



1 Response of water vapour D-excess to land-
2 atmosphere interactions in a semi-arid
3 environment

4

5 Stephen D. Parkes¹, Matthew F. McCabe^{1,2}, Alan D. Griffiths³, Lixin Wang⁴, Scott Chambers³,

6 Ali Ershadi^{1,2}, Alastair G. Williams³, Josiah Strauss², Adrian Element³

7

8 ¹ Water Desalination and Reuse Centre, King Abdullah University of Science and Technology
9 (KAUST), Jeddah, Saudi Arabia

10 ² Department of Civil and Environmental Engineering, University of New South Wales, Sydney,
11 Australia

12 ³ Australian Nuclear Science and Technology Organization, Sydney, New South Wales,
13 Australia

14 ⁴ Department of Earth Sciences, Indiana University–Purdue University Indianapolis (IUPUI),
15 Indianapolis

16

17 Corresponding Author: Stephen Parkes, stephen.parkes@kaust.edu.sa

18 Keywords: Stable isotopes, D-excess, water vapour, land-atmosphere coupling

19



20 Key points:

- 21 • Examine the influence of local land-atmosphere coupling on water vapour isotopes
- 22 • Diurnal cycle of D-excess in water vapour is determined by an interplay between large
23 scale moisture sources and nocturnal processes
- 24 • The D-excess of the evaporation fluxes impose negative forcing on the ambient vapour
- 25 • Nocturnal D-excess values are determined by surface exchange and turbulent mixing

26



27 **Abstract**

28 The stable isotopic composition of water vapour provides information about moisture sources and
29 processes that is difficult to obtain with traditional measurement techniques. Recently, it has been
30 proposed that the D-excess ($d_v = \delta^2\text{H} - 8 \times \delta^{18}\text{O}$) of water vapour can provide a diagnostic tracer
31 of continental moisture recycling. However, D-excess exhibits a diurnal cycle that has been
32 observed across a variety of ecosystems and may be influenced by a range of processes beyond
33 regional scale moisture recycling, including local evaporation (ET) fluxes. There is a lack of
34 measurements of D-excess in evaporation (ET) fluxes, which has made it difficult to assess how
35 ET fluxes modify the D-excess in water vapour (d_v). With this in mind, we employed a chamber
36 based approach to directly measure D-excess in ET (d_{ET}) fluxes. We show that ET fluxes imposed
37 a negative forcing on the ambient vapour and could not explain the higher daytime d_v values. The
38 low d_{ET} observed here was sourced from a soil water pool that had undergone an extended drying
39 period, leading to low D-excess of the soil moisture. A strong correlation between daytime d_v and
40 locally measured relative humidity was consistent with an oceanic moisture source, suggesting
41 that remote hydrological processes were the major contributor to daytime d_v variability. During
42 the early evening, ET fluxes into a shallow nocturnal inversion layer caused a lowering of the d_v
43 values near the surface. In addition, transient mixing of vapour with a higher D-excess from above
44 the nocturnal inversion modified these values, causing large within night variability. These results
45 indicate d_{ET} can generally be expected to show large spatial and temporal variability and to depend
46 on the soil moisture state. For long periods between rain events, common in semi-arid
47 environments, ET would be expected to impose negative forcing on the surface d_v . The variability
48 of D-excess in ET fluxes therefore needs to be considered when using d_v to study moisture



49 recycling and during extended dry periods may act as a tracer of the relative humidity of the
50 oceanic moisture source.

51

52 **1 Introduction**

53 Climate change has the potential to significantly impact surface and atmospheric water budgets.
54 Our best understanding of future exchanges between the atmospheric water cycle and the land
55 surface on a regional to global scale, is likely to be gained through application of numerical models
56 (Decker et al. 2015; Evans and McCabe, 2010; Harding and Snyder, 2012; Wei et al. 2012).
57 Consequently, continual improvement of available models is essential, but is contingent upon
58 ongoing validation and evaluation of model performance over a broad range of landscapes and
59 climate types (McCabe et al. 2016). To do this effectively, a range of datasets are necessary to
60 evaluate the variety of processes represented within these models. Unfortunately, datasets that
61 evaluate land-atmosphere exchanges at the process level are rare.

62 Water is composed of a number of stable isotopologues that have sufficient abundance to be
63 measured in atmospheric water vapour ($^1\text{H}_2^{16}\text{O}$, $^1\text{H}^2\text{H}^{16}\text{O}$, $^1\text{H}_2^{18}\text{O}$ and $^1\text{H}_2^{17}\text{O}$). The deviation of
64 the isotope ratios, reported as

$$65 \quad \delta = \left[\frac{R_{\text{sample}}}{R_{\text{VSMOW}}} - 1 \right] \% \quad (1)$$

66 where R is the isotope ratio ($^2\text{H}/^1\text{H}$ or $^{18}\text{O}/^{16}\text{O}$) and VSMOW (Vienna Standard Mean
67 Ocean Water) is the international standard used for reporting water isotope ratios, have potential
68 to evaluate land-atmosphere exchanges by discriminating different processes based on their



69 isotopic signature (Berkelhammer et al. 2013; Lee et al. 2009; Noone et al. 2013; Risi et al. 2013).
70 Isotopic ratios of water vapour ($\delta^2\text{H}$ and $\delta^{18}\text{O}$) can therefore provide information that is either
71 complimentary or even unobtainable when using conventional measurement techniques.

72 The utility of water isotope ratios for tracing sources of moisture derives from the characteristic
73 isotopic fractionation that occurs when water undergoes a phase change, causing light water
74 molecules to preferentially accumulate in the vapour phase. Soil moisture is typically enriched in
75 heavy isotopes relative to the ocean (Gat, 1996), so water vapour derived from land surface
76 evaporation is expected to have a different isotopic composition to moisture evaporated from the
77 ocean. This has led to a number of studies using stable isotope ratios of precipitation to partition
78 between oceanic and land derived sources (Froehlich et al. 2008; Tian et al. 2001). However, land-
79 atmosphere exchange is not restricted to periods of precipitation, and there are relatively few
80 studies examining the role of land-atmosphere exchange on ambient humidity budgets using stable
81 isotope observations of vapour (Aemisegger et al. 2014; Risi et al. 2013).

82 In addition to the source of moisture, the magnitude of the isotopic fractionation that occurs as
83 water evaporates is related to the temperature of the liquid surface and the humidity gradient
84 between the evaporating surface and the atmosphere (Craig and Gordon, 1965). The temperature
85 dependent equilibrium exchange between the liquid and vapour is the largest contributor to
86 isotopic fractionation during evaporation, with the fractionation for ^2H approximately a factor of
87 8 greater than ^{18}O . The effect of kinetic fractionation associated with moisture diffusing from the
88 thin laminar layer of vapour in equilibrium with the water surface to the turbulent atmosphere
89 above is influenced by the relative humidity of the atmosphere and wind speed (Merlivat and
90 Jouzel, 1979). The kinetic fractionation factors for ^2H and ^{18}O are similar, causing the ratio of



91 ^2H to ^{18}O in the evaporating vapour to decrease as kinetic effects increase with decreasing
92 relative humidity. This phenomenon has been observed for evaporative conditions over the
93 Mediterranean sea (Gat et al. 2003; Pfahl and Wernli, 2009) and the Great Lakes in Northern USA
94 (Gat et al. 1994; Vallet-Coulomb et al. 2008).

95 The D-excess parameter ($\text{D-excess} = ^2\text{H} - 8 \times ^{18}\text{O}$) (Dansgaard, 1964), quantifies the non-
96 equilibrium isotopic fractionation. A reproducible relationship between the D-excess and relative
97 humidity near the ocean surface has been observed across a wide range of locations (Kurita, 2011;
98 Pfahl and Wernli, 2008; Steen-Larsen et al. 2015; Uemura et al. 2008). Given this, it has been
99 suggested that for precipitation, the D-excess is a good tracer of sea surface evaporative conditions
100 (Masson-Delmotte et al. 2005; Merlivat and Jouzel, 1979). However, this view has recently been
101 challenged due to the role local and regional scale land-atmosphere coupling has in modifying the
102 D-excess of atmospheric humidity over diurnal (Lai and Ehleringer, 2011; Simonin et al. 2014;
103 Welp et al. 2012; Zhao et al. 2014) and synoptic timescales (Aemisegger et al. 2014). As evidence
104 for the role ET plays in modifying the D-excess of water vapour (d_v), a diurnal cycle of the d_v near
105 the land surface across a range of land surface types has been observed (Berkelhammer et al. 2013;
106 Simonin et al. 2014; Welp et al. 2012). The diurnal cycle shows higher values in the day, which
107 has been proposed to be driven by entrainment (Lai and Ehleringer, 2011; Welp et al. 2012), local
108 evapotranspiration sources (Simonin et al. 2014; Zhao et al. 2014) and meteorological conditions
109 affecting the D-excess of the evaporative fluxes (d_{ET}) (Welp et al. 2012; Zhao et al. 2014), coupled
110 with low nocturnal values resulting from equilibrium exchange between liquid and vapour pools
111 (Simonin et al. 2014) and dew fall (Berkelhammer et al. 2013). For synoptic scales, Aemisegger
112 *et al.* (2014) showed that moisture recycling from the land surface had a significant impact on d_v



113 for *in-situ* measurements in Switzerland. The studies of the d_v diurnal cycles have largely relied
114 on isotopic models to assess the contribution of ET fluxes, but a lack of d_{ET} measurements make
115 it difficult to draw robust conclusions.

116 The evidence provided by these studies suggest that d_v is a tracer of moisture recycling both on
117 diurnal and synoptic time scales as well as informing upon the dynamics of surface moisture
118 budgets in the atmospheric boundary layer (ABL). However, as outlined by Welp *et al.* (2012),
119 ET and entrainment fluxes both increase as the ABL grows through the previous days residual
120 layer, which can make interpreting the role of local moisture recycling on d_v difficult. To overcome
121 this, Simonin *et al.* (2014) used a trajectory model to simulate the D-excess of vapour evaporated
122 over the ocean. As the d_v was greater than the modelled oceanic moisture source, it was assumed
123 that high daytime values were supported by local ET fluxes. Zhao *et al.* (2014) suggested that
124 since, on cloudy days, no diurnal cycle was observed for the d_v , that ET fluxes played a dominant
125 role. Whilst these studies provide compelling evidence for the role of ET driving the diurnal cycle
126 of d_v , no measurements of d_{ET} were made. To date the only measurements of d_{ET} have been
127 presented by Huang *et al.* (2014) over a maize crop in north west China. Interestingly, their direct
128 measurements conflicted with previous interpretations and showed that the d_{ET} invoked a negative
129 forcing on d_v , even though a strong diurnal cycle of high values in the day and low values at night
130 were observed. In order to better interpret the role of local moisture recycling on the diurnal cycle
131 of d_v , measurements of d_{ET} are required to assess if the negative forcing is consistent in different
132 ecosystems.

133 The aim of this work is to provide much needed d_{ET} measurements to investigate how ET fluxes
134 modulate the d_v diurnal cycle. To do this, chamber based measurements of the isotopic composition



135 of the ET fluxes were combined with *in-situ* measurements of water vapour isotope ratios,
136 meteorological and radon concentration observations. The data was collected in a region of the
137 semi-arid Murray Darling basin in south-eastern Australia. These data represent the first such
138 collection of the ^2H , ^{18}O and D-excess in water vapour from this region of Australia. The
139 augmentation of the chamber based measurements with in-situ observations provide a framework
140 to directly assess the role local ET fluxes have has on ambient vapour D-excess.

141 2 Methods

142 2.1 Site Description

143 During the austral autumn of 2011, a field campaign covering the period April 27 to May 11 was
144 conducted at the Baldry Hydrological Observatory (BHO) (-32.87, 148.54, 460 m above sea level)
145 located in the central-west of New South Wales, Australia (Figure 1). The climate of the region is
146 characterised as semi-arid with no clear wet season, a mean annual rainfall of 600 mm, and a mean
147 annual temperature of 24.2°C (source Australian Bureau of Meteorology, 2015,
148 <http://www.bom.gov.au/>). The BHO grassland flux tower was the central site of the measurements
149 and was located in a natural grassland paddock of dimensions approximately 900m (north-south)
150 by 300m (west-east), with a gentle slope decreasing in elevation by approximately 20 m from
151 southeast to northwest. The flux tower was located 650 m from the road to the south and 200 m
152 from a reforested paddock to the west. The forest site to the west and southwest was reforested in
153 2001 with *Eucalyptus camaldulensis*, *Eucalyptus crebra* and *Corymbia maculate*. At the time of
154 the campaign these trees were approximately 10 m tall. All other adjacent paddocks and most of
155 the surrounding region had similar surface characteristics to the grassland measurement site.



156 **2.2 Water stable isotope analyses**

157 **2.2.1 *In-situ* water vapour calibration and sampling**

158 Stable isotope ratios were monitored using two different *in-situ* analysers. *In-situ* measurements of
159 the isotopic composition in water vapour were made using a Wavelength Scanning Cavity Ring
160 Down Spectrometer (WS-CRDS L115-I, Picarro Inc., Sunnyvale, CA, USA), while the chambers
161 were interfaced to an Off Axis Integrated Cavity Output Spectrometer (OA-ICOS, DLT100, Los
162 Gatos Research (LGR), Mountain View, CA, USA) to determine the isotopic composition of the
163 ET flux. Using an automated continuous flow calibration system (built in-house), we
164 simultaneously determined the calibration coefficients for the two analysers. Calibration
165 experiments were designed to determine the water vapour mixing ratio cross-sensitivity of the
166 isotope ratios and linearity of the $\delta^2\text{H}$ and ^{18}O measurements. Due to logistical constraints, the
167 calibration system was not transported into the field, so corrections were determined by
168 compositing multiple calibration experiments run before and after the campaign.

169 During the campaign, a second portable calibration system was employed to monitor time
170 dependent drift of the Picarro analyser (CTC HTC-Pal liquid autosampler; LEAP Technologies,
171 Carrboro, NC, USA). Two standards spanning expected water vapour ^2H (-49.1 and -221.9‰)
172 and ^{18}O (-9.17 and -27.57‰) ranges were injected at approximately 18 mmol mol^{-1} on three
173 occasions during the campaign.

174 The uncertainty of the isotope measurements of the two analysers was estimated by applying
175 mixing ratio cross-sensitivity and linearity corrections to all calibration measurements collected
176 prior, during and after the campaign. For the Picarro instrument, measurement uncertainty was 0.8,
177 0.2 and 1.9‰ for $\delta^2\text{H}_v$, $\delta^{18}\text{O}_v$ and d_v , respectively. No calibrations were performed for the LGR in



178 field, so the measurement uncertainty was estimated from calibration measurements made before
179 and after the campaign, which was 0.9, 0.4 and 3.3‰ for ^2H , ^{18}O and d_v .

180 Although no calibration experiments were run on the LGR during the campaign, simultaneous *in-*
181 *situ* measurements were made with the Picarro when chamber measurements were not operated.
182 During the day, the average difference was 0.3 (± 1.7), 0.1 (± 0.6) and 0.2 (± 5.1)‰ for ^2H , ^{18}O
183 and d_v , respectively. At night, while the Picarro was able to maintain a steady cavity and optical
184 housing temperature, the LGR cavity temperature dropped by up to 8°C. In response to the drop
185 in cavity temperature, night time LGR measurements of $\delta^{18}\text{O}$ and d_v and to a lesser extent the $\delta^2\text{H}$,
186 were physically unrealistic and discarded from subsequent analyses. Morning chamber based
187 measurements were therefore restricted to after 9:00 am, when the LGR cavity temperature had
188 stabilised and *in-situ* measurements were again in agreement with the Picarro.

189 A schematic diagram illustrating the sampling design for water vapour is shown in Figure 2. Half-
190 hourly vertical profiles of humidity and isotopes were sampled by drawing air to the *in-situ*
191 analyser through 10 mm O.D. PTFE tubing, located at 5 heights on a 7.5 m tower (0.5, 1, 2, 5 and
192 7.5 m Above Ground Level). The instrument was interfaced to a 5 inlet manifold that enabled
193 sequential sampling of the different heights. A vacuum pump (MV 2 NT, Vacuubrand, Wertheim,
194 Germany) was used to draw air from all inlets to the analyser. To avoid condensation, sample tubes
195 and intakes were wrapped in 15 W m⁻¹ heat tape, insulated by Thermobreak pipe and placed inside
196 100 mm PVC pipe. The sample tube temperature was controlled using a Resistance Thermometer
197 Detector (RTD) coupled to a CAL3300 temperature controller (CAL controls Ltd., Grayslake, IL,
198 USA). The inlets at each height were constructed from inverted funnels with mesh filters. In this
199 study we present block hourly averages of all measurements collected at all heights.



200 2.2.2 Flux chambers

201 To separate the isotopic signatures of the ET flux components, flux chambers were deployed on
202 both bare soil and vegetated plots to determine the isotopic signature of the evaporative fluxes. An
203 open chamber was designed with a high volume to footprint ratio to avoid the chamber mixing
204 ratio rapidly reaching the dew point temperature (causing condensation) and to minimise impacts
205 on the evaporation environment. A schematic of the chamber design is shown in Figure 3. Four
206 flanged metal collars were inserted ~10 cm into the soil column two days before the beginning of
207 the campaign. All vegetation was removed from bare soil plots when the metal collars were
208 inserted into the soil. A single chamber cover was constructed of 4 mm G-UVT Plexiglass (Image
209 Plastics, Padstow, Australia), selected for its higher transmittance of UV and blue light. The
210 dimensions of the chamber were 0.1 x 0.1 x 0.8 m (width x length x height), with the inlets and
211 outlets at 0.1 and 0.7 m above the surface, respectively. All sampling tube was 10 mm PTFE. The
212 inlet to the chamber was connected to tubing that drew in air from 1.5 m above the ground surface.
213 The outlet was connected to a flowmeter (VFA-25, Dwyers, Michigan City, IN, USA) that
214 regulated the air flow at 10 L.min⁻¹ and was driven by a two-stage diaphragm pump. A T-piece
215 was connected to the LGR, which bled off the required air flow. All tubing between chamber and
216 the analyser were wrapped in heating tape (15 W m⁻²) and foam insulation.

217 To monitor the internal chamber environment, an air temperature and humidity probe (HMP155,
218 Vaisala, Vantaa, Finland) was mounted inside the chamber. To monitor the attenuation of the
219 incoming radiation by the chamber, photosynthetic flux density was measured (LI-190R, Licor,
220 Lincoln, NE, USA) inside and outside the chamber. Ten second averages of the temperature,
221 relative humidity and photosynthetic flux density were stored in a datalogger (CR1000, Campbell



222 Scientific, Logan, UT, USA). In the supplementary section we use these ancillary measurements
 223 to assess the impact of the change in the evaporative conditions caused by the chamber on the
 224 isotopic composition of the ET flux. The largest contributor to uncertainty caused by changing the
 225 evaporative environment was the temperature, although these affects were small compared to the
 226 overall variability of the chamber derived ET isotopic compositions.

227 **2.2.3 Isotopic composition of ET flux from chamber measurements**

228 Using the open chamber system, we determined the isotopic composition of the ET flux using the
 229 Keeling mixing model (Keeling, 1958; Wang et al. 2013) given by

$$230 \quad \delta_{chamber} = q_{BG} \frac{(\delta_{BG} - \delta_{ET})}{q_{chamber}} + \delta_{ET} \quad (2)$$

231 where q_{BG} is the water vapour mixing ratio entering the chamber through the inlet and δ_{BG}
 232 its isotopic composition, $q_{chamber}$ is the mixing ratio in the chamber and δ_{ET} is the isotopic
 233 composition of the ET flux. For the Keeling method, δ_{ET} is determined from the intercept of $\delta_{chamber}$
 234 against $1/q_{chamber}$. A key assumption of the Keeling method is that the isotopic composition of the
 235 background vapour and the evaporation flux remain constant during the chamber measurements.
 236 For chamber measurements longer than 5 minutes, non-linear Keeling plots were commonly
 237 observed, indicating a change in isotopic composition of one of the sources of vapour. We
 238 therefore restricted the Keeling analysis of the chamber measurement to a maximum of 5 minutes
 239 after an increase in the concentration was observed by the analyser. The analysis was also restricted
 240 to periods where the H_2O mixing ratio was increasing, so that analysis was generally performed
 241 on 2-5 minutes of data. In addition, only chamber measurements where the correlation between
 242 $\delta_{chamber}$ and $1/q_{chamber}$ was significant ($p < 0.001$) were included in this analysis. A few chamber



243 measurements where obvious non-linearity or very small changes in q_{chamber} occurred were also
 244 subjectively removed. Of a total of 105 chamber measurements made from the 4 vegetation plots
 245 during the campaign, 99 measurements of the $^2\text{H}_{\text{ET}}$, and 97 measurements of $^{18}\text{O}_{\text{ET}}$ and d_{ET} were
 246 retained. For the bare soil plots, 84 of the 86 chamber measurements were retained for the $^2\text{H}_{\text{ET}}$,
 247 and 77 of the $^{18}\text{O}_{\text{ET}}$ and d_{ET} . The eight plots were sampled 2 to 4 times each day on all days except
 248 the first two days of the campaign, and the 2nd and 5th of May. Sampling was restricted to between
 249 9:00 and 17:00 LST (local solar time) as the large temperature dependence of the LGR at low
 250 ambient temperatures limited the accuracy of the chamber measurements.

251 Results from the vegetated plots were used to determine the isotopic composition of the ET flux
 252 and determine how it influences surface vapour. The bare soil plots were used to determine the
 253 isotopic composition of the soil evaporation flux and to provide an estimate of the isotopic
 254 composition of the soil water at the evaporation front. The isotopic composition of the water at the
 255 evaporation front (δ_L) was determined by rearranging the Craig and Gordon model:

$$256 \quad \delta_L = \frac{\delta_E(1-RH)+RH \delta_A+\epsilon+\epsilon_k}{\alpha} \quad (3)$$

257 where the isotopic composition of the evaporation flux (δ_E) is taken from the bare soil
 258 chamber measurements, relative humidity (RH) normalised to the surface temperature determined
 259 from infrared surface temperature measurements (section 2.3), and the ambient vapour isotope
 260 composition (δ_A) determined from Picarro *in-situ* measurements. Equilibrium fractionation and
 261 enrichment factors (ϵ , $\epsilon_k = (-1)\%$) were calculated from the surface temperature measurements
 262 using the equations of Horita and Wesolowski (1994), while the kinetic enrichment factor (ϵ_k) was
 263 determined as in Gat (1996), but using the parameterisation of the exponent of the diffusion



264 coefficients described by Mathieu and Bariac (1996) and the diffusion coefficients determined by
265 Merlivat (1978).

266 **2.2.4 Iso-Forcing of ET**

267 The isotopic composition of the near-surface atmospheric water vapour is modified by surface ET
268 fluxes. The impact of ET fluxes on surface vapour isotopes varies over diurnal timescales with the
269 strength of vertical mixing in the ABL or over synoptic timescales as background moisture
270 conditions change. The magnitude and isotopic composition of the ET flux as well as the amount
271 of water vapour in the atmosphere also have an influence. The ET iso-forcing (I_{ET}) represents a
272 useful quantity to study the influence of ET fluxes on the surface vapour and is defined as:

$$273 \quad I_{ET} = \frac{F_{ET}}{H_2O} (\delta_{ET} - \delta_A) \quad (8)$$

274 where F_{ET} is the ET flux in $\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ taken from an eddy covariance tower, H_2O is the
275 ambient mixing ratio in $\text{mol}\cdot\text{air}\cdot\text{mol}\cdot\text{H}_2\text{O}^{-1}$ measured by the local meteorological tower, and δ_{ET}
276 and δ_A are the isotopic compositions of the evaporation flux and ambient water vapour,
277 respectively (Lee et al. 2009).

278 For each chamber measurement, a surface iso-forcing was calculated for $\delta^2\text{H}$, δ^{18} and D-excess
279 from the determined ET isotopic composition, the hourly averaged ET flux, mixing ratio and δ_A
280 values. The importance of the surface fluxes in modifying surface vapour isotope composition was
281 investigated for diurnal and synoptic timescales.



282 **2.2.5 Plant and soil sampling**

283 Grass samples were collected three times a day for the duration of the campaign. They were
284 sampled randomly within 100 m of the instrumentation. Each sample consisted of approximately
285 10 grass leaves, which were placed in 12ml Exetainer vials (Labco, Ceredigion, UK). The grass
286 samples were assumed to represent bulk leaf water. Soil samples were collected every 2 days
287 throughout the campaign by sampling from the top 5cm of the soil column. They were collected
288 in 50ml glass bottles. Soil and plant samples were stored in a fridge (4°C), before they were
289 distilled using a method similar to West *et al.* [2006] and analysed on a Delta V Advantage Isotope
290 Ratio Mass Spectrometer (Thermo Fisher Scientific Corporation, Massachusetts, United States).
291 For ^2H analysis, water samples were introduced into a H-Device containing a chromium reactor,
292 while for the ^{18}O analysis, water samples were equilibrated with CO_2 on a Gas Bench II
293 chromatography column (Thermo Fisher Scientific Corporation, Massachusetts, USA) before
294 being transferred to the IRMS for analysis.

295 **2.3 ET Fluxes and Meteorological measurements**

296 An eddy covariance system comprising a Campbell Scientific 3D sonic anemometer (CSAT-3,
297 Campbell Scientific, Logan, UT, USA) along with a LiCOR 7500 (Li-7500, LiCor Biosciences,
298 Lincoln, NB, USA) analyser was installed at an elevation of 2.5 m. The system was located
299 approximately 10 m from the stable isotope observation tower and sampled at 10 Hz, with flux
300 averages output at 30 minute intervals. The ET fluxes from the eddy covariance tower are used to
301 quantify the Iso-Forcing of ET on the overlying atmosphere.

302 A meteorological tower was co-located with the eddy covariance system, acting as an aid to the
303 interpretation of measurements. The tower comprised a Kipp and Zonen CNR4 radiometer,



304 Apogee infrared surface temperature, RIMCO rainguage, Vaisala HMP75C temperature and
305 humidity probe, RM Young wind sentry (wind speed and direction), Huskeflux ground heat flux
306 plate and Vaisala BaroCap barometric pressure sensor. Both meteorological tower data and eddy-
307 covariance data were inspected visually to detect and remove spikes. The low-frequency eddy
308 covariance data (30 minute resolution) were corrected for coordinate rotation (Finnigan et al. 2003)
309 and density effects (Leuning, 2007) using the PyQC software tool (available from
310 code.google.com/p/eddy).

311 **2.4 Radon-222 measurements**

312 The naturally occurring radioactive gas radon (^{222}Rn) is predominantly of terrestrial origin and its
313 only atmospheric sink is through radioactive decay (Zahorowski et al. 2004). The surface flux
314 density of radon is relatively constant in space and time, and since the half-life is much greater
315 than ABL mixing timescales, it is an ideal tracer of the strength of vertical mixing within the ABL
316 (Chambers et al. 2014; Griffiths et al. 2013; Williams et al. 2010). Hourly radon concentrations
317 were measured by an Alpha Guard (Saphymo GmbH, Frankfurt, Germany) placed in a ~20 L
318 enclosure. The enclosure was purged at ~15 L.min⁻¹ with a vacuum pump (2107 Series, Thomas,
319 Wisconsin, USA) that sampled from a height of 2 m through 10 mm O.D. PTFE tubing. Radon
320 measurements were used to aid the interpretation of the diurnal variations in vertical mixing (see
321 Griffiths et al. 2013).

322 **3 Results**

323 **3.1 Meteorological observations**

324 The two-week campaign was conducted under predominantly calm meteorological conditions. The
325 last rain event was 10 days prior to the campaign, after which clear skies saw the soil dry to a



326 moisture content close to minimum values observed for the site (Figure 4). In the middle of the
327 campaign (May 2nd), a cold front moved across south eastern Australia, producing cloudy
328 conditions and 1.4 mm of precipitation at the site. No change in soil moisture was observed
329 following the rain event over the 0-10 cm soil layer.

330 The wind direction was quite variable over the course of the campaign (see figure S2a and b).
331 Figure S3 shows that from 27th to 30th April, the dominant daytime wind direction was mainly
332 from the east. The predominant wind direction after May 3rd was from the south, except on the
333 7th and 8th when the wind was from the west and with a fetch from the adjacent forest. At other
334 times the fetch did not include the forested site. Daily maximum temperatures on clear days ranged
335 from 16 to 23°C, whilst night time minimum temperatures fell to between 8 and -4°C. The coldest
336 nights were observed from May 7th onwards, with temperatures falling below zero. On clear nights
337 the surface temperature fell below the dew point temperature as the air reached saturation,
338 indicating dew fall. Apart from the night of the 27-28th April and the cloudy nights between 1st
339 and 3rd May, dew or frost was observed at the site each morning, although dewfall and frost
340 formation was much heavier from the 7th May onwards.

341 Radon concentrations were low during the day, when the convective boundary layer reached its
342 maximum height, with high concentrations at night, when radon emissions from the surface were
343 confined within the shallow nocturnal boundary layer. The accumulation at night was quite
344 variable and indicated a varying degree of nocturnal stability, mixing depth and the occurrence of
345 transient mixing events (Griffiths et al. 2013). There was general agreement between high
346 nocturnal radon concentrations and low wind speeds, but no direct relationship. The lack of a direct
347 relationship indicates that radon can provide additional information about nocturnal mixing and



348 surface exchange that compliments standard meteorological measurements (Chambers et al.
349 2015a, 2015b; Williams et al. 2013).

350 The eddy covariance derived ET fluxes were in general quite low, reflecting the low soil moisture
351 content. The ET flux did show a marked increase the day after the small rain event on May 2nd
352 and noticeably smaller fluxes were observed after the first night that frost was observed. The health
353 of the grass visibly deteriorated over the last 4 days, coinciding with frost formation.

354 **3.2 Relationship between ^2H and ^{18}O of the different water pools**

355 A summary of the isotopic composition of all observed and modelled water pools are presented
356 in Figure 6. The global meteoric water line (MWL) is shown (Craig, 1961), along with the local
357 MWL from the nearby town of Lithgow (Hughes and Crawford, 2013). The local MWL is to the
358 left of the global MWL, illustrating the characteristically high D-excess of precipitation in the
359 region (Crawford et al. 2013; Hughes and Crawford, 2013). The ambient vapour observations align
360 closely with the local MWL, but with a distribution that fell both to the left and right of the local
361 MWL. Alignment between observations and the MWL show that equilibrium fractionation was
362 the dominant process modifying $\delta^2\text{H}$ and $\delta^{18}\text{O}$ in water vapour, while non-equilibrium kinetic
363 processes shift observations away from the MWL and are more easily observed for d_v
364 measurements.

365 Plant and soil water pools were enriched relative to the vapour and distributed to the right of the
366 MWL, indicating evaporative enrichment. The low slope for vegetation samples is consistent with
367 a number of previous studies across a range of ecosystems (Cernusak et al. 2002; Simonin et al.
368 2014; Zhao et al. 2014). ET flux isotopic compositions determined from the chambers was



369 enriched relative to the vapour and was distributed to the right of the MWL with a slope of 3.2.
370 Similar isotopic compositions were measured from bare soil and vegetated chambers, with mean
371 values and standard deviations (1 σ) of -47.1 (± 13) and -50.2 (± 11) for ^2H , -5.03 (± 3.8) and -6.3
372 (± 2.7)‰ for ^{18}O , and -6.3 (± 23) and -0.12 (± 15)‰ for D-excess, respectively. Only ^2H showed a
373 significant difference ($p < 0.01$) between the two surface types. The ET isotopic compositions did
374 not show a diurnal cycle, which may result from measurements being restricted mainly to the
375 middle of the day after the large changes in relative humidity and temperature. In addition, no
376 observable trend was seen for synoptic timescales, indicating isotopic compositions of the local
377 moisture source did not vary. Soil water isotopes at the evaporation front (u_L) were much more
378 enriched and had a much lower D-excess (50 ± 12 , 31 ± 3.8 and -131 ± 22 ‰ for $\delta^2\text{H}$, $\delta^{18}\text{O}$ and D-
379 excess) than the average soil moisture between 0 and 5 cm (-15 ± 4.2 , 2.6 ± 2.5 and -36 ± 17 ‰ for
380 $\delta^2\text{H}$, $\delta^{18}\text{O}$ and D-excess). These relatively enriched signatures indicate evaporative enrichment of
381 this water pool.

382 The low slope for the ^2H vs ^{18}O relationship of ET fluxes and liquid pools, illustrates the role of
383 non-equilibrium processes in the evaporation processes. These non-equilibrium processes are
384 likely to have occurred since the last rain event causing the observed enrichment and low D-excess
385 of the soil water pool. The similar isotopic compositions of ET fluxes from bare soil and vegetation
386 chambers indicate that soil evaporation was the dominant process contributing to total ET. The
387 extremely low D-excess values of the moisture at the evaporation front produced fluxes with a D-
388 excess which was low compared to the local ambient vapour. These results illustrate the
389 importance of the soil moisture state for the effect local ET fluxes have on atmospheric vapour D-
390 excess values.



391 3.3 *In-situ* water vapour isotopes and ET iso-forcing

392 The observed water vapour mixing ratio and stable isotope composition are shown in Figure 7.
393 The ^2H and ^{18}O variability was similar, reflecting changes in both the synoptic and local
394 meteorology. Leading up to the rain event on May 2nd, relatively moist conditions were observed
395 as air was transported from the warmer ocean off the east coast of Australia (see wind direction in
396 figure S3). After May 5th, drier conditions were observed caused by the lower dew point of air
397 being transported from the colder sea surface to the south of continental Australia. These moisture
398 source regions were confirmed by backward air trajectories calculated using the Stochastic Time-
399 Inverted Lagrangian Transport Model (STILT; Lin *et al.* (2003) not shown). These two time
400 periods are hereinafter referred to as the “wet period” (before May 2nd) and “dry period” (after
401 May 5th). The wet period had relatively moister conditions and coincided with more enriched
402 water isotopes and less diurnal variability. In the later part of the campaign, a reproducible diurnal
403 cycle for ^2H and $\delta^{18}\text{O}$ was observed, characterised by a sharp increase at sunrise before decreasing
404 from mid-morning, when vertical mixing increased until the next sunrise. These observations
405 emphasise the complex relationship between stable isotope observations in water vapour and both
406 local and synoptic scale meteorology.

407 The dominant feature in the d_v dataset was a robust diurnal cycle of high values in the day and low
408 values at night, consistent with what has been observed across a growing number of locations
409 (Bastrikov *et al.* 2014; Berkelhammer *et al.* 2013; Simonin *et al.* 2014; Welp *et al.* 2012; Zhao *et al.*
410 2014). For the wetter conditions at the start of the campaign, the daytime values were on average
411 lower than those observed for the dry period. Whilst the nocturnal d_v was consistently lower during
412 the night, nocturnal measurements were variable both from night to night and within individual



413 nights. However, no clear difference was observed between nocturnal observations during the wet
414 and dry periods. The differences between the contrasting daytime measurements of the wet and
415 dry periods indicate a role of large scale processes, whilst the lack of contrast in the nocturnal
416 observation show the importance of local processes.

417 The I_{ET} shown in Figure 7 shows large variability on each day, but was always positive for ^2H
418 and ^{18}O and mostly negative for D-excess. I_{ET} is most sensitive to the magnitude of the ET fluxes,
419 producing the greatest forcing on ambient vapour in the middle of the day. Examination of the
420 time series in Figure 7, shows that during the day, ^2H and $\delta^{18}\text{O}$ often decreased, even though the
421 I_{ET} was positive. Likewise, high d_v values observed during the day were associated with negative
422 I_{ET} and the highest daytime d_v observed during the dry period did not correspond to the least
423 negative I_{ET} . These observations illustrates that local ET fluxes were not very important for the
424 trends in d_v over the diurnal cycle or even over synoptic timescales.

425 **3.4 Relationship between water vapour isotopes and local meteorology**

426 The relationships between local meteorological variables and water vapour isotopes were
427 examined to interpret which local processes contributed to the isotope variability. The slope,
428 intercept, coefficient of determination and p-value are shown for the correlations between selected
429 meteorological variables and the isotopes in Table 1. These statistics are shown for both the hourly
430 observations and for the average daytime values (between 11:00 and 16:00 LST). Selecting the
431 daytime measurements removes the effect of the processes associated with the transition between
432 the stable nocturnal and daytime convective boundary layer. It also allows removal of nocturnal
433 periods, where local surface equilibrium exchange and dewfall are known to affect the isotopic
434 composition of ambient vapour. The correlations determined using only measurements in the



435 middle of the day therefore provide a better indicator of how local meteorology and its isotopic
436 composition modified ambient water vapour isotope ratios from day to day.

437 For ^2H , a weak but significant correlation with daytime I_{ET} was observed ($R^2=0.45$, $p<0.05$). The
438 correlation with I_{ET} potentially indicates that local ET fluxes were important for water vapour ^2H ,
439 but the slope for the relationship was negative. As I_{ET} was positive for ^2H , the case where local
440 ET fluxes drive daytime ^2H in water vapour should yield a positive relationship when the ET flux
441 ^2H is constant. The ^{18}O and ^2H showed weak but significant correlations with air temperature
442 ($R^2=0.24$ and 0.04 , respectively) and the mixing ratio ($R^2=0.2$ for both isotopes) for hourly
443 measurements, while ^2H also showed a weak significant correlation with RH ($R^2=0.09$). The weak
444 relationships with the local meteorology reinforces the role of larger scale precipitation processes
445 and atmospheric mixing that occurs as the moisture is transported to the site.

446 As already stated, the diurnal cycle was the main mode of variability for d_v , which led to significant
447 correlations with air temperature and RH for the hourly observations. While the local air
448 temperature and RH could modify d_{ET} on diurnal timescales and in turn local d_v , the chamber
449 measurements showed relatively constant d_{ET} over diurnal and synoptic timescales. These
450 correlations may therefore result from the coincident variation of the d_v , RH and air temperature
451 with the daily growth and decay of the ABL.

452 The daytime average d_v showed significant correlations with the air temperature, RH, ET flux and
453 mixing ratio. The relationship between daytime average ET fluxes and d_v was weak ($R^2=0.3$) and
454 positive, but as negative iso-forcing was determined for D-excess a negative relationship should
455 have been observed. Likewise, the slope of the linear regression between air temperature and d_v is



456 negative, counter to what theory would predict for local or remote moisture sources. The strongest
457 relationship was observed with daytime RH ($R^2 = 0.74$), with the negative slope consistent with
458 an expected increase in d_v with RH decreasing at the evaporation source. The strong relationship
459 of d_v with the daytime RH could indicate an important role for the evaporation conditions at remote
460 moisture sources, as is discussed below in Section 4.1.

461 **3.5 Diurnal variability of vapour isotopes**

462 Diurnal composites for the meteorology, radon concentrations and *in-situ* isotope observations are
463 shown in Figure 8, divided into the dry and wet periods defined above. At sunrise (approximately
464 06:30 LST), surface heating leads to initiation of vertical mixing, shown by the topping out of
465 Radon concentrations. The temperature and ET flux start to increase and RH decreases after
466 sunrise. Radon concentrations were still relatively high compared to the middle of day, suggesting
467 that strong convective mixing had not started and that a shallow mixing volume was still in place.
468 ET was injecting water vapour into a relatively shallow mixing volume at this time, which caused
469 a corresponding increase in near-surface humidity. Similarly for ^2H and $\delta^{18}\text{O}$, the observed spike
470 immediately after sunrise was likely caused by ET fluxes with an enriched heavy isotope
471 composition, possibly from re-evaporation of dewfall. During the dry period, when more dewfall
472 occurred, the ^2H and $\delta^{18}\text{O}$ increased more steeply. Assuming the post sunrise inversion layer was
473 of similar depth to the wet period, which radon concentrations indicate was the case, the steeper
474 increase in the stable isotopes can be explained by a greater ET flux from the larger amount of
475 surface condensation into a lower background water vapour volume. As indicated by the rapidly
476 decreasing Radon concentrations, vigorous vertical mixing entraining air from the residual layer
477 of the previous day begins to dilute the early morning fluxes around mid-morning, causing the



478 ^2H , $\delta^{18}\text{O}$ and the mixing ratio to first stabilise and then decrease. ET fluxes rapidly increased
479 during this period, but were not large enough to offset the dilution from the dry air being mixed
480 down from above, or was the iso-forcing of surface fluxes strong enough to stop the depletion of
481 surface humidity isotopes.

482 Similarly to the mixing ratio, ^2H and ^{18}O , the d_v increased after sunrise, but the greatest rate of
483 increase was observed slightly later when strong vertical mixing commenced, as shown by the
484 close agreement with the rapid Radon decrease. The iso-forcing for the D-excess was negative,
485 evidence that d_v increased due to encroachment mixing as the new mixed layer grew in depth and
486 not ET fluxes. The dry period showed a greater increase in d_v during the morning transition, which
487 was probably caused by a shift in the d_v of the background water vapour and greater differences
488 between the d_v of the residual and nocturnal layer moisture.

489 In the afternoon, d_v decreased back to values similar to those observed prior to sunrise. At this
490 time, ET fluxes decreased with solar insolation and consequently convective mixing decayed.
491 Radon shows the effect that a reduction in vertical mixing has on the concentration of tracers
492 emitted from the surface. So while ET decreased, small fluxes were still observed well after 18:00
493 when large changes in ^{18}O and d_v were observed. Hence, as the I_{ET} was positive and negative for
494 ^{18}O and D-excess, respectively, small ET fluxes into a poorly mixed surface layer may have led
495 to these observed changes.

496 During the night, dew fall led to a ^2H and ^{18}O decrease as heavy isotopes were removed in
497 condensing water vapour, which was especially the case during the dry period when greater surface
498 cooling and dew fall was observed. As expected, dew fall did not have a large impact upon the d_v ,



499 as dew formation is an equilibrium process. The composites of both the dry and wet period
500 nocturnal measurements do not show any clear nocturnal trends, remaining stable between the end
501 of the evening transition and the start of the morning transition. However, inspection of individual
502 nights in Figure 8, shows considerable variability, and a regression of the nocturnal d_v
503 measurements with radon concentrations produces a significant negative relationship ($p < 0.001$, R^2
504 = 0.31). High radon concentrations are associated with the most stable conditions, which would
505 enhance the effect of surface exchange in the early evening. Low radon concentrations on the other
506 hand, would be associated with periods of atmospheric turbulence in which high d_v air from above
507 the nocturnal inversion is mixed down towards the surface. This process would therefore tend to
508 move d_v back towards the higher values observed during the day.

509

510 **4 Discussion**

511 As has been previously observed (Steen-Larsen et al. 2013; Welp et al. 2012) and predicted by
512 isotopic models (Gat, 1996), our water vapour $\delta^2\text{H}$ and $\delta^{18}\text{O}$ observations showed different
513 temporal trends than d_v , indicating they are controlled by different atmospheric and hydrological
514 processes. Results presented here illustrate that the diurnal cycle was the dominant mode of
515 variability for d_v , consistent with previous studies for a range of ecosystems (Simonin et al. 2014;
516 Welp et al. 2012; Zhao et al. 2014). However, supported by our chamber based measurements of
517 d_{ET} , interpretation of the data showed that D-excess variability was controlled by local
518 meteorological conditions and surface exchange at night, and larger scale processes in the middle
519 of the day.



520 **4.1 Daytime d_v variability and remote hydrological processes**

521 As shown in Table 1, the results illustrate a strong correlation between daytime RH and d_v . The
522 slope of the relationship was $-0.52\%.\%^{-1}$, which is within the range of slopes determined from
523 coastal locations or shipborne platforms for the Mediterranean or different ocean basins (between
524 -0.43 and $-0.53\%.\%^{-1}$) (Kurita, 2011; Pfahl and Wernli, 2008; Steen-Larsen et al. 2014, 2015;
525 Uemura et al. 2008). These studies indicate a robust relationship between d_v and RH for
526 evaporation from large water bodies. This strong robust relationship appears not to be restricted to
527 coastal locations or measurements over the ocean surface, as Aemisegger *et al.* (2014) used a
528 trajectory model to investigate continental moisture recycling in Europe, and found a similar
529 relationship between d_v and RH of remote moisture sources during the cold season ($-0.57\%.\%^{-1}$).
530 However, their warm season observations produced a relationship with a lower slope ($-0.19\%.\%^{-1}$).
531 They proposed that because moisture recycling is weakest during winter, d_v retained the
532 signature of the evaporation conditions at the oceanic moisture source. However, moisture
533 recycling increased in the summer and attenuated the relationship. The agreement between their
534 winter measurements and our data indicates that the daytime d_v is at least partly determined by the
535 RH of the oceanic moisture source, and d_v may preserve the signature of evaporation from the
536 ocean surface.

537 Aemisegger *et al.* (2014) proposed that the relationship between the RH and the d_v could be used
538 to determine the D-excess of the moisture source. They argued that when air was saturated with
539 water vapour (when RH is 100%), the d_v is equal to the moisture source, as only equilibrium
540 processes take place and no change in the D-excess between liquid and vapour occurs. By
541 extrapolating the regression between d_v and RH to 100% RH, the D-excess of the moisture source



542 can be estimated, and gives a value of -8‰. This is approximately the same as the D-excess
543 determined for ocean water off the east coast of Australia by Xu *et al.* (2012) using a global ocean
544 model. In contrast to recent literature (Simonin et al. 2014; Welp et al. 2012; Zhao et al. 2014),
545 this suggests that although the common diurnal cycle was observed, daytime observations are
546 potentially a tracer of oceanic evaporation conditions, in particular RH.

547 Whilst we have shown a relationship between the RH and d_v consistent with an oceanic vapour
548 source, longer data collections should be examined to evaluate the consistency of the relationship.
549 Datasets collected over longer time periods and over the land surface have not reproduced the
550 strong relationship shown here. For a study of six mid-latitude sites in China and the USA, Welp
551 *et al.* (2012) used much longer datasets and observed slopes between daytime d_v and RH ranging
552 from -0.01 and -0.36‰ %⁻¹. Following on from the warm season interpretation of Aemisegger *et*
553 *al.* (2014), the lower slopes indicate the importance of moisture recycling from the land surface.
554 Whilst local geographical features such as lakes or seas could alter this relationship, the longer
555 length of their datasets include a greater range of atmospheric conditions and soil moisture states
556 that modify the observed relationship. Here we present data from comparatively quiescent
557 meteorological conditions, where the dominant moisture source is the ocean surrounding the
558 Australian continent. Wetter periods may see an increase in the local and remote moisture
559 recycling that could influence the relationship between local RH and d_v (Aemisegger et al. 2014).
560 However, during dry soil moisture states and for locations such as semi-arid Australia with a
561 relatively close ocean basin, the d_v may indeed be a tracer of oceanic evaporative environments.

562 The discussion of daytime d_v variability above has not considered the isotopic signature of the
563 entrainment flux from above the convective boundary layer introducing an additional moisture



564 source from the free troposphere. It would be expected to have a depleted isotopic composition
565 compared to the surface and explains the continuing decrease in ^2H and ^{18}O values after midday
566 when convective mixing extends to the top of the capping inversion. The drying trend observed
567 during the day indicates that the entrainment flux was larger than the surface ET fluxes, which has
568 been previously shown using large-eddy simulations (Huang et al. 2011) and observations (Davis
569 et al. 1997). Likewise, the greater entrainment flux would be expected to cause the surface vapour
570 to become depleted in heavy isotopes (Lee et al. 2012).

571 The drying and depleting trend of surface vapour illustrates that entrainment fluxes may influence
572 moisture right down to the surface, but whether this moisture flux impacts on d_v is unclear. In
573 contrast to ^2H and $\delta^{18}\text{O}$ observations, d_v showed no clear trend throughout the middle of the day,
574 but remained high. The expected sign of the iso-forcing of moisture entrained from the free
575 troposphere on d_v is uncertain, as very few measurements exist in the free troposphere. Of the
576 measurements that have been collected, Galewsky and Samuels-Crow (2014) collected *in-situ*
577 observations of water vapour on the Chilean Altiplano and showed periods of extremely high d_v
578 for descending air and lower d_v for ascending air from the ABL. He and Smith (1999) also showed
579 high d_v in the free-troposphere in North America collected from an aircraft platform. These papers
580 suggest moisture entrained from the free troposphere could cause an increase in d_v , but
581 discrimination between the background moisture and free tropospheric moisture is not possible
582 here. Nevertheless, d_v values stabilised during the day and did not show a clear trend until vertical
583 mixing began decaying later in the afternoon, so d_v values of moisture entrained from the free
584 troposphere are not likely to have had a large impact on surface d_v as the two sources of moisture
585 may have had a similar d_v .



586 4.2 Controls of d_{ET}

587 From the chamber measurements of d_{ET} , we have shown that the ET fluxes do not contribute to
588 the high values observed during the day. The ET fluxes actually imposed a negative iso-forcing on
589 d_v , which contrasts with interpretations made in previous studies investigating d_v variability on
590 diurnal time scales (Simonin et al. 2014; Zhao et al. 2014). However, it is expected that the sign
591 and magnitude of the D-excess iso-forcing would vary both spatially and temporally.

592 The measurements presented were collected for a dry soil moisture state with low ET fluxes. The
593 d_{ET} and the associated iso-forcing would be expected to vary with soil moisture state. After a rain
594 event, the soil dries through evaporation of moisture with a D-excess greater than the soil moisture
595 pool, causing the soil moisture D-excess to decrease following a pseudo-Rayleigh process (Barnes
596 and Allison, 1988). The D-excess of moisture at the evaporation front modelled from the bare soil
597 chambers was extremely low ($-130 \pm 22\%$). Assuming the most recent precipitation had a similar
598 D-excess to the intercept of the local MWL (16.2%), the soil moisture at the evaporation front
599 must have undergone significant evaporative drying. Therefore, immediately after a rainfall event,
600 d_{ET} would be much higher and likely to impose a positive forcing on d_v . As the soil dries, there is
601 likely a tipping point when the ET fluxes switch from positive to negative D-excess iso-forcing.
602 This will have implications for studies attempting to use d_v as a tracer of continental moisture
603 recycling, as the large spatial variability of rainfall and the associated soil moisture state would
604 therefore lead to large spatial and temporal variability for d_{ET} . Although, the strongest moisture
605 recycling is expected for the wettest soils, when d_{ET} is higher and ET fluxes are largest, variability
606 in d_{ET} may still be important.



607 Relative magnitudes of evaporation and transpiration fluxes are important for d_{ET} , as the two
608 processes draw on moisture from different depths within the soil column and are fundamentally
609 different, so fluxes are likely to have different D-excess values. Deeper in the soil column,
610 evaporation has a smaller impact on the soil moisture, so transpiration would be expected to have
611 a higher D-excess than soil evaporation. In agreement with this idea and consistent with the
612 importance of identifying the isotopic composition of the evaporation front outlined by Dubbert et
613 al. (2013), measurements of the D-excess of soil moisture in the 0-5 cm soil layer were much
614 higher than the modelled D-excess for the evaporating front. Similarly we observed slightly higher
615 d_{ET} from vegetated chambers, even when the vegetation was of poor health. It would therefore be
616 expected that for periods of improved vegetation health, increased transpiration fluxes would lead
617 to a higher d_{ET} . The land surface and cover type would therefore constitute an important variable
618 influencing the D-excess of moisture recycling. Further studies investigating how ET partitioning
619 and drying of the soil moisture reservoir following irrigation or precipitation events would lead to
620 a better understanding of how moisture recycling influences the ambient d_v on continental and
621 local scales.

622 **5 Conclusions**

623 To determine how local ET fluxes modified water vapour stable isotopes and in particular the D-
624 excess, in-situ observations were collected in a semi-arid region of south-eastern Australia. The
625 observed diurnal cycle for the D-excess of water vapour, with high values during the day and low
626 values at night, reflected findings from previous analyses. With the addition of chamber based
627 measurements of isotopic compositions in evaporative fluxes, it was shown that the local ET fluxes
628 exhibited a negative forcing on the ambient water vapour D-excess that could not explain the high



629 daytime values. A strong negative relationship was observed between the locally measured relative
630 humidity and vapour D-excess during the daytime, consistent with relationships observed for
631 oceanic moisture sources. In the evening and night time however, ET fluxes into the shallow
632 nocturnal inversion layer were responsible for lowering the D-excess of water vapour near the
633 surface. In addition, a negative correlation between D-excess and radon concentrations at night
634 indicated that transient nocturnal mixing events shifted the D-excess back towards the higher
635 values observed during the day, with the most stable (least turbulent) nights producing the lowest
636 D-excess values. In the morning, encroachment and entrainment of high D-excess air from above
637 caused D-excess of the surface vapour to increase back to the synoptic values.

638 Overall, it was found that the magnitude of the D-excess diurnal cycle was controlled
639 predominantly by interplay between synoptic forcing and local ABL processes and was modified
640 further by nocturnal surface exchange processes and turbulent mixing. The low D-excess of the
641 ET fluxes determined from flux chambers in this study illustrated that the impact of large scale
642 moisture recycling may be both spatially and temporally variable, depending on the soil moisture
643 state. This has implications for studies using D-excess to investigate moisture recycling.

644 **6 Acknowledgements**

645 Stephen Parkes was supported by the Atmospheric Mixing and Pollution Transport (AMPT)
646 project at the Australian Nuclear Science and Technology Organization (ANSTO). The Baldry
647 Hydrological Observatory field campaign was supported by Australian Research Council
648 Discovery grants DP0987478 and DP120104718. Matthew McCabe and Stephen Parkes
649 acknowledge the support of the King Abdullah University of Science and Technology. We thank
650 Peter Graham, Cecilia Azcurra, Dr. Jin Wang and Yingzhe Cai for their assistance during the



651 campaign. We also appreciate the support of Diana and Jason Tremain for access to the Baldry
652 Hydrological Observatory and surrounding farmland, Chris Dimovski for performing plant and
653 soil water extractions, and Barbara Neklapilova analysis of plant and soil water samples.

654 For access to the data used in this paper contact Dr Stephen Parkes by email
655 (stephen.parkes@kaust.edu.sa).

656

657

658 **7 References**

659 Aemisegger, F., Pfahl, S., Sodemann, H., Lehner, I., Seneviratne, S. I. and Wernli, H.: Deuterium
660 excess as a proxy for continental moisture recycling and plant transpiration, *Atmos. Chem. Phys.*,
661 14(8), 4029–4054, doi:10.5194/acp-14-4029-2014, 2014.

662 Barnes, C. J. and Allison, G. B.: Tracing of water movement in the unsaturated zone using stable
663 isotopes of hydrogen and oxygen, *J. Hydrol.*, 100(1-3), 143–176, doi:10.1016/0022-
664 1694(88)90184-9, 1988.

665 Bastrikov, V., Steen-Larsen, H. C., Masson-Delmotte, V., Griбанov, K., Cattani, O., Jouzel, J. and
666 Zakharov, V.: Continuous measurements of atmospheric water vapour isotopes in western Siberia
667 (Kourovka), *Atmos. Meas. Tech.*, 7(6), 1763–1776, doi:10.5194/amt-7-1763-2014, 2014.

668 Berkelhammer, M., Hu, J., Bailey, A., Noone, D. C., Still, C. J., Barnard, H., Gochis, D., Hsiao,
669 G. S., Rahn, T. and Turnipseed, A.: The nocturnal water cycle in an open-canopy forest, *J.*
670 *Geophys. Res. Atmos.*, 118(17), 10,210–225,242, doi:10.1002/jgrd.50701, 2013.

671 Cernusak, L. A., Pate, J. S. and Farquhar, G. D.: Diurnal variation in the stable isotope composition
672 of water and dry matter in fruiting *Lupinus angustifolius* under field conditions, *Plant. Cell*
673 *Environ.*, 25(7), 893–907, doi:10.1046/j.1365-3040.2002.00875.x, 2002.

674 Chambers, S. D., Wang, F., Williams, A. G., Xiaodong, D., Zhang, H., Lonati, G., Crawford, J.,
675 Griffiths, A. D., Ianniello, A. and Allegrini, I.: Quantifying the influences of atmospheric stability
676 on air pollution in Lanzhou, China, using a radon-based stability monitor, *Atmos. Environ.*, 107,
677 233–243, doi:10.1016/j.atmosenv.2015.02.016, 2015a.

678 Chambers, S. D., Williams, A. G., Crawford, J. and Griffiths, A. D.: On the use of radon for
679 quantifying the effects of atmospheric stability on urban emissions, *Atmos. Chem. Phys. Discuss.*,
680 14(18), 25411–25452, doi:10.5194/acpd-14-25411-2014, 2014.



- 681 Chambers, S. D., Williams, A. G., Crawford, J. and Griffiths, A. D.: On the use of radon for
682 quantifying the effects of atmospheric stability on urban emissions, *Atmos. Chem. Phys.*, 15(3),
683 1175–1190, doi:10.5194/acp-15-1175-2015, 2015b.
- 684 Craig, H.: Isotopic Variations in Meteoric Waters, *Sci.*, 133 (3465), 1702–1703,
685 doi:10.1126/science.133.3465.1702, 1961.
- 686 Craig, H. and Gordon, L. I.: Deuterium and oxygen-18 variations in the ocean and marine
687 atmosphere, in *Stable isotopes in oceanographic studies and paleotemperatures*, Proceedings,
688 Spoleto, Italy., edited by E. Tongiogi, pp. 9–130, Pisa, Italy., 1965.
- 689 Crawford, J., Hughes, C. E. and Parkes, S. D.: Is the isotopic composition of event based
690 precipitation driven by moisture source or synoptic scale weather in the Sydney Basin, Australia?,
691 *J. Hydrol.*, 507(0), 213–226, doi:http://dx.doi.org/10.1016/j.jhydrol.2013.10.031, 2013.
- 692 Dansgaard, W.: Stable isotopes in precipitation, *Tellus*, 16(4), 436–468, doi:10.1111/j.2153-
693 3490.1964.tb00181.x, 1964.
- 694 Davis, K. J., Lenschow, D. H., Oncley, S. P., Kiemle, C., Ehret, G., Giez, A. and Mann, J.: Role
695 of entrainment in surface-atmosphere interactions over the boreal forest, *J. Geophys. Res. Atmos.*,
696 102(D24), 29219–29230, doi:10.1029/97JD02236, 1997.
- 697 Decker, M., Pitman, A. and Evans, J.: Diagnosing the seasonal land–atmosphere correspondence
698 over northern Australia: dependence on soil moisture state and correspondence strength definition,
699 *Hydrol. Earth Syst. Sci.*, 19(8), 3433–3447, doi:10.5194/hess-19-3433-2015, 2015.
- 700 Dubbert, M., Cuntz, M., Piayda, A., Maguás, C. and Werner, C.: Partitioning evapotranspiration
701 – Testing the Craig and Gordon model with field measurements of oxygen isotope ratios of
702 evaporative fluxes, *J. Hydrol.*, 496, 142–153, doi:10.1016/j.jhydrol.2013.05.033, 2013.
- 703 Evans, J. P. and McCabe, M. F.: Regional climate simulation over Australia’s Murray-Darling
704 basin: A multitemporal assessment, *J. Geophys. Res. Atmos.*, 115(D14), n/a–n/a,
705 doi:10.1029/2010JD013816, 2010.
- 706 Finnigan, J. J., Clement, R., Malhi, Y., Leuning, R. and Cleugh, H. A.: A Re-Evaluation of Long-
707 Term Flux Measurement Techniques Part I: Averaging and Coordinate Rotation, *Boundary-Layer*
708 *Meteorol.*, 107(1), 1–48, doi:10.1023/A:1021554900225, 2003.
- 709 Froehlich, K., Kralik, M., Papesch, W., Rank, D., Scheifinger, H. and Stichler, W.: Deuterium
710 excess in precipitation of Alpine regions – moisture recycling, *Isotopes Environ. Health Stud.*,
711 44(1), 61–70, doi:10.1080/10256010801887208, 2008.
- 712 Galewsky, J. and Samuels-Crow, K.: Summertime Moisture Transport to the Southern South
713 American Altiplano: Constraints from In Situ Measurements of Water Vapor Isotopic
714 Composition, *J. Clim.*, 28(7), 2635–2649, doi:10.1175/JCLI-D-14-00511.1, 2014.
- 715 Gat, J. R.: OXYGEN AND HYDROGEN ISOTOPES IN THE HYDROLOGIC CYCLE, *Annu.*
716 *Rev. Earth Planet. Sci.*, 24(1), 225–262, doi:10.1146/annurev.earth.24.1.225, 1996.
- 717 Gat, J. R., Bowser, C. J. and Kendall, C.: The contribution of evaporation from the Great Lakes to
718 the continental atmosphere: estimate based on stable isotope data, *Geophys. Res. Lett.*, 21(7), 557–
719 560, doi:10.1029/94GL00069, 1994.
- 720 Gat, J. R., Klein, B., Kushnir, Y., Roether, W., Wernli, H., Yam, R. and Shemesh, A.: Isotope



- 721 composition of air moisture over the Mediterranean Sea: an index of the air–sea interaction pattern,
722 *Tellus B*, 55(5), 953–965, doi:10.1034/j.1600-0889.2003.00081.x, 2003.
- 723 Griffiths, A. D., Parkes, S. D., Chambers, S. D., McCabe, M. F. and Williams, A. G.: Improved
724 mixing height monitoring through a combination of lidar and radon measurements, *Atmos. Meas.*
725 *Tech.*, 6(2), 207–218, doi:10.5194/amt-6-207-2013, 2013.
- 726 Harding, K. J. and Snyder, P. K.: Modeling the Atmospheric Response to Irrigation in the Great
727 Plains. Part I: General Impacts on Precipitation and the Energy Budget, *J. Hydrometeorol.*, 13(6),
728 1667–1686, doi:10.1175/JHM-D-11-098.1, 2012.
- 729 He, H. and Smith, R. B.: Stable isotope composition of water vapor in the atmospheric boundary
730 layer above the forests of New England, *J. Geophys. Res. Atmos.*, 104(D9), 11657–11673,
731 doi:10.1029/1999JD900080, 1999.
- 732 Horita, J. and Wesolowski, D. J.: Liquid-vapor fractionation of oxygen and hydrogen isotopes of
733 water from the freezing to the critical temperature, *Geochim. Cosmochim. Acta*, 58(16), 3425–
734 3437, doi:http://dx.doi.org/10.1016/0016-7037(94)90096-5, 1994.
- 735 Huang, J., Lee, X. and Patton, E. G.: Entrainment and budgets of heat, water vapor, and carbon
736 dioxide in a convective boundary layer driven by time-varying forcing, *J. Geophys. Res. Atmos.*,
737 116(D6), D06308, doi:10.1029/2010JD014938, 2011.
- 738 Huang, L. and Wen, X.: Temporal variations of atmospheric water vapor δD and $\delta^{18}O$ above an
739 arid artificial oasis cropland in the Heihe River Basin, *J. Geophys. Res. Atmos.*, 119(19),
740 2014JD021891, doi:10.1002/2014JD021891, 2014.
- 741 Hughes, C. E. and Crawford, J.: Spatial and temporal variation in precipitation isotopes in the
742 Sydney Basin, Australia, *J. Hydrol.*, 489(0), 42–55,
743 doi:http://dx.doi.org/10.1016/j.jhydrol.2013.02.036, 2013.
- 744 Keeling, C. D.: The concentration and isotopic abundances of atmospheric carbon dioxide in rural
745 areas, *Geochim. Cosmochim. Acta*, 13(4), 322–334, doi:10.1016/0016-7037(58)90033-4, 1958.
- 746 Kurita, N.: Origin of Arctic water vapor during the ice-growth season, *Geophys. Res. Lett.*, 38(2),
747 n/a–n/a, doi:10.1029/2010GL046064, 2011.
- 748 Lai, C.-T. and Ehleringer, J.: Deuterium excess reveals diurnal sources of water vapor in forest air,
749 *Oecologia*, 165(1), 213–223, doi:10.1007/s00442-010-1721-2, 2011.
- 750 Lee, X., Griffis, T. J., Baker, J. M., Billmark, K. A., Kim, K. and Welp, L. R.: Canopy-scale kinetic
751 fractionation of atmospheric carbon dioxide and water vapor isotopes, *Global Biogeochem.*
752 *Cycles*, 23(1), GB1002, doi:10.1029/2008GB003331, 2009.
- 753 Lee, X., Huang, J. and Patton, E.: A Large-Eddy Simulation Study of Water Vapour and Carbon
754 Dioxide Isotopes in the Atmospheric Boundary Layer, *Boundary-Layer Meteorol.*, 145(1), 229–
755 248, doi:10.1007/s10546-011-9631-3, 2012.
- 756 Leuning, R.: The correct form of the Webb, Pearman and Leuning equation for eddy fluxes of
757 trace gases in steady and non-steady state, horizontally homogeneous flows, *Boundary-Layer*
758 *Meteorol.*, 123(2), 263–267, doi:10.1007/s10546-006-9138-5, 2007.
- 759 Lin, J. C., Gerbig, C., Wofsy, S. C., Andrews, A. E., Daube, B. C., Davis, K. J. and Grainger, C.
760 A.: A near-field tool for simulating the upstream influence of atmospheric observations: The



- 761 Stochastic Time-Inverted Lagrangian Transport (STILT) model, *J. Geophys. Res.*, 108, 4493,
762 doi:10.1029/2002JD003161, 2003.
- 763 Masson-Delmotte, V., Jouzel, J., Landais, A., Stievenard, M., Johnsen, S. J., White, J. W. C.,
764 Werner, M., Sveinbjörnsdóttir, A. and Fuhrer, K.: GRIP Deuterium Excess Reveals Rapid and
765 Orbital-Scale Changes in Greenland Moisture Origin, *Sci.*, 309 (5731), 118–121,
766 doi:10.1126/science.1108575, 2005.
- 767 Mathieu, R. and Bariac, T.: A numerical model for the simulation of stable isotope profiles in
768 drying soils, *J. Geophys. Res. Atmos.*, 101(D7), 12685–12696, doi:10.1029/96JD00223, 1996.
- 769 McCabe, M. F., Ershadi, A., Jimenez, C., Miralles, D. G., Michel, D. and Wood, E. F.: The
770 GEWEX LandFlux project: evaluation of model evaporation using tower-based and globally
771 gridded forcing data, *Geosci. Model Dev.*, 9(1), 283–305, doi:10.5194/gmd-9-283-2016, 2016.
- 772 Merlivat, L.: Molecular diffusivities of H₂¹⁶O, HD¹⁶O, and H₂¹⁸O
773 in gases, *J. Chem. Phys.*, 69(6), 2864, doi:10.1063/1.436884, 1978.
- 774 Merlivat, L. and Jouzel, J.: Global climatic interpretation of the deuterium-oxygen 18 relationship
775 for precipitation, *J. Geophys. Res. Ocean.*, 84(C8), 5029–5033, doi:10.1029/JC084iC08p05029,
776 1979.
- 777 Noone, D., Risi, C., Bailey, A., Berkelhammer, M., Brown, D. P., Buening, N., Gregory, S.,
778 Nusbaumer, J., Schneider, D., Sykes, J., Vanderwende, B., Wong, J., Meillier, Y. and Wolfe, D.:
779 Determining water sources in the boundary layer from tall tower profiles of water vapor and
780 surface water isotope ratios after a snowstorm in Colorado, *Atmos. Chem. Phys.*, 13(3), 1607–
781 1623, doi:10.5194/acp-13-1607-2013, 2013.
- 782 Pfahl, S. and Wernli, H.: Air parcel trajectory analysis of stable isotopes in water vapor in the
783 eastern Mediterranean, *J. Geophys. Res. Atmos.*, 113(D20), n/a–n/a, doi:10.1029/2008JD009839,
784 2008.
- 785 Pfahl, S. and Wernli, H.: Lagrangian simulations of stable isotopes in water vapor: An evaluation
786 of nonequilibrium fractionation in the Craig-Gordon model, *J. Geophys. Res. Atmos.*, 114(D20),
787 D20108, doi:10.1029/2009JD012054, 2009.
- 788 Risi, C., Noone, D., Frankenberg, C. and Worden, J.: Role of continental recycling in intraseasonal
789 variations of continental moisture as deduced from model simulations and water vapor isotopic
790 measurements, *Water Resour. Res.*, 49(7), 4136–4156, doi:10.1002/wrcr.20312, 2013.
- 791 Simonin, K. A., Link, P., Rempe, D., Miller, S., Oshun, J., Bode, C., Dietrich, W. E., Fung, I. and
792 Dawson, T. E.: Vegetation induced changes in the stable isotope composition of near surface
793 humidity, *Ecohydrology*, 7(3), 936–949, doi:10.1002/eco.1420, 2014.
- 794 Steen-Larsen, H. C., Johnsen, S. J., Masson-Delmotte, V., Stenni, B., Risi, C., Sodemann, H.,
795 Balslev-Clausen, D., Blunier, T., Dahl-Jensen, D., Ellehøj, M. D., Falourd, S., Grindsted, A.,
796 Gkinis, V., Jouzel, J., Popp, T., Sheldon, S., Simonsen, S. B., Sjolte, J., Steffensen, J. P., Sperlich,
797 P., Sveinbjörnsdóttir, A. E., Vinther, B. M. and White, J. W. C.: Continuous monitoring of summer
798 surface water vapor isotopic composition above the Greenland Ice Sheet, *Atmos. Chem. Phys.*,
799 13(9), 4815–4828, doi:10.5194/acp-13-4815-2013, 2013.
- 800 Steen-Larsen, H. C., Sveinbjörnsdóttir, A. E., Jonsson, T., Ritter, F., Bonne, J.-L., Masson-
801 Delmotte, V., Sodemann, H., Blunier, T., Dahl-Jensen, D. and Vinther, B. M.: Moisture sources

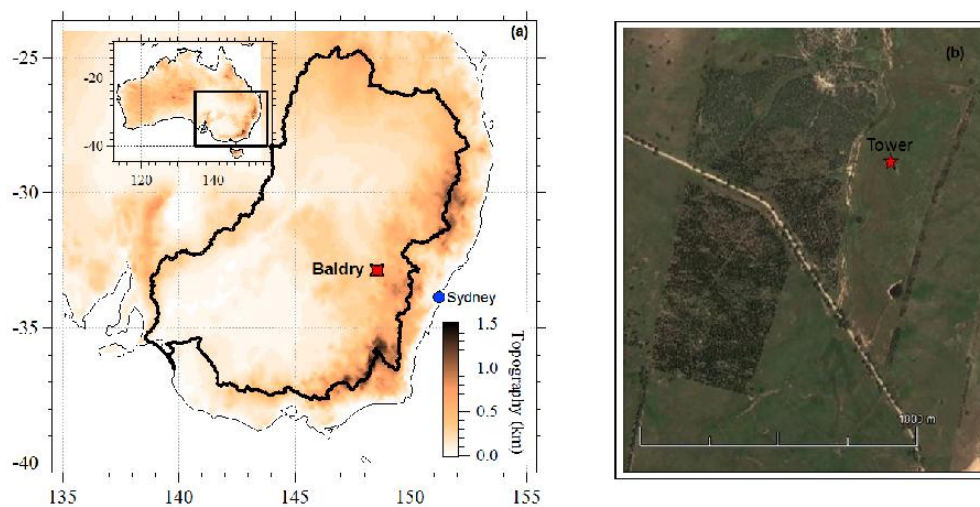


- 802 and synoptic to seasonal variability of North Atlantic water vapor isotopic composition, J.
803 Geophys. Res. Atmos., 120(12), 2015JD023234, doi:10.1002/2015JD023234, 2015.
- 804 Steen-Larsen, H. C., Sveinbjörnsdóttir, A. E., Peters, A. J., Masson-Delmotte, V., Guishard, M.
805 P., Hsiao, G., Jouzel, J., Noone, D., Warren, J. K. and White, J. W. C.: Climatic controls on water
806 vapor deuterium excess in the marine boundary layer of the North Atlantic based on 500 days of
807 in situ, continuous measurements, Atmos. Chem. Phys., 14(15), 7741–7756, doi:10.5194/acp-14-
808 7741-2014, 2014.
- 809 Tian, L., Masson-Delmotte, V., Stievenard, M., Yao, T. and Jouzel, J.: Tibetan Plateau summer
810 monsoon northward extent revealed by measurements of water stable isotopes, J. Geophys. Res.
811 Atmos., 106(D22), 28081–28088, doi:10.1029/2001JD900186, 2001.
- 812 Uemura, R., Matsui, Y., Yoshimura, K., Motoyama, H. and Yoshida, N.: Evidence of deuterium
813 excess in water vapor as an indicator of ocean surface conditions, J. Geophys. Res. Atmos.,
814 113(D19), n/a–n/a, doi:10.1029/2008JD010209, 2008.
- 815 Vallet-Coulomb, C., Gasse, F. and Sonzogni, C.: Seasonal evolution of the isotopic composition
816 of atmospheric water vapour above a tropical lake: Deuterium excess and implication for water
817 recycling, Geochim. Cosmochim. Acta, 72(19), 4661–4674, doi:10.1016/j.gca.2008.06.025, 2008.
- 818 Wang, L., Niu, S., Good, S. P., Soderberg, K., McCabe, M. F., Sherry, R. A., Luo, Y., Zhou, X.,
819 Xia, J. and Caylor, K. K.: The effect of warming on grassland evapotranspiration partitioning using
820 laser-based isotope monitoring techniques, Geochim. Cosmochim. Acta, 111, 28–38,
821 doi:10.1016/j.gca.2012.12.047, 2013.
- 822 Wei, J., Dirmeyer, P. A., Wisser, D., Bosilovich, M. G. and Mocko, D. M.: Where Does the
823 Irrigation Water Go? An Estimate of the Contribution of Irrigation to Precipitation Using MERRA,
824 J. Hydrometeorol., 14(1), 275–289, doi:10.1175/JHM-D-12-079.1, 2012.
- 825 Welp, L. R., Lee, X., Griffis, T. J., Wen, X.-F., Xiao, W., Li, S., Sun, X., Hu, Z., Val Martin, M.
826 and Huang, J.: A meta-analysis of water vapor deuterium-excess in the midlatitude atmospheric
827 surface layer, Global Biogeochem. Cycles, 26(3), GB3021, doi:10.1029/2011GB004246, 2012.
- 828 Williams, A. G., Chambers, S. and Griffiths, A.: Bulk Mixing and Decoupling of the Nocturnal
829 Stable Boundary Layer Characterized Using a Ubiquitous Natural Tracer, Boundary-Layer
830 Meteorol., 149(3), 381–402, doi:10.1007/s10546-013-9849-3, 2013.
- 831 Williams, A. G., Zahorowski, W., Chambers, S., Griffiths, A., Hacker, J. M., Element, A. and
832 Werczynski, S.: The Vertical Distribution of Radon in Clear and Cloudy Daytime Terrestrial
833 Boundary Layers, J. Atmos. Sci., 68(1), 155–174, doi:10.1175/2010JAS3576.1, 2010.
- 834 Xu, X., Werner, M., Butzin, M. and Lohmann, G.: Water isotope variations in the global ocean
835 model MPI-OM, Geosci. Model Dev., 5(3), 809–818, doi:10.5194/gmd-5-809-2012, 2012.
- 836 Zahorowski, W., Chambers, S. D. and Henderson-Sellers, A.: Ground based radon-222
837 observations and their application to atmospheric studies, J. Environ. Radioact., 76(1–2), 3–33,
838 doi:http://dx.doi.org/10.1016/j.jenvrad.2004.03.033, 2004.
- 839 Zhao, L., Wang, L., Liu, X., Xiao, H., Ruan, Y. and Zhou, M.: The patterns and implications of
840 diurnal variations in the d-excess of plant water, shallow soil water and air moisture, Hydrol. Earth
841 Syst. Sci., 18(10), 4129–4151, doi:10.5194/hess-18-4129-2014, 2014.



842

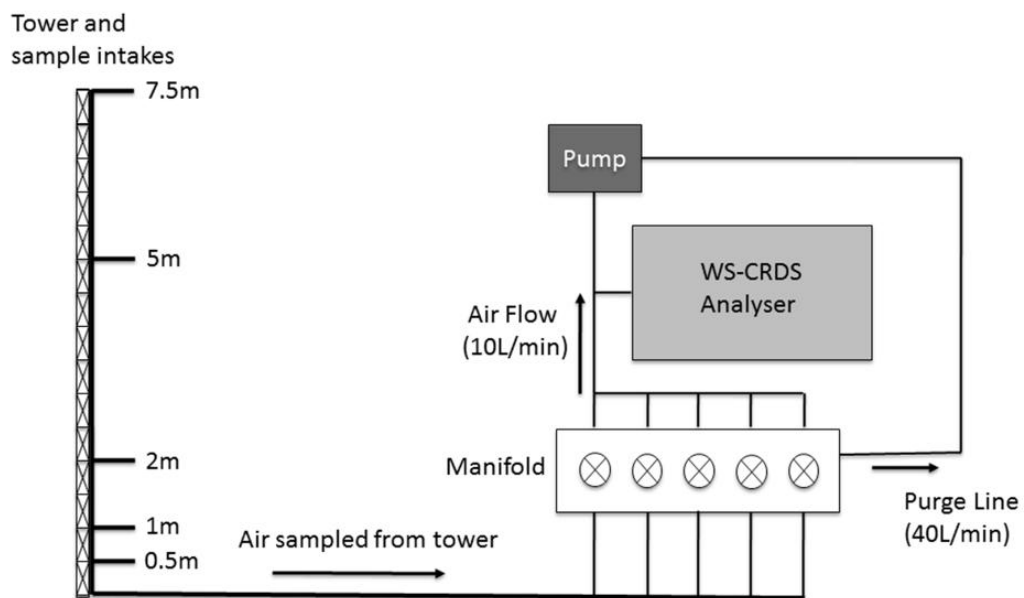
843 **8 Figure Captions**



844

Figure 1: a) Location of the Baldry Hydrological Observatory, the heavy black border shows the limits of the Murray-Darling Basin, (b) location of the field site used for the campaign.

845



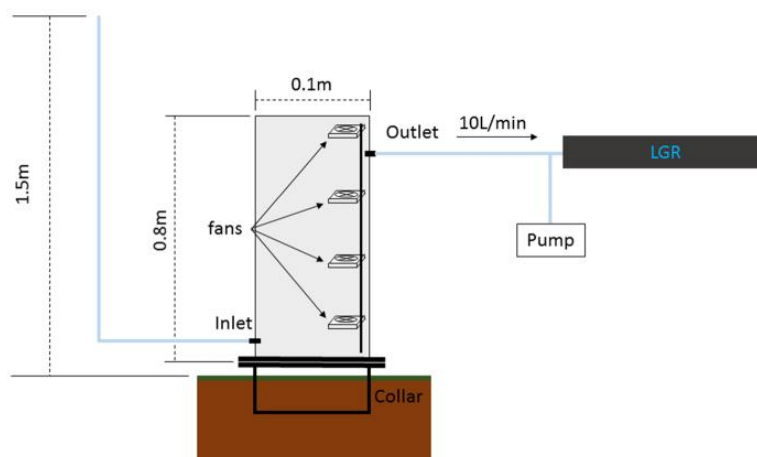
846

847 Figure 2: Sampling system for the automated in-situ sampling of water vapour isotopes from the

848 tower.

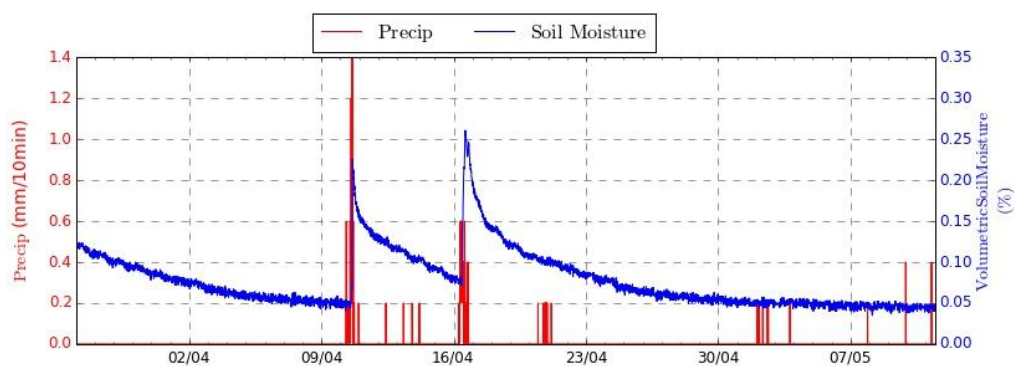


849



850

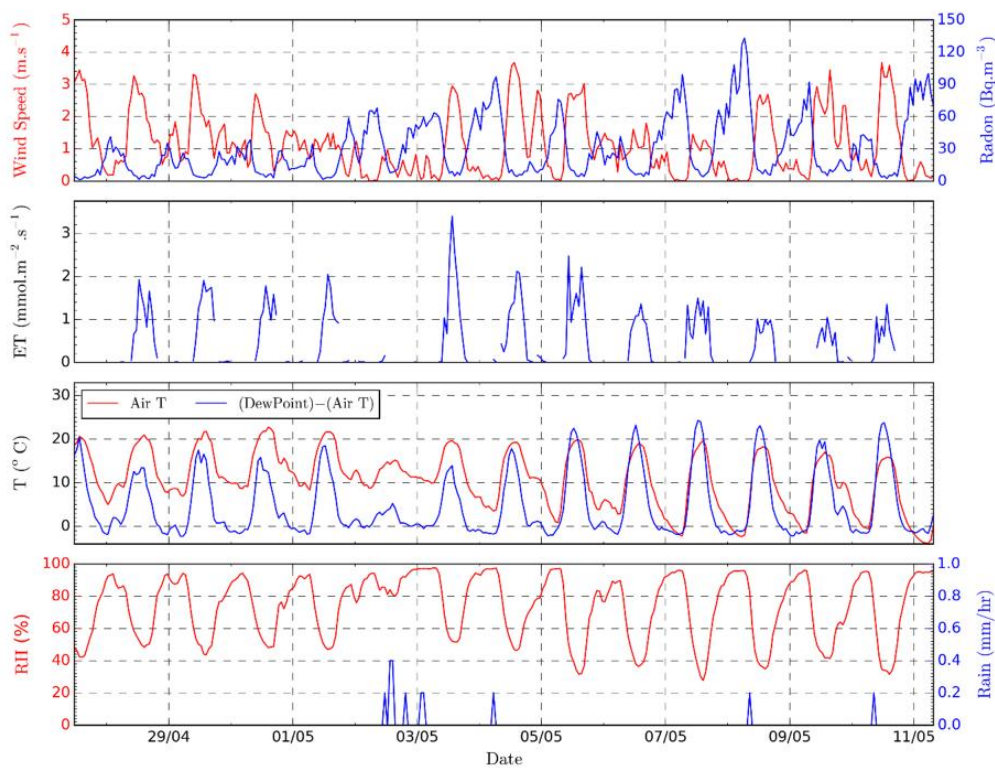
851 Figure 3: Chamber design used for determining isotopic compositions of ET fluxes.



852



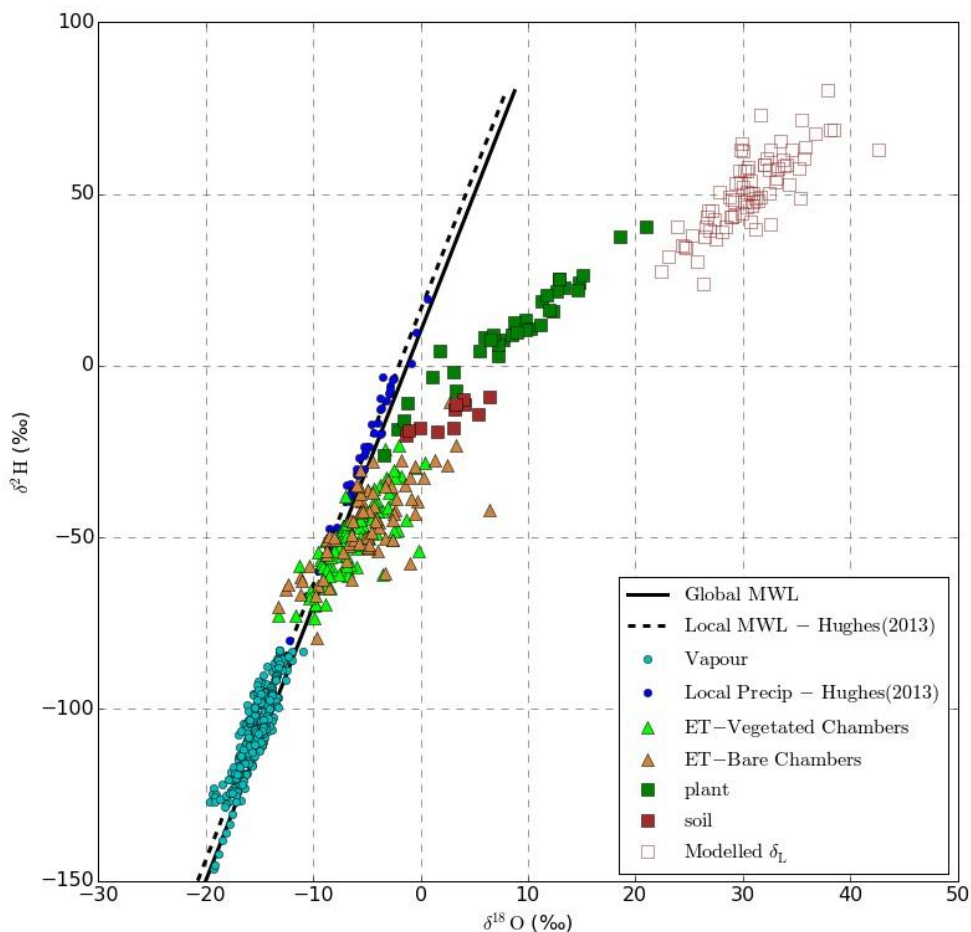
Figure 4: Precipitation and 0-10 cm soil moisture for the month leading up to and including the field campaign.



853

Figure 5: Meteorological and radon measurements collected throughout the field campaign. Meteorological measurements are block hourly averages calculated from 15-minute observations. Small rain events on the 4th, 8th and 10th May were most likely dew fall rather than precipitation.

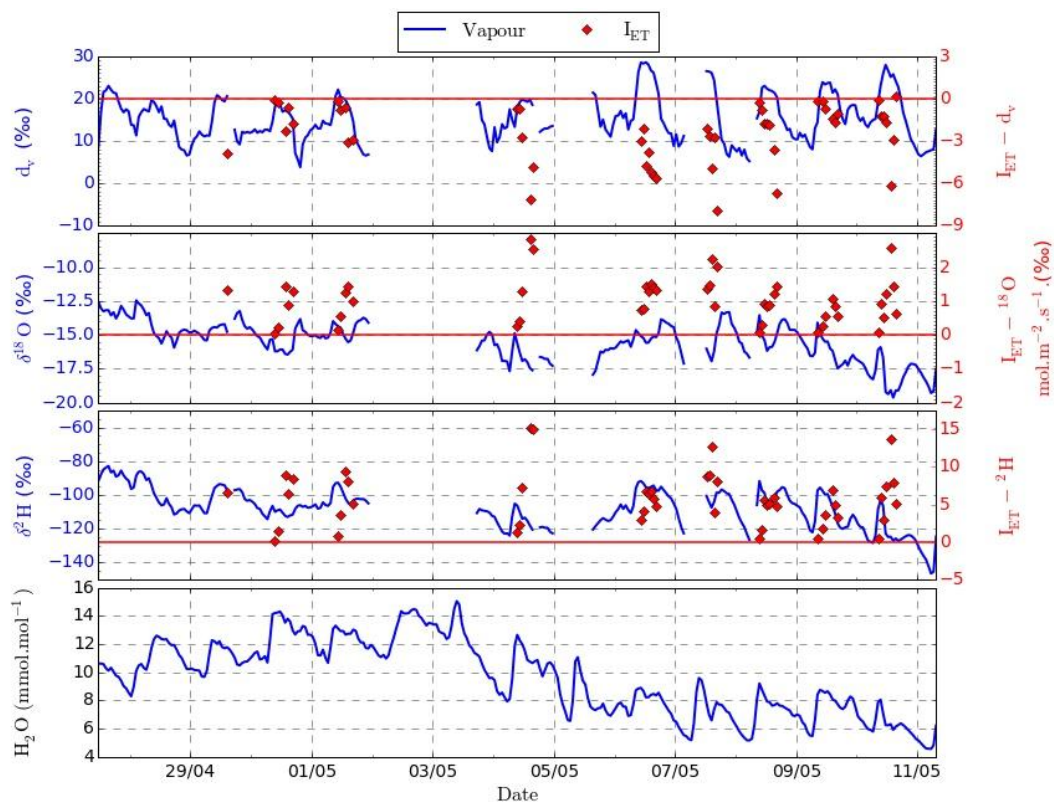
854



855

Figure 6: Relationship between ^2H and ^{18}O for observed and modelled water pools. Linear regressions are shown for local and global meteoric water lines (MWL). Data from Hughes and Crawford (2013) are for monthly cumulative rainfall samples between 2005 and 2008.

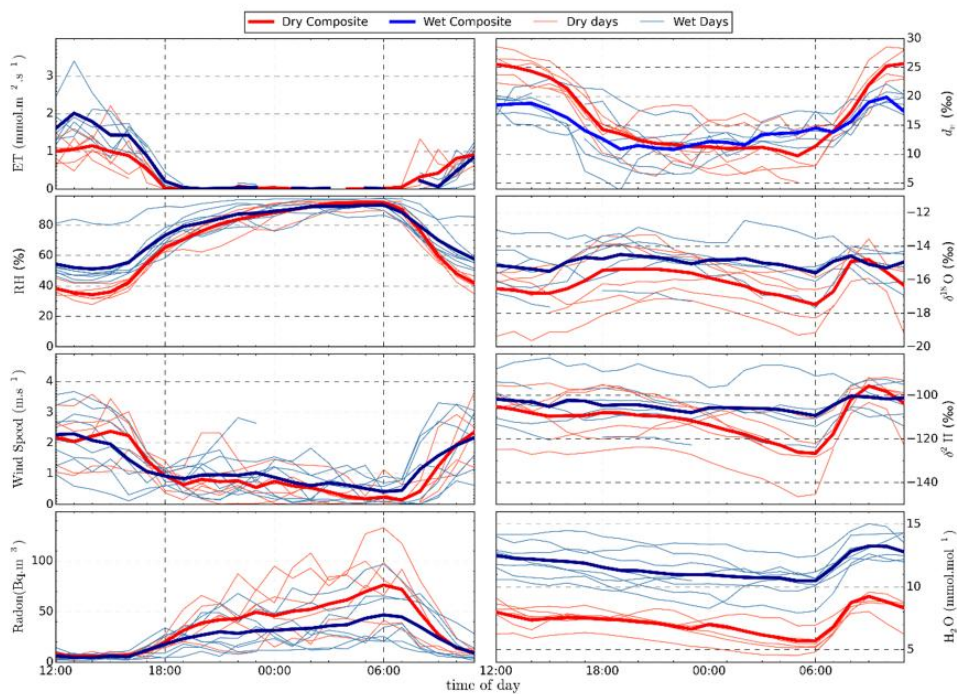
856



857

Figure 7: Time series of hourly water vapour mixing ratio, isotopic composition and ET iso-forcing (I_{ET}).

858



859

Figure 8: Data plotted by time of day and divided into dry and wet periods (see text in section 3.5). Diurnal composites are shown for dry (red) and wet (blue) periods.

860



861 Table 1: Correlation between meteorological variables and the isotopic composition of water
 862 vapour. Values outside the brackets are statistics for the hourly observations. Inside the brackets
 863 are correlation statistics for average values calculated between 11:00 and 15:00 LST, hence
 864 representing activity during a convective boundary layer. Significant correlations are shown in bold;
 865 $p < 0.001$ for hourly observations and $p < 0.05$ for the daytime averages (due to the smaller number
 866 of points).

		T	RH	ET	H₂O	I_{ET}^a
² H	Slope	0.83 (0.51)	-0.17 (0.23)	1.4 (6.1)	2.1 (0.85)	-1.1 (-3.0)
	Intercept	-120 (-140)	-95 (-110)	-110 (-110)	-110 (-130)	-99 (-83)
	R²	0.24 (0.13)	0.09 (0.02)	0.001 (0.04)	0.2 (0.04)	0.2 (0.45)
	p	<0.001 (0.3)	<0.001 (0.7)	0.32 (0.6)	<0.001 (0.5)	0.002 (0.05)
¹⁸ O	Slope	0.046 (0.44)	-0.01 (-0.01)	-0.37 (1.8)	0.27 (0.29)	-0.7 (-1.9)
	Intercept	-16 (-24)	-16 (-20)	-15 (-18)	-18 (-19)	-15 (-14)
	R²	0.04 (0.30)	0.004 (0.2)	0.02 (0.16)	0.2 (0.2)	0.14 (0.32)
	p	<0.001 (0.08)	0.26 (0.19)	0.05 (0.26)	<0.001 (0.15)	0.008 (0.11)
<i>d_v</i>	Slope	0.51 (-1.4)	-0.21 (-0.52)	0.01 (-0.16)	0.15 (-1.3)	-1.4 (-2.4)
	Intercept	-9.9 (48)	31 (44)	-15 (-18)	14 (35)	21 (20)
	R²	0.40 (0.48)	0.62 (0.74)	0.22 (0.30)	0.004 (0.71)	0.06 (0.08)
	p	<0.001 (0.02)	<0.001 (<0.01)	0.05 (0.01)	0.26 (<0.01)	0.01 (0.44)

867 ^aIso-forcing correlations were calculated for simultaneous vapour and chamber measurements.
 868 Hourly averaged values were used for both.

869