %  
37 and 5.5 % of the total nutrient loading to the lakes, indicating the significance of LGD  
38 in delivering disproportionate DIN into the lake. The primary productivity of the lake  
39 water is calculated to be  $0.41 \text{ mmol C m}^{-2} \text{ d}^{-1}$ . This study presents the first attempts to  
40 evaluate the LGD and hydrologic partitioning in the glacial lake by coupling  
41 radioactive and stable isotopic approaches and the findings advance the understanding  
42 of nutrient budgets and primary productivity in the proglacial lakes of QTP. The study  
43 is also instructional in revealing the hydrogeochemical processes in proglacial lakes  
44 elsewhere.



45 **Keywords:** Proglacial lake;  $^{222}\text{Rn}$ ; lacustrine groundwater discharge; hydrologic  
46 partitioning; nutrient budgets; primary productivity

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## 48 1. Introduction

49 High altitude and latitude areas are intensively influenced by the melting of  
50 glaciers due to climatic warming. Of particular importance are the proglacial areas,  
51 such as proglacial lakes and moraines, because they are particularly affected by  
52 climatic change induced glacier retreating and thawing of permafrost (Barry 2006,  
53 Heckmann et al. 2015, Slaymaker 2011). The proglacial lakes are usually located  
54 close to ice front of a glacier, ice cap or ice sheet, with the vicinity to the ice front  
55 sometimes defined as the areas with subrecent moraines and formed by the last  
56 significant glacier advances at the end of the Little Ice Age (Barry 2006, Harris et al.  
57 2009, Heckmann et al. 2015, Slaymaker 2011). Proglacial lakes are located in the  
58 transition zones from glacial to non-glacial conditions, providing natural laboratories  
59 to explore hydrological processes, biogeochemical cycles and geomorphic dynamics  
60 under current climatic conditions (Dimova et al. 2015, Heckmann et al. 2015).  
61 Mountainous proglacial lakes, formed by glacial erosion and filled by melting glaciers,  
62 are widely distributed in the Qinghai-Tibet Plateau (QTP), especially along the  
63 substantial glacier retreating areas of Himalaya Mountains (MT.), Qilian MT.,



64 Tienshan MT., etc. Characterized by higher elevations, small surface areas but  
65 relatively large depths, mountainous proglacial lakes in QTP lack systematic  
66 field-based hydrological studies due to their remote locations and difficulty in  
67 conducting field work (Bolch et al. 2012, Farinotti et al. 2015, Yao et al. 2012).

68 There has been extensive recognition of the importance of groundwater discharge  
69 to various aquatic systems for decades (Dimova and Burnett 2011, Johannes 1980,  
70 Valiela et al. 1978). Very recently, the topic of ‘lacustrine groundwater discharge  
71 (LGD)’, which is comprehensively defined as groundwater exfiltration from lake  
72 shore aquifers to lakes (Blume et al. 2013, Lewandowski et al. 2015, Lewandowski et  
73 al. 2013, Rosenberry et al. 2015), has been introduced. LGD is analogous of in  
74 submarine groundwater discharge (SGD) in coastal environments. LGD plays a vital  
75 role in lake hydrologic partitioning, which is defined as the separation of groundwater  
76 discharge/exfiltration, riverine inflow, riverine outflow infiltration, surface  
77 evaporation and precipitation for the hydrological cycle of the lake (Good et al., 2015).

78 LGD also serves as an importance component in delivering solutes to lakes since  
79 groundwater is usually concentrated in nutrients, CH<sub>4</sub>, dissolved inorganic/organic  
80 carbon (DIC/DOC) and other geochemical components (Belanger et al. 1985, Dimova  
81 et al. 2015, Lecher et al. 2015, Paytan et al. 2015). Nutrients and carbon loading from  
82 groundwater greatly influences ratios of dissolved inorganic nitrogen (DIN) to



83 dissolved inorganic phosphate (DIP) (referred as N: P ratios thereafter), ecosystem  
84 structure and the primary productivity of the lake aquatic system (Belanger et al. 1985,  
85 Hagerthey and Kerfoot 1998, Nakayama and Watanabe 2008).

86 LGD studies utilize various methods including direct seepage meters (Lee 1977,  
87 Shaw and Prepas 1990), geo-tracers such as radionuclides, stable  $^2\text{H}$  and  $^{18}\text{O}$  isotopes  
88 (Gat 1995, Kluge et al. 2007, Kraemer 2005, Lazar et al. 2008), heat and temperature  
89 (Liu et al. 2015, Sebok et al. 2013), numerical modeling (Smerdon et al. 2007, Winter  
90 1999, Zlotnik et al. 2009, Zlotnik et al. 2010) and remote sensing (Anderson et al.  
91 2013, Lewandowski et al. 2013, Wilson and Rocha 2016). Recently, some researchers  
92 started to investigate groundwater dynamics in peri- and proglacial areas, mostly  
93 based on the approaches of numerical modeling (Andermann et al. 2012, Lemieux et  
94 al. 2008a, Lemieux et al. 2008b, Lemieux et al. 2008c, Scheidegger and Bense 2014).

95 However, the quantification of groundwater and surface water exchange in proglacial  
96 lakes is still challenging due to limited hydrogeological data and extremely seasonal  
97 variability of aquifer permeability (Callegary et al. 2013, Dimova et al. 2015, Xin et al.  
98 2013).

99  $^{222}\text{Rn}$ , a naturally occurring inert gas nuclide highly concentrated in groundwater,  
100 can be more applicable in fresh aquatic systems and has been widely used as a tracer  
101 to quantify groundwater discharge in fresh water lakes (Corbett et al. 1997, Dimova et



102 al. 2015, Dimova and Burnett 2011, Dimova et al. 2013, Kluge et al. 2007, Kluge et al.  
103 2012, Luo et al. 2016, Schmidt et al. 2010) and terrestrial rivers and streams  
104 (Batlle-Aguilar et al. 2014, Burnett et al. 2010, Cook et al. 2006, Cook et al. 2003). Of  
105 particular interest are investigations of temporal  $^{222}\text{Rn}$  distribution in lakes, since it  
106 can be used to quantify groundwater discharge and reflect the locally climatological  
107 dynamics (Dimova and Burnett 2011, Luo et al. 2016). Temporal radon variations  
108 give high resolution estimates of groundwater discharge to lakes over diel cycles,  
109 allowing evaluation of LGD and the associated chemical loadings. However, there has  
110 been no study of radon-based groundwater discharge in mountainous proglacial lakes,  
111 especially for those lakes in the QTP.

112 This study aims to investigate the groundwater surface water interactions for the  
113 proglacial lake of Ximen Co, by estimating the LGD and evaluating the hydrologic  
114 partitioning of the lake. LGD is estimated with  $^{222}\text{Rn}$  mass balance model, and the  
115 hydrologic partitioning of the lake is obtained with the three endmember model  
116 coupling the mass balance of water, stable isotopes and Cl<sup>-</sup>. Then, LGD derived  
117 nutrients are estimated and the nutrient budgets of the lake are depicted. Finally,  
118 primary productivity of the lake water is calculated based on the nutrient budgets.  
119 This study, to our knowledge, makes the first attempt to quantify the LGD,  
120 hydrologic partition, and groundwater borne nutrients of the proglacial lake in QTP



121 and elsewhere via the approach integrating multiple tracers. This study provides  
122 insights of hydrologic partitioning in a typical mountainous proglacial lake under  
123 current climate condition and reveals groundwater borne chemical loadings in this  
124 proglacial lake in QTP and elsewhere.

125

## 126 **2. Methodology**

### 127 2.1 Site descriptions

128 The Nianbaoyeze MT., located at the eastern margin of the QTP and being the  
129 easternmost part of NW-SW trending Bayan Har Shan, is situated at the main water  
130 divide of the upper reaches of Yellow River and Yangtze River (Figure 1). With a peak  
131 elevation of 5369 m, the mountain rises about 500-800 m above the surrounding  
132 peneplain and displays typical Pleistocene glacial landscapes such as moraines,  
133 U-shaped valleys and cirques (Lehmkuhl 1998, Schlutz and Lehmkuhl 2009,  
134 Wischnewski et al. 2014). The present snow line is estimated to be at an elevation of  
135 5100 m (some updated references) (Schlutz and Lehmkuhl 2009). Controlled by the  
136 South Asia and East Asia monsoons, the mountain has an annual precipitation of 975  
137 mm in the southern part and 582 mm in the northwestern part, with 80 % occurring  
138 during May and October (Yuan et al. 2014, Zhang and Mischke 2009). The average  
139 temperature gradient is about 0.55 °C per 100 m, and the closest weather station,



140 locating in Jiuzhi town (N: 33.424614°, E: 101.485998) at the lower plains of the  
141 mountain, recorded a mean annual temperature of 0.1 °C. Snowfalls occur in nearly  
142 10 months of the entire year and there is no free-frost all year around (Böhner 1996,  
143 2006, Schlutz and Lehmkuhl 2009). The precipitation, daily bin-averaged wind speed  
144 and temperature in Aug, 2015 were recorded to be 90 mm, .7 m s<sup>-1</sup> and 9.5 °C from  
145 Jiuzhi weather station (Figure 2). The water surface evaporation was recorded to be  
146 1429.8 mm in 2015 from Jiuzhi weather station.

147 Among the numerous proglacial lakes developed in the U-shaped valleys of the  
148 Nianbaoyeze MT., Ximen Co lake is located at the northern margin of the mountain  
149 with an elevation of 4030 m asl, and is well studied and easily accessible (Lehmkuhl  
150 1998, Schlutz and Lehmkuhl 2009, Yuan et al. 2014, Zhang and Mischke 2009). The  
151 lake was formed in a deep, glacially eroded basin with a catchment area of 50 km<sup>2</sup>,  
152 and has a mean and a maximum depth of 40 m and 63.2 m, and a surface area of 3.6  
153 km<sup>2</sup>. The vegetation around the lake is dominated by pine meadows with dwarf shrubs,  
154 rosette plants and alpine cushion (Schlutz and Lehmkuhl 2009, Yuan et al. 2014,  
155 Zhang and Mischke 2009). Mostly recharged by the glacial and snowpack melting  
156 water and regional precipitation, the lake is stratified with an epilimnion depth about  
157 4.4 m in the summer time. The lake is usually covered by ice in the winter time  
158 (Zhang and Mischke 2009). The superficial layer within the U-shaped valley is



159 characterized by peat, clay and fluvial gravels with a depth about 1-3.5 m.  
160 Discontinuous and isolated permafrost is present at the slope of the valley above the  
161 elevation of about 4150 m. The maximum frozen depth is about 1.5 m for the seasonal  
162 frozen ground around the lake. The seasonal frozen ground serves as an unconfined  
163 aquifer during the unfrozen months from July to October, and groundwater discharges  
164 into the epilimnion of the lake (Schlutz and Lehmkuhl 2009, Wang 1997, Zhang and  
165 Mischke 2009).

166

## 167 2.2 Sampling and field analysis

168 The field campaign to Ximen Co Lake was conducted from August, 2015, when it is  
169 warm enough to take the water samples of different origins as the studied site is  
170 seasonally frozen. A  $^{222}\text{Rn}$  continuous monitoring station was setup at the southeast  
171 part of the lake, which is fairly flat for setting up our tent and monitoring system.  
172 Surface water samples were collected around the lake, rivers at the upstream and  
173 downstream. Porewater samples were collected at one side of the lake as the other  
174 side is steep and rocky. The basic water quality parameters of conductivity (EC),  
175 dissolved oxygen (DO), TDS, ORP, pH in the water were recorded with the  
176 multi-parameter meter (HANNA, Co.). Relative humidity was recorded with a  
177 portable thermo-hydrometer (KTH-2, Co.). Lake water samples were taken with a



178 peristaltic pump into 2.5 L glass bottles for  $^{222}\text{Rn}$  measurement with the Big Bottle  
179 system (DurrIDGE, Co.). Surface water samples were filtered with 0.45  $\mu\text{m}$  filters  
180 (Advantec, Co.) in situ and taken into 5 ml, 15 ml, 15 ml and 50 ml Nalgene  
181 centrifugation tubes for stable isotope, major anion, cation and nutrient analysis.  
182 Porewater samples were taken from the lakes shore aquifers with a push point sampler  
183 (M.H.E, Co.) connected to peristaltic pump (Solinst, Co.) (Luo et al. 2014, Luo et al.  
184 2016). 100 ml raw surface water or porewater was titrated with 0.1  $\mu\text{M}$   $\text{H}_2\text{SO}_4$   
185 cartridge (Hach, Co.) in situ to measure total alkalinity (Hasler et al. 2016, Warner et  
186 al. 2013, White et al. 2016). Porewater was filtered with 0.45  $\mu\text{m}$  syringe filters  
187 (Advantec, Co.) in situ and taken into 5 ml, 15 ml, 15 ml and 50 ml Nalgene  
188 centrifugation tubes for stable isotope, major anion, cation and nutrient analysis. 250  
189 ml porewater was taken for  $^{222}\text{Rn}$  measurement with RAD7  $\text{H}_2\text{O}$  (DurrIDGE, Co.)  
190 Samples for major cation analysis were acidified with distilled  $\text{HNO}_3$  immediately  
191 after the sampling.

192  $^{222}\text{Rn}$  continuous monitoring station was set up at the northwest of the lake, close  
193 to the downstream of the lake (Figure 1b). Lake water (about 0.5 m) was pumped with  
194 a DC pump (12 V) driven by lithium batteries (100 Ah) and sprinkled into the  
195 chamber of RAD7 AQUA with a flow rate  $> 2 \text{ min L}^{-1}$ , where  $^{222}\text{Rn}$  in water vapor  
196 was equilibrated with the air  $^{222}\text{Rn}$ . The vapor in the chamber was delivered into two



197 large dry units (Drierite, Co) to remove the moisture and circulated into RAD7  
198 monitor, where  $^{222}\text{Rn}$  activities were recorded every 5 mins. A temperature probe  
199 (HOBO<sup>®</sup>) was insert into the chamber to record the temperature of the water vapor.  
200 The monitoring was performed from 11: 31 am, Aug 22<sup>nd</sup> to 6: 30 am, Aug 24<sup>th</sup>, 2015.  
201 During the period of 1:50-4:30 pm on Aug 22<sup>nd</sup>, a sudden blizzard occurred, leading  
202 to an hourly precipitation about 0.6 mm to the lake area. Daily and hourly  
203 climatological data such as wind speed, air temperature and precipitation were  
204 retrieved from the nearest weather station in Jiuzhi town (N: 33.424614°, E:  
205 101.485998). Water level and temperature fluctuations were recorded with a  
206 conductivity-temperature-depth diver (Schlumberger, Co.) fixed at about 20 cm below  
207 the lake surface and calibrated with local atmospheric pressure recorded by a  
208 baro-diver (Schlumberger, Co.) above the lake. To correct for dissolved  $^{226}\text{Ra}$   
209 supported  $^{222}\text{Rn}$ , one radium sample was extracted from 100 L lake water with  $\text{MnO}_2$   
210 fiber as described elsewhere (Luo et al. 2014, Moore 1976).

211

### 212 2.3 Chemical analysis

213 Major ions were measured with ICS-1100 (Dionex. Co.) in the Department of Earth  
214 Sciences, the University of Hong Kong. The uncertainties of the measurements are  
215 less than 5 %. Nutrients, DIN and DIP were analyzed with flow injection analysis



216 equipped with auto-sampler (Lachat. Co.) in the School of Biological Sciences, the  
217 University of Hong Kong. Stable  $^{18}\text{O}$  and  $^2\text{H}$  isotopes were measured with  
218 MOA-ICOS laser absorption spectrometer (Los Gatos Research (LGR) Triple Isotope  
219 Water Analyzer (TIWA-45EP)) at State Key Laboratory of Marine Geology, Tongji  
220 University, Shanghai. The stable isotopic standards and the recovery test has been  
221 fully described elsewhere (Luo et al., 2017). The measurement uncertainty is better  
222 than 0.1 % for  $^{18}\text{O}$  and 0.5 % for  $^2\text{H}$ .  $^{226}\text{Ra}$  was detected with RAD7 with the method  
223 described elsewhere (Kim et al. 2001, Lee et al. 2012)

224

#### 225 2.4 Radon transient model

226 Previous studies employed a steady state radon-222 mass balance model to  
227 quantify LGD to lentic system such as lakes and wetlands (Dimova and Burnett 2011,  
228 Luo et al. 2016). This model assumes that radon input derived from groundwater  
229 inflow, diffusion and river inflow are balanced by the radon losses of atmospheric  
230 evasion, decay and river outflow. However, recently studies revealed that the steady  
231 state is mainly reached after 2-15 days of constant metrological conditions, and  
232 mostly lentic system can be not be treated as steady state due to rapid radon-222  
233 degassing to the atmosphere driven by wind-induced turbulence (Gilfedder et al.,  
234 2015; Dimova and Burnett, 2011).



235 Ximen Co lake is demonstrated to be highly stratified with an epilimnion of 4.4  
236 m (Zhang and Mischke 2009). The lake was formed by glacier erosion and the  
237 lakebed is characterized by granite bedrock with a thin sedimentary clay layer.  
238 Previous studies have indicated that sediment with a thickness of 0.7-3.3 m has been  
239 developed on the bedrock and forms the lake shore aquifer, which consists of clay,  
240 soils and gravels (Schlutz and Lehmkuhl 2009). Porewater sampled in the aquifer  
241 immediately behind the lake shore can well represent groundwater discharging into  
242 the lake, as suggested previously (Lewandowski et al. 2015, Rosenberry et al. 2015,  
243 Schafran and Driscoll 1993). LGD has been widely considered to occur within the  
244 first few meters of the lake shore (Lee et al. 1980, Rosenberry et al. 2015, Schafran  
245 and Driscoll 1993) and groundwater is considered to predominately discharge into the  
246 epilimnion since deep groundwater flow is highly limited by the Precambrian bedrock  
247 (Einarsdottir et al., 2016). Therefore,  $^{222}\text{Rn}$  mass balance model is established to  
248 quantify LGD to the epilimnion from the lake shore. Due to negligible hydrological  
249 connection between the epilimnion and hypolimnion, LGD for the lake can be  
250 quantified with  $^{222}\text{Rn}$  mass balance model for the epilimnion.

251 The governing equation of radon-222 transient mass balance model within a 1 x  
252 1 x z cm (where z is the depth in cm) can be expressed as (Gilfedder et al. 2015):

253 
$$z \frac{\partial I_w}{\partial t} = F_{gw} + (I_{^{226}\text{Ra}} - I_w) \times z \times \lambda_{^{222}} + F_{diff} - F_{atm} \quad (1)$$



254 where  $F_{gw}$ ,  $F_{diff}$ ,  $F_{atm}$  [ $\text{Bq m}^{-2} \text{d}^{-1}$ ] are  $^{222}\text{Rn}$  loadings from LGD, water-sediment  
255 diffusion and water-air evasion, respectively;  $z$  [m] is the lake water level depth  
256 recorded by the diver.  $\lambda_{222}$  is the decay constant of  $^{222}\text{Rn}$  with a value of  $0.186 \text{d}^{-1}$ .  
257  $\lambda_{222} \times I_{226Ra}$  and  $\lambda_{222} \times I_w$  account for the production and decay of  $^{222}\text{Rn}$  [ $\text{Bq m}^{-2} \text{d}^{-1}$ ]  
258 in the water column, respectively.  $I_w$  and  $I_{226Ra}$  [ $\text{Bq m}^{-2}$ ] represent  $^{222}\text{Rn}$  and  $^{226}\text{Ra}$   
259 inventories in the epilimnion, and are expressed as:  $I_w = H \times C_w$  and  
260  $I_{226Ra} = H \times C_{226Ra}$ , respectively; where  $H$  [m] is the depth of the epilimnion;  $C_w$  and  
261  $C_{226Ra}$  is the  $^{222}\text{Rn}$  and  $^{226}\text{Ra}$  activity [ $\text{Bq m}^{-3}$ ], respectively.

262 The model is valid under the following assumptions: 1) The epilimnion is well  
263 mixed which is the actual condition for most natural boreal and high altitude glacial  
264 lakes (Åberg et al. 2010, Zhang and Mischke 2009). 2)  $^{222}\text{Rn}$  input from riverine  
265 water inflow, and loss from the lake water outflow and infiltration into the lake shore  
266 aquifer is negligible compared to the groundwater borne  $^{222}\text{Rn}$ , because  $^{222}\text{Rn}$   
267 concentration of groundwater is 2-3 orders of magnitude larger than that lake water  
268 (Dimova and Burnett 2011, Dimova et al. 2013). Generally,  $^{222}\text{Rn}$  in the epilimnion is  
269 sourced from LGD and decay input from parent isotope of  $^{226}\text{Ra}$  under secular  
270 equilibrium, and is mainly lost via atmospheric evasion and radioactive decay.

271  $F_{atm}$  is the key sinking component of the transient model and is finally a function of  
272 wind speed and water temperature, both of which are temporal variant variables



273 (Supplementary information). Lake water level  $z$  is also a temporal variant variable  
274 which represents the fluctuations of water volume of the epilimnion. This equation is  
275 discretized by the forward finite difference method, and the groundwater flux at each  
276 time step can be solved as follow

$$277 \quad [{}^{222}\text{Rn}_{t+\Delta t}] = \frac{[z \times {}^{222}\text{Rn}_t + [F_{diff} + F_{gw} - F_{atm} - {}^{222}\text{Rn}_t \times \lambda \times z] \times \Delta t]}{z} \quad (2)$$

278 where  ${}^{222}\text{Rn}_{t+\Delta t}$  and  ${}^{222}\text{Rn}_t$  [Bq m<sup>-3</sup>] is the <sup>222</sup>Rn activity at current time step and at  
279 the previous time steps, respectively, and  $\Delta t$  [min] is the time step. With the inverse  
280 calculation based on Equation (4), the groundwater inflow at each time step can be  
281 obtained. However, large errors of the final LGD calculation will be induced by even  
282 a small amount of noise in the measured <sup>222</sup>Rn data due to the  ${}^{222}\text{Rn}_{t+\Delta t} - {}^{222}\text{Rn}_t$  term  
283 being with the measure uncertainty. To reduce the random errors of the  
284 measured <sup>222</sup>Rn concentrations, the time window with a width of 1 hour is proposed to  
285 smooth the curve (Supplementary information).

286

### 287 3. Results

#### 288 3.1 Time series data

289 Figure 2 shows the basic climatological parameters of the lake catchment during  
290 the campaign month. There are discrete rainfall events occurring throughout the  
291 month with an average rainfall of 3.1 mm d<sup>-1</sup>. The temperature throughout the month  
292 ranges from 5.0 – 12.5 °C within an average of 9.3 °C. The daily averaged wind speed



293 generally ranges from  $0.7 - 2.5 \text{ m s}^{-1}$ , with an average of  $1.7 \text{ m s}^{-1}$ .  $^{222}\text{Rn}$  temporal  
294 distribution and other time series data are shown in Figure 3a and listed in  
295 Supplementary Table 1. Generally,  $^{222}\text{Rn}$  concentration varies from  $32.2$  to  $273 \text{ Bq m}^{-3}$ ,  
296 with an average of  $144.2 \pm 27.7 \text{ Bq m}^{-3}$ .  $^{222}\text{Rn}$  over the monitoring period shows  
297 typical diel cycle, much higher at nighttime and lower in the day time. Figures 3b-3d  
298 shows the time series data of temperature (5 mins interval), nearshore lake water level  
299 (1 min interval), and wind speed (1 hour interval). Temperature and lake water level  
300 also show typical diel cycles, but with antiphase fluctuations with each other.  
301 Temperature is higher during the daytime and lower at nighttime. However a sudden  
302 decrease of temperature was recorded due to the sudden blizzard (Figure 3b). Water  
303 level is higher at nighttime and lower during the daytime, with a strong fluctuation  
304 due to the turbulence caused by the blizzard (Figure 3c). The variability might reflect  
305 the dynamics of groundwater input and surface water inflow. The air temperature of  
306 the lake area is in phase with the water temperature. Wind speed is normally higher  
307 during the daytime and lower at nighttime (Figure 3d).

308 The variation of  $^{222}\text{Rn}$  is nearly in antiphase with the fluctuations of lake water  
309 temperature and air temperature, indicating that the dominated controlling factors  
310 of  $^{222}\text{Rn}$  fluctuations are water temperature and wind speed (Figure 3a). This  
311 phenomenon is reasonable as lake water  $^{222}\text{Rn}$  is predominately lost via atmospheric



312 evasion, which is the function of wind speed and water temperature (Dimova et al.  
313 2015, Dimova and Burnett 2011, Dimova et al. 2013). High water temperature and  
314 wind speed leads to elevated atmospheric evasion and causes the decline of  $^{222}\text{Rn}$   
315 concentration in the lake water. However, there is a sudden reduction of radon activity  
316 from 2: 00 pm to 4: 00 pm on Jul 22<sup>nd</sup>, 2015, when the snow event led to a sudden  
317 decrease of water temperature, increase of wind speed, and large surface water  
318 turbulence as indicated by water level fluctuations (Figures 3a-3d).  $^{222}\text{Rn}$  in the  
319 porewater is 2-3 orders of magnitude larger than  $^{222}\text{Rn}$  in the lake water, suggesting  
320 that  $^{222}\text{Rn}$  is an ideal tracer to estimate the LGD (Supplementary Table 1).  $^{222}\text{Rn}$   
321 concentrations in surface water range from 22.2 to 209 Bq m<sup>-3</sup>, with an average of  
322 92.5 Bq m<sup>-3</sup> (n = 12), which is in the range of  $^{222}\text{Rn}$  continuous monitoring results,  
323 suggesting reliable  $^{222}\text{Rn}$  measurements (Supplementary Table 2).

324

### 325 3.2 Geochemical results

326 The results of major ions, nutrients and stable isotopes in different water end  
327 members are shown in Figures 4 and 5.  $\text{Cl}^-$  ranges from 0.6 to 2.1 mg L<sup>-1</sup> in the  
328 surface water (including riverine inflow water, lake water and downstream water), 0.4  
329 to 2.7 mg L<sup>-1</sup> in porewater and has a much higher concentration of 5.9 mg L<sup>-1</sup> in  
330 rainfall water.  $\text{Na}^+$  ranges from 1.6 to 3.4 mg L<sup>-1</sup> in the surface water, 1.2 to 4.4 mg



331  $\text{L}^{-1}$  in porewater and has a concentration of  $4.4 \text{ mg L}^{-1}$  in rainfall water.  $\text{SO}_4^{2-}$  ranges  
332 from  $1.2$  to  $2.3 \text{ mg L}^{-1}$  in the surface water,  $0.4$  to  $1.7 \text{ mg L}^{-1}$  in porewater and has a  
333 significant low concentration of  $0.01 \text{ mg L}^{-1}$  in rainfall water.  $\text{Ca}^{2+}$  ranges from  $3.0$  to  
334  $12.4 \text{ mg L}^{-1}$  in lake water,  $3.4$  to  $12.5 \text{ mg L}^{-1}$  in porewater and has a significant high  
335 concentration of  $20.5 \text{ mg L}^{-1}$  in rainfall water. Other concentrations of major ions are  
336 listed in Supplementary Table 2. As shown in Figure 4d and Supplementary Table 2,  
337  $\delta^{18}\text{O}$  in the lake water ranges from  $-13.06 \text{ ‰}$  to  $-12.11 \text{ ‰}$ , with an average of  $-$   
338  $12.41 \text{ ‰}$  ( $n = 7$ ), and  $\delta^2\text{H}$  ranges from  $-91.83 \text{ ‰}$  to  $-87.47 \text{ ‰}$ , with an average of  $-$   
339  $89.0 \text{ ‰}$  ( $n = 7$ ).  $\delta^{18}\text{O}$  in the riverine inflow water ranges from  $-13.44 \text{ ‰}$  to  $-13.29 \text{ ‰}$ ,  
340 with an average of  $-13.37 \text{ ‰}$  ( $n = 2$ ), and  $\delta^2\text{H}$  ranges from  $-93.25 \text{ ‰}$  to  $-91.92 \text{ ‰}$ ,  
341 with an average of  $-92.59 \text{ ‰}$  ( $n = 2$ ).  $\delta^{18}\text{O}$  in the downstream water ranges from  $-$   
342  $12.51 \text{ ‰}$  to  $-12.18 \text{ ‰}$ , with an average of  $-12.35 \text{ ‰}$  ( $n = 3$ ), and  $\delta^2\text{H}$  ranges from  $-$   
343  $88.96 \text{ ‰}$  to  $-87.1 \text{ ‰}$ , with an average of  $-87.98 \text{ ‰}$  ( $n = 3$ ).  $\delta^{18}\text{O}$  in the porewater  
344 ranges from  $-12.66 \text{ ‰}$  to  $-11.52 \text{ ‰}$ , with an average of  $-11.97 \text{ ‰}$  ( $n = 8$ ), and  $\delta^2\text{H}$   
345 ranges from  $-91.3 \text{ ‰}$  to  $-82.87 \text{ ‰}$ , with an average of  $-85.5 \text{ ‰}$  ( $n = 8$ ). DIN in the  
346 surface water (including riverine inflow water, lake water and downstream water)  
347 range from  $6.6$  to  $16.9 \text{ }\mu\text{M}$ , with an average of  $10.3 \text{ }\mu\text{M}$ , and DIP from  $0.36$  to  $0.41$   
348  $\text{ }\mu\text{M}$ , with an average of  $0.38 \text{ }\mu\text{M}$ . The concentrations of DIN for the porewater range  
349 from  $0.7$  to  $358.8 \text{ }\mu\text{M}$ , with an average of  $92.8 \text{ }\mu\text{M}$ , and DIP from  $0.18$  to  $0.44 \text{ }\mu\text{M}$



350 with an average of 0.31  $\mu\text{M}$  (Figure 5).

351

#### 352 4. Discussion

##### 353 4.1 Proglacial hydrologic processes and geochemical implications

354 Generally, major ion concentrations in the lake water and porewater of Ximen  
355 Co lake are significantly lower than those in main rivers, streams and other tectonic  
356 lakes in the QTP (Wang et al. 2010, Wang et al. 2016b, Yao et al. 2015), and are  
357 similar to those of snow and glaciers (Liu et al. 2011), suggesting that the lake water  
358 is mainly originated from glacier and snow melting. Ion concentrations in the lake and  
359 porewater of Ximen Co lake are much lower than those of rainfall collected in Jiuzhi  
360 town. This suggests that lake water is less influenced by precipitation (Figures 4a-4c).  
361 The concentrations of major ions in the porewater are high compared to the lake water,  
362 indicating weathering affects from the aquifer grains. The ratios of  $\text{Ca}^{2+}/\text{Na}^{+}$  in the  
363 porewater and groundwater is  $>1$ , also suggesting influences of weathering digenesis  
364 of major ions from the seasonal frozen ground at the lake shore aquifer (Wang et al.  
365 2010, Weynell et al. 2016, Yao et al. 2015).

366 The isotopic compositions of the lake water and porewater are significantly  
367 isotopic depleted, with values close to the compositions of glaciers and surface snow  
368 in the QTP, suggesting the lake is dominantly recharged from snow and glacier



369 melting (Cui et al. 2014, Wang et al. 2016a, Zongxing et al. 2015). The relation of  
370  $\delta^{18}\text{O}$  versus  $\delta^2\text{H}$  for the lake water is  $\delta^2\text{H} = 4.25 \times \delta^{18}\text{O} - 35.99$ , with a slope much  
371 lower than that of the global meteoric water line (GMWL) (Figure 4d), suggesting the  
372 effects of lake surface evaporation. The relation of  $\delta^{18}\text{O}$  versus  $\delta^2\text{H}$  for the porewater  
373 is  $\delta^2\text{H} = 6.93 \times \delta^{18}\text{O} - 2.67$ , overall on GMWL (Figure 4d). Deuterium excesses is  
374 defined as  $\Delta\text{D} = \delta\text{D} - 8 \times \delta^{18}\text{O}$  (Dansgaard 1964). The value of  $\Delta\text{D}$  is dependent on  
375 air mass origins, altitude effect and the kinetic effects during evaporation (Hren et al.  
376 2009). Global meteoric water has a  $\Delta\text{D}$  of + 10 ‰. In QTP, glacier/snowpack melting  
377 water usually has large positive  $\Delta\text{D}$ , while the precipitations derived from warm and  
378 humid summer monsoon has lower  $\Delta\text{D}$  (Ren et al. 2017, Ren et al. 2013). In this study,  
379  $\Delta\text{D}$  of surface water, lake and porewater ranges from + 37.1 to + 41.2 ‰, closed to  
380 the glacier melting water but much larger than that of the local precipitation of +  
381 29.72 ‰. This indicates the stream and lake water are mainly originated from  
382 glacial/snowpack melting rather than precipitation (Gat 1996, Lerman et al. 1995,  
383 Wang et al. 2016a). The slopes of  $\delta^{18}\text{O}$  versus  $\delta^2\text{H}$  in lake water and porewater are  
384 4.25 and 6.93, both of which are lower than that of GMWL due to surface evaporation.  
385 Lake water is more intensively influenced by evaporation compared to porewater. The  
386 plots of  $\delta^{18}\text{O}$  versus  $\text{Cl}^-$ , and  $\delta^2\text{H}$  versus  $\text{Cl}^-$  are well clustered for porewater end  
387 member (orange area), lake water end member (blue area), riverine inflow water end



388 member (yellow area), and precipitation water (Figures 4e and 4f), suggesting stable  
389  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  isotopes and  $\text{Cl}^-$  can serve as tracers to quantify the hydrologic  
390 partitioning of the lake by setting three endmember models.

391 The concentrations of DIN and DIP are all within the ranges of other glacial  
392 melting water and proglacial lake water (Hawkings et al. 2016, Hodson 2007, Hodson  
393 et al. 2005, Hudson et al. 2000, Tockner et al. 2002). Briefly, rainfall and upstream  
394 lake water such as YN-4 has the highest DIN concentration, indicating the glacier  
395 melting and precipitation could be important DIN sources in proglacial areas  
396 (Anderson et al. 2017, Dubnick et al. 2017). DIN in porewater is overall higher  
397 compared to the lake water, suggesting the porewater to be DIN effective source; and  
398 DIP concentrations is higher in the lake water compared to porewater, suggesting the  
399 porewater is a DIP sink (Figure 5). The N: P ratios in the lake water and porewater are  
400 averaged to be 27.1 and 320.5, respectively, both much larger than the Redfield Ratio  
401 (N: P = 16:1) in water and organism in most aquatic system and within the range of  
402 other proglacial lakes (Anderson et al. 2017). This also suggests that the lake water  
403 and porewater are under phosphate limited condition. N: P ratio in the rainfall water is  
404 30.4, similar to the lake water. The average N: P ratio of porewater is much higher  
405 than that of lake water, indicating DIN enrichment in the lake shore aquifers (Figure 5).  
406 In pristine groundwater,  $\text{NO}_3^-$  is the predominated form of N and is highly mobile



407 within the oxic aquifers, leading to much higher DIN concentrations in the porewater;  
408 DIP has high affinity to the aquifer grains, resulting in much lower DIP concentrations  
409 in the porewater (Lewandowski et al. 2015, Rosenberry et al. 2015, Slomp and Van  
410 Cappellen 2004). Thus, in analogous to surface runoff from glacier/snowpack melting,  
411 LGD can be also regarded as an important source for the proglacial lakes. Because of  
412 very high DIN and N: P ratios in the porewater, a relatively small portion of LGD  
413 delivers considerable nutrients into the glacial lake, shifting the aquatic N: P ratios  
414 and affecting the proglacial aquatic ecosystem (Anderson et al. 2017).

415

#### 416 4.2 Estimation of LGD

417 Figure 6a shows all the sinks and sources of radon with the epilimnion of the lake.  
418 Within  $^{222}\text{Rn}$  transient mass balance model, the dominant  $^{222}\text{Rn}$  loss is atmospheric  
419 degassing/evasion. Generally,  $^{222}\text{Rn}$  degassing rate is the function of the radon-222  
420 concentration gradient at the water-air interface and the parameter of gas piston  
421 velocity  $k$ , which is finally the function of wind speed and water temperature (Dimova  
422 and Burnett 2011, Gilfedder et al. 2015). To evaluate  $^{222}\text{Rn}$  evasion rate, this study  
423 employs the widely used method proposed by MacIntyre et al. (1995) which is also  
424 detailed described in Supplementary Information. Based on the field data of  $^{222}\text{Rn}$   
425 concentration in the lake water, wind speed and temperature log, the radon degassing



426 rate is calculated in a range of 0.8 to 265.2 Bq m<sup>2</sup> d<sup>-1</sup>, with an average 42.0 of Bq m<sup>2</sup>  
427 d<sup>-1</sup>.

428 In addition to the atmospheric loss and sedimentary diffusion inputs, <sup>222</sup>Rn is also  
429 sinked via radioactive decay, and sourced from decay of parent isotope of <sup>226</sup>Ra. The  
430 decay loss of <sup>222</sup>Rn fluctuates in phase with the distribution of <sup>222</sup>Rn concentration  
431 monitored by RAD 7 AQUA. The equations to estimate benthic fluxes are shown in  
432 supplementary information. The decay loss is calculated to be 26.4 to 223.4 Bq m<sup>-2</sup> d<sup>-1</sup>,  
433 with an average of 118.0 ± 22.7 Bq m<sup>-2</sup> d<sup>-1</sup>. <sup>226</sup>Ra concentration is 0.01 Bq m<sup>-3</sup> for the  
434 lake water. Under secular equilibrium, the <sup>226</sup>Ra decay input can be calculated by  
435 multiplying <sup>226</sup>Ra concentration in the lake water with  $\lambda_{222}$  (Corbett et al. 1997, Kluge  
436 et al. 2007, Luo et al. 2016). <sup>226</sup>Ra decay input is calculated to be 0.83 Bq m<sup>-2</sup> d<sup>-1</sup>,  
437 which is significantly low compared to other <sup>222</sup>Rn sources to the epilimnion.

438 With the obtained sinks and sources of <sup>222</sup>Rn in the lake, and the constants given in  
439 Table 1, LGD rate can be obtained by dividing the groundwater derived <sup>222</sup>Rn with its  
440 concentration in groundwater endmember. The obtained LGD rate, ranges from -23.7  
441 mm d<sup>-1</sup> to 90.0 mm d<sup>-1</sup>, with an average of 10.3 ± 8.2 mm d<sup>-1</sup> (Figure 7) The LGD rate  
442 range is relatively less than the daily lake water level variations (≈ 50 mm), indicating  
443 that the lake water level variation could be a combined effect of surface runoff and  
444 LGD (Hood et al. 2006). The negative values of LGD rate reflect the return



445 groundwater flow due to infiltration into the porewater. Normally, the dominant  
446 values are positive, indicating LGD rate is significant compared to water infiltrations  
447 into lakeshore aquifer. The temporal variation of LGD rate could be attributed to the  
448 fluctuations of the hydraulic gradient in the proglacial areas (Hood et al. 2006, Levy  
449 et al. 2015). As indicated by  $\Delta D$  (mostly  $> 10$ ) of surface water, the lake and the  
450 upstream water is considered to be mainly recharged from glacial/snowpack melting  
451 rather other precipitations.

452 To assess the magnitude of uncertainty of  $^{222}\text{Rn}$  transient model, the sensitivity of  
453 estimated LGD to changes in other variables is examined. A sensitivity coefficient  $f$  is  
454 proposed to evaluate this uncertainty according to Langston et al. (2013)

$$455 \quad f = (\Delta F_{LGD} / F_{LGD}) / (\Delta y_i / y_i) \quad (3)$$

456 where  $\Delta F_{LGD}$  is the amount of change in  $F_{LGD}$  from the original value.  $\Delta y_i$  is the  
457 amount of change in the other variable of  $y_i$  from the original value. Thus, higher  $f$   
458 indicates a large uncertainty of final LGD estimate. The uncertainty mainly stems  
459 from  $^{222}\text{Rn}$  measurements in different water endmembers, the atmospheric loss and  
460 water level record. The uncertainties of  $^{222}\text{Rn}$  measurement are about 10 % and 15-20  
461 % in groundwater and lake water endmember, respectively. The uncertainty of  
462 atmospheric loss is derived from uncertainty of  $^{222}\text{Rn}$  in lake water (with an  
463 uncertainty of 15-20 %), temperature (with an uncertainty  $\approx 5$  %) and wind speed



464 (with an uncertainty  $\approx 5\%$ ). Thus, the final LGD estimate has an uncertainty of  
465 35-40 %.

466

### 467 **4.3 Hydrologic partitioning**

468 Compared to the groundwater labeled radionuclide of  $^{222}\text{Rn}$ , stable  $^{18}\text{O}/^2\text{H}$   
469 isotopes are advantageous in the investigation of evaporation processes due to their  
470 fractionations from water to vapor and have been widely used to investigate the  
471 hydrologic cycle of lakes in various environments (Gat 1995, Gibson et al. 1993,  
472 Gonfiantini 1986, Stets et al. 2010). With the field data of stable isotopic composition  
473 and  $\text{Cl}^-$  concentrations in different water end members, groundwater input, surface  
474 water input, lake water outflow and infiltration, and evaporation can be partitioned by  
475 coupling stable isotopic mass balance model with  $\text{Cl}^-$  mass balance model (Figure 6b).

476 The model, consisting of the budgets of stable isotopes and  $\text{Cl}^-$ , and water masses  
477 for the epilimnion, is used to quantify riverine inflow, lake water outflow and  
478 infiltration, and evaporation (Gibson et al. 2016, LaBaugh et al. 1995, LaBaugh et al.  
479 1997). The model is valid under the following assumptions: (1) constant density of  
480 water; (2) no long-term storage change in the reservoir; (3) well-mixed for the  
481 epilimnion (Gibson 2002, Gibson et al. 2016, Gibson and Edwards 2002, LaBaugh et  
482 al. 1997). The above assumptions are reasonably tenable during the short monitoring



483 period. The model can be fully expressed as

484 
$$F_{in} + F_{LGD} + F_p = F_E + F_{out} \quad (4)$$

485 
$$F_{in} \times \delta_{in} + F_{LGD} \times \delta_{gw} + F_p \times \delta_p = F_E \times \delta_E + F_{out} \times \delta_L \quad (5)$$

486 
$$F_{in} \times [Cl^-]_{in} + F_{LGD} \times [Cl^-]_{gw} + F_p \times [Cl^-]_p = F_{out} \times [Cl^-]_L \quad (6)$$

487 where  $F_{in}$  [ $\text{mm d}^{-1}$ ] is the surface water inflow to the lake;  $F_{LGD}$  [ $\text{mm d}^{-1}$ ] is LGD rate.

488  $F_p$  [ $\text{mm d}^{-1}$ ] is the mean daily rainfall rate during the sampling period.  $F_E$  [ $\text{mm d}^{-1}$ ] is

489 the lake evaporation.  $F_{out}$  [ $\text{mm d}^{-1}$ ] is the lake water outflow via runoff and

490 infiltration into the lake shore aquifer.  $\delta_{in}$ ,  $\delta_{gw}$ ,  $\delta_E$  and  $\delta_p$  are the isotopic

491 compositions of surface water inflow, LGD, and evaporative flux, respectively. The

492 values of  $\delta_{in}$ ,  $\delta_{gw}$ , and  $\delta_p$  are obtained from field data and the composition of  $\delta_E$  are

493 calculated as shown in supplementary information.  $[Cl^-]_{in}$ ,  $[Cl^-]_{gw}$ ,  $[Cl^-]_L$  and

494  $[Cl^-]_p$  are the chloride concentrations in the inflow water, porewater, lake water and

495 precipitation, respectively.

496 The components of the mass balance model can be obtained from the field data of

497 isotopic composition and  $Cl^-$  concentrations in different water endmembers. The

498 average  $^{18}O$  composition -13.37 ‰ of riverine inflow water is taken as the value of

499 the input parameter  $\delta_{in}$ .  $\delta^{18}O$  and  $\delta^2H$  in the groundwater endmember and lake water

500 end member are calculated to be -12.41 ‰ and -87.18 ‰, respectively.  $\delta^{18}O$  and  $\delta^2H$

501 in the rainfall are measured to be -5.47 ‰ and -24.98 ‰, respectively. With the



502 measured values of  $\delta_L$ ,  $h$ ,  $\delta_{in}$ , and the estimated  $\varepsilon$  and  $\delta_a$ , the isotopic composition  
503 of  $\delta_E$  is calculated to be -35.11 ‰, which is in line with the results of alpine and  
504 arctic lakes elsewhere (Gibson 2002, Gibson et al. 2016, Gibson and Edwards 2002).  
505 The values of  $[Cl^-]_{in}$ ,  $[Cl^-]_{gw}$ , and  $[Cl^-]_L$  are calculated to be 0.91 mg L<sup>-1</sup>, 1.48  
506 mg L<sup>-1</sup> and 1.02 mg L<sup>-1</sup>, respectively. All the parameters used in the model are shown  
507 in Table 2.

508 According to Equations 4-6, the uncertainties of calculations of  $F_{in}$ ,  $F_{out}$  and  $E$  are  
509 mainly derived from the uncertainty of  $F_{LGD}$  and the compositions of Cl<sup>-</sup>, δD and δ<sup>18</sup>O  
510 in different water endmembers as suggested in previous studies (Genereux 1998,  
511 Klaus and McDonnell 2013). The compositions of Cl<sup>-</sup>, δD and δ<sup>18</sup>O in surface water,  
512 groundwater endmembers have an uncertainty of 5 %. The uncertainty of  $\delta_E$  is  
513 reasonably assumed to be  $\approx 20$  %. Thus, considering the uncertainty propagation of  
514 all the above parameters, the uncertainties of  $F_{in}$ ,  $F_{out}$  and  $E$  would be scaled up to  
515 70-80 % of the final estimates.

516

#### 517 4.4 The hydrologic partitioning of the glacial lake

518 Based on the three endmember model of <sup>18</sup>O and Cl<sup>-</sup>, the riverine inflow rate was  
519 calculated to be  $135.6 \pm 119.0$  mm d<sup>-1</sup>, and the lake outflow rate is estimated to be  
520  $141.5 \pm 132.4$  mm d<sup>-1</sup>; the evaporation rate is calculated to be  $5.2 \pm 4.7$  mm d<sup>-1</sup>. The



521 summary of the hydrologic partitioning of the lake is shown in Figure 8a. Generally,  
522 the proglacial lake is mostly recharged by the riverine inflow from the snowpack or  
523 the glacier melting. The groundwater discharge contributes about only 7.0 % of the  
524 total water input to the lake, indicating groundwater input does not dominate water  
525 input to the proglacial lake. The lake water is mainly lost via surface water outflow  
526 and infiltration to the lake shore aquifers. The evaporation constitutes relatively small  
527 ratio ( $\approx 3.5\%$ ) of total water losses. The annual evaporation rate was recorded to be  
528 1429.8 mm (equivalent to  $3.92\text{ mm d}^{-1}$ ) in 2015 by the Jiuzhi weather station, lower  
529 than the obtained evaporation in this study. This may be due to much higher  
530 evaporation in August during the monitoring period. The recent review on LGD rate  
531 by Rosenberry et al. (2015) suggests that the median of LGD rate in the literatures is  
532  $7.4\text{ mm d}^{-1}$  ( $0.05\text{ mm d}^{-1}$  to  $133\text{ mm d}^{-1}$ ), which is about  $2/3$  of LGD rate in this study.  
533 This difference may be due to the hydrogeological setting of the lake shore aquifer.  
534 This aquifer is formed by grey loam, clayey soil and sand (Lehmkuhl 1998, Schlutz  
535 and Lehmkuhl 2009), which is with relatively high permeability. Previous studies  
536 have indicated that groundwater forms a key component of proglacial hydrology  
537 (Levy et al. 2015). However, there have been limited quantitative studies of  
538 groundwater contribution to hydrologic budget of proglacial areas. This study  
539 summarizes the groundwater discharge studies over the glacial forefield areas. Brown



540 et al. (2006) investigated the headwater streams at the proglacial areas of Taillon  
541 Glacier in French and found that groundwater contributes 6-10 % of the stream water  
542 immediate downwards of the glacier. Using water mass balance model, Hood et al.  
543 (2006) shows that groundwater inflow is substantial in the hydrologic partitioning of  
544 the proglacial Lake O'Hara in front of Opabin Glacier in Canada and comprised of 30  
545 -74 % of the total inflow. Roy and Hayashi (2008) studied the proglacial lakes of  
546 Hungabee lake and Opabin lake at glacier forefield of Opabin Glacier and found that  
547 groundwater component is predominant water sources of the lakes and consisted of  
548 35-39 % of the total water input of the lakes. Langston et al. (2013) further  
549 investigated a tarn immediate in front of Opabin Glacier and indicated the tarn is  
550 predominantly controlled by groundwater inflow/outflow, which consisted of 50-100 %  
551 of total tarn volume. Magnusson et al. (2014) studied the streams in the glacier  
552 forefield of Dammagletscher, Switzerland and revealed that groundwater contributed  
553 only 1-8 % of the total surface runoff. Groundwater contribution in this study is  
554 similar to those obtained the mountainous proglacial areas in Europe, but much lower  
555 than those obtained in the proglacial areas of polar regions. It is concluded that  
556 proglacial lakes/streams in front of mountainous glaciers are mainly recharged by  
557 surface runoff from glacier/snowpack melting. This might be due to well-developed  
558 stream networks and limited deep groundwater flow (Brown et al. 2006, Einarsdottir



559 et al. 2017, Magnusson et al. 2014). However, proglacial tarns and lakes in the polar  
560 areas are predominantly controlled by groundwater discharge, due to less connectivity  
561 of surface runoff and high shallow and deep groundwater connectivity (Hood et al.  
562 2006, Langston et al. 2013, Roy and Hayashi 2008).

563 4.5 LGD derived nutrient loadings, nutrient budget and ecological implications

564 Compared to extensive studies of SGD derived nutrient loadings in the past decade  
565 (Luo and Jiao 2016, Slomp and Van Cappellen 2004), studies of LGD derived nutrient  
566 loadings have received limited attention, even given the fact that groundwater in lake  
567 shore aquifers is usually concentrated in nutrients (Lewandowski et al. 2015,  
568 Rosenberry et al. 2015). Even fewer studies focus on chemical budgets in the  
569 proglacial lakes which are often difficult to access for sampling. Groundwater borne  
570 DIN and DIP across the sediment-water interface in this study are determined with an  
571 equation coupling the advective or LGD-derived, and diffusive solute transport  
572 (Hagerthey and Kerfoot 1998, Lerman et al. 1995)

573 
$$F_j = -nD_j^m \frac{dC_j}{dx} + v_{gw} C_j \quad (7)$$

574 where  $-nD_j^m \frac{dC_j}{dx}$  is the diffusion input and  $v_{gw} C_j$  is the LGD derived fluxes,  $F_j$   
575 [ $\mu\text{M m}^{-2} \text{d}^{-1}$ ] is the mol flux of nutrient species  $j$  (representing DIN or DIP).  $n$  is the  
576 sediment porosity.  $D_j^m$  is the molecular diffusion coefficient of nutrient species  $j$ ,



577 which is given to be  $4.8 \times 10^{-5} \text{ m}^2 \text{ d}^{-1}$  for DIP (Quigley and Robbins 1986), and  $8.8 \times$   
578  $10^{-5} \text{ m}^2 \text{ d}^{-1}$  for DIN (Li and Gregory 1974), respectively.  $C_j$  [ $\mu\text{M}$ ] is the concentration  
579 of nutrient species  $j$ .  $x$  [m] is the sampling depth.  $v_{\text{gw}}$  is LGD rate estimated by  $^{222}\text{Rn}$   
580 mass balance model and has a value of  $10.3 \pm 8.2 \text{ mm d}^{-1}$ .  $\frac{dC_j}{dx}$  is the concentration  
581 gradient of nutrient species  $j$  across the water-sedimentary interface.

582 Substituting the constants and the field data of DIN and DIP in to Equation 6, LGD  
583 derived nutrient loadings are calculated to be  $954.3 \mu\text{mol m}^{-2} \text{ d}^{-1}$  and  $3.2 \mu\text{mol m}^{-2} \text{ d}^{-1}$   
584 for DIN and DIP, respectively. Riverine inflow brings  $1195.0 \mu\text{mol m}^{-2} \text{ d}^{-1}$  DIN,  $52.9$   
585  $\mu\text{mol m}^{-2} \text{ d}^{-1}$  DIP into the lake. Lake water outflow derived nutrient loss is estimated  
586 to be  $1439.9 \mu\text{mol m}^{-2} \text{ d}^{-1}$  and  $54.7 \mu\text{mol m}^{-2} \text{ d}^{-1}$  for DIN and DIP, respectively.  
587 Nutrients in the lake can be also sourced from atmospheric deposit (mostly in form of  
588 precipitation). With the nutrient concentrations in the rain water during the monitoring  
589 period, the wet deposit is calculated to be  $76 \mu\text{mol m}^{-2} \text{ d}^{-1}$  and  $2.5 \mu\text{mol m}^{-2} \text{ d}^{-1}$ , for  
590 DIN and DIP, respectively. The loadings of DIN to the lakes are mainly from surface  
591 runoff and LGD, which comprised of 42.9 % and 53.7 % of the total DIN loadings..  
592 Groundwater derived DIP input, however, constitutes only 6.3 % of the total DIP  
593 inputs to the lake, indicating groundwater borne DIP is less contributive to the  
594 nutrient budget of the lake compared to DIN. Very recent studies on polar regions  
595 have indicated that the glacier/snowpack water is the main N sources to the proglacial



596 lakes (Anderson et al. 2013, Dubnick et al. 2017). However, they do not consider the  
597 contribution of groundwater borne N, in spite of the high groundwater connectivity in  
598 the proglacial areas (Roy and Hayashi 2008). This study stresses that groundwater  
599 borne DIN could be comparable to the surface runoff derived DIN.

600 Based on nutrient results, the lake is considered to be an oligotrophic lake and  
601 under phosphate limited condition. Thus, the primary production (PP) is therefore  
602 considered to be controlled by the DIP loadings. The sum of DIN and DIP inputs  
603 minus the sum of the calculated DIN and DIP outputs leads to surpluses of 785.4  
604  $\mu\text{mol m}^{-2} \text{d}^{-1}$  and  $3.9 \mu\text{mol m}^{-2} \text{d}^{-1}$  for DIN and DIP, respectively. The surpluses are  
605 expected to be consumed by the phytoplankton and converted into the PP under the  
606 Red Field ratio (C: N: P = 106: 16: 1), leading to a PP of  $0.41 \text{ mmol C m}^{-2} \text{d}^{-1}$ . The  
607 nutrient budgets for DIN and DIP are summarized in Figures 8b and 8c. The estimated  
608 primary productivity is lower than most temperate eutrophicated and ologotrophic  
609 lakes (Cole et al. 1998, Smith 1979), and comparable to some high latitude or altitude  
610 lakes (Richerson et al. 1986, Sterner 2010).

#### 611 4.6. Implications, prospective and limitations

612 Mountainous proglacial lakes are readily developed in glacier forefields of QTP and  
613 other high mountainous glacial such as Europe Alps and Pamir at central Asian  
614 (Heckmann et al. 2016). The proglacial lakes are always trapping system of sediment



615 and sinks for water and chemical originated from glacier/snowpack melting and  
616 groundwater. In analogous to cosmogenic isotopes such as  $^{10}\text{Be}$  serving as a tool to  
617 quantify the sediment sources, approaches integrating  $^{222}\text{Rn}$  and stable isotopes  
618 provides both qualitatively and quantitatively evaluations of groundwater  
619 contributions and hydrologic partitioning in these remote and untapped lacustrine  
620 systems. Thus, it is expected that the multiple aqueous isotopes is considered to be  
621 effective tools to investigate the LGD and hydrologic partitioning in other proglacial  
622 lakes. This study is mainly limited by the relatively short sampling and monitoring  
623 period. As a special hydrologic regime, the lake shore aquifers of the proglacial lakes  
624 are experiencing frozen-unfrozen transition seasonally, and the dominant recharge of  
625 glacial melting could be fluctuated significantly due to air temperature variation.  
626 Therefore, future groundwater and hydrological studies can be extended to longtime  
627 sampling and monitoring of stable isotopes and  $^{222}\text{Rn}$  in different water endmembers  
628 to reveal the seasonally hydrological and hydrogeological dynamics and their impacts  
629 on local biogeochemical cycles and ecological systems. Special concerns would be  
630 placed on how surface/groundwater interactions and the associated biogeochemical  
631 processes in response to the seasonal frozen ground variations and glacier/snowpack  
632 melting intensity.

633



634 **5. Conclusion**

635 A  $^{222}\text{Rn}$  continuous monitoring is conducted at Ximen Co Lake, a proglacial lake  
636 located at the east QTP. A dynamic  $^{222}\text{Rn}$  mass balance model constrained by radium  
637 mass balance and water level fluctuation is used to quantify temporal distribution of  
638 LGD of the lake. The obtained LGD over the monitoring time ranges from  $-23.7$   
639  $\text{mm d}^{-1}$  to  $80.9 \text{ mm d}^{-1}$ , with an average of  $10.3 \pm 8.2 \text{ mm d}^{-1}$ . Thereafter, a three  
640 endmember model consisting of the budgets of water, stable isotopes and  $\text{Cl}^-$  is used  
641 to depict the hydrologic partitioning of the lake. Riverine inflow, lake water outflow  
642 via surface runoff, and surface evaporation are estimated to be  $135.6 \text{ mm d}^{-1}$ ,  $141.5$   
643  $\text{mm d}^{-1}$  and  $5.2 \text{ mm d}^{-1}$ , respectively. LGD derived nutrient loading is estimated to be  
644  $785.4 \mu\text{mol m}^{-2} \text{ d}^{-1}$  and  $3.2 \mu\text{mol m}^{-2} \text{ d}^{-1}$  for DIN and DIP, respectively. Upon  
645 depicting nutrient budget within the lake, the primary productivity is estimated to be  
646  $0.41 \text{ mol C m}^{-2} \text{ d}^{-1}$ . This study also implicates that LGD constitutes relatively small  
647 portion of the proglacial hydrologic partitioning, however, delivers nearly a half of the  
648 nutrient loadings to the proglacial lake.

649 This study presents the first attempt to quantify LGD and the associated nutrient  
650 loadings to the proglacial lake of QTP. To our knowledge, there is almost no study on  
651 the groundwater-lake water interaction in the high altitude proglacial lakes in QTP.  
652 This study demonstrates that  $^{222}\text{Rn}$  based approach can be used to investigate the



653 groundwater dynamics in the high altitude proglacial lakes. The method is  
654 instructional to similar studies in other proglacial lakes in the QTP and elsewhere.

655

### 656 **Acknowledgements**

657 This study was supported by a grant from the National Natural Science Foundation of  
658 China (NSFC, No.41572208) and the Research Grants Council of Hong Kong Special  
659 Administrative Region, China (HKU17304815). The authors thank Mr. Buming Jiang  
660 for his kind help in the field works during the campaign and Ergang Lian for his help  
661 in stable isotope analysis. The authors thank Jessie Lai for her help in FIA analysis in  
662 School of Biological Sciences, HKU. Supporting data are included as in the files of  
663 supplementary information 2 and 3; Climatological data are purchased through  
664 <http://www.weatherdt.com/shop.html>; any additional data may be obtained from L.X.  
665 (email: xinluo@hku.hk);

666

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#### 1006 **Figure captions**

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1008 **Figure 1** The geological and topographic map of the Yellow River Source Region,  
1009 Nianbaoyeze glacial mountains (a), and the sampling settings of the Ximen Co Lake  
1010 (b), with the bathymetry map of the lake (d). (c) Photograph of the Ximen Co Lake  
1011 and the surrounding geomorphic settings looking northeast direction on 22 Aug 2015,  
1012 showing the late-laying snowpack in the U-shaped valleys of the north part of  
1013 Nianbaoyeze MT.

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1015 **Figure 2** The climatological parameters (wind speed, air temperature, and  
1016 precipitation) in the Aug, 2015 recorded from Jiuzhi weather station.

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1018 **Figure 3** The temporal distributions of  $^{222}\text{Rn}$  (**a**), water temperature (**b**), water level  
1019 fluctuation recorded by the divers (**c**), and hourly wind speed and air temperature  
1020 recorded in Jiuzhi weather station (**d**).

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1022 **Figure 4** The cross plots of  $\text{Cl}^-$  versus  $\text{Na}^+$  (**a**),  $\text{SO}_4^{2-}$  versus  $\text{Cl}^-$  (**b**),  $\text{Ca}^{2+}$  versus  $\text{Cl}^-$   
1023 (**c**); The relations of  $^2\text{H}$  versus  $^{18}\text{O}$  (**d**),  $\text{Cl}^-$  versus  $^2\text{H}$  (**e**), and  $\text{Cl}^-$  versus  $^{18}\text{O}$  (**f**).

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1025 **Figure 5** Cross plots of  $^{222}\text{Rn}$  versus DIN (**a**) and DIP (**b**).

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1027 **Figure 6** The conceptual model of  $^{222}\text{Rn}$  transient model (**a**), and three endmember  
1028 model (**b**).

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1030 **Figure 7** The results of the final LGD derived from  $^{222}\text{Rn}$  transient model.

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1032 **Figure 8** The hydrologic partition of the proglacial lake of Ximen CO (**a**), and the  
1033 budgets of DIN (**b**) and DIP (**c**).

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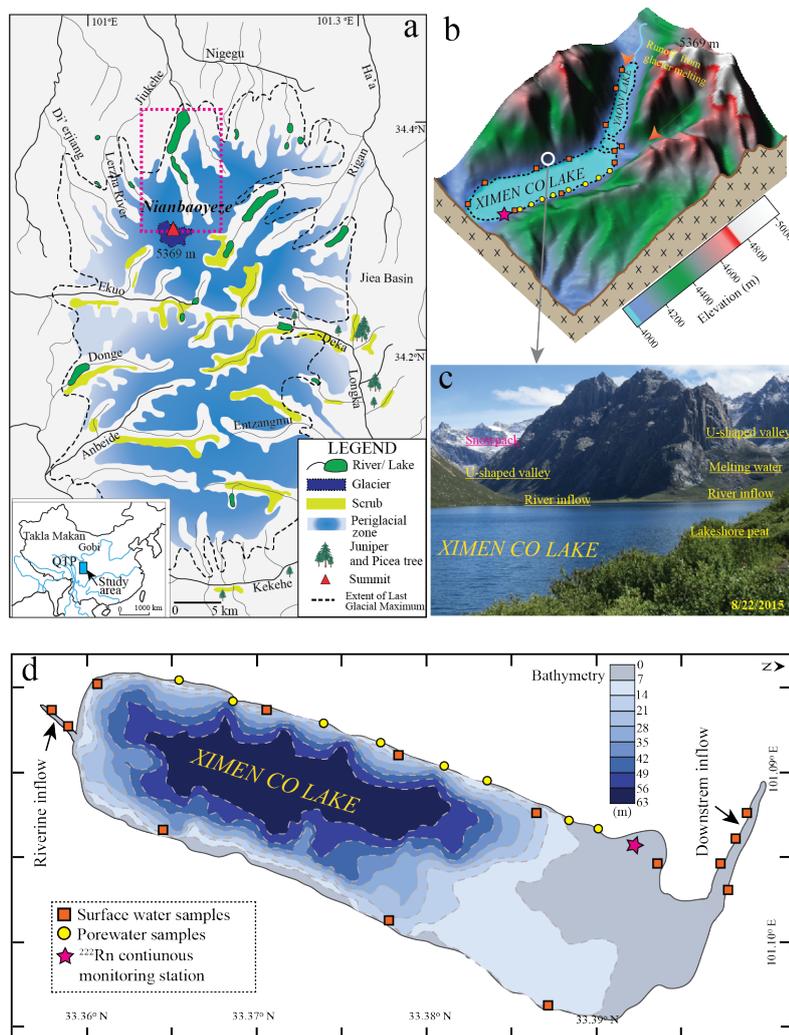
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1044 Figure 1



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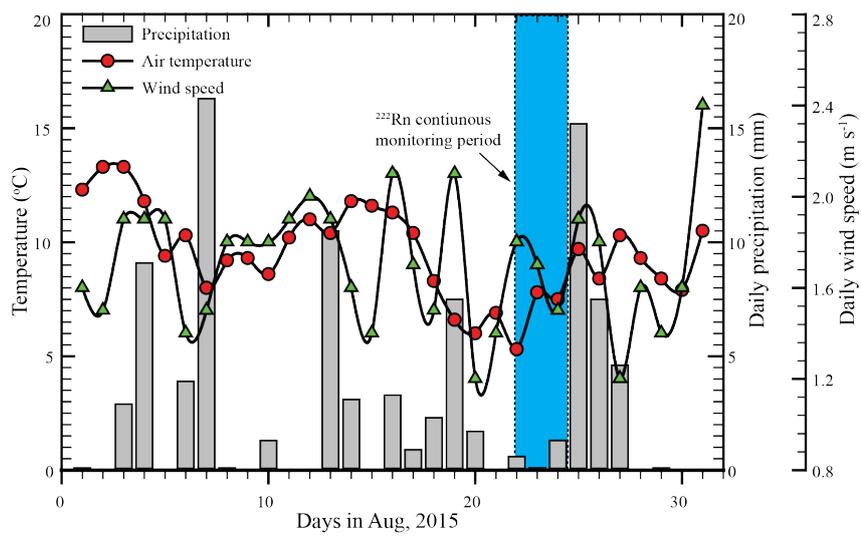
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1050 Figure 2



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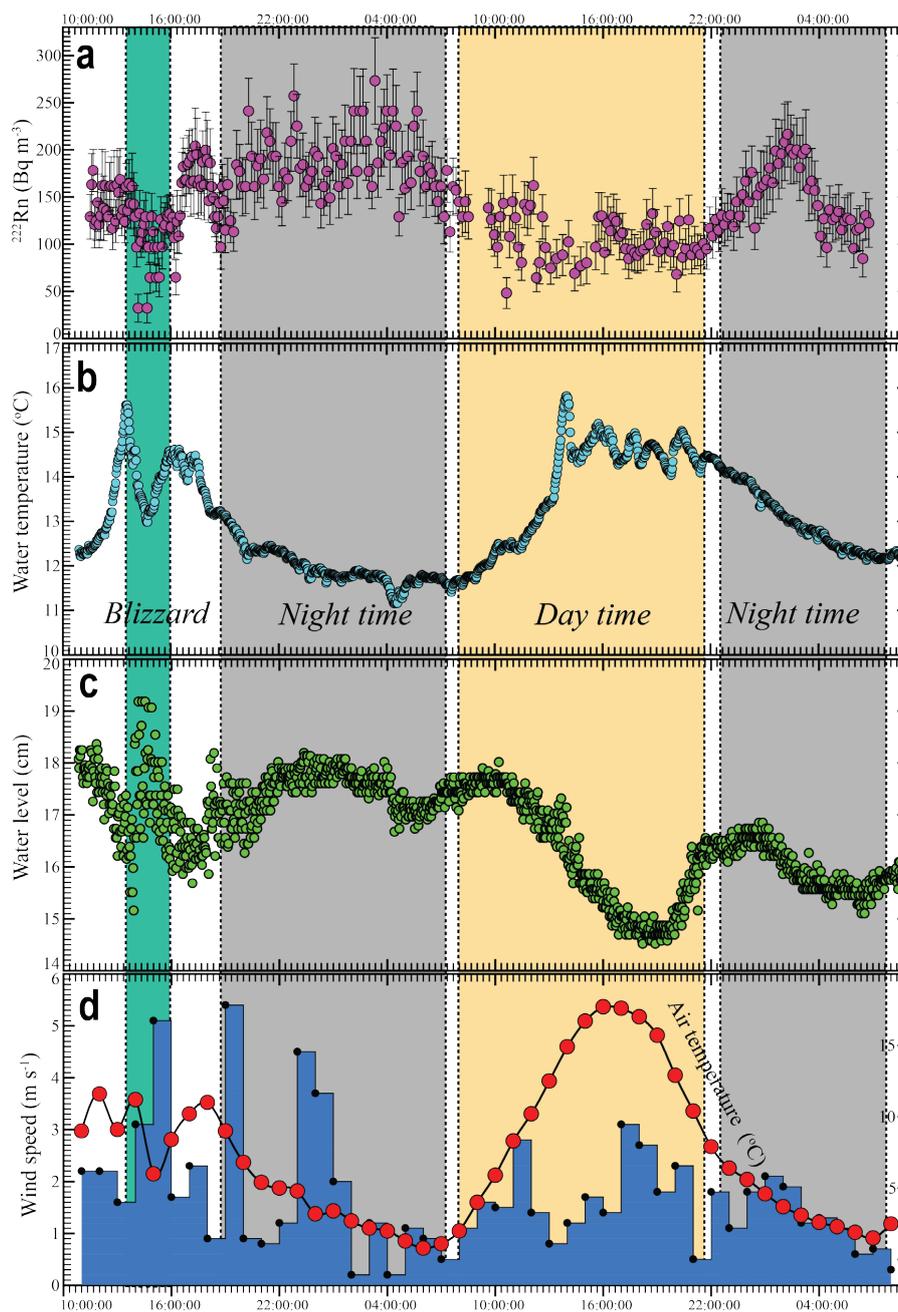
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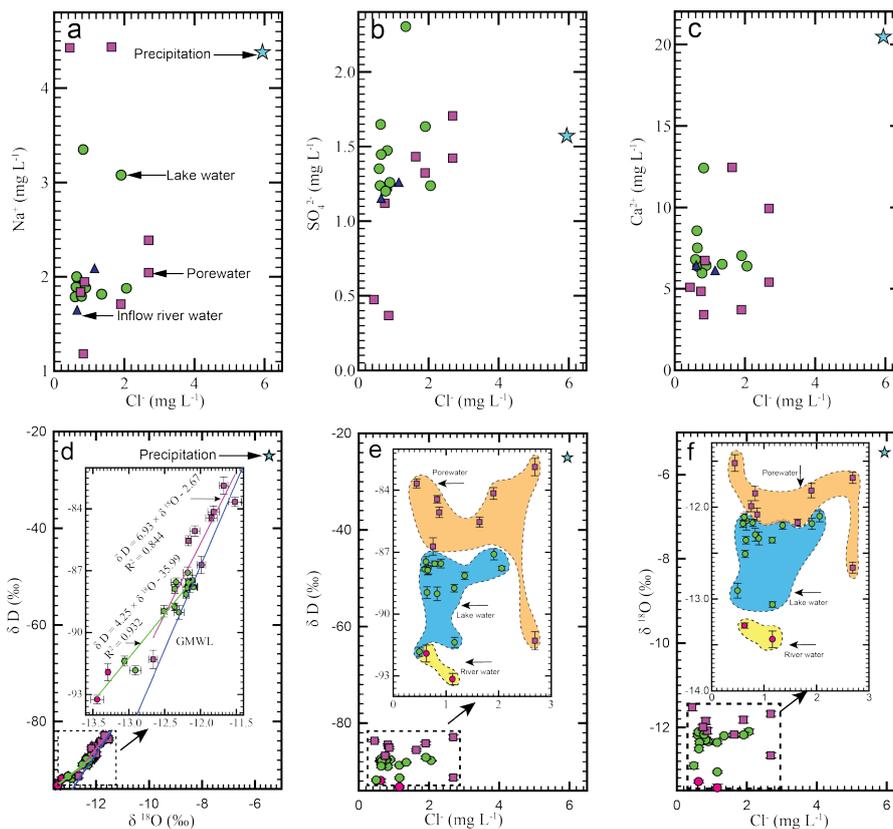


1063 Figure 3



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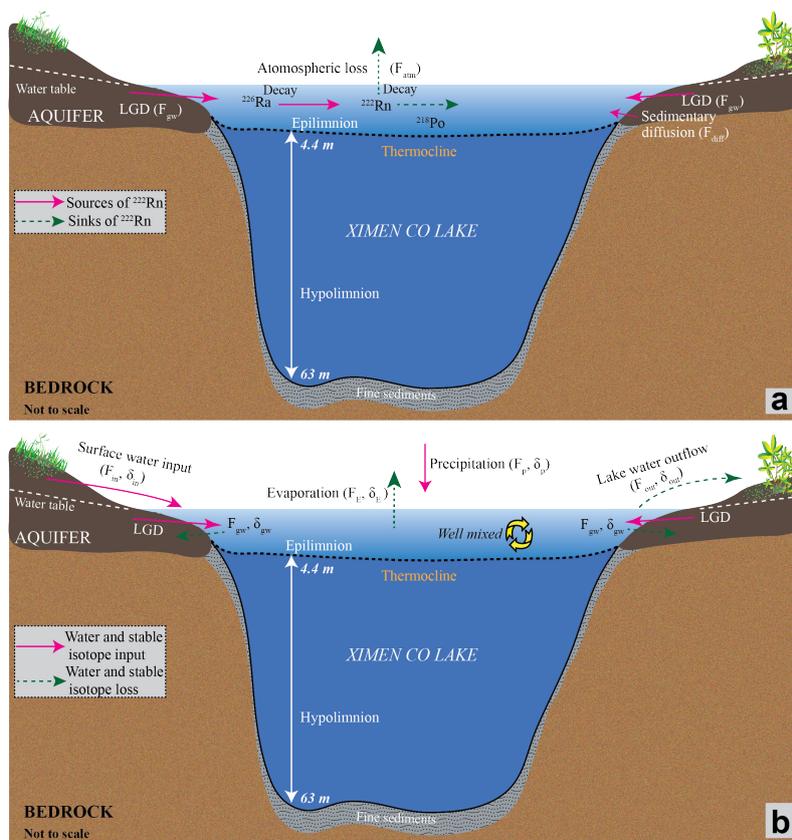
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1075 Figure 6



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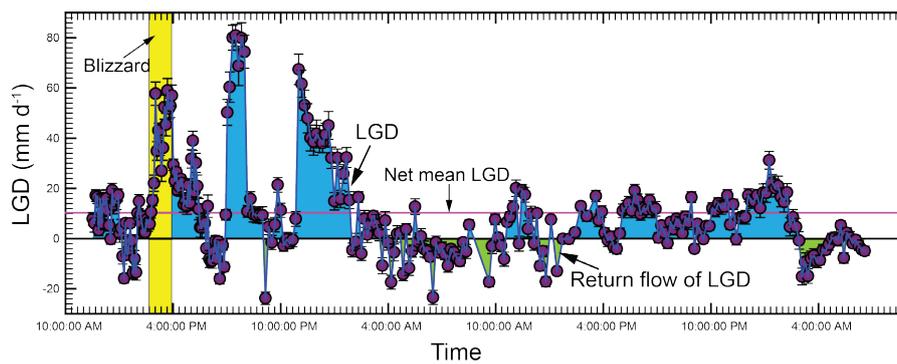
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1084 Figure 7



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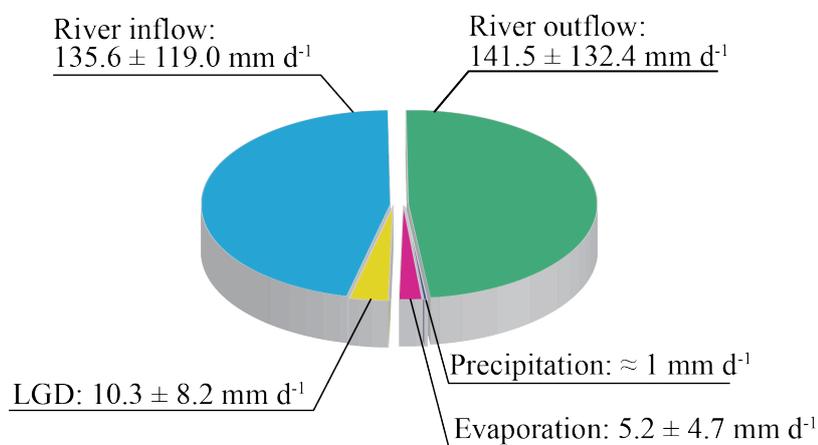
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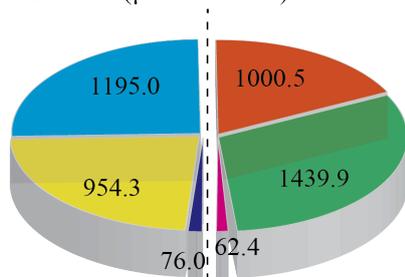


1099 Figure 8

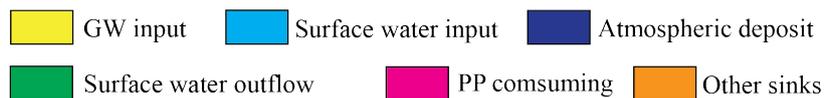
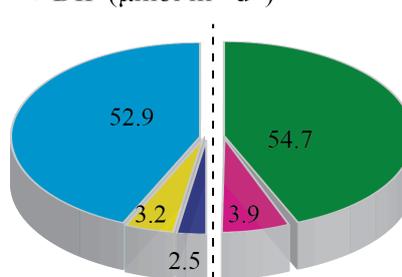
**a** Hydrologic partitioning



**b** DIN ( $\mu\text{mol m}^{-2} \text{d}^{-1}$ )



**c** DIP ( $\mu\text{mol m}^{-2} \text{d}^{-1}$ )



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**Table 1** Parameters used to establish the mass balance model of  $^{222}\text{Rn}$  in Ximen Co Lake.

Parameters	Units or values	Estimated Uncertainty (%)	Evaluation
Wind speed ( $\omega_m$ )	0.2 - 5.4 m s <sup>-1</sup>	n.a	From Jiuzhi weather stations
Water-air temperature	11.2 - 15.6 °C	n.a	Recorded with probe in the chamber; sensitive to temperature results
Molecular diffusion of $^{222}\text{Rn}$ in water ( $D_m$ )	$9.2 \times 10^{-6}$ - $1.0 \times 10^{-5}$ cm <sup>2</sup> s <sup>-1</sup>	n.a	$1.16 \times 10^{-6}$ at 20 °C; adjustable for temperature
Molecular diffusion of $^{222}\text{Rn}$ in sediments ( $D_s$ )	$2.2 \times 10^{-6}$ - $2.5 \times 10^{-5}$ cm <sup>2</sup> s <sup>-1</sup>	n.a	Adjusted for temperature, sediment porosity
Dynamic viscosity ( $\mu$ )	$1.1 \times 10^{-3}$ - $1.3 \times 10^{-3}$ cm <sup>2</sup> s <sup>-1</sup>	n.a	Calculated based on water temperature, density and salinity
Schmidt number ( $S_c$ )	1078.6 - 1371.6 [-]	n.a	Calculated as the ratio of $\nu$ to $D_m$
Water depth ( $H$ )	4.4 m	n.a	Epilimnion depth of Ximen Co Lake
Decay constant $^{222}\text{Rn}$ ( $\lambda_{222}$ )	0.186 d <sup>-1</sup>	n.a	Constant
Groundwater endmember $^{222}\text{Rn}$ ( $C_{gw}$ )	$11200 \pm 1200$ Bq m <sup>-3</sup>	8 sites dependant for	Measured; final result for water flux inversely proportional to $^{222}\text{Rn}$
Lake water endmember $^{222}\text{Rn}$ ( $C_l$ )	21.6 - 418.8 Bq m <sup>-3</sup>	15-25%	Measured with RAD 7 AQUA
Atmospheric $^{222}\text{Rn}$ ( $C_a$ )	$1.5 \pm 1.0$ Bq m <sup>-3</sup>	20-25%	Measured or assumed value, model not sensitive to radon in air variation
$K_{\text{air/water}}$	0.29 - 0.33 [-]	n.a	Calculated based on temperature in the chamber and salinity in lake water
Porosity $n$	0.31	n.a	Assumed based on literatures
Tortuosity $\theta$	2.05	n.a	Calculated based on porosity
Piston velocity ( $K$ )	0.004 - 1.11 m d <sup>-1</sup>	20-25%	Calculated from Equation 3 in supplementary information
$^{226}\text{Ra}$ concentration in lake waters ( $C_{226\text{Ra}}$ )	0.01 Bq m <sup>-3</sup>	≈10%	Measured with RAD7
Diffusive flux of $^{222}\text{Rn}$ ( $F_{\text{diff}}$ )	0.68 - 213.5 Bq m <sup>-2</sup> d <sup>-1</sup>	n.a	Calculated from Equation 9 in supplementary information
Atmospheric flux of $^{222}\text{Rn}$ ( $F_{\text{atm}}$ )	0.7 - 213.5 Bq m <sup>-2</sup> d <sup>-1</sup>	n.a	Calculated from Equation 1 in supplementary information
Groundwater flux of $^{222}\text{Rn}$ ( $F_{\text{gw}}$ )	14.7 - 349.8 Bq m <sup>-2</sup> d <sup>-1</sup>	n.a	Calculated from Equation 1
Inventories of $^{222}\text{Rn}$ ( $I$ )	Bq m <sup>-2</sup>	n.a	Measured with RAD7 AQUA
Groundwater discharge ( $Q_{\text{gw}}$ )	$10.3 \pm 8.2$ (3.5-38.6) mm d <sup>-1</sup>	n.a	Calculated from Equation 1



**Table 2** Input parameters for the three endmember model of Ximen Co. Lake

Input parameter	Description	Values (using $^{18}\text{O}$ as a tracer)	Parametric sources
$h$	Relatively humidity	0.63	Measured by the humidity meter
$T$ (°C)	Water temperature	15.66	Monitored with divers
$\delta_{\text{surface}} (^{18}\text{O})$ ‰	Surface water isotopic compositions	-12.45	Average value of surface inflow samples
$\delta_{\text{gw}} (^{18}\text{O})$ ‰	Groundwater isotopic compositions	-11.97	Average value of porewater samples
$\delta_L (^{18}\text{O})$ ‰	Lake water isotopic compositions	-12.54	Average value of Ximen Co. Lake water samples
$F_{\text{gw}}$ (mm/d)	LGD rates	14.18	Calculated based on $^{222}\text{Rn}$ mass balance model
$\varepsilon^* (^{18}\text{O})$ ‰	Effective equilibrium isotopic enrichment factor	10.12	Equations 13-14 in supplementary information
$C_k (^{18}\text{O})$ ‰	Kinetic constant for $^{18}\text{O}$	14.2	Constants based on evaporating experiment
$\varepsilon_k (^{18}\text{O})$ ‰	Kinetic enrichment factor	5.2	From Equation 15 in supplementary information
$\varepsilon$ (‰)	Total isotopic enrichment factor	15.33	The sum of $\varepsilon^*$ and $C_k$
$\alpha^* (^{18}\text{O})$ ‰	Effective isotopic equilibrium factor	1.01	$\alpha^* = 1 + \varepsilon^*$
$\delta_a (^{18}\text{O})$ ‰	Isotopic composition of ambient air	-23.12	Estimated with $\delta_{\text{in}}$ and $\delta_a$
$\delta_{\text{in}} (^{18}\text{O})$ ‰	Isotopic composition of surface inflow water	-13.41	Average value of surface inflow water
$\delta_E (^{18}\text{O})$ ‰	Isotopic compositions of evaporating vapor	-35.1	From Equation 12 in supplementary information
$[CT]_{\text{in}}$ (mgL $^{-1}$ )	Chloride concentrations in surface inflow water	0.91	Filed data
$[CT]_L$ (mgL $^{-1}$ )	Chloride concentrations in lake water	1.02	Filed data
$[CT]_{\text{gw}}$ (mgL $^{-1}$ )	Chloride concentrations in groundwater	1.48	Filed data