



1 **Hydrochemical and isotopic evidences for deciphering conceptual model of groundwater**
2 **salinization processes in a coastal plain, north China**

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8

9 **Abstract**

10 Groundwater is the important water resource for agricultural irrigation, urban and tourism development and
11 industrial utilization in the coastal region of north China. In the past five decades, coastal groundwater
12 salinization in the Yang-Dai River coastal plain has become more serious than ever before under natural
13 climate change and anthropogenic activities. It is pivotal for the scientific management of coastal water
14 resources to accurately understand groundwater salinization processes and its inducement. Hydrochemical
15 and stable isotopic ($\delta^{18}\text{O}$ and $\delta^2\text{H}$) analysis for the different water bodies (surface water, groundwater,
16 geothermal water, and seawater) were applied to provide a better understanding of the processes of
17 groundwater salinization in the Quaternary aquifers. Saltwater intrusion is the major aspect and can be
18 caused by vertical infiltration along the riverbed at the downstream areas of rivers during the tide/surge
19 period, and lateral inflow into fresh aquifer derived from intensively pumping groundwater. Seawater
20 proportion can reach ~13% in the well field. High mineralized geothermal water (TDS up to 10.6 g/L) with
21 the indicator of paleoseawater relics (lower Cl/Br ratios relative to modern seawater) overflows into the
22 cold Quaternary aquifers. Groundwater salinization can also be exacerbated by the anthropogenic activities
23 (e.g., irrigation return-flow with solution of fertilizers, domestic wastewater discharge). Additionally, the
24 interaction between surface water and groundwater can make the groundwater freshening or salinizing in
25 different sections to locally modify the groundwater hydrochemistry. The cease of the well field and
26 establishment of anti-tide dam in the Yang River estuary area have effective function to contain the
27 development of saltwater intrusion. This study can guide the future water management practices, and
28 provide research approaches and foundation for further investigation of seawater intrusion in this and
29 similar region.

30 **1. Introduction**

31 Coastal region is the key area for the world social and economic development. Approximately 40% of
32 the world's population lives within 100 kilometers of the coast (UN Atlas, 2010). The worldwide coastal



33 area has become increasingly urbanized with 14 of the world's 17 largest cities located along coasts (Creel
34 L., 2003). China has 18,000 km of continental coastline, about 164 million people (about 12% of total
35 Chinese population) live in 14 coastal provinces, and nearly 80% of them distribute in the three coastal
36 economic regions, namely Beijing-Tianjin-Hebei economic region, the Yangtze River delta economic
37 region and the Pearl River delta economic region (Shi, 2012). The rapid economic development and the
38 growing population in the coastal region have increased demands for fresh water, meanwhile been
39 confronted with the threat from waste and sewage water discharge into coastal ecosystems.

40 Coastal groundwater resources play crucial roles on the social economic and ecologic function in the
41 global coast system (IPCC, 2007). Coastal groundwater system connects the ocean and the continental
42 hydro-ecological systems (Moore, 1996; Ferguson and Gleeson, 2012). Groundwater as an important
43 freshwater resource could be over extracted due to that the periods of highest demand (e.g., agricultural
44 irrigation and tourist seasons) are often the periods of lowest recharge rates (Post, 2005). In addition to
45 occurrence of some environmental issues, such as land subsidence, contaminants transport, the
46 over-exploitation of groundwater can readily result in seawater intrusion in the coastal area. Seawater
47 intrusion has become a global issue and the related studies can be found from the coastal aquifer system of
48 different countries, such as Israel (Sivan et al., 2005; Yechieli et al., 2009; Mazi et al., 2014), Spain (Price
49 and Herman, 1991; Pulido-Leboeuf, 2004; Garing et al., 2013), France (Barbecot et al., 2000; de Montety
50 et al., 2008), Italy (Giambastiani et al., 2007; Ghiglieri et al., 2012), Morocco (Bouchaou et al., 2008; El
51 Yaouti et al., 2009), USA (Gingerich and Voss, 2002; Masterson, 2004; Langevin et al., 2010), Australia
52 (Zhang et al., 2004; Narayan et al., 2007; Werner, 2010), China (Xue et al., 2000; Han et al., 2011, 2015),
53 Vietnam (An et al., 2014), Indonesia (Rahmawati et al., 2013), India (Radhakrishna, 2001; Bobba, 2002),
54 Brazil (Gico Montenegro et al., 2006; Cary et al., 2015), etc. Werner et al. (2013) gave an excellent review
55 on seawater intrusion processes, investigation and management. A variety of approaches have been used to
56 investigate seawater intrusion, including head measurement, geophysical methods, geochemical methods
57 (environmental tracers combined hydrochemical and isotope data), conceptual and mathematical modeling
58 (see reviews by Jones et al., 1999; Werner et al., 2013).

59 Seawater/saltwater intrusion is a complicated hydrogeological process due to the impact of aquifer
60 properties, anthropogenic activities (e.g., intensive groundwater pumping, irrigation practices), recharge
61 rate, variable density flow between the estuary and adjacent fresh groundwater system, tidal/surge activity
62 and global climate change (Ghassemi et al., 1993; Robinson et al., 1998; Smith and Turner, 2001; Simpson



63 and Clement, 2004; Narayan et al., 2007; Werner and Simmons, 2009; Wang et al., 2015). Brockway et al.
64 (2006) reported the negative relationship between saltwater intrusion length and river discharge.
65 Understanding the complex interactions between groundwater and surface water, groundwater and seawater
66 is essential for the effective management of water resources (Sophocleus, 2002; Mondal et al., 2010). There
67 was a vastly different result based on numerical simulations for the additional distance of intrusion in the
68 Nile Delta Aquifer of Egypt and in the Bay of Bengal under the same sea level rise (Sherif and Singh,
69 1999). Bobba (2002) also employed numerical simulations to demonstrate an apparent risk of saltwater
70 intrusion in the Godavari delta, India due to sea level rise. Westbrook et al. (2005) defined the hyporheic
71 transition zone of mixing between river water and groundwater influenced by tidal fluctuations and the
72 contaminant distribution. Modelling seawater intrusion in the Burdekin Delta irrigation area, North
73 Queensland (Australia) show that seawater intrusion is far more sensitive to pumping rates and recharge
74 than to aquifer properties (e.g., hydraulic conductivity), and compared to the effects of groundwater
75 pumping, the effect of tidal fluctuations on saltwater intrusion can be neglected (Narayan et al., 2007).
76 However, rare studies have focused on delineating the interactions among surface-ground-sea-waters in
77 estuarine environment and the effects of vertical infiltration of seawater into the off-shore aquifer through
78 river channel vs. the lateral landward migration of the freshwater-saltwater interface.

79 The data of China's marine environment bulletin released on March 2015 by State Oceanic
80 Administration People's Republic of China showed that the major bays, including Bohai Bay, Liaodong
81 Bay, Hangzhou Bay, are polluted seriously with the inorganic nitrogen and active phosphate as the major
82 pollutants (SOA, 2015). Seawater intrusion in China is the most serious around the Circum-Bohai-Sea
83 region. The escalating seawater intrusion in the future may be not a simple problem related to groundwater
84 salinization in this region. It is likely to be more difficult to remediate groundwater pollution caused by the
85 contaminated seawater. This study will take the Yang-Dai River coastal plain in Qinhuangdao City, Hebei
86 province of north China, as an example to investigate groundwater salinization processes and interactions
87 among surface water, groundwater and seawater, and the seawater intrusion caused by groundwater
88 exploitation in Zaoyuan well field. Qinhuangdao is an important port and tourist city of northern China. In
89 the past 30 years, many previous studies had done to investigate distribution of seawater intrusion and its
90 influence factors using hydrochemical analysis of groundwater (Xu, 1986; Yang, 1994, 2008; Chen and Ma,
91 2002; Sun and Yang, 2007; Zhang, 2012) and numerical simulations (Han, 1990; Bao, 2005; Zuo, 2009).
92 This study is a continuation of previous investigations of the coastal plain aquifers in Qinhuangdao.



93 Hydrochemical and stable isotopic compositions of collected water samples were analyzed for making up
94 the knowledge gap of surface-, ground-and sea-water interactions in this region. This study aims to describe
95 the conceptual model of the complex processes for the groundwater salinization of the coastal aquifers, to
96 reveal the major aspects responsible for the increasing groundwater salinity in the coastal aquifers, and to
97 obtain a conceptual model for deciphering the groundwater flow system of the study area. The results will
98 be helpful for the further numerical simulations of coastal groundwater system. It is very significant for
99 water resources management in the coastal plain.

100 **2. Study area**

101 The Yang-Dai River coastal plain (Fig. 1) covers approximately 200km² in the west side of Beidaihe
102 District of Qinhuangdao City, the northeastern Hebei Province. It connects the eastern section of Yanshan
103 Mountain and surrounded by mountains. The southern boundary of the study area is Bohai Sea. The plain
104 become low from northwest to southeast and fan-shaped distribution of the piedmont- coastal inclined
105 alluvial plain. Elevation ranges from 390 in the west and to 40-100 m in the north, and 25-40 in the east,
106 and 25-1m in the south coastal region, with the average slope of 0.008. Zaoyuan well field, located in the
107 southern edge of alluvial fan, was built in 1959 (Xu, 1986) as major water supply for this region. It is 4.3
108 km from the southeastern well field to Yang River estuary.

109 **2.1 Climate and hydrology**

110 The study area is in a warm and semi-humid monsoon climate. On the basis of a 56-a record in
111 Qinhuangdao area, the mean annual rainfall is estimated to be 640 mm, the average annual temperature is
112 about 11°C, and mean potential evaporation of 1469 mm. 75% of the total annual rainfall falls in
113 July-September (Zuo, 2006). The average annual tide level is 0.86m (meters above Yellow Sea base level),
114 the highest tide is 2.48m, and the lowest is -1.43m.

115 The Yanghe River and Daihe River originated from the Yanshan Mountains are the major surface
116 water body in this area (Fig. 1). The river soared when heavy rains happened with short peak duration,
117 whereas it became minimal flow or drying during the dry season. The Yang River is about 100 km long
118 with the catchment area of 1029 km², and the average annual runoff of 1.11×10^8 m³/a (Han, 1988). Dai
119 River has the length 35 km and catchment area of 290 km², with annual runoff of 0.27×10^8 m³/a and
120 average gradient of 11.4‰. The two rivers flow into the southern Bohai Sea.



121 **2.2 Geological and hydrogeological setting**

122 Groundwater in this area mainly include fissure water in the bedrock and water in the Quaternary
123 porous media. The bedrock fissure water is distributed in the north platform area. Its water abundance is
124 mainly depended on the degree of weathering and the nature and regularity of fault zone. The strata
125 outcropping in the west, north and eastern edge of the plain includes the Archean gneiss, Proterozoic mixed
126 granite and Jurassic sandstone, shale and so on. The ex-Quaternary, which is exposed in the offshore area of
127 the region, is mainly the Archean metamorphic granite, which is widely distributed. The mineral
128 composition includes mainly quartz, feldspar, and biotite. The Quaternary sediments of the plain are mostly
129 underlain by the Archean gneisses and Proterozoic mixed granites. The basement faults under the
130 Quaternary cover mainly include the NE-trending fault and the NW-trending fault. The Quaternary aquifer
131 system of the Yang-Dai River coastal plain is a complete groundwater system from the piedmont to the
132 coast (see P-P' cross-section of Figure 2). Geological technics control the development and deformation of
133 the sediments, the distribution of hot springs and geothermal anomalies. Fault zones are also the main
134 channel for deep-water cycle and thermal convection.

135 The Quaternary sediments are widely distributed in the area. The bottom of the Holocene in most
136 areas has clay or clay layers, which make the groundwater in the coastal zone under confined or
137 semi-confined status. There are no regional, continuous aquitards between several aquifers. The thickness
138 of the Quaternary strata has a range of 5-80 m, mostly 20-40 m, and up to more than 100 m near the
139 coastline. The aquifer is mainly composed of medium sand, coarse sand and gravel with thickness of 10-20
140 m and water table depth of 1-4 m in the phreatic aquifer, and thickness of 10-30 m and water table depth of
141 1-5 m in the confined aquifer (Zuo, 2006). In the yearly peak season of agricultural water, the groundwater
142 level decline sharply and reaches the lowest water table in April-May period, and become highest in
143 January-February. The main sources of aquifer recharge are from rainfall infiltration, river water and
144 irrigation return-flow, lateral subsurface runoff from the piedmont area. Apart from the phreatic water
145 evaporation, groundwater pumping it the main pathway of groundwater discharge for agricultural,
146 industrial, tourism and sanatorium's utilization. The general flow direction of groundwater is from
147 northwest to south. Naturally, groundwater discharges into the river and the Bohai Sea.

148 The geothermal water near the fault zone discharges into shallow Quaternary sediments, which is the
149 overlying strata in geothermal anomalous area (Hui, 2009). The temperature of thermal water range is
150 27-57 °C in this low-to-medium temperature geothermal field (Zeng, 1991). The thickness of the overlying



151 strata is varied from 24.6 to 58.8 m and consists of alluvial sand, gravel, clayey loam, clay and silt. The
152 thermal water stored in the Archeozoic granite and metamorphic rocks, which are composed of migmatite,
153 gneiss, and amphibole plagioclase (Pan, 1990). Major deep fracture zones are the good passage for the
154 geothermal water movement (Yang, 2011). The heated groundwater in the deep zones could upward
155 transport along the fault and mix with the cold groundwater in the Quaternary aquifers (Shen et al., 1993).

156 **2.3 Environmental issues and seawater intrusion history**

157 The shallow groundwater pumped from the Quaternary aquifer occupies 94% of the total groundwater
158 exploitation, which is used for agricultural irrigation (accounts for 52% of the total groundwater use),
159 industrial (32%) and domestic water (16%) (Meng, 2004). Many large and medium-sized reservoirs were
160 built in the 1960s and 1970s and resulted in that the surface water was intercepted and the downstream
161 runoff dropped sharply, even became dry in drought years. With the intensification of human
162 socio-economic activities and growing urbanization, coupled with extended drought years (severe drought
163 during 1976-1989 in north China) (Wilhite, 1993; Han et al., 2015), increased groundwater exploitation to
164 meet the ever-growing fresh water demands has resulted in groundwater level declining and seawater
165 intrusion (SWI) in the coastal aquifers.

166 The pumping rate in the Zaoyuan well field was gradually increased from 1.25 million m³/a in the
167 early 1960s to 3.5 million m³/a in the late 1970s, and beyond 10 million m³/a in the 1980s. During
168 1966-1989, the major agricultural planting in this region is paddy field with big water consumption. The
169 groundwater pumping time is mainly from May to October with pumping rate of 7~80,000 m³/d, which
170 was over-exploited and resulted in formation of groundwater level declining depression. In May 1986, the
171 groundwater level in the depression center, which is located in Zaoyuan-Jiangying, was decreased to below
172 -2 m.a.s.l. (meters above sea level), with the depression area, which has groundwater level below the sea
173 level, covered 28.2 km². Since 1990, the rapid development of township enterprises in the 1980s (mainly
174 refer to paper mills), groundwater over-exploitation in the west area (i.e. the groundwater pumping rate for
175 paper mill development reached 55,000 m³/d in 2002) resulted in the groundwater level depressions around
176 Liushouying and Fangezhuang. The lowest groundwater level in the depression center in 1991 was up to
177 -11.6 m.a.s.l., and -17.4 m.a.s.l. in 2002. After the implementation of “Transferring Qing River water to
178 Qinhuangdao” project since 1992, the intensity of groundwater pumping became slowed down. The
179 depression center moved to Liushouying area. The groundwater level of the depression center was



180 recovered to -4.3 m.a.s.l. in July 2006. The shape of the depression was elliptical with the major axis of the
181 EW direction. The depression area developed to 132.3km² in May 2004.

182 The groundwater quality of this area has become gradually salinized since the early 1980s. Chloride
183 concentrations increased year by year. As early as 1979, seawater intrusion occurred in the Zaoyuan well
184 field. The intrusion area with groundwater chlorine concentration greater than 250 mg/L has been
185 developed to 21.8 km² in 1984, and 32.4 km² in 1991, 52.6 km² in 2004, 57.3 km² in 2007 (Zuo, 2006;
186 Zang et al., 2010). The chloride concentration of groundwater pumped from this well field changed from 90
187 mg/L in 1963, 218 mg/L in 1978, 567 mg/L in 1986, 459 mg/L in 1995, and 1367 mg/L in 2002 (Zuo,
188 2006), 812 mg/L in July 2007 (this study). The distance of seawater intrusion into the inland reached 6.5
189 km inland in 1991, and developed to 8.75 km in 2008 (Zang et al., 2010). At the early 1990s, 16 of 21
190 pumping wells in the well field have been abundant due to the salinized water quality (Liang et al., 2010).
191 370 of 520 pumping wells has been abundant in the Yang-Dai River coastal plain during 1982-1991 (Zuo,
192 2006).

193 3. Methods

194 Totally 80 water samples were collected from the Yang-Dai River coastal plain, including 58
195 groundwater samples, 19 river water samples (from 12 sites), and 3 seawater samples during three sampling
196 campaigns, namely, June 2008, September 2009 and August 2010. Groundwater samples pumped from 28
197 productive wells with well depth of 6-110m, including 7 deep wells, which has well depth more than 60m.
198 The water sampling sites can be shown in Figure.1. In this study, we investigated cold groundwater from
199 the productive wells. However, the geothermal water existing around Danihe cannot be ignored. The related
200 data can be available and referenced from Zeng (1991) due to that we cannot obtain the hot water samples
201 from the current geothermal field.

202 Measurements of some physical-chemical parameters (i.e. pH, temperature, and electrical conductivity
203 (EC)) were conducted in situ using portable meter (WTW Multi 3500i). All water samples were filtered to
204 0.45 μm membrane filters before collection for analysis of hydrochemical composition. Two aliquots in
205 polyethylene 100mL bottles at each site were collected: for major cation and anion analysis, respectively.
206 Samples for cation analysis (Na, K, Mg and Ca) were added 6 N HNO₃ to prevent precipitation. Water
207 samples were sealed and stored at 4 °C until determination. Biocarbonates were determined by titration
208 within 12 hours after sampling. The concentrations of cations and some trace elements (i.e. B and Sr) were



209 analyzed by inductively coupled plasma-optical emission spectrometry (ICP-OES) on filtered samples in
 210 the chemical laboratory of the Institute of Geographic Sciences and Natural Resources Research (IGSNRR),
 211 Chinese Academy Sciences (CAS). Concentrations of major anions (i.e. Cl, SO₄, NO₃ and F) were analyzed
 212 by a High Performance Ion Chromatograph (SHIMADZU, LC-10ADvp) at the IGSNRR, CAS. The ion
 213 balance errors of the chemical results are less than 8%. The hydrochemical and physical data are shown in
 214 Table 1. The stable isotopes ($\delta^{18}\text{O}$ and $\delta^2\text{H}$) of water samples were measured by a Finnigan MAT 253 mass
 215 spectrometer after on-line pyrolysis with a Thermo Finnigan TC/EA in the Stable Isotopes Laboratory of
 216 the IGSNRR, CAS. The results of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values shown in Table 1 were expressed in ‰ relative to
 217 international standards (V-SMOW (Vienna Standard Mean Ocean Water)). The analytical precision for $\delta^2\text{H}$
 218 is $\pm 2\text{‰}$ and for $\delta^{18}\text{O}$ is $\pm 0.5\text{‰}$.

219 Saturation indices for common minerals (i.e. calcite, dolomite, and gypsum) were calculated using
 220 PHREEQC version 2.8 (Parkhurst and Appelo, 1999) to understand the saturation status of these minerals
 221 in the aquifer. Ionic delta values were calculated to further investigate the hydrogeochemical behavior that
 222 take place in the aquifer and modify groundwater hydrochemistry. The ionic delta values express
 223 enrichment or depletion of each ion's concentration relative to its theoretical concentration calculated from
 224 the Cl⁻ concentration of the sample for a conservative freshwater-seawater mixing system (Fidelibus et al.,
 225 1993; Appelo, 1994). The delta values have been used as effective indicators of coastal groundwater
 226 undergoing freshening or salinizing processes, accompanied by related water-rock interaction (prevalingly
 227 cation exchange). Cl⁻ can be regarded as a conservative tracer for the calculations mentioned below. The
 228 seawater contribution for each sample can be expressed by a fraction of seawater (f_{sw}), which can be
 229 calculated using (Appelo and Postma, 2005):

$$230 \quad f_{sw} = \frac{C_{Cl,sam} - C_{Cl,f}}{C_{Cl,sw} - C_{Cl,f}} \quad (1)$$

231 where $C_{Cl,sam}$, $C_{Cl,f}$ and $C_{Cl,sw}$ refer to the Cl⁻ concentration in the sample, freshwater, and seawater,
 232 respectively. Based on the f_{sw} value, the theoretical concentration ($C_{i,mix}$) of each ion in a water sample can
 233 be calculated by:

$$234 \quad C_{i,mix} = f_{sw} \cdot C_{i,sw} + (1 - f_{sw}) \cdot C_{i,f} \quad (2)$$

235 $C_{i,sw}$, and $C_{i,f}$ refer to the measured concentration of the ion i in the seawater and freshwater,
 236 respectively. The ionic delta value (ΔC_i) of ion i can be obtained by:



$$\Delta C_i = C_{i,sam} - C_{i,mix} \quad (3)$$

238 $C_{i,sam}$ - the measured concentration of the ion i in the water sample.

239 4. Results

240 4.1 Groundwater dynamics

241 Due to the different groundwater pumping rate and patterns, the variation trend of groundwater level
242 has been different in the east and west areas of the Yang-Dai River coastal plain. In the east part, owing to
243 the intensive exploitation in the Zaoyuan well field, the groundwater level was gradually declined to be
244 lower than the sea level during the 1980s. The center of groundwater level depression was located in
245 Zaoyuan-Jiangying region, with the groundwater level lower than -3 m.a.s.l. The local government
246 commenced to reduce the exploitation after 1992. The groundwater level decreased slowly after 1995, even
247 started to recovery in some wells as a result of pumping reduction. During the extreme drought year (1999),
248 the consequential increased water demand made the groundwater level declined again in the east region. In
249 the late 1980s, the groundwater level at the west region was still more than 0 m.a.s.l. But in the late 1990s,
250 due to the fast development of the local paper mills as the big water consumers, the groundwater level
251 dropped year by year and had big falling amplitude after 2000, resulting in the overall transfer of
252 groundwater depression center to the western region (Liushouying-Fangezhuang). The groundwater level in
253 this center was up to -14 m.a.s.l. in May 2002.

254 Based on the data from the three monitoring wells, the seasonal variation of groundwater level in this
255 area can be seen from Figure 3. After 2000, the groundwater level in the east of the Yang-Dai River coastal
256 plain was mainly affected by the groundwater pumping for agricultural and domestic water use. During
257 March and June of each year, the shallow groundwater pumping as the major water source for irrigation has
258 resulted in the fast dropped water level occurred between April and June, down to the lowest level of water
259 throughout the year. As the rainy season started in July, groundwater pumping began to decrease.
260 Groundwater level rise rapidly with the infiltration of irrigation return-flow and rainfall, lateral subsurface
261 runoff from the surrounding aquifers. After the end of the rainy season (July to September), the water level
262 continues to rise gently and reach the annual maximum water level during January and February. With the
263 amount of recharge is reduced along with the increase of domestic water pumping, water level circularly
264 slow down to the next agricultural peak. In addition to groundwater over-exploitation, climate
265 change-induced recharge reduction in recent three decades has been also part of the cause of groundwater



266 level declining, resulting in the seawater intrusion. The annual average rainfall varied from 639.7 mm
267 (1954 - 1979) to 594.2 mm (1980-2010). It obviously finds that there is a significant decrease in rainfall
268 over the last 30 years (Zhang, 2012). In general, the groundwater runoff intensity gradually decreases from
269 the piedmont to the coastal region.

270 4.2 Water stable isotopes ($\delta^2\text{H}$ and $\delta^{18}\text{O}$)

271 19 water samples collected from Yang River and Dai River have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from
272 -10.1 to -0.6‰ (mean= -5.4‰) and from -71~-11‰ (mean = -43‰), respectively. It seems that the stable
273 isotopes composition have significant seasonal variation. For Yang River, 3 surface water sampled in
274 relatively dry season (June 2008) were characterized by $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -5.5 to -1.1‰
275 (mean= -3.0‰) and from -49~-15‰ (mean = -31‰), respectively. Whereas 6 water samples sampled in
276 wet season (August 2009 and September 2010) have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -10.1 to -2.4‰
277 (mean= -6.6‰) and from -71~-21‰ (mean = -48‰), respectively. As to Dai River, in dry season, 3 surface
278 water samples are characterized by $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -3.9 to -0.6‰ (mean= -2.6‰) and
279 from -44~-11‰ (mean = -32‰), respectively; and in wet season, 7 surface water samples have $\delta^{18}\text{O}$ and
280 $\delta^2\text{H}$ values ranging from -9.7 to -1.2‰ (mean= -6.6‰) and from -69~-12‰ (mean = -49‰), respectively.
281 The water samples collected from Yang River and Dai River have similar stable isotopes composition.

282 56 groundwater samples are characterized by $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -11.0 to -4.2‰
283 (mean= -6.5‰) and from -76~-39‰ (mean = -50‰), respectively. Among them, 43 shallow groundwater
284 samples have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -11.0 to -4.2‰ (mean = -6.6‰) and from -76~-39‰
285 (mean = -50‰), respectively; 13 deep groundwaters have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -7.8 to -5.1‰
286 (mean = -6.3‰) and from -58~-43‰ (mean = -50‰), respectively. For the shallow groundwater, during the
287 dry season, 12 water samples have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values ranging from -7.2 to -4.2‰ (mean = -5.7‰) and
288 from -56~-39‰ (mean = -48‰), respectively; during the wet season, 31 water samples are featured by
289 $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values with a range of -11.0 ~ -5.3‰ (mean = -6.9‰) and -76 ~ -43‰ (mean = -51‰),
290 respectively. For the deep groundwater, during the dry season, 3 water samples have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values
291 ranging from -5.3 to -5.1‰ (mean = -5.2‰) and from -47~-45‰ (mean = -46‰), respectively; during the
292 wet season, 10 water samples are featured by $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values with a range of -7.8 ~ -5.2‰ (mean =
293 -6.6‰) and -58 ~ -43‰ (mean = -51‰), respectively.

294 The local meteoric water line (LMWL, $\delta^2\text{H}=6.6 \delta^{18}\text{O}+0.3$, $n=64$, $r^2=0.88$) is based on $\delta^2\text{H}$ and $\delta^{18}\text{O}$



295 mean values of the monthly rainfall between 1985 and 2003 at Tianjin station some 120 km SW of
296 Qinhuangdao City. The data were obtained from International Atomic Energy Agency/World
297 Meteorological Organization (IAEA/WMO, 2006). Due to the similar climatic and coastal conditions
298 between Tianjin and Qinhuangdao, this meteoric water line can be regarded as the local meteoric water line
299 (LMWL) in this study. From Figure 4, it can be seen that surface water have more wide range of $\delta^{18}\text{O}$ and
300 $\delta^2\text{H}$ values relative to groundwater. Water samples collected in wet season have more wide range of $\delta^{18}\text{O}$
301 and $\delta^2\text{H}$ values relative to water sampled in dry season. Most of water samples plot below the LMWL.
302 Seawater represents an end-member enriched in isotopes and plots far below the LMWL.

303 4.3 Water salinity and major dissolved ions

304 TDS (total dissolved solids) concentrations of the surface water samples from Dai River have a range
305 of 0.3g/L~31.4g/L with 22-78% Na^+ , 56-4% Ca^{2+} of total cations and 36-91% Cl^- of total anions,
306 $\text{Ca}\cdot\text{Na}\cdot\text{Mg}\cdot\text{Cl}\cdot\text{HCO}_3$ to $\text{Na}\cdot\text{Cl}$ water type from the upstream to the downstream locations. The Cl^-
307 concentrations varied from about 70 mg/L in the upstream to 16700 mg/L near the coastline. For Yang
308 River, the collected water samples have TDS concentrations of 0.3-26.1 g/L with percentages (33-91%) of
309 Cl^- concentrations (63.2-14953.5 mg/L) from the up-reach to the down-reach locations, with water types
310 changed from $\text{Ca}\cdot\text{Na}\cdot\text{HCO}_3\cdot\text{Cl}\cdot\text{SO}_4$, $\text{Ca}\cdot\text{Mg}\cdot\text{Cl}\cdot\text{SO}_4\cdot\text{HCO}_3$ to $\text{Na}\cdot\text{Cl}$. The nitrate contents range from 2.8 to
311 65.2 mg/L in the surface water samples.

312 Groundwater hydrochemistry can be modified the comprehensive effects from geological, climatic,
313 hydrogeological processes and anthropogenic activities. In the early 1960s, groundwater pumped from the
314 Zaoyuan well field was featured by the $\text{Ca}\cdot\text{HCO}_3$ water type and chloride concentrations of 90-130 mg/L.
315 In the early 1970s, individual wells appear slightly salinized. It has been deteriorated rapidly since the early
316 1980s. The chloride concentration of groundwater from water supply wells was 90mg/L in 1963, 218 mg/L
317 in 1975, 385 mg/L in 1984, 456.3 mg/L in 1986, 459.5 mg/L in 1995, 928.3 mg/L in 2000, 1367 mg/L in
318 2002, and 1290.4 mg/L in 2005 (Zang et al., 2010). In this study, the shallow groundwater is characterized
319 by TDS concentrations of 0.4-4.8 g/L with the percentage of Cl^- (34-77%), Na^+ (12-85%), Ca^{2+} (5-69%)
320 and water types varied from $\text{Ca}\cdot\text{HCO}_3\cdot\text{Cl}$, $\text{Ca}\cdot\text{Na}\cdot\text{Cl}$, $\text{Na}\cdot\text{Ca}\cdot\text{Cl}$ to $\text{Na}\cdot\text{Cl}$, which can be seen from Piper
321 plot (Figure 5). The deep groundwater is featured by TDS concentrations of 0.3-2.8g/L, which is dominated
322 by Ca (up to 77%) in the upstream area and Na (up to 85%) near coast, with water type distributed in series
323 of $\text{Ca}\cdot\text{Cl}\cdot\text{HCO}_3$, $\text{Ca}\cdot\text{Na}\cdot\text{Cl}$ and $\text{Na}\cdot\text{Mg}\cdot\text{Cl}$ (Figure 5). The TDS of groundwater from the well field reaches



324 3.31 g/L with Na-Cl water type in the well G15. The relative high fracture of seawater occurs in the shallow
325 well G10 and deep well G2, with f_{sw} values of 12.95% and 5.35%, respectively. The nitrate contents have a
326 range of 2.0-178.5 mg/L (mean 90.1 mg/L) for shallow groundwater, and 2.0-952.1 mg/L (mean 232.1
327 mg/L) for the deep groundwater, respectively, most of which seriously exceeds the WHO drinking water
328 standard (50 mg/L).

329 There is a geothermal field around Danihe-Luwangzhuang area (Fig. 1). Hydrochemical features of
330 thermal water are very distinct from cold water. The previous investigation has identified the buried
331 geothermal water with high TDS in the fracture/fissure of deep metamorphic rock (Zeng, 1991). Due to the
332 pumping wells for pumping thermal water were protected and not permitted to be sampled, we have to
333 collect some data associated this geothermal field from the previous research. The geothermal field is
334 controlled by the fault distribution under confined state. The thermal water flows along the fault zone and
335 enters the Quaternary aquifer, forming hot salt water distributed around the spill point and expanded
336 towards downstream. It can result in the similar hydrochemical characteristics between Quaternary salt
337 groundwater and deep original thermal waters from bedrocks. The geothermal water is characterized by
338 Ca•Na-Cl water type, 6.2-10.6 g/L of TDS and 7.4-8.7 of pH values. Cl^- concentrations range from 5.4 to
339 6.5 g/L, Na^+ from 1.7 to 2.0 g/L, Ca^{2+} from 1.6 to 1.9 g/L, F from 3.0 to 3.6 mg/L, Sr from 6.73 to 89.8
340 mg/L, Li from 0.43 to 1.58 mg/L, and SiO_2 from 44.0 to 48.3 mg/L (Hui, 2009). The groundwater samples,
341 collected from the wells G8, G19, and G9 with different depths, are featured by Ca•Na-Cl water type with
342 relative high TDS ranges (0.8-1.4 g/L, 1.3-1.6 g/L, and 1.5-2.8 g/L, respectively) and Sr contents (1.1-1.9
343 mg/L, 4.9-7.1 mg/L, and 7.3-11.6 mg/L, respectively).

344 5. Discussions

345 5.1 Groundwater flow system and hydrochemical features

346 Generally, Quaternary groundwater system in the Yang-Dai River coastal plain is recharged by
347 precipitation, irrigation return flow, river infiltration and lateral subsurface runoff from mountain-front
348 region. Due to the natural geological function and human pumping activities, there have been interactions
349 between groundwater and geothermal waters around Danihe area or between groundwater and seawater in
350 the coastal area. The groundwater geochemical features are controlled by the complex hydrogeological
351 conditions and these hydrological processes. The different sources of water bodies are characterized by
352 different of stable isotopic and hydrochemical compositions, determining the groundwater salinization



353 processes in this area.

354 The stable isotopes of O and H can be used to describe the groundwater origin and to identify the
355 mixing processes between different water bodies. The slope of the best-fit regression line for collected
356 groundwater samples (dashed line in Fig. 4) given as $\delta^2\text{H}=4.4\times\delta^{18}\text{O}-21.7$, which is significantly lower than
357 either the local or global meteoric water lines. The deviation of groundwater and surface water lines from
358 the LMWL has evidenced evaporative processes occurred during water infiltration and surface runoff. The
359 composition of stable isotopes in groundwater samples collected in the relatively dry season has been
360 narrower and enricher than that collected in the wet season. It could be resulted from evaporation processes
361 during the infiltration of local irrigation return-flows in the dry season.

362 The composition of stable isotopes for thermal groundwater could be originated from precipitation,
363 however, its ^{14}C age dating between 3.4-12.8ka with tritium content of less than 2 TU (Zeng, 1991),
364 indicating thermal waters might be formed under cooler climate condition than present climate. The
365 composition of stable isotopes of thermal groundwater are more depleted than that of cold groundwater,
366 even lower than the cold groundwater from mountain-front area, indicating the thermal groundwater could
367 be mainly originated from NW mountain area, where has higher elevation. The elevation range of recharge
368 area for Danihe geothermal field is from 1200 to 1500 m.a.s.l obtained by Zeng (1991).

369 Fresh groundwater has depleted $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values relative to seawater. Theoretically, the mixing of
370 fresh groundwater and seawater should show a straight line connecting the two end members. Obviously,
371 some surface water samples (e.g. S1, S2, S3, S7, S12) are the mixture with seawater. In this study area,
372 there are three end members (namely, fresh groundwater, thermal groundwater and seawater), which has
373 been evidenced by the previous studies (Han, 1988; Zeng, 1991). Thus, the diagram of $\delta^{18}\text{O}$ vs. Cl⁻ (Fig. 6)
374 can be used to identify the mixing pattern among three end members. Fig. 6 shows the mixing lines
375 between shallow fresh groundwater (G4) and seawater, between deep fresh groundwater (G25) and
376 seawater, between shallow fresh groundwater and thermal water, and between deep fresh groundwater and
377 thermal water. The shallow groundwater samples (e.g. G15, G10, G11, G14) collected from or around the
378 Zaoyuan well field are characterized by mixing with seawater. The deep groundwater samples (e.g. G13,
379 G2, G16, G14') collected from the coastal zone are also resulted from mixing with seawater. The sampling
380 site of deep groundwater sample G29 is located between thermal field and the coastline and obviously
381 affected by both of mixing processes. The groundwaters (e.g. G9, G19), sampled from the area affected by
382 geothermal field are mixture between fresh cold groundwater with thermal waters. The mixing fraction (f_{sw})



383 of seawater has a range of 1.2~13.0% for the shallow brackish groundwater, and 2.6~6.0% for the deep
384 brackish groundwater. f_{sw} reaches the highest percentage of 13% in the well G10, which is located in the
385 north part of the well field.

386 At the late 1950s, groundwater pumped from the Zaoyuan well field was characterized by Ca-HCO₃
387 type water with Cl concentrations ranging from 130 to 170 mg/L. The hydrochemical data investigated in
388 1986 (Han, 1986) showed that there were mainly five water types in this study area, including Ca-HCO₃
389 type with TDS less than 0.5g/L distributed in the mountain-front area, Ca•Na-Cl•SO₄, Ca•Na•Mg-SO₄•Cl,
390 or Na•Ca-Cl•SO₄ type water with TDS 0.4-0.7g/L distributed around the Zaoyuan well field and
391 Wanggezhuang, Ca•Na-Cl type water with TDS 0.4-1.8g/L distributed around the geothermal field
392 (Luwangzhuang) and Duzhai, Na-HCO₃ or Na-HCO₃•Cl type water with TDS 0.5-0.9g/L distributed the
393 SW area close to the coastal zone, and Cl-Na type water with TDS 0.4-2.4g/L distributed in the coastal
394 zone. Due to the disturbance of human activities, the current groundwater hydrochemistry has become more
395 complex than that before. Compared salty water distributed 2 km away from coastline in the late 1950s, the
396 distance has increased to about 7 km away from coastline. The Cl-Ca•Na or Cl-Ca type water type mainly
397 distributed in the area affected by geothermal field, such as G5, G8, G19, G29, and G24, indicating the
398 salinizing process during the mixing between cold groundwater and the thermal waters. In the upstream
399 area, the groundwater samples (e.g. G7, G23, G25) have feature of Ca•Mg•Na-Cl•SO₄, Ca-Cl•SO₄, and
400 Ca-Cl•HCO₃ type, not the Ca-HCO₃ type in the 1980s. It suggests that the salinized composition has
401 resulted from the anthropogenic pollution. The groundwater samples (e.g. G10, G11, G15, G26) collected
402 from the well field show the feature of Cl-Na• (Ca) type water with TDS 1.2-4.8 g/L. The samples (e.g. G1,
403 G2, G3, G4, G12, G22, G14, G14') collected from the coastal zone show the water type of Na-Cl or
404 Na•Ca-Cl•SO₄ or Ca•Na•Mg-Cl•SO₄, indicating that, apart from seawater intrusion, the anthropogenic
405 pollution also plays important role on modifying the groundwater chemistry.

406 The seawater from Bohai Sea has relatively higher NO₃⁻ concentrations (810 mg/L in this study, up to
407 1092 mg/L in the coastal seawater of Dalian, Han et al., 2015) due to wastewater discharge into the sea.
408 NO₃⁻ concentration of groundwater in the well field increased from 5.4 mg/L in May 1985 to 146.8~339.4
409 mg/L in Aug 2010, while the concentration of seawater in this area changed from 57.4 mg/L in May 1985 to
410 810.1 mg/L in Aug 2010. The diagram (Fig. 7) of Cl⁻ vs. NO₃⁻ concentration of groundwater can be used to
411 identify the different mixing trend in this study area, including the mixing process with contaminated
412 seawater, and the anthropogenic NO₃⁻-sources (e.g. domestic/industrial wastewater discharge, NO₃⁻-bearing



413 fertilizer input through precipitation infiltration and the irrigation return-flow) in the inland area. It can be
414 seen from Fig. 7 that the major source of NO_3^- in groundwater is from anthropogenic input, with the
415 exception of G10 and G15 mixing with seawater in the well field. The deep groundwater (e.g. G9, G14') is
416 also contaminated by higher NO_3^- concentrations, which is likely associated with the abandoned wells.
417 According to one investigation by Zang et al.(2010), 14 of 21 pumping wells in the Zaoyuan well field
418 have been abandoned due to the salinized water quality, and 307 pumping irrigation wells (occupied 2/3
419 total pumping wells for irrigation) have been abandoned. However, the local department has not made any
420 measure to deal with those abandoned wells.

421 5.2 Groundwater salinization processes

422 5.2.1 Development of seawater intrusion and associated hydrochemical behavior

423 The intensively pumping groundwater from the Quaternary aquifer of Yang-Dai River coastal plain
424 has resulted in the development of groundwater depression cones from Zaoyuan well field to Fangezhuang,
425 with the aggravation of seawater intrusion in this region. In the 1950s, the seawater intrusion in the study
426 area was only occurred within 2 km distance from the coastline, and it expanded to over 5 km distance in
427 the 1980s. In 1986, the groundwater depression cone centered in the Zaoyuan well field was characterized
428 by 6 meters depths below the sea level, with the water level -3 m.a.s.l. The enclosed area by 0 m.a.s.l water
429 level contours covered 10 km². The original Ca-HCO₃ type water changed to Ca•Na-Cl type. Apart from
430 the intensive exploitation fresh groundwater from coastal aquifer, the successive drought (1976-1989) also
431 played important roles on controlling the groundwater recharge and exacerbating seawater intrusion in the
432 coastal area of north China (Wilhite, 1993; Han et al., 2015). In this study area, the annual mean
433 precipitation was 668.7 mm during 1954-1995, the Cl concentrations was ranging from 130 to 170 mg/L in
434 the Zaoyuan well field. Whereas the annual mean precipitation reduced to 559.7 mm during 1996-2011,
435 resulting in Cl concentration in the well field up to 550 mg/L in May 1986, and 812 mg/L in July 2006. It
436 has seriously threatened the safety of water supply in this region. The seawater intrusion in the coastal
437 aquifer shows the wedge-shaped body and has vertically characterized by freshwater in the upper part and
438 salt water in the lower part of shallow aquifer. Since 2002, with the establishment of anti-tide dam in the
439 Yang River estuary area, it has good effect on preventing the horizontal pouring seawater into riverway.
440 Thus, the seawater intrusion is mainly caused by lateral inflow of seawater in the aquifer.

441 According to the guidelines of drinking water standards from China Environmental Protection



442 Authority (GB 5749-2006) or US EPA or WHO, the guideline of chloride concentration for drinking water
443 is 250 mg/L. Most groundwater distributed in the seawater intrusion area cannot be used for irrigation, the
444 source of drinking water and industrial utilization. It has enhanced the scarcity of fresh water resources in
445 this region by vicious cycle of groundwater level decline → seawater intrusion → groundwater salinization
446 → groundwater level decline again. This will also influence the surface water runoff. How to judge the
447 criterion of seawater intrusion? Generally, 250 mg Cl/L can be regarded as the intruded standard, and more
448 than 1000 mg Cl/L as the serious intrusion (Jiang and Li, 1997; Zhuang et al., 1999). Some studies took the
449 TDS (>1000 mg/L) as the intruded standard (e.g., Xue et al., 1997; Zhang and Peng, 1998). Water type can
450 also be used as the intruded standard, such as Ca-Cl type water occurs during seawater intrusion into
451 freshwater aquifers, and Na-HCO₃ type water displays during flushing of the mixing zone by freshwater
452 (Appelo and Postma, 1993). Additionally, the multi-hydrochemical ionic ratios can also provide important
453 confirmation of hydrogeochemical processes modifying groundwater chemistry during seawater intrusion
454 (Vengosh et al., 1997; Jones et al., 1999). However, the frequent anthropogenic activities modified coastal
455 hydrologic dynamics and hydrogeochemical characteristics to great extent. For instance, with except of
456 modern seawater, the sources of chloride in groundwater system could be derived from paleoseawater relics
457 in aquifers, infiltration of agricultural return flow with fertilizer solutions, and discharge of industrial and
458 domestic wastewater.

459 Hydrogeochemical facies evolution diagram (HFE-D) proposed by Giménez-Forcada (2010) can be
460 used to analyze the phase of seawater intrusion or freshening and its dynamics. From Fig. 8, it can be seen
461 that most brackish groundwaters (e.g., G11, G16, G17, G20, G25, G28, G29) have evolved in the series of
462 Ca-HCO₃ → Ca-Cl → Na-Cl facies under the intrusion period. Locally, several water samples (e.g., G1,
463 G12, G26) collected from the interfluvial area have been characterized by freshening process. Deep
464 groundwater G11 sampled from the well field shows being under salinizing period in the relatively dry
465 season, and under freshening period in the relatively wet season. In the coastal zone, the river water has
466 obvious mixing trend between end members. Some shallow groundwaters (e.g., G2, G4, G10, G13, G15)
467 are close to the mixing line between end members on this figure, indicating significant mixing with
468 seawater. The groundwaters (G10, G11, G15, G26) collected from the productive wells of the Zaoyuan
469 well field display the different processes occurring in salinizing or freshening stages, indicating that the
470 heterogeneous hydrogeological conditions could be responsible for this distinguished patterns.

471 The calculated results of saturated indices (Fig. 9) show that that SI_{cal} and SI_{dol} have some deviation



472 from equilibrium (-0.4 to +0.5 for SI_{cal} , and -0.5 to +0.5 for SI_{dol}). The distribution of SI_{cal} and SI_{dol} is
 473 related to the sampling period. In the wet season, most of water samples are characterized by $SI_{cal} < 0$ and
 474 $SI_{dol} < 0$, suggesting they are under unsaturated for these minerals, while in the dry season, most of water
 475 samples are under saturated with respect to these minerals. In contrast, all sampled groundwater had
 476 negative saturation indices with respect to gypsum ($SI_{gyp} < 0$), indicating that these water samples are
 477 under-saturated with respect to gypsum. The plots (Fig. 10 a-f) of ionic molar ratios (Na/Cl, SO_4/Cl , Mg/Ca,
 478 $Ca/(SO_4+HCO_3)$, Ca/SO_4 , and $(Ca+Mg)/Cl$) can be used to further reveal the groundwater salinized
 479 processes and dominated hydrochemical behavior. The brackish groundwaters in this study area have an
 480 enriched Ca^{2+} (i.e., the ratio of $Ca/(HCO_3+SO_4) > 1$ with low ratios of Na/Cl and SO_4/Cl as the seawater
 481 proportion in the mixture increases. As shown in Fig. 10a, Na/Cl ratios of brackish groundwater affected by
 482 seawater intrusion are usually lower than the ratio (0.86) of modern seawater. The high Na/Cl ratios (> 1)
 483 could be typical of anthropogenic sources (i.e., domestic waste waters). When seawater intrudes into
 484 coastal freshwater aquifers, Ca^{2+} on the clay-bearing sediments can be replaced by Na^+ :



486 This process can decrease the Na/Cl ratios and increase the $(Ca+Mg)/Cl$ ratios. The dolomitization process
 487 can be described by the transformation reaction (Appelo and Postma, 2005):



489 It can result in Ca-enrichment over Mg in solution and that Mg/Ca ratios decreases. This process may also
 490 be characterized by Ca-Cl water type.

491 To explain the enrichment in Ca^{2+} relative to SO_4^{2-} concentrations, observed in most water samples
 492 (Fig. 10e), gypsum dissolution ($SI_{gyp} < 0$) can be coupled by cation exchange reactions under the interaction
 493 with clay stratum and calcite precipitation with incongruent dissolution of dolomite and gypsum.
 494 Additionally, due to the ORP values ranging from 3 to 74 mV for 18 of 22 water samples collected in
 495 August 2010, the sulfate reduction under anaerobic conditions may be responsible for relatively high
 496 Ca/SO_4 and low SO_4/Cl ratios. Generally, low Na/Cl, SO_4/Cl and high $Ca/(HCO_3+SO_4) (> 1)$ ratios are
 497 further indicator of the arrival of seawater intrusion.

498 ΔNa is negative in most samples of this area (Fig. 11a and Fig. 12a). The depletion of Na^+ could be
 499 caused by the inverse cation exchange taken place with the clay sediments. This exchange produces Ca
 500 release to the solution during the seawater intrusion. The positive ΔCa and ΔMg may be due to the
 501 dissolution of calcite, dolomite and gypsum present in the aquifer strata. Water flushing during aquifer



502 recharge can result in positive ΔNa and negative ΔCa and/or ΔMg (Fig. 11a). For some water samples, the
503 Ca enrichment is not accompanied by Na depletion, which could be caused by dolomitization (Ca
504 enrichment with Mg depletion) (Fig. 12b). The excess of SO_4 compared to conservative mixing (Fig. 11d)
505 can be explained by redissolution of the precipitated gypsum along the mixing front.

506 *5.2.2 Mixing between thermal and cold groundwater*

507 Sea level rose by about 100 m since the end of the last glacial period (18,000 years, 18 ka BP) and
508 stabilized around 5 ka BP in the eastern China (Yang, 1996). The marine sediments could not be found in
509 the geothermal field, indicating the transgression in the geologic history did not occur around Danihe area
510 (Zeng, 1991). However, the fracture and structural fissure developed well in this study area became the
511 major subsurface pathway of seawater intrusion. The previous studies have revealed that the geothermal
512 waters in this area are characterized by the features of residual seawater and modern precipitation (Zeng,
513 1991; Hui, 2009). The results of ^{14}C age dating for the geothermal waters in this area are ranging from 3.4
514 ka to 12.8 ka with lower tritium contents (less than 2TU) (Zeng, 1991). The Piper plot (Fig. 5) shows
515 CaNa-Cl type water for the geothermal waters. It is noteworthy that the geothermal water from Danihe
516 geothermal field has higher Sr concentrations (up to 89.8 mg/L) relative to that in seawater (5.4-6.5 mg/L in
517 this study), due to the Sr-bearing minerals (i.e., celestite, strontianite) with Sr contents of 300-2000 mg/kg
518 present in the bedrock (Hebei Geology Survey, 1987). The mixture waters sampled from the geothermal
519 field in this study also have the higher Sr concentrations relative to seawater, i.e., G9 with Sr concentrations
520 ranging from 7.4 to 11.6 mg/L, and G19 from 4.9 to 7.1 mg/L. The diagram of chloride versus strontium
521 concentrations of different water samples (Fig. 13) shows that the groundwater samples (e.g., G9, G19)
522 collected from the geothermal field have obviously been characterized by closing to mixing line between
523 fresh cold- and thermal-groundwater. Some waters (G16, G20, G29) sampled from the downstream area
524 also close to this mixing line, indicating the thermal water overflows into the coastal aquifers in different
525 depths. The water samples collected from the well field are located between two mixing lines (Fig. 13),
526 suggesting the groundwater in the well field simultaneously suffered from the mixing with thermal water
527 and obvious seawater intrusion. Additionally, the points of water samples (G5, G8, G9, and G19), collected
528 from the geothermal field, mainly occurs on the HFE-D (Fig. 8) in the *I2* (MixCa-Cl) and *I6* (Ca-Cl) facies
529 zones, indicating these waters have been modified by the reverse base-exchange reactions.

530 As both Cl and Br are not affected by water-rock interactions and usually behave conservatively, the



531 Cl/Br ratio can be used as a reliable tracer to study the processes of evaporation and salinization of water
532 (Edmunds, 1996; Jones et al., 1999). Standard seawater (Cl/Br molar ratio=650.8) may be distinguished
533 from relics of evaporated seawater (normally less than 669.3), input of evaporite dissolution (more than
534 2256) and anthropogenic pollution (e.g., sewage effluents, Cl/Br ratios up to 1805; Vengosh and Pankratov,
535 1998) or agricultural return-flows with low Cl/Br ratios (Jones et al., 1999). It can be seen from Fig. 14 that
536 the points of the thermal waters lie lower than the ratio line of standard seawater, indicating that they are
537 affected by mixing with relics of evaporated seawater. The points of cold groundwaters (G9, G19) sampled
538 from the geothermal field display between the seawater and the thermal waters, indicating these cold waters
539 are mixture between cold groundwater and the thermal water, which has relics of evaporated seawater.
540 However, it cannot exclude adding the Br inputs into groundwater system through the pesticides
541 application of the pronounced agricultural activity (Davis et al., 1998), this effect could lower the Cl/Br
542 ratios of the groundwaters. The groundwater sample G10 in the well field shows the feature of high Cl/Br
543 ratio in Fig. 14, indicating obvious anthropogenic inputs (e.g., discharge domestic wastewater) occurring in
544 the shallow aquifers around the well field.

545 *5.2.3 Interaction between surface- and ground-water*

546 Coastal zones encompass the complex interaction among different waters (i.e., river water, seawater,
547 groundwater, rainfall water). The interaction between surface- and ground-water in the Yang-Dai River
548 coastal plain is usually ignored by the previous studies. However, understanding how surface water
549 interacts with the groundwater is essential for managing freshwater resources. Groundwater depression
550 cone below the sea level has formed in the early 1980s. Due to the irrigation supported by transfer of
551 surface water from the upper and middle stream of Yang-Dai River, the amount of surface water discharged
552 into the Bohai Sea declined to great extent. Under the tide effects, seawater can be poured into the estuary
553 of the downstream section of the rivers, resulting in the river bed filled with saltwater, which can cause
554 mixing between river water and seawater. The results of water chemistry analysis from two river sections
555 show that the distribution of salt water reached more than 10 km above the estuary of the Yang River, and
556 about 4 km above the estuary of the Dai River (Han, 1988). It led to that the seawater simultaneously
557 intruded into the coastal aquifers through not only the lateral subsurface flow from coastline to the inland
558 but also vertical infiltration from the riverbed to both sides of the river. The hazard caused by the latter
559 pattern had been more serious than the former pattern, before the establishment of anti-tide dam at the



560 estuary of Yang River. Currently, the seawater-intruded distance towards inland has been controlled within
561 4 km away from the coastline.

562 The stable isotope compositions of different water samples (Fig. 4) display that the points of most
563 surface water samples are deviated from the LMWL to the right, indicating that these waters are likely to be
564 subject to evaporation to different degrees. The points of surface water samples (S1, S2, S3, and S7) in
565 Fig.4 close to the compositions of local seawater, indicating the pronounced mixing process with seawater
566 for these surface waters. However, the samples S2, S6, S8, and S9 have the depleted compositions of stable
567 isotopes, probably resulting from the exchange between them and local groundwater. S12, located at the
568 estuary area, has variable compositions due to the sampling seasons. HFE-D shows that most of surface
569 water samples are close to the mixing line between end members (freshwater and seawater). S9 is
570 significantly characterized by salinization process probably due to the interaction with ambient
571 groundwater. It can be seen from the relationship between ionic delta values and seawater proportions for
572 the water samples (Fig. 11) that G1, G11, G12, G14', G26, G29, due to these wells close to the river or
573 located at the flat interfluvium, may be dominated by the obvious freshening process. While G2, G3, and G16
574 under the salinizing process could be subject to the vertical infiltration of saltwater in the river. The points
575 of surface waters (S1, S2, S3, and S7) on Fig. 4 and Fig. 8 are distributed along the mixing line between
576 fresh- and sea-water end members. It is due to the direct mixture occurs in the riverway. S12 sampled from
577 the Dai River estuary may be contaminated by the wastewater discharge with higher Sr concentration
578 relative to seawater. By contrast, surface water from the Dai River have higher seawater proportions
579 compared with that from Yang River, owing to that the local government did not establish anti-tide dam in
580 the Dai River estuary. G1, G2, G3, G10 and G13 collected from coastal zone are obviously mixed with
581 seawater with closing to the mixing line between seawater and freshwater in Fig. 13.

582 5.3 Conceptual model of groundwater flow patterns

583 Generally, groundwater in this study area is mainly originated from precipitation, river infiltration,
584 lateral subsurface runoff, upflow of geothermal waters and seawater intrusion in the coastal area. The
585 associated hydrological processes driven by the natural (hydrologic, geologic, climatic) changes and
586 anthropogenic activities have resulted in groundwater salinization processes, along with the complex
587 hydrogeochemical characteristics of groundwater system. Groundwater changes from Ca-HCO₃ type water
588 in the piedmont area to the Na-Cl type water in the coastal area.



589 A conceptual model groundwater flow system in the Yang-Dai River coastal plain can be summarized
590 in Fig. 15. Four subsurface processes, including seawater intrusion, return-flow of agricultural irrigation,
591 mixing with geothermal water and interaction between surface water and groundwater/seawater, could be
592 responsible for the groundwater salinization in this area. Two aspects of seawater intrusion, identified by
593 depleted ΔNa and enriched ΔCa with Ca-Cl type water and $\text{Ca}/(\text{HCO}_3+\text{SO}_4)>1$ and lower Na/Cl and
594 SO_4/Cl relative to these ratios of seawater, can be delineated, namely vertical infiltration of saltwater inflow
595 towards the inland estuary and lateral inflow of seawater driven by over-pumping groundwater from fresh
596 aquifers. Irrigation return-flow from local groundwater can cause groundwater nitrate pollution (up to 340
597 mg/L NO_3^- in groundwater of this area) due to the infiltration and dissolution of fertilizers. Geothermal
598 water with high TDS, F, and Sr concentrations flows into the Quaternary aquifers, mixing with cold
599 groundwater, and transports to the downstream area of Yang River Basin. Additionally, the interaction
600 between surface- and ground-water can cause seasonal flushing local groundwater in the upstream
601 interfluvium or lead to saltwater infiltration affected by tide/surge along the riverbed at the estuary.

602 **6. Conclusions**

603 It has been recognized that groundwater in the Quaternary aquifers of the Yang-Dai River coastal plain
604 is the important water resource for agricultural irrigation, urban and tourism development and industrial
605 utilization. Natural climate change (e.g., continuous drought, overflow of geothermal water) and human
606 activities have made the problem of groundwater salinization in this area increasingly prominent, even
607 resulting in the closure of the Zaoyuan well field. Based on the analysis of hydrochemical and stable
608 isotopic compositions of different water bodies, including surface water, cold groundwater, geothermal
609 water, and seawater, we delineated the groundwater flow system and groundwater salinization processes.
610 Seawater intrusion is the main aspect responsible for the groundwater salinization in the coastal zone,
611 including the vertical saltwater infiltration along the riverbed into aquifers, which is affected by the
612 tide/surge process, and the lateral seawater intrusion caused by pumping for fresh groundwater. The
613 overflow of the high mineralized thermal water into the Quaternary aquifers along the fault zone mixes
614 with the cold groundwater and makes it salinized. The thermal water has characterized by lower Cl/Br
615 ratios and higher Sr concentrations relative to seawater. It cannot be ignored that the salinization or nitrate
616 pollution from the anthropogenic activities (e.g., agricultural irrigation return-flow with solution of
617 fertilizers). Additionally, the interaction between surface- and ground-water can also affect the groundwater



618 salinization in this area. Different approaches of hydrochemical analysis, such as Piper plot, HFE-D, major
619 ionic ratios (Na/Cl , SO_4/Cl , Ca/SO_4 , $(\text{Ca}+\text{Mg})/\text{Cl}$, $\text{Ca}/(\text{SO}_4+\text{HCO}_3)$, Cl/Br) and Sr, were used in this study
620 to identify the different hydrogeochemical reactions and freshening or salinizing processes in the
621 Quaternary aquifers.

622 Groundwater salinization has become a prominent water environment problem in the coastal area of
623 north China, which has caused the further paucity of fresh water resources and has become the bottleneck
624 of urban development to a certain extent. Since the 1990s, the local government has begun to pay attention
625 to the development of seawater intrusion, and the irrational exploitation has been restricted. The Zaoyuan
626 well field has ceased to pump groundwater since 2007. The anti-tide dam has been established in the Yang
627 River estuary area in 2002, effectively intercepting the seawater pouring into riverway during the tide/surge
628 period. These actions have made the rate of intrusion slowed down. The joint use of surface water and
629 groundwater with reasonable exploitation program is essential and economical for the local water resources
630 management. However, the quantitative understanding to the vertical and lateral saltwater intrusion into
631 fresh aquifers should be obtained from further continuous groundwater monitoring and numerical
632 groundwater flow and transport modeling. This study would be benefit the local agricultural development
633 and groundwater resources management.

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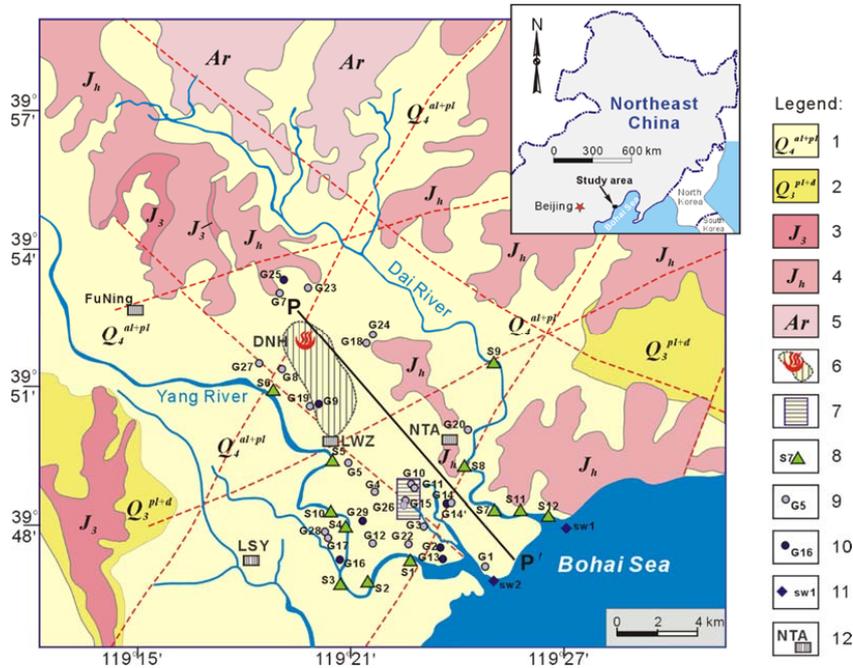


Figure 1. Map for showing geological background and water sampling sites in the study area
 Explanation for the legend: 1-Holocene (al- alluvial, pl- pluvial) sediments; 2- Upper Pleistocene (dpl- proluvial-deluvial) sediments; 3- Jurassic andesite; 4- Jurassic migmatitic granite; 5- Archaean group gneiss; 6-Geothermal field location and its influence area; 7- Zaoyuan well field; 8- surface water sampling site; 9- shallow groundwater sampling site; 10- deep groundwater sampling site; 11- seawater sampling site; 12- village/county site (DNH-Danihe; NTA-Niutouai; LWZ-Luwangzhuang; LSY-Liushouying). The red dashed lines are showing the buried fault distribution in this area.

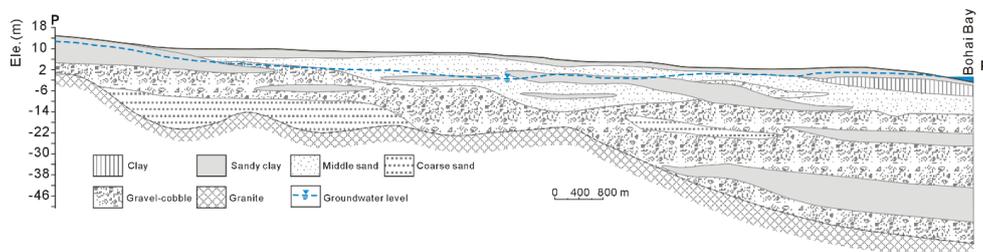


Figure 2. Hydrogeological cross-section of Yang-Dai River Plain (P-P' in Fig. 1) (modified from Han, 1988)

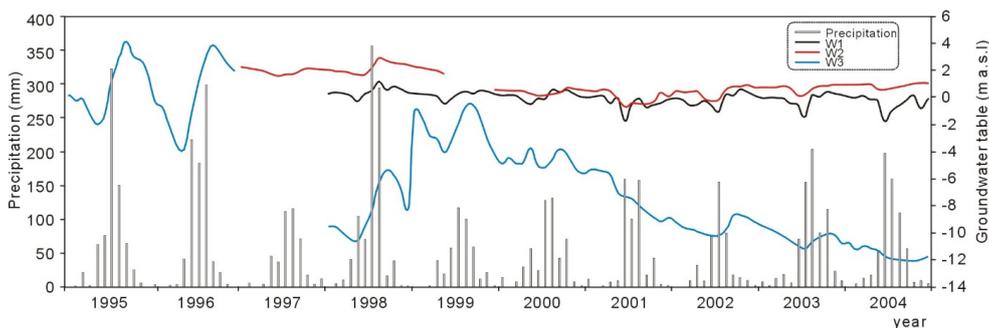


Figure 3. Distribution of precipitation and dynamics of groundwater table in the study area

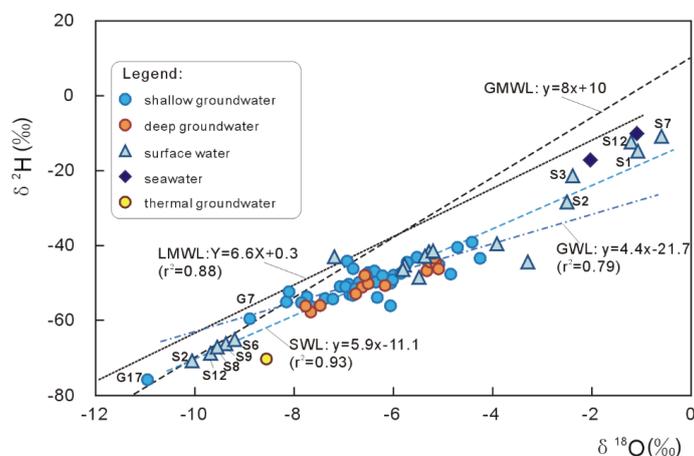


Figure 4. Stable isotope compositions of different water samples collected from the study area
 LMWL - local meteoric water line; GMWL – global meteoric water line (Craig, 1961); GWL – groundwater line; SWL – surface (river) water line.

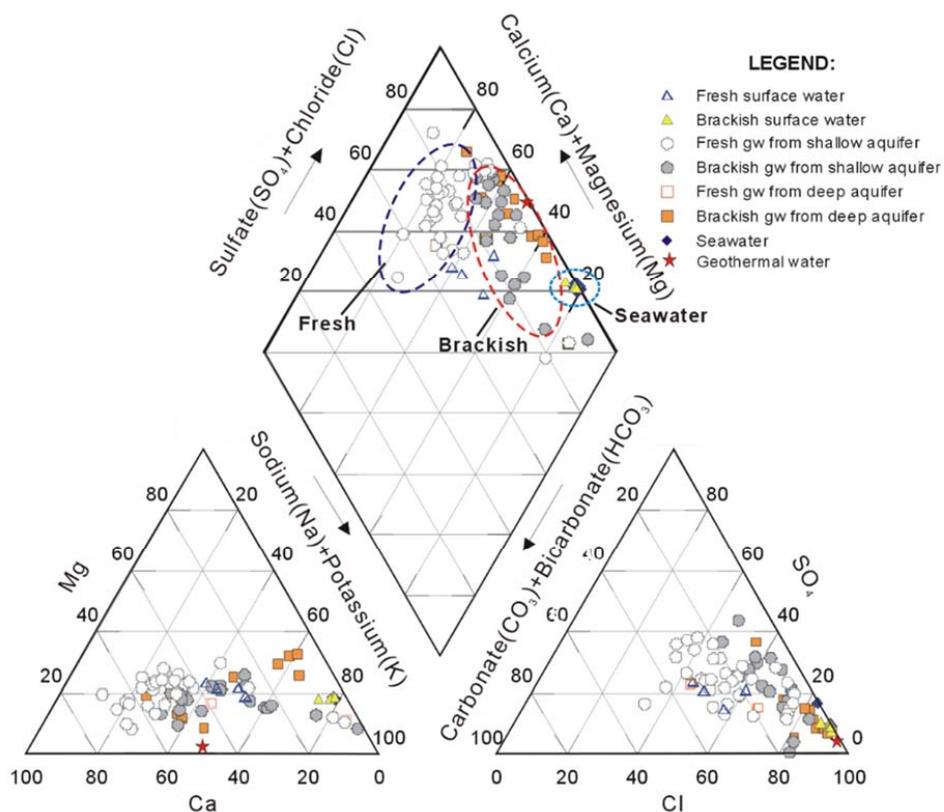


Figure 5. Piper plot of different water samples

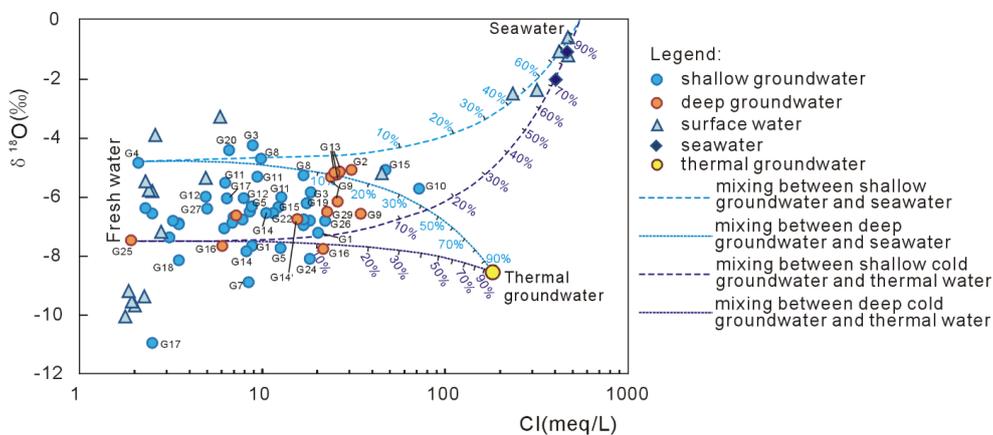


Figure 6. Relationship between chloride content and isotopic signature of different water samples as a means to differentiate mixing processes in the area. The data of thermal water are from Zeng (1991).

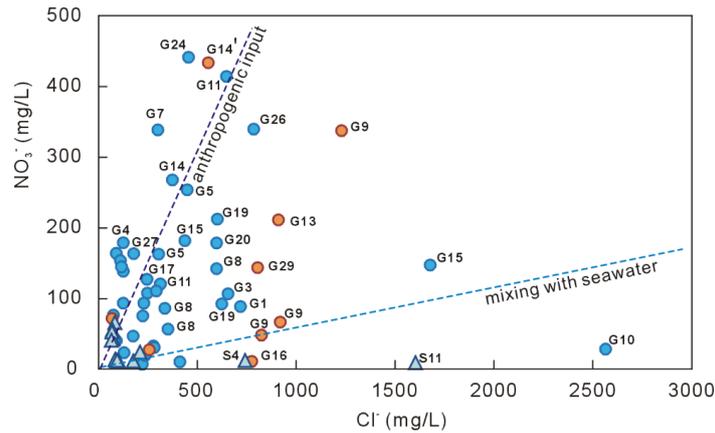


Figure 7. Plots of chloride versus nitrate concentrations. Symbols are same as in Fig. 6.

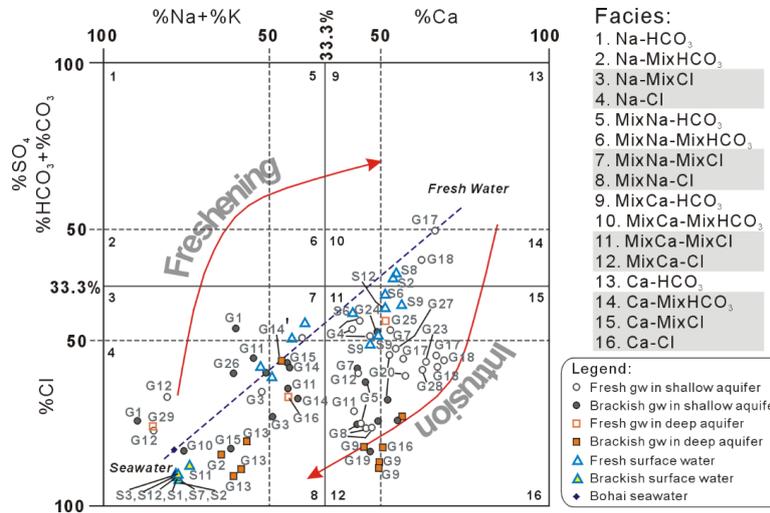


Figure 8. Hydrogeochemical facies evolution diagram (HFE-D) for the collected water samples

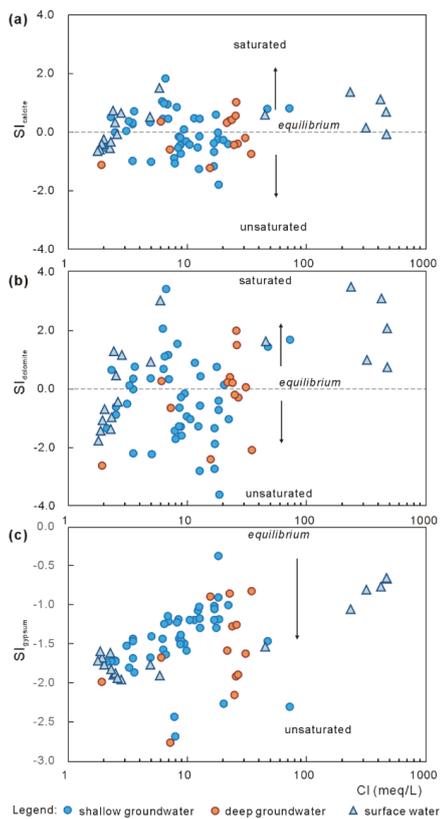


Figure 9. Variation of saturation indices with respects to calcite (a), dolomite (b) and gypsum (c) versus chloride concentrations

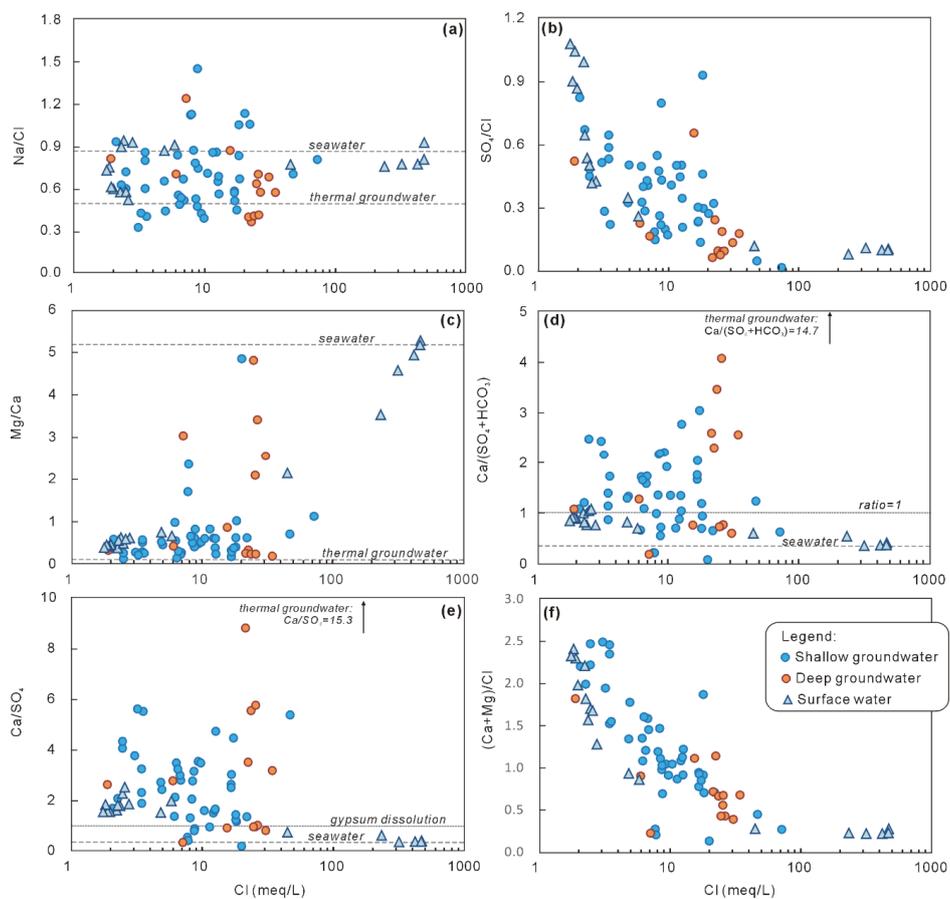


Figure 10. Molar ratios of major ions versus chloride concentrations for different water samples from the study area

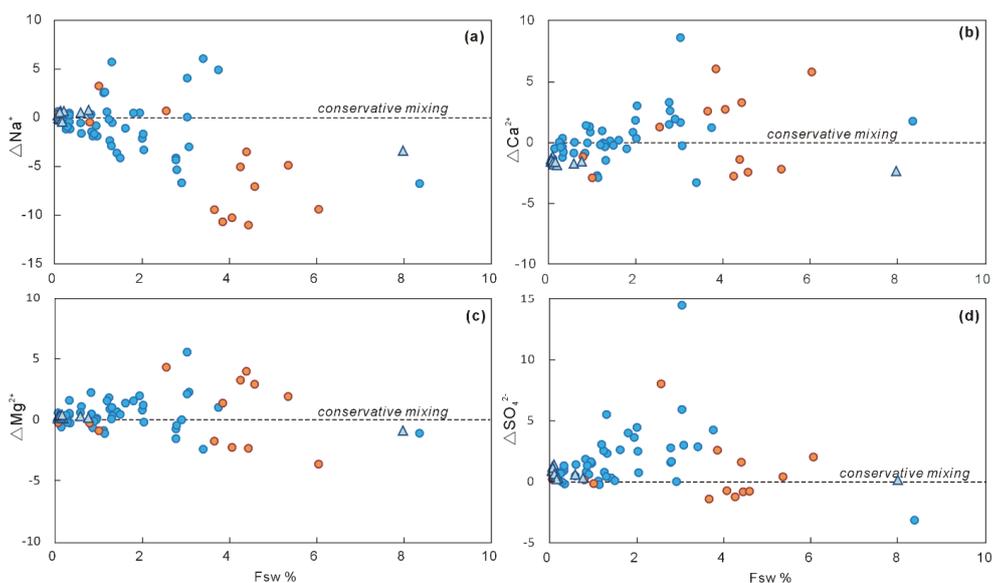


Figure 11. ΔNa^+ (a), ΔCa^{2+} (b), ΔMg^{2+} (c) and ΔSO_4^{2-} (d) versus calculated seawater percentage ($F_{sw}\%$)

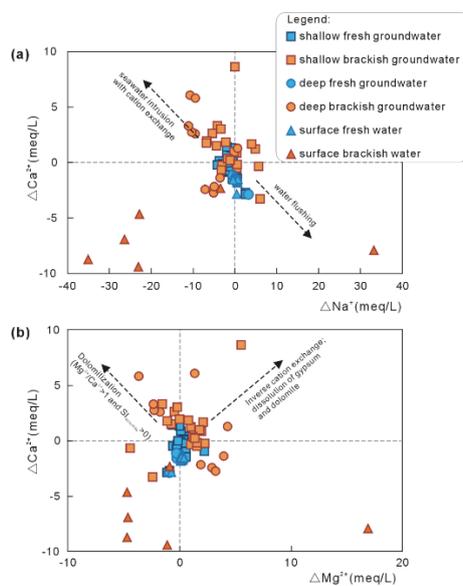


Figure 12. Distribution of cationic deltas (a)- ΔNa^+ versus ΔCa^{2+} ; (b) ΔMg^{2+} versus ΔCa^{2+} for each sample (in meq/L)

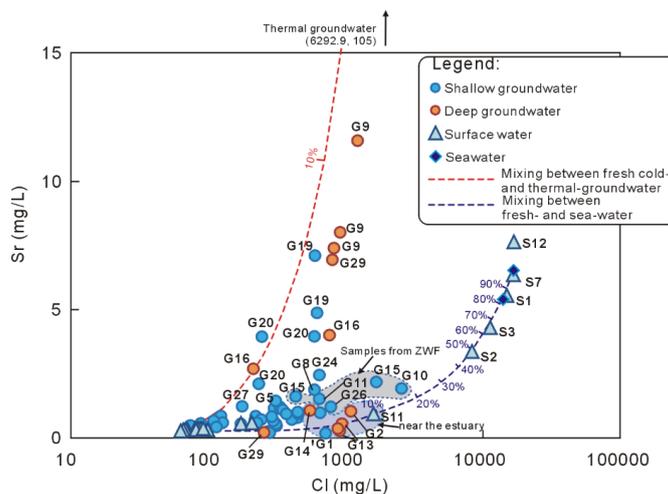


Figure 13. Chloride versus strontium concentrations of different water samples
 ZWF- Zaoyuan well field

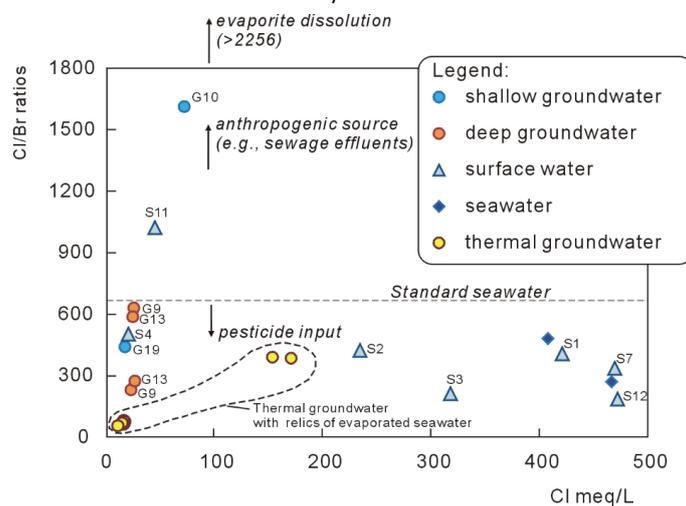


Figure 14. Chloride versus bromide to chloride molar ratios. Data of 8 thermal groundwater samples are from Zeng (1991) and Hui (2002).

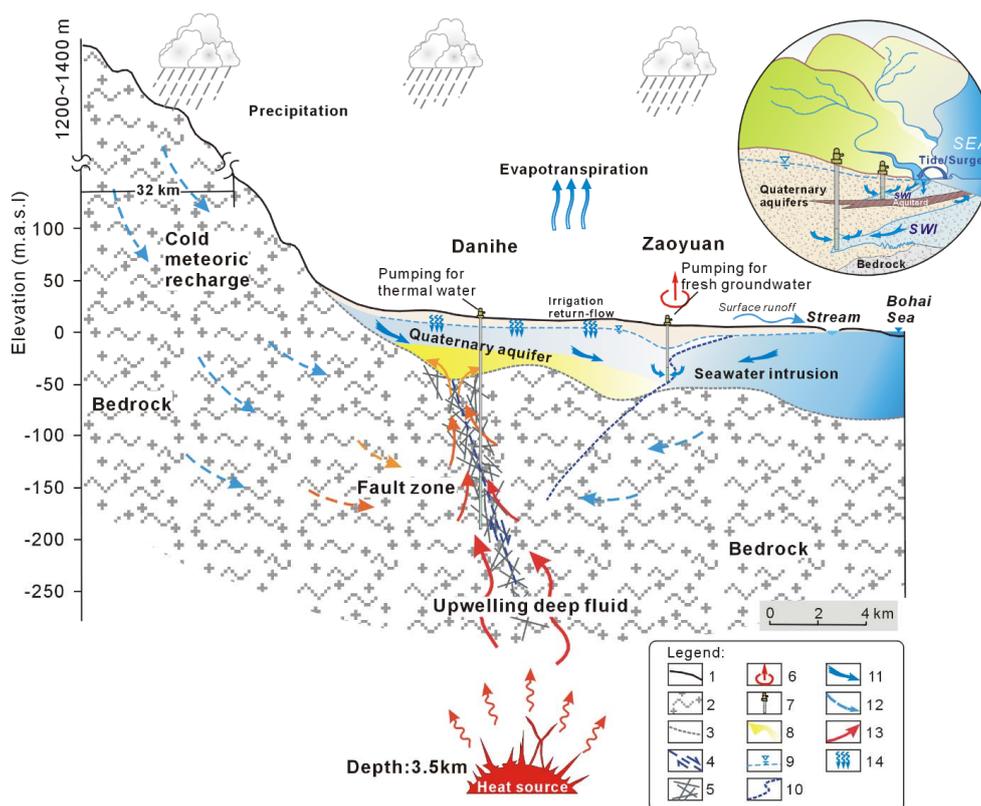


Figure 15. Conceptual model of groundwater flow system in the Yang-Dai River coastal plain
 Explanation: 1- Land surface; 2- Bedrock; 3- Boundary between Quaternary sediments and bedrock; 4-Fault; 5- Permeable fracture zone; 6- Concentrated groundwater pumping zone; 7- Pumping wells; 8- Zone affected by upflow of geothermal fluids; 9- Groundwater table; 10- Potential interface between fresh- and salt-water; 11-Groundwater flow direction in Quaternary aquifers; 12- Groundwater flow in bedrocks; 13- Geothermal groundwater flow direction; 14- Irrigation return-flow.



Table 1. Physical, hydrochemical and isotopic data of the water samples collected from the Yang-Dai River coastal plain

WaterType	ID	Sampling Time	Ele. m	WellDepth m	WaterTable Depth(m)	EC $\mu\text{S/cm}$	pH	T $^{\circ}\text{C}$	ORP mV	DO mg/L	Cl ⁻ mg/L	NO ₃ ⁻ mg/L	SO ₄ ²⁻ mg/L	HCO ₃ ⁻ mg/L	Ca ²⁺ mg/L	Na ⁺ mg/L	K ⁺ mg/L	Mg ²⁺ mg/L	Sr ²⁺ mg/L	δH ‰	$\delta^{18}\text{O}$ ‰	
Shallow Groundwater Samples																						
	G4	Aug.2010	5	15	4.2	741	6.5	20.5	6	3.9	124.3	92.6	88.7	136.9	70.3	69.2	2.3	21.8	0.67	-50	-6.9	
	G27	Aug.2010	3	12	10.4	1014	6.4	14.8	16	3.8	177.5	163.0	121.0	128.0	122.0	50.6	2.6	33.4	1.22	-50	-6.4	
	G12	Aug.2010	8	10	1152	7.2	25.0	9	5.9	276.9	31.9	68.8	148.8	158	148.8	158	201.9	11.1	16.2	0.25	-51	-6.8
	G17	Aug.2010	5	25	624	7.1	19.1	13	3.3	88.8	39.5	58.3	233.3	98.4	41.4	4.3	7.5	0.33	0.71	-55	-11.0	
	G18	Aug.2010	5	23	1.7	934	7.4	14.9	12	7.1	124.3	137.8	98.7	235.2	133.3	48.3	2.2	23.2	0.71	-55	-7.1	
	G1	Sep.2009	6	8	2.4	1673	8.3	19.4		220.1	2.0	148.3	207.9	84.3	119.6	17.1	49.9	0.66	-51	-7.1		
	G4	Sep.2009	6	15	4.6	1295	7.9	13.9		124.3	178.5	108.7	92.4	104.4	64.4	2.7	35.9	0.85	-44	-6.9		
	G5	Sep.2009	11	16	6.5	1544	7.8	13.3		303.4	162.3	108.2	38.5	124.5	103.7	1.1	39.1	1.14	1.14	-47	-6.5	
	G23	Sep.2009	15	30	3.0	566	7.9	14.9		88.8	163.4	54.0	51.9	97.4	34.5	0.5	15.6	0.55	-48	-6.6		
	G7	Sep.2009	17	8	1.7	901	8.4	14.9		81.7	41.2	74.2	69.3	64.0	33.2	1.0	16.5	0.41	-47	-6.4		
	G8	Sep.2009	4	13	4.3	1621	7.8	14.0		334.6	85.4	90.1	69.3	132.5	91.9	1.4	38.5	1.19	-44	-5.3		
	G11	Sep.2009	4	13	1237	8.6	18.4		222.9	74.5	98.7	30.8	87.4	80.1	1.8	29.0	0.84	-43	-5.5			
	G12	Sep.2009	8	10	1114	7.9	15.0		174.0	46.0	76.4	107.8	86.2	73.6	4.3	27.2	0.53	-48	-6.0			
	G28	Sep.2009	6	23	6.8	1748	8.2	14.6		127.8	22.5	38.4	107.8	88.2	33.3	1.3	14.0	0.58	-54	-7.4		
	G18	Sep.2009	9	23	5.5	2850	7.8	13.7		110.1	152.9	76.7	53.9	120.3	23.1	1.1	20.5	0.65	-50	-6.7		
	G20	Sep.2009	12	30	1602	7.7	19.5		246.9	107.1	135.9	107.8	158.2	82.8	1.7	26.3	3.94	-50	-6.1			
	G17	Sep.2009	6	33	7.9	819	8.1	14.3		243.5	126.4	141.7	161.7	176.3	105.5	1.8	24.6	0.71	-53	-6.9		
	G3	Jun.2008	4	20	1573	7.3	14.5		315.0	184.4	54.9	67.3	152.1	3.8	33.2	0.46	-43	-4.2				
	G4	Jun.2008	6	6	688	7.5	14.1		74.6	75.8	83.2	60.4	58.2	45.2	1.2	20.6	0.47	-48	-4.8			
	G5	Jun.2008	5	11	1455	7.4	14.7		310.9	119.5	92.7	52.2	121.3	95.4	0.9	29.9	1.43	-54	-6.3			
	G8	Jun.2008	4	30	1402	7.3	14.7		349.7	55.9	81.5	85.1	118.0	88.1	1.2	36.9	1.08	-40	-4.7			
	G12	Jun.2008	8	10	1285	7.6	19.3		281.0	29.8	56.8	74.1	9.7	205.7	11.0	13.7	0.19	-50	-6.0			
	G17	Jun.2008	6	33	1462	8.1	15.0		227.9	92.6	123.8	175.7	179.2	72.1	1.1	15.9	0.57	-56	-6.1			
	G18	Jun.2008	9	23	1125	7.9	18.8		115.5	144.2	44.4	90.6	103.7	31.8	1.4	13.7	0.41	-53	-6.8			
	G20	Jun.2008	12	30	1210	9.1	26.3		234.3	18.2	90.6	230.6	122.0	81.4	2.6	22.4	2.10	-39	-4.4			
	G1	Aug.2010	6	11	2750	7.4	17.3	143	1.5	312.4	120.0	337.1	205.4	112.0	294.4	29.9	41.3	0.54	-56	-7.7		
	G3	Aug.2010	4	8	2490	5.9	14.7	74	1.4	654.3	106.0	283.7	86.3	128.2	283.6	6.6	78.6	0.97	-47	-5.8		
	G15	Aug.2010	4	30	2290	6.6	15.4	3	3.3	435.7	181.3	263.7	244.1	164.3	164.3	242.8	6.6	60.8	1.60	-50	-6.3	
	G26	Aug.2010	4	18	4080	6.4	15.2	145	1.3	784.6	339.4	341.4	485.2	192.3	538.2	36.0	72.1	1.20	-46	-6.8		
	G11	Aug.2010	6	13	3490	6.5	15.0	47	1.3	646.1	414.1	402.7	402.6	204.8	441.8	34.9	76.3	1.51	-52	-6.8		
	G20	Aug.2010	12	30	1513	6.7	18.8	4	4.3	596.4	177.9	245.0	184.6	268.6	225.4	2.0	29.2	3.95	-51	-7.0		
	G5	Aug.2010	5	16	1500	6.2	15.6	7	5.9	447.3	253.3	304.7	79.3	204.1	188.6	0.8	47.5	0.78	-54	-7.7		
	G8	Aug.2010	4	30	2.1	1563	6.3	15.7	120	4.9	596.4	141.4	188.1	104.2	196.3	220.8	3.2	38.7	1.87	-44	-5.3	
	G7	Aug.2010	5	8	1438	6.8	19.3	24	6.1	299.6	338.3	192.7	128.0	164.2	151.8	0.3	50.0	0.66	-60	-8.9		
	G19	Aug.2010	8	11	2380	6.5	16.1	42	3.2	600.0	211.9	192.1	119.1	241.4	199.5	6.6	42.4	7.10	-52	-6.8		
	G24	Aug.2010	4	21	4400	6.8	15.5	4	5.9	646.1	952.1	813.3	223.3	484.1	349.6	3.4	117.4	2.44	-52	-8.1		
	G22	Sep.2009	2	20	2170	8.0	17.8			408.3	2.0	408.3	130.9	109.2	226.8	21.3	54.2	0.70	-48	-6.5		
	G10	Sep.2009	3	18	9770	7.5	15.1			2563.1	27.7	25.3	870.2	181.4	1340.0	98.1	123.0	1.92	-44	-5.7		
	G14	Sep.2009	2	8.5	1886	8.5	16.6			291.1	109.8	216.4	92.4	117.8	164.7	19.3	46.5	0.82	-55	-7.9		
	Brackish Groundwater																					



G24	Sep.2009	11	21	5.3	1003	7.8	13.6	454.4	441.5	128.1	115.5	252.3	165.6	0.6	36.2	0.93	-48		
G19	Sep.2009	12	11	4.7	2560	8.1	14.6	622.0	91.5	115.6	69.3	214.9	180.7	6.0	49.4	4.87	-6.2		
G1	Jun.2008	6	8		3150	8.5	15.0	717.1	87.7	285.6	98.8	9.1	527.4	26.7	26.6	0.17	-54		
G11	Jun.2008	6	13		2370	7.2	12.2	451.2	211.3	162.0	145.1	201.2	22.0	52.6	1.00	-49	-6.0		
G14	Jun.2008	3	8.5		2050	7.4	18.6	372.8	267.3	206.1	49.4	136.4	171.3	12.7	49.5	0.93	-48		
G15	Jun.2008	4	18		5990	7.5	15.6	1675.6	146.8	110.1	47.9	246.5	764.7	20.2	104.9	2.17	-45		
Deep Groundwater samples:																			
G25	Aug.2010	4	95		444	6.6	23.5	41	4.4	68.1	71.0	48.2	89.3	52.5	35.9	1.1	10.4	0.25	-56
G16	Sep.2009	3	110		1214	7.9	19.9	214.4	66.1	100.1	76.3	97.8	2.8	19.5	268	-8	-7.7		
G29	Sep.2009	6	60		1291	8.1	20.9	255.6	26.3	57.5	69.3	8.0	205.9	12.3	14.6	0.20	-51	-6.6	
G29	Aug.2010	6	60	6.2	3220	7.2	24.1	12	5.6	803.4	143.0	284.7	178.6	386.8	189.3	4.3	771.6	6.95	-50
G16	Aug.2010	3	110		1733	7.3	21.5	16	4.8	766.8	5.5	67.1	205.4	246.2	197.8	4.5	37.9	4.00	-56
G9	Jun.2008	9	104		3110	7.8	15.0	823.6	47.4	110.5	85.1	255.1	222.2	5.1	36.7	7.40	-47	-5.3	
G8	Sep.2009	9	104		3190	8.5	20.4	917.4	65.8	116.6	61.6	279.8	245.1	5.7	40.6	8.02	-51	-6.2	
G9	Aug.2010	9	104		4600	6.3	24.3	18	4.0	1228.3	337.4	296.3	92.3	392.2	455.4	5.5	45.3	11.59	-48
G14*	Aug.2010	5	110	2.1	2850	6.1	22.9	120	1.8	563.8	433.7	491.4	134.0	186.0	312.8	52.3	96.6	1.06	-53
G13	Aug.2010	3	90		3230	8.9	19.2	36	2.4	908.8	210.6	231.8	92.3	92.0	413.5	29.0	115.7	0.26	-45
G13	Jun.2008	3	90		3180	7.9	13.5	945.4		122.2	52.2	51.4	351.9	25.0	105.4	0.55	-45	-5.1	
G13	Sep.2009	3	90	2.6	3070	8.1	15.3	882.0		91.9	38.5	36.5	362.9	26.0	105.3	0.35	-43	-5.2	
G2	Jun.2008	2	60		3780	7.7	18.5	1093.4	200.3	96.1	67.3	484.1	25.8	103.1	1.03	46	-5.1		
River water samples:																			
Fresh water samples																			
Dai River	S9	Aug.2010	511	7.2	22.2	22	5.5	80.3	65.2	107.9	86.3	72.1	29.9	3.7	16.7	0.34	-66	-9.4	
Dai River	S12	Aug.2010	485	7.5	25.8	18	7.3	71.3	41.9	83.7	83.4	54.3	27.6	4.0	15.1	0.30	-69	-9.7	
Dai River	S8	Aug.2010	495	7.3	22.1	24	5.8	68.6	54.9	96.8	89.3	62.3	27.2	4.1	15.9	0.32	-67	-9.6	
Yang River	S6	Aug.2010	507	7.0	23.3	17	4.5	66.0	51.9	80.5	107.2	61.6	32.2	3.8	16.8	0.30	-65	-9.2	
Yang River	S2	Aug.2010	435	7.3	7.3	6	4.9	63.2	40.2	92.2	101.2	59.3	29.9	4.3	14.0	0.26	-71	-10.1	
Yang River	S5	Sep.2009	718	8.4	24.4			85.2	12.9	62.0	107.8	46.1	52.0	6.4	17.4	0.36	-45	-5.8	
Yang River	S4	Sep.2009	2630	8.1	24.7			733.1	6.6	142.2	115.5						-42	-5.3	
Yang River	S6	Sep.2009	718	8.3	25.6			99.4	6.6	57.5	107.8	44.3	59.8	5.0	16.4	0.30	-43	-7.2	
Dai River	S9	Sep.2009	560	8.0	23.6			88.8	6.8	60.5	92.4	57.2	33.2	3.3	16.7	0.36	-46	-5.8	
Dai River	S8	Sep.2009	1013	8.2	23.6			174.0	6.4	82.2	92.4	52.1	98.1	6.0	23.6	0.55	-43	-5.4	
Yang River	S6	Jun.2008	1166	7.5	12.6			81.7	12.6	71.3	112.5	53.6	47.5	4.9	18.0	0.31	-49	-5.5	
Dai River	S10	Jun.2008	1255	9.1	28.0			208.9	23.1	73.7	175.7	60.7	123.2	10.8	24.2	0.57	-44	-3.3	
Dai River	S9	Jun.2008	1163	7.8	11.8			92.3	7.5	52.2	90.6	54.7	31.1	3.6	19.5	0.35	-40	-3.9	
Brackish and salt water samples																			
Yang River	S3	Sep.2009	34800	7.8	15.5			11289.4	1684.7	115.5	251.0	5658.7	231.3	690.1	4.29	21	-2.4		
Dai River	S12	Sep.2009	47100	7.5	23.7			16766.3	2416.5	77.0	412.1	10074.3	398.2	1306.0	7.64	-12	-1.2		
Dai River	S11	Sep.2009	52500	8.5	23.4			1601.5	2.8	258.7	84.7	79.7	801.4	32.9	102.7	0.93	-41	-5.2	
Yang River	S1	Jun.2008	39800	8.7	24.2			14953.5	2035.9	134.5	313.0	7498.0	270.5	928.3	5.55	-15	-1.1		
Yang River	S2	Jun.2008	20200	8.8	28.5			8328.3	912.1	189.4	233.4	4094.2	147.9	495.9	3.37	-28	-2.5		
Dai River	S7	Jun.2008	49500	8.4	11.5			16677.1	2261.3	113.9	349.0	8730.0	326.4	1084.0	6.36	-11	-0.6		
Sea water:	SW1	Aug.2010	45600	7.8	25.5	83	4.90	14768.3	810.1	4047.0	148.8	312.7	8326.4	293.6	1007.0	5.79	3.8	1.1	
Sea water:	SW1	Sep.2009	47700	7.8	24.2			16568.0	2394.3	92.4	352.2	8922.0	322.2	1107.0	6.45	-10	-1.1		
Sea water:	SW2	Sep.2009	39500	7.3	23.2			14484.8	1926.6	107.8	313.2	7214.0	267.9	916.9	5.43	-17	-2.0		