



1 **A post-wildfire response in cave dripwater chemistry**

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23 **Abstract**

24 Cave environments are sensitive because surface environmental changes can affect both the  
25 isotopic composition and solute concentration of infiltrating cave dripwaters. These changes  
26 subsequently affect the speleothem geochemical record. One such agent of change is wildfire,  
27 however its effect on karst processes remains poorly understood. Using dripwater data from a  
28 shallow cave, at a forested site in southwest Australia, we provide a unique analysis of the post-  
29 wildfire effects on dripwater  $\delta^{18}\text{O}$  and solute concentrations. We analyse how wildfires affect  
30 on local controls, i.e. vegetation cover, evapotranspiration and carbonate water-rock  
31 interactions, effects cave dripwater hydrology and geochemistry. We compare our post-  
32 wildfire data with modelled drip water  $\delta^{18}\text{O}$ , regional groundwater chemistry, and a second  
33 cave dripwater dataset, to determine the extent to which wildfire affects cave dripwater  
34 composition. We find in our forested, shallow cave site, by effecting surface evaporation and  
35 transpiration rates wildfire can have a multi-year impact on subsurface hydrology and  
36 dripwater chemistry. Here we open a new avenue for speleothem science in fire-prone regions,  
37 focusing on the geochemical records of speleothems as potential paleo-fire archives.

38



## 39 **1 Introduction**

40 Caves are observatories that contain invaluable geochemical archives of past-climates;  
41 preserved in the form of speleothems (stalagmites, stalactites and flowstones). The existing  
42 paradigm in speleothem science has largely focused on establishing paleoclimate proxies in  
43 stalagmites (McDermott et al., 2001; Treble et al., 2008; Woodhead et al., 2010). While these  
44 proxies have been useful for reconstructing paleoclimates, their interpretations may hold a  
45 predisposed bias towards using these proxies as indicators of paleo-climates only, instead of  
46 including local environmental factors.

47 One way of verifying these paleoclimate proxies has been to conduct dripwater monitoring  
48 studies. To date dripwater monitoring has predominantly focused on understanding the extent  
49 to which climatic controls affect dripwater  $\delta^{18}\text{O}$  (Lachniet, 2009) and solute concentration  
50 (Fairchild and Treble, 2009). Further, such studies have largely been restricted to mid to high  
51 latitude climate regions where precipitation is larger than evapotranspiration ( $P > \text{AET}$ ).  
52 However, the exception – studies in water-limited regions – tell a different story. Here  
53 dripwater chemistry is influenced to a greater extent by environmental factors such as  
54 evaporation (Pape et al., 2010; Cuthbert et al., 2014; Rutledge et al., 2014) and transpiration  
55 (Tremaine and Froelich, 2013; Treble et al., submitted).

56 While evaporation and transpiration are climate induced, wildfires – common in water-limited  
57 regions - are an agent of change than can dramatically alter evaporation and transpiration rates  
58 by destroying vegetation. The potential impacts of vegetation loss due to fire are both short-  
59 term and long-term. The short-term impacts include: (1) an increase in evaporation rates due  
60 to changes in albedo and/or lack of shading (Silberstein et al., 2008); (2) a reduction in  
61 transpiration; (3) a reduction in soil microbial and root  $\text{CO}_2$  production (Coleborn et al., 2016);  
62 (4) a decrease in-cave air  $\text{CO}_2$  which can influence Mg/Ca and Sr/Ca compositions (Wong and



63 Banner, 2010); (5) the addition of plant ash to the soil profile, increasing concentrations of Ca,  
64 K, Mg and S (Grove et al., 1986; Yusiharni and Gilkes, 2012a); and (6) altered infiltration  
65 patterns (González-Pelayo et al., 2010). Long-term impacts include: (1) the spatial  
66 redistribution of nutrients (Abbott and Burrows, 2003); (2) regrowth impacts on water balance  
67 and nutrient flux (Treble et al., submitted); and (3) a reduction in total soil CO<sub>2</sub> due to the  
68 destruction of CO<sub>2</sub> sequestering microbial communities and plant roots, both significant  
69 sources of soil CO<sub>2</sub> (Coleborn et al., 2016). Despite wildfires regularly impacting water-limited  
70 regions, its impact on δ<sup>18</sup>O and solute concentrations in cave dripwater has not been reported.

71 Here, we analyse dripwater compositions from five years of drip monitoring (August 2005 -  
72 March 2011) in a shallow cave system in southwest Australia following an intense wildfire in  
73 February 2005. We compare this dripwater data to the regional groundwater geochemistry and  
74 published cave dripwater from another cave in the southwest Australia (Treble et al. 2015).  
75 Our analysis provides a unique analysis of the impact of wildfire on (1) cave dripwater,  
76 challenging the existing use of δ<sup>18</sup>O and solute concentrations for paleoclimate proxies in  
77 water-limited regions; (2) local water balance and; (3) bioproductivity during the initial post-  
78 fire period, in which vegetation is removed, as well as during the subsequent period of forest  
79 regeneration.

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## 81 **2 Site description**

82 Our study was conducted in Yonderup Cave in Yanchep National Park (lat 31.5475 S, long  
83 115.690833 E), 20 km north of Perth, southwestern Australia (Fig. 1A). This region has a  
84 Mediterranean climate characterised by dry hot summers and colder wet winters with an  
85 average yearly surface temperature of 15.1 °C and a 25 year average annual rainfall of 664 mm  
86 with 85% falling between May and October. Yonderup cave is located in the Tamala Limestone



87 Formation, a porous, partially lithified calcareous coastal dune sand. This karst process is said  
88 to be “syngenetic” with karstification occurring simultaneously with lithification of the host  
89 rock (Jennings, 1964; Fairchild and Baker, 2012).

90 Yonderup Cave is situated in a tuart forest (*Eucalyptus gomphocephala*), with mature tuart  
91 trees 30 m high, and an understory of shrubs and trees standing 5-10 m high, Sheoak trees  
92 (*Allocasuarina fraseriana*) approx. 5-15 m high, and Balga trees (*Banksia attenuata*, *Banksia*  
93 *menziesii*, *Banksia grandis*, *Allocasuarina fraseriana*, *Xanthorrhoea pressii*). Tree roots are  
94 exposed in the cave, both in the roof (fine roots), and cave floor (thick tap roots). In February  
95 2005, the area above the cave was burnt in an intense wildfire (Fig. 1B, Department of Parks  
96 and Wildlife, pers. comm., 2015), substantially modifying vegetation above the cave including  
97 death of mature trees and complete removal of canopy and understorey in some areas.

98 Over the period of August 2005 to March 2011, two drip sites in Yonderup Cave (Site 1a and  
99 Site 2a), were monitored for their chemical and hydrological variations. These two sites are  
100 22.8 m apart (Site 2a east of Site 1a, ~ 1 m slope towards the East), located at similar depths  
101 below the surface (~ 4 m) within the same chamber (~ 7 m height) and partially separated by a  
102 large boulder fall-in. We use an existing cave survey to determine the location of each cave  
103 drip site relative to the ground surface (Fig. 1). We also compare Mg/Ca and Sr/Ca  
104 concentrations between drip sites to make sure each drip has an independent flow path (Fig. 2).  
105 A soil depth survey was conducted within 5 m of each site (Sup. Table 4), along with visual  
106 vegetation/ground surface observations post-fire. Soil depths were measured every meter with  
107 a dynamic soil penetrometer in north, south, east and west directions and averaged soil depth  
108 above each site were calculated.

109 Site 1a, 30 m from the cave entrance, has a drip source within a large cluster of soda-straw  
110 stalactites known as the ‘Wheatfield’ (Sup. Figure 2B). This circular feature is approximately



111 1 m across and as it appears in an otherwise very sparsely decorated part of the ceiling, suggests  
112 that it represents a focused flow path into the cave. The land surface above this site is flat with  
113 70% covered by shallow heterogeneous soil (average 124 mm thickness) and the remaining  
114 surface is bedrock exposure (approx. 30%). A dead tuart tree, killed and collapsed during the  
115 2005 wildfire, lies directly above Site 1a, entirely removing the upper canopy directly above  
116 Site 1a as no other canopy trees are close enough to provide canopy cover.

117 In contrast, Site 2a situated 50 m from the cave entrance is in a highly decorated part of the  
118 cave known as the ‘Cathedral’ characterised by large icicle shaped stalactites. Above Site 2a,  
119 the soil cover is thicker (200 mm) and more homogenous with no bedrock exposure, and no  
120 trees directly above, however there is a partial canopy cover from adjacent trees ~15 m away.

121

### 122 **3 Data collection**

123 Cave dripwater was collected from 1L high-density polyethylene (HDPE) collection vessels at  
124 the two sites Site 1a and Site 2a between August 2005 and March 2011 at approximately bi-  
125 monthly intervals. The water was separated into three aliquots: two aliquots were filtered with  
126 0.45 µm mixed-cellulose filters into two 50 ml polypropylene bottles for major and minor ion  
127 determination; the third was stored with zero-headspace in a 12 ml amber glass bottle for stable  
128 isotopes. All aliquots were refrigerated below 5°C until analysis. Anion concentrations (Cl and  
129 SO<sub>4</sub>) were determined using a Dionex DX-600 ion chromatograph with self-regenerating  
130 suppressor on one aliquot. The second aliquot was acidified to 2% HNO<sub>3</sub> in the collection  
131 bottle and used for cation concentrations (Ca, K, Mg, Na, Si and Sr) using a Thermo Fisher  
132 inductively coupled plasma-atomic emission spectroscopy (ICP-AES) ICAP7600. An internal  
133 standard with concentrations approximating the cave waters was included in each cation batch  
134 to check for between-run reproducibility.



135 Dripwaters collected between August 2005 and May 2008 were analysed for  $\delta^{18}\text{O}$  using  
136 Isotope-Ratio Mass Spectrometry (IRMS) at the Australian National University (see Treble et  
137 al., 2013 for method). The remaining dripwaters were analysed for  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  at ANSTO  
138 using the Cavity Ring Down Spectroscopy (CRDS) method. Additionally, as there was  
139 sufficient remaining water in the stored aliquots analysed by IRMS for Site 2a, these were also  
140 re-analysed using CDRS to obtain a complete time series for  $\delta^2\text{H}$ . After Jan 2007 dripwater  
141 volume at Site 1a became insufficient to collect all three aliquots. Collections of aliquots were  
142 prioritised in the following order: 1) stable isotopes; 2) cations and; 3) anions.

143 At each cave visit for dripwater sampling, drip rates were manually recorded using a stopwatch  
144 and the level of water accumulated in the bottles was recorded to the nearest 100 ml. Weekly  
145 discharge was estimated using a drip volume of 0.2 ml per drip (Collister and Matthey, 2008).  
146 When timing drip intervals became impractical, only the bottle level was recorded. The period  
147 of overlapping measurements by the two drip recording methods was used to calculate weekly  
148 discharge based on the bottle volume recordings for the periods where the stopwatch method  
149 could not be used.

150 To distinguish dry and wet periods we applied a residual mass curve (RMC), (Hurst, 1951) to  
151 monthly P - AET data. This is the cumulative sum of the monthly anomaly calculated from the  
152 22 year mean and used to generate a time series of cumulative potential water surplus or deficit  
153 starting from Jan 2000, highlighting trends in above average or below average P - AET, we  
154 refer to this calculation as cumulative water balance (CWB) throughout the rest of this paper.

155 Unpublished monthly  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  rainfall data (2005 – 2011) from Perth were obtained from  
156 the Australian Nuclear Science and Technology Organisation (ANSTO; unpublished).  
157 Modelled regional precipitation (P) and actual evapotranspiration (AET or FEW is the sum of  
158 soil evaporation and transpiration by vegetation based on Priestly-Taylor equations) from the



159 Australian Water Availability Project (AWAP) (Raupach et al., 2009; Raupach et al., 2011)  
160 were used with monthly parameters, to determine  $P - AET$ . AWAP precipitation ( $P$ ), actual  
161 evapotranspiration ( $AET$ ) and rainfall  $\delta^{18}O$  data was then used as input data in a forward model  
162 (detailed in the next section) to predict cave dripwater  $\delta^{18}O$  composition. Predictions are based  
163 solely on  $P - AET$  data which are then compared to the dripwater observations.

164 Monthly rainfall  $\delta^{18}O$  and  $\delta^2H$  compositions were amount weighted and fitted with a linear  
165 regression (Hughes and Crawford, 2012) and compared to the long-term groundwater mean  
166 obtained from Turner and Thorpe (2001) and the cave drip data to determine whether  
167 evaporation has affected cave dripwater isotopic composition (see section 4).

168 Post-fire solute and  $\delta^{18}O$  data from Yonderup dripwater are also compared to other relevant  
169 published data. These include, long term Perth rainfall  $\delta^{18}O$  from Turner and Thorpe and Cook  
170 (2001), and local Yanchep rainfall solute data from Hingston and Gailitis, (1976), and  
171 published dripwater data from Golgotha Cave, located 300 km south of Yanchep. Golgotha  
172 Cave has been monitored since 2005 (Treble et al., 2013; Treble et al., 2015; Treble et al.,  
173 submitted; Mahmud et al., 2015). The climate at Golgotha Cave is also Mediterranean, but  
174 receives rainfall of approx. 795 mm/a, which is 23% higher than Yanchep. Both caves are  
175 located within the Tamala Limestone Formation, however the caves vary in depths: Golgotha  
176 Cave is significantly deeper than Yonderup approx. 30 - 35 m. Golgotha Cave is covered by a  
177 more extensive forest of mixed marri/karri (*Eucalyptus calophylla* / *Eucalyptus diversicolor*)  
178 trees and this site has not experienced an intense wildfire since 1992 and no prescribed burns  
179 since 2006.

180





181 **4 Forward model**

182 To compare our measured post-fire dripwater  $\delta^{18}\text{O}$  to the values expected in the case where  
183 there is no fire, we forward model dripwater  $\delta^{18}\text{O}$  based on a monthly discretised water balance  
184 and a mass balance for  $\delta^{18}\text{O}$  using the same method as Baker et al., (2010). The model permits  
185 the mixing of the isotopic composition of incoming monthly rainwater in the soil and epikarst  
186 where inputs are governed by monthly rainfall isotopic composition and rainfall amount, and  
187 outputs by monthly AET and infiltration into a seepage reservoir and into fracture-fed flow.  
188 The latter are determined by hydrological thresholds (see Baker and Bradley, 2010). This  
189 appropriately reflects our knowledge of the geology of the Tamala Limestone, which retains a  
190 high primary porosity (the seepage reservoir) whilst fractures focus flow to specific regions of  
191 the cave ceiling (Treble et al., 2013; Mahmud et al., in review).

192 Water entering the seepage reservoir has a storage time that is expressed as a Gaussian  
193 distribution. This is prescribed with a maximum age of 3 years, reflecting the shallow depth of  
194 our cave system, and a mean and standard deviation that can be specified. In contrast, the  
195 fracture-fed flow is instantaneously passed through the system (i.e. with a travel time of less  
196 than one month). Thresholds for water movement to the seepage reservoir and fracture flow  
197 can be specified to reflect the observed variability of dripwater discharge. The seepage and  
198 fracture-fed components are mixed in a karst store before discharging to the cave with the  
199 modelled dripwater  $\delta^{18}\text{O}$  composition. By request the authors can supply the forward model as  
200 an excel spread sheet.

201 We performed sensitivity analyses on the model, changing the mean and standard deviation of  
202 the seepage reservoir flow time and the threshold values for recharge to the seepage reservoir  
203 and fracture flow, as follows. In order to reflect the observed lag in drip hydrology response to  
204 piston flow (Tooth and Fairchild, 2003) and seasonal recharge patterns we optimized our model



205 to fit our observations of a mean 9-month delay and a standard deviation of 3 months. We then  
206 varied seepage and fracture thresholds and ran multiple scenarios in order to replicate a  
207 response in dripwater isotopic composition which could be accounted for by climate and  
208 bedrock limits alone. Results of various scenarios are explained in section 5.2.



## 209 **5 Results**

210 A time series of monthly  $P - AET$ , cumulative water balance (CWB), discharge, dripwater  
211  $\delta^{18}O$ , and ion concentrations for Sites 1a and 2a from August 2005 – March 2011 are shown in  
212 Figure 3.

213

### 214 **5.1. Water balance**

215 Firstly, we observe a distinct seasonality in the potential water availability ( $P - AET$ ) (Fig. 3A),  
216 where winter months generate an excess ( $P > AET$ ), while summer months generate a deficit  
217 ( $P < AET$ ) in in the soil water availability. Further, CWB shows three distinct trends throughout  
218 the monitoring period: 1) a decline over the period of January 2006 to June 2006, consistent  
219 with very low excess in  $P - AET$ ; 2) an overall rise from June 2006 to February 2010; 3) a  
220 decrease in  $P - AET$  from February 2010 to September 2010. Site 1a and 2a display moderate  
221 and similar discharge rates, at the start of the monitoring period that continue until July 2006;  
222 Site 1a an average of  $90 \text{ ml} \pm 21$  per week and Site 2a an average of  $92 \text{ ml} \pm 23$  per week. This  
223 coincides with infiltration indicated by positive CWB (Fig. 3B). In July 2006, Site 1a  
224 dramatically increases discharge five-fold to  $468 \text{ ml/week}$  on one cave visit, but had decreased  
225 to  $55 \text{ ml} \pm 3 \text{ ml}$  per week on the subsequent visit two weeks later and was completely dry, three  
226 months later. This site has not re-activated since (Department Parks and Wildlife, pers. comm.).  
227 Site 2a shows much less variation in discharge overall, but contains smooth long-term trends.  
228 Two periods of higher discharge are observed in August 2005 to May 2006 (average  $92 \text{ ml} \pm$   
229  $23 \text{ ml}$  per week) and April 2008 to February 2009 (average  $93 \text{ ml} \pm 29 \text{ ml}$  per week), coinciding  
230 with positive trends in CWB.

231

232



233 **5.2 Water isotopes**

234 Dripwater  $\delta^{18}\text{O}$  from Site 1a (Fig. 3C) shows no seasonal pattern but we see a steady increase  
235 of 1‰ to January 2007, then a further steeper rise of 1.5‰ in June 2007, after which the drip  
236 ceases. Dripwater  $\delta^{18}\text{O}$  from Site 2a presents an overall increasing trend rising from -3‰ to +  
237 0.7‰ over the monitoring period with a six-month quasi-seasonal signal (approx. 2‰ range)  
238 that peaks from May/June to November/December generally coinciding with months when  
239 infiltration from rainfall occurs. We hypothesise that the  $P < \text{AET}$  environment in drier summer  
240 months isotopically enriches soil water, but this only arrives at the cave when seepage  
241 thresholds are exceeded in periods of  $P > \text{AET}$  (winter months).

242

243 We undertook forward modelling of the dripwater  $\delta^{18}\text{O}$  data from our rainfall isotopic data.  
244 Our sensitivity analyses showed that seepage threshold values greater than 30 mm resulted in  
245 the seasonal cessation of dripwater, which is not observed at our sites, thus the seepage  
246 threshold is  $< 30$  mm. Next we varied the fracture threshold between 10 and 1000 mm, the  
247 wide range reflecting our uncertainty over this parameter. Given seepage flow is more  
248 dominant in these highly porous highly calcarenite host rocks (Treble et al., 2003) we set our  
249 chosen forward model to have a low threshold of 15 mm for seepage flow and a higher fracture  
250 threshold of 100 mm to allow for dominant seepage flow but also some fracture flow in time  
251 of high recharge. We set a  $9 \pm 3$  month transit time for water from the surface to dripwater to  
252 account for seasonal  $P - \text{AET}$  recharge forcing (Fig. 4). The mean of our forward-modelled  
253 dripwater  $\delta^{18}\text{O}$  (15 – 100 mm) is -3.98‰, less than the mean of Perth rainfall (-3.1‰) and  
254 higher than the mean regional groundwater composition (-4.68‰). The time series of modelled  
255 dripwater  $\delta^{18}\text{O}$  (Fig. 3C) falls from -4.7 to -5.2 ‰ from the beginning of monitoring until  
256 January 2007, but rises to -3.3‰ in January 2009, then rises to around -2‰ until Jan 2010



257 when it finally declines to -3.2‰ at the end of the monitoring period. This chosen 15 mm –  
258 100 mm threshold scenario (Fig. 4) and all other modelled scenarios show a gentle increasing  
259 trend, suggesting enrichment – a possible drying trend. Even when incorporating this slight  
260 increasing  $\delta^{18}\text{O}$  trend, the difference between both our observed sites and modelled  $\delta^{18}\text{O}$   
261 dripwater threshold scenarios at statistically significant (Table 1; detailed below).

262 The slopes in the measured  $\delta^{18}\text{O}$  time-series at Site 1a (0.13‰ /month) and Site 2a (0.05‰  
263 /month) are both higher than that of the climatically-driven forward modelled dripwater  $\delta^{18}\text{O}$   
264 (0.03‰ /month) and there are considerable differences between the sites in their response. In  
265 all meaningful modelled scenarios (Fig. 3), taking into account climate and bedrock hydrologic  
266 processes, observed dripwater compositions are significantly more enriched than forward  
267 modelled  $\delta^{18}\text{O}$ . This is demonstrated through our t-test which set out to determine if the  
268 difference between observed  $\delta^{18}\text{O}$  at each site and forward modelled  $\delta^{18}\text{O}$  is statistically  
269 significant. The results of our test (Table 1) show both sites are significantly different from  
270 modelled  $\delta^{18}\text{O}$ . It is also well-above the mean  $\delta^{18}\text{O}$  for cave dripwater at Golgotha Cave (-  
271 3.9‰; Treble et al, 2013). Figure 5 shows that cave dripwater isotopic enrichment occurs along  
272 a least squares regression (LSR) that falls within the standard error ( $\pm 0.45\%$ ) of the slope for  
273 the local meteoric water line (LMWL, weighted LSR). This confirms a *type 1* scenario  
274 suggested by Cuthbert et al., (2014) where a single water source dominates cave dripwater and  
275 shows increased near-surface evaporation in a humid subsurface environment ( $> 95\%$  relative  
276 humidity).

277

### 278 **5.3 Water Solutes**

279 There are significant differences in solute concentrations and trends between the two sites (Fig.  
280 3E, 3F, 3G and 3H). Solute concentrations are typically higher at Site 1a versus Site 2a and



281 they demonstrate opposite trends post-fire. At Site 1a, Cl, Ca, Mg and Sr decline overall,  
282 although this trend is step-wise for Ca, and reverses for Cl ~ 6 months before the drip ceases.  
283 The trends in these solutes are inconsistent with that expected from climate and the observed  
284 CWB (Fig 3A). Solute concentrations decrease with decreasing CWB, where as a decrease in  
285 CWB should result in an increase in solute concentrations, as a result of up-concentration of  
286 solutes in response to evaporation. We also note that initial Cl and other solute concentrations  
287 at Site 1a are twice that at Site 2a, and there are tree roots visible above Site 1a and not at Site  
288 2a, indicating these may be effecting dripwater composition by concentrating solutes.  
289 Furthermore at Site 2a Cl and other solute concentrations show a direct relationship to CWB  
290 (i.e. increasing solute concentration with decreasing CWB and decrease with increasing CWB).

291 Trends in SO<sub>4</sub> and K are more subtle than for other solutes, K only shows a slight decline prior  
292 to Site 1a ceasing to drip and SO<sub>4</sub> appears to be slightly increasing, although our SO<sub>4</sub> time  
293 series is shorter. Similarly, trends in SO<sub>4</sub> and K for Site 2a are subtle, although they are slightly  
294 increasing through time. K and SO<sub>4</sub> concentrations are again, 2 - 3 times higher at Site 1a versus  
295 Site 2a and are considerably higher than those recorded at Golgotha Cave (Table 2) whilst  
296 concentrations for other solutes between these cave sites are more comparable.

297 In summary, while post-fire solute concentrations at Yonderup, are typically higher at Site 1a  
298 versus Site 2a, they demonstrate opposite trends post-fire. While Site 2a solutes demonstrate a  
299 direct relationship with climate driven CWB, at Site 1a, they show a more complex  
300 relationship, which we argue is largely influenced by changes in a combination of local factors  
301 including vegetation change, discharge and evaporation post-fire.

302



303

## 304 **6 Discussion**

305

### 306 **6.1 Post-fire hydrology**

307 Discharge at Site 1a is inconsistent with CWB: discharge rose as rainfall fell below the long-  
308 term mean ( $P < AET$ ) (Fig. 3A and 3B), suggesting that Site 1a received a localised increase  
309 post-fire in available water despite the declining input from rainfall. In contrast, discharge at  
310 Site 2a post-fire is related more directly to the cumulative water balance (CWB), with higher  
311 discharge coinciding with periods of higher water surplus suggesting a simple hydrological  
312 response driven by infiltration. These post-fire observations are consistent with dripwater Cl  
313 and other solutes at Yonderup. Chloride is a chemically conservative solute (Graedel and  
314 Keene, 1996), hence its concentration in dripwater will primarily be a function of physical  
315 processes such as dilution, mixing (Tooth and Fairchild, 2003; Tremaine and Froelich, 2013)  
316 and potentially evaporation and transpiration (Treble et al., in prep).

317 We propose that Cl concentrations at Site 2a increases during the period of declining CWB  
318 due to increased evapotranspiration and that this trend is reversed when rainfall infiltration  
319 increases ( $P > AET$ ) causing dilution (Fig. 3E). Similarly, the rise in discharge at Site 1a  
320 coincided with falling Cl concentrations suggesting dilution. However, since this cannot be due  
321 to increased rainfall, a different mechanism is required to explain the dilution of Cl at Site 1a.  
322 During the fire, a large Tuart tree, growing above Site 1a (within 5 m according to our survey),  
323 was destroyed. Transpiration by deeply-rooted trees in this environment has been implicated  
324 in producing high Cl concentrations in the unsaturated zone (Turner et al., 1987) and the impact  
325 of this process on dripwater at Golgotha Cave has been demonstrated, with dripwater Cl  
326 concentrations being at least five times higher than rainfall concentrations owing to ion  
327 exclusion during transpiration. This coupled with the initial concentration of Cl at Site 1a (Tuart



328 tree) being double that of Site 2a (no Tuart tree) we believe transpiration is largely controlling  
329 Cl concentrations at this site. Thus we propose that a reduction in localised transpiration, and  
330 consequently an increase in discharge following the fire, is driving the dilution that lead to a  
331 decline in Cl at Site 1a. Moreover, post-fire Cl values at the shallow Yonderup Cave, are more  
332 than double baseline Cl concentrations from groundwater and cave sites such as Golgotha Cave  
333 (Table 2). The Cl response at both sites strongly suggests fire can alter dripwater Cl by altering  
334 transpiration and evaporation rates. The impact of this on the dripwater chemistry is discussed  
335 further and a conceptual model devised later in this section.

336

## 337 **6.2 Post-fire carbonate chemistry**

338 Similar to Cl, declining concentrations of carbonate metals (Mg, Ca and Sr) at Site 1a reflect  
339 dilution that we interpret as a reduction in post-fire transpiration owing to the death of the tuart  
340 tree. We interpret that the rising trend at Site 2a reflects increasing solute concentrations due  
341 to a rise in post-fire evapotranspiration owing to vegetation recovery and evaporation of near-  
342 surface water stores (Fig. 3F, 3G and 3H respectively). Additionally, Ca concentrations at Site  
343 1a decrease in a non-linear fashion (Fig. 3G), with a step-like decrease each autumn suggesting  
344 that a further mechanism, in addition to dilution, is contributing to falling dripwater Ca  
345 concentrations at Site 1a.

346 There are a number of mechanisms that could influence post-fire Ca concentrations, including:  
347 (1) increased evaporation in the near-surface inducing prior calcite precipitation (PCP) shown  
348 by the diagnostic model for water-rock interactions that includes PCP in Figure 6 (Fairchild et  
349 al., 2000; Sinclair 2011; Treble et al., 2015). (2) A decrease in the amount of microbial and  
350 root respiration, which have both been implicated as a significant source of CO<sub>2</sub> (Coleborn et  
351 al., 2016; Treble et al., 2015) and hence a decrease in carbonic acid and a decrease in dissolved





352 Ca from lower carbonate mineral dissolution by carbonic acid. (3) The addition of plant ash  
353 (Yusiharni and Gilkes, 2012a) and highly soluble CaO, produced by the burning of exposed  
354 surface rock to the fire (Yusiharni and Gilkes, 2012b) to the soil profile, could influence Ca  
355 availability through leaching over time in dripwater. The relative importance of these three  
356 processes is difficult to assess in our data; processes (1) and (2) cannot be constrained as our  
357 in-cave  $p\text{CO}_2$  data is limited while for (3) the addition of plant ash and CaO would increase Ca  
358 availability in cave dripwater, however it is unclear whether this would be reflected in  
359 dripwater composition as (1) and (2), which are not constrained can alter Ca concentrations, as  
360 they influence cave dripwater after (3). For example: the death of the tuart tree may have  
361 reduced in-cave  $p\text{CO}_2$  from roots – a key driver of Ca concentrations – thus reducing Ca  
362 concentrations in the dripwater. However as previously stated it is difficult to constrain a  
363 particular mechanism that is controlling the composition of Ca at Site 1a.

364 Sulphate and K post-fire at Site 1a show abnormally high  $\text{SO}_4$  and K (Fig. 3H and 3I), and an  
365 increase in  $\text{SO}_4$  concentrations which is counterintuitive to the initial dilution signal (owing to  
366 reduced transpiration) interpreted for the other solutes. While at Site 2a,  $\text{SO}_4$  and K show an  
367 increase similar to other solutes and can be explained by the inferred increase in  
368 evapotranspiration. The  $\text{SO}_4$  and K response at Site 1a suggests an additional process is  
369 affecting  $\text{SO}_4$  and K concentrations post-fire (Fig. 3H and 3I). The majority of aboveground  
370  $\text{SO}_4$  is predominantly stored within the lower to middle storey of the forest (O'Connell and  
371 Grove, 1996) and post-fire soils have been found to contain 23% more S and 16% more K than  
372 pre-fire soils due to biomass-sourced ash deposition (Grove et al., 1986). Thus differences in  
373 biomass above the two sites may explain the high concentrations at Site 1a and differences in  
374 between drip site  $\text{SO}_4$  and K concentrations post-fire. We suggest dripwater  $\text{SO}_4$  and K at Site  
375 1a maintains relatively high constant concentrations (Fig. 3H and 3I) due to increased source  
376 input (from burnt vegetation and litter) until the increase in near surface evaporation from 2007



377 onwards that drives even higher concentrations mimicking the Cl and  $\delta^{18}\text{O}$  trend. This is  
378 consistent with Site 2a, which has much less biomass above the site and thus lower initial  $\text{SO}_4$   
379 and K concentrations, further these solutes show a steady increasing trend that is consistent  
380 with increased evapotranspiration post-fire displayed in other solutes. We compare our  
381 Yonderup results to Golgotha to put  $\text{SO}_4$  and K concentrations into context. Lower  
382 concentrations of  $\text{SO}_4$  at Golgotha Cave, which last experienced a wildfire in 1992 and a  
383 prescribed burn in 2006, strongly suggest that Yonderup has comparatively high concentrations  
384 of  $\text{SO}_4$  and K (Table 2). Given the high intensity of the wildfire and the large amount of  
385 resulting ash deposited, we suggest that these high concentrations reflect the intensity of the  
386 fire and the biomass burnt above Yonderup (Grove et al., 1986).

387

### 388 **6.3 A multi-proxy fire signal in dripwater**

389 Here we propose post-fire scenarios for both sites at Yonderup (refer to conceptual model Fig.  
390 7) that account for the observed dripwater chemistry that have been shown to be altered by  
391 local, changes in the surface soil/vegetation/hydrology initiated by an intense wildfire. A  
392 relatively straight-forward relationship between CWB, discharge and Cl concentrations at Site  
393 2a, suggests increased concentration of solutes (Ca, Mg, Sr, K and  $\text{SO}_4$ ) in response to an  
394 increase in near-surface evaporation (Fig. 3C and 3D). At Site 1a, we hypothesize both the  
395 death of the tuart tree in the wildfire - which resulted in a reduction of transpiration - and a  
396 subsequent increase in rainfall infiltration lead to an initial dilution of solutes causing a  
397 decrease in Cl, Mg, Ca and Sr concentrations. This is supported by the modelled  $\delta^{18}\text{O}$  agreeing  
398 with dripwater  $\delta^{18}\text{O}$  until 2007 (Fig. 3).

399 Tuart tree transpiration at the surface and subsurface ranges from  $-0.86 \pm 0.11$  MPa in the  
400 summer to  $-0.35 \pm 0.02$  MPa in the winter (Drake et al., 2011). We suspect this potential created



401 by transpiration was creating a positive hydraulic gradient towards Site 1a and maintaining  
402 high concentrations of Cl (the conservative solute) in dripwater. Post-fire a multitude of  
403 processes such as the increase in discharge, the decrease in transpiration, the increase in near-  
404 surface evaporation and post-fire ash deposition explain the various trends observed in  
405 dripwater chemistry. Trends such as the gradual decline in concentrations of Cl, Mg and Sr that  
406 reflect dilution and the decrease in transpiration and the consistently high SO<sub>4</sub> and K  
407 concentrations that are largely attributed to post-fire ash deposition (Grove et al., 1986).  
408 Accordingly, the combination of a lower hydraulic gradient and a higher evaporation rate –  
409 reflected by the sharp increase in δ<sup>18</sup>O post Jan 2007 - resulted in depletion of the near-surface  
410 reservoir feeding Site 1a, and subsequently an inactivation of the drip site.

411 At Site 2a there was a lower-middle understory above the site that was burnt during the fire.  
412 We propose that the loss of this lower-middle understory increased the near-surface  
413 evaporation signal and initially decreased transpiration but as lower storey vegetation  
414 recovered an increase in transpiration and a high evaporation rate (still due to a low albedo and  
415 limited forest litter) resulted in the increasing concentration of solutes (Cl, Mg, Ca, and Sr) and  
416 δ<sup>18</sup>O (Fig. 3E, 3F, 3G and 3H) at Site 2a. In contrast at Site 1a various factors including:  
417 dilution, the reduction of canopy cover from the tuart tree coupled with a reduction in albedo  
418 (blackened soil, Fig. 1) from the fire, may have contributed to higher surface temperatures  
419 resulting in evaporation-driven high δ<sup>18</sup>O values. However, the reduction in transpiration due  
420 to the death of tree eventually caused a decline in solutes.

421 Hence we propose that differences in surface vegetation above sites influence site specific drip  
422 chemistry, for example; Site 1a has a larger driver of transpiration - the tuart tree - which died  
423 during the fire and subsequently influences discharge post-fire, while at Site 2a the fire impact  
424 is limited to understory shrub vegetation and discharge is much less influenced post-fire.



425 Furthermore, a larger abundance of biomass - the tuart tree - at Site 1a and an increase in  
426 leaching as a result of increased discharge can account for high concentrations of SO<sub>4</sub> and K at  
427 Site 1a, while less biomass above Site 2a, a stable low discharge and predominantly near-  
428 surface evaporation and transpiration dominated response accounts for lower concentrations of  
429 SO<sub>4</sub> and K at Site 2a. While this heterogeneity in response between the two sites at Yonderup  
430 is significant, so too is the comparison between Yonderup drip chemistry and the drip chemistry  
431 at Golgotha, which – more generally - highlights that, despite site specific variation, an intense  
432 wildfire has multi-year effects on dripwater composition in shallow caves post-fire.

433

## 434 **7 Conclusions and avenues for future research**

435 We isolate a post-wildfire response by comparing a recently burnt cave monitoring site with  
436 baseline values of local pre-fire, groundwater and nearby cave monitoring data. We provide a  
437 novel analysis of the multi-year impacts wildfire has on cave dripwater. Our analysis shows a  
438 strong hydrologic relationship between surface environments and shallow caves that are  
439 located within the tree rooting zone. This finding is especially important in water-limited  
440 environments ( $P < ET$ ) as the overlying vegetation can exert controls on the cave  
441 hydrogeochemical environment. The post-wildfire dripwater response is clearest in  $\delta^{18}O$  and  
442 Cl due to their sensitivity to changes in near-surface evaporation and SO<sub>4</sub> and K due to their  
443 leaching from the ash. Other solutes such as Mg, Sr and Ca support these conclusions when  
444 using a multi-proxy approach.

445 We propose a conceptual model for a multi-year post-wildfire cave dripwater response in  
446 forested water-limited regions. This involves a 5 - 10 yr response of: 1) higher  $\delta^{18}O$  and Cl in  
447 cave dripwater due to increased evaporation and decreased shading soon after the wildfire; 2)  
448 increased K and SO<sub>4</sub> due to the leaching of biomass-sourced ash; and 3) increased variability



449 in Mg, Sr and Ca due to changes in evaporation, transpiration and water-rock interactions post-  
450 fire. We would expect a full recovery within 10 - 20 years after the wildfire and a return to pre-  
451 fire isotopic, major and trace element concentrations as a result of revegetation growth (Treble  
452 et al., submitted) and re-establishment of the pre-fire albedo.

453 Our post-wildfire dripwater response from  $\delta^{18}\text{O}$  shows ~ 2‰ increase - relative to forward  
454 modelled climate  $\delta^{18}\text{O}$ , regional groundwater and Golgotha Cave  $\delta^{18}\text{O}$  data. If preserved at  
455 equilibrium this is equivalent to some of the largest changes seen in the Quaternary record.  
456 This highlights the significance of a fire signal that could be misinterpreted as climate  
457 variability especially when combined with other proxies that would also be affected by fire,  
458 such as speleothem growth rate. Furthermore, the potential preservation of a unique fire  
459 signature of  $\delta^{18}\text{O}$ , S and Cl in speleothems, combined with trace elements (Mg, Sr), opens a  
460 new avenue for research into paleo-fire records.

461



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472

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597 Table 1. Summary of two sample t-test of Site 1a and 2a  $\delta^{18}\text{O}$  values in Figure 2D against  
598 forward modeled  $\delta^{18}\text{O}$ .

Variable	Number of samples	Mean	Standard deviation	P - value	Significance (P < 0.05)
Site 1a	23	-2.80	0.94	0.002	Yes (P > 0.05)
Site 2a	41	-1.56	0.83	<0.001	Yes (P < 0.05)
Modelled	72	-3.58	1.23		

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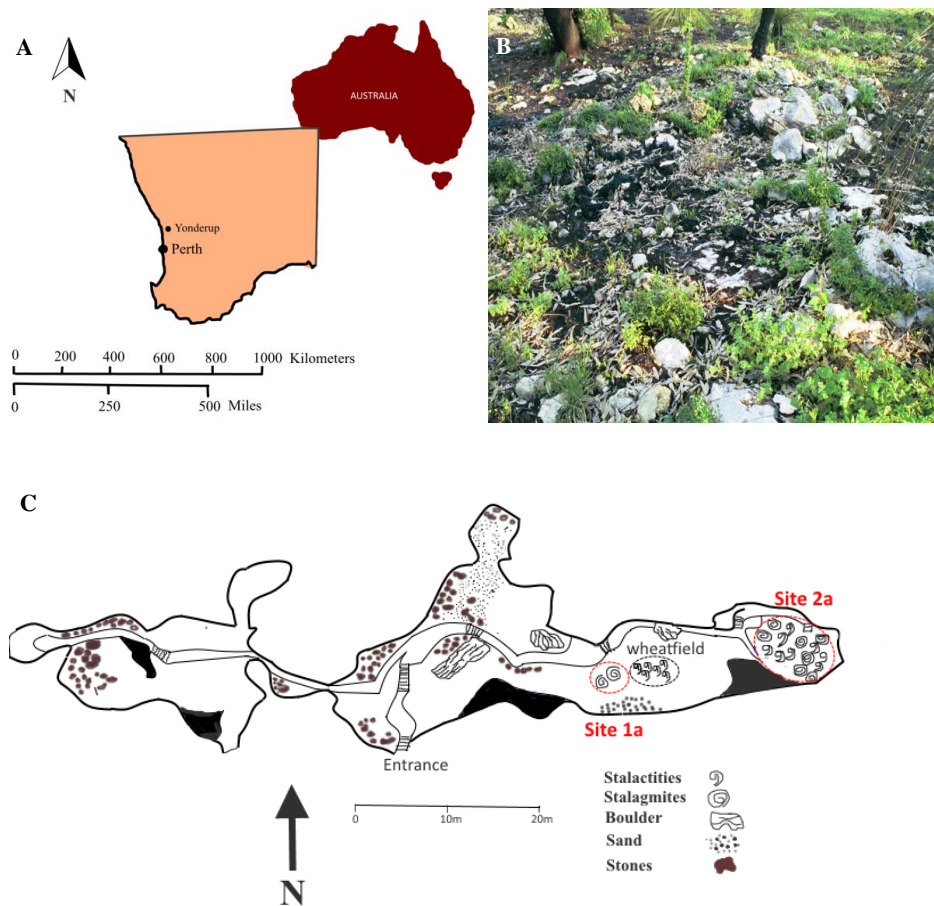


602 Table 2. A summary table comparing the hydrogeochemistry of, (A) shallow Yonderup drip sites burnt in a 2005 wildfire to; (B) deeper  
 603 Golgotha cave sites burnt in a 1992 wildfire and a 2006 prescribed burn; (C) unpublished Perth rainfall data during the monitoring period  
 604 (ANSTO); (D) rainfall isotopic composition (Turner and Thorpe, 2001); (E) groundwater isotopic composition (Turner and Thorpe, 2001); (F)  
 605 Yanchep rainfall solute composition (Hingston and Gailitis, 1976).

606 Location		Discharge (ml/day)	Ca (mmol/L)	Mg (mmol/L)	Sr (mmol/L)	Cl (mmol/L)	SO <sub>4</sub> <sup>2+</sup> (mmol/L)	K (mmol/L)	δ <sup>18</sup> O (per mil)	n
<b>(A) Yonderup</b>										
Site 1A	Median	12.2	1.2	0.46	0.01	8.1	0.25	0.15	-3.2	27
	SD	2.6	0.6	0.13	0.2 × 10 <sup>-2</sup>	0.6	0.02	0.03	1.23	
Site 2A	Median	6.1	1.42	0.37	0.01	5.91	0.14	0.05	-1.48	49
	SD	4.8	0.2	0.04	0.5 × 10 <sup>-3</sup>	1.75	0.05	0.8 × 10 <sup>-2</sup>	0.84	
<b>(B) Golgotha</b>										
Site 1A	Median	63	1.2	0.3	0.2 × 10 <sup>-2</sup>	3.35	8.0 × 10 <sup>-2</sup>	0.02	-4.1	85
	SD	4.7	0.22	0.03	0.183 × 10 <sup>-3</sup>	0.18	0.4 × 10 <sup>-2</sup>	0.3 × 10 <sup>-2</sup>	0.3	
Site 1B	Median	40	1.1	0.3	0.2 × 10 <sup>-2</sup>	3.41	8.0 × 10 <sup>-2</sup>	2.3 × 10 <sup>-2</sup>	-3.9	82
	SD	2.0	0.25	0.03	2.39 × 10 <sup>-4</sup>	0.25	0.5 × 10 <sup>-2</sup>	0.5 × 10 <sup>-2</sup>	0.1	
Site 2A	Median	47	1.1	0.32	0.2 × 10 <sup>-2</sup>	2.76	0.13	3.8 × 10 <sup>-2</sup>	-3.9	77
	SD	10	0.25	0.02	0.1 × 10 <sup>-3</sup>	0.22	0.8 × 10 <sup>-2</sup>	0.5 × 10 <sup>-2</sup>	0.1	
Site 2B	Median	67	1.22	0.33	0.2 × 10 <sup>-2</sup>	4.49	9.8 × 10 <sup>-2</sup>	3.8 × 10 <sup>-2</sup>		84
	SD	85	0.27	0.03	0.2 × 10 <sup>-3</sup>	0.24	0.6 × 10 <sup>-2</sup>	0.7 × 10 <sup>-2</sup>		
Site 2E	Median	524	1.9	0.31	0.1 × 10 <sup>-3</sup>	4.11	0.085	3.8 × 10 <sup>-2</sup>		51
	SD	45	0.2	0.02	0.2 × 10 <sup>-3</sup>	0.17	0.4 × 10 <sup>-2</sup>	0.4 × 10 <sup>-2</sup>		
<b>(C) ANSTO Rainfall</b>	Mean								-3.1	64
<b>(D) CSIRO long term rainfall (Perth)</b>	Mean								-3.85	165
<b>(E) Regional groundwater (Perth)</b>	Mean								-4.68	43
<b>(F) Yanchep Rainfall Solutes</b>	Mean		0.008			0.01	0.1 × 10 <sup>-2</sup>	0.4 × 10 <sup>-2</sup>	0.4	



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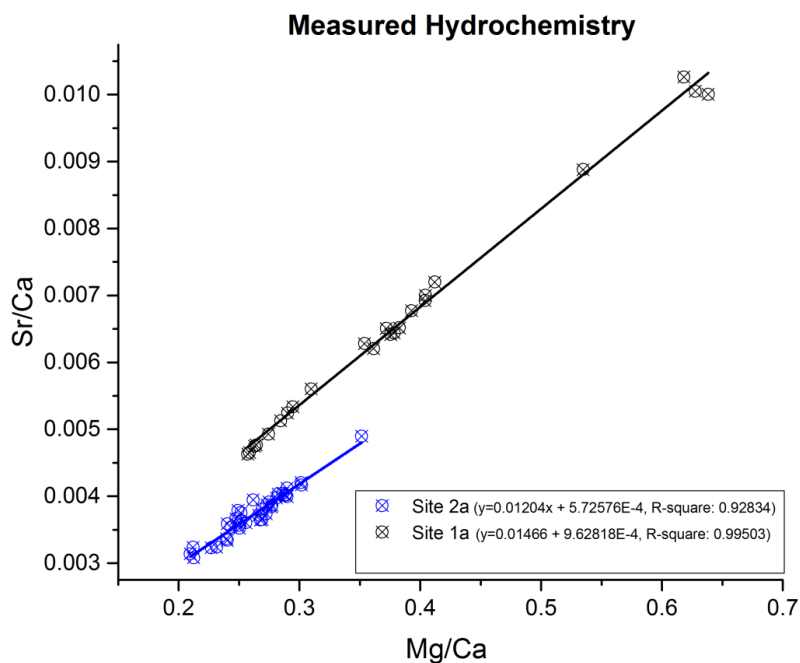
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609 Figure 1. Shows the geographical location of our study site (A), a post fire photograph of the  
610 area (B) taken in August 2005 a photo of recovering shrubs and grass post wildfire, and (C) a  
611 map of Yonderup cave to scale originally surveyed by Watts and Henley, (1973).

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616 Figure 2. Relationship between Sr/Ca vs Mg/Ca for both Site 1a (black) and Site 2a (blue), this

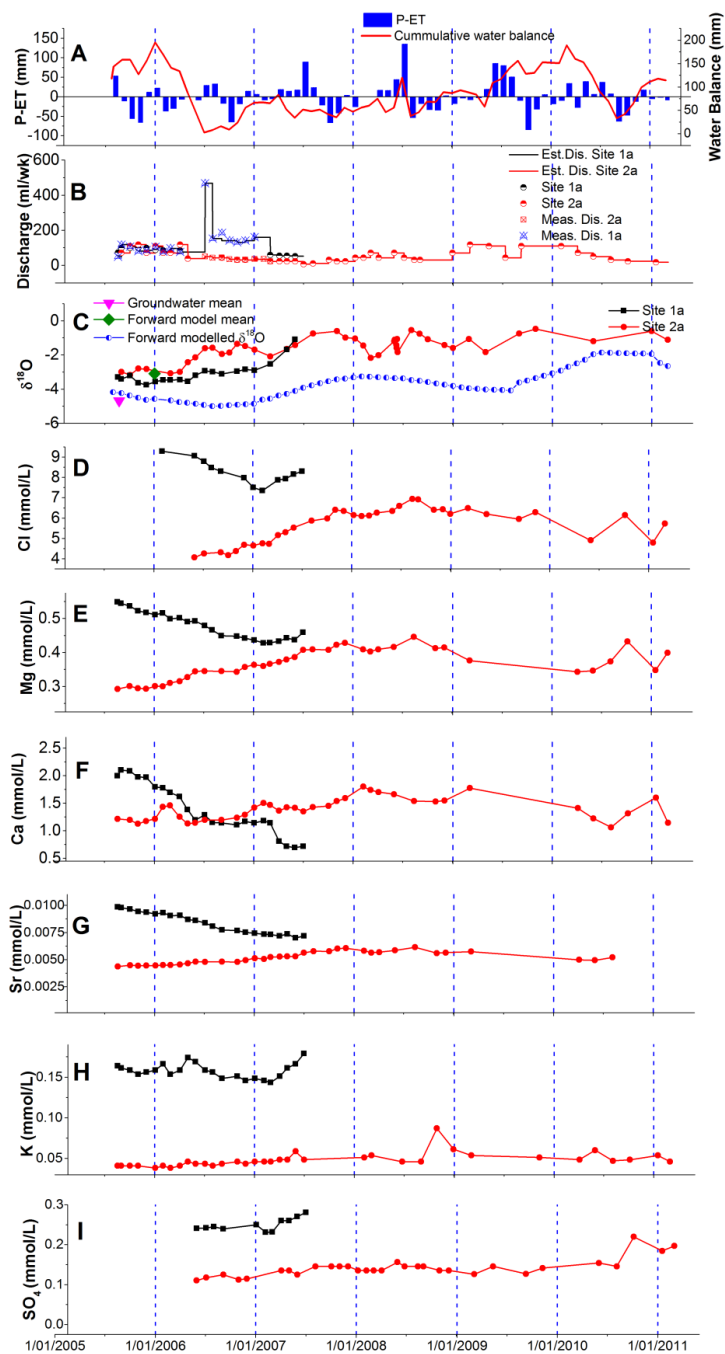
617 indicates that both sites have independent flow paths.

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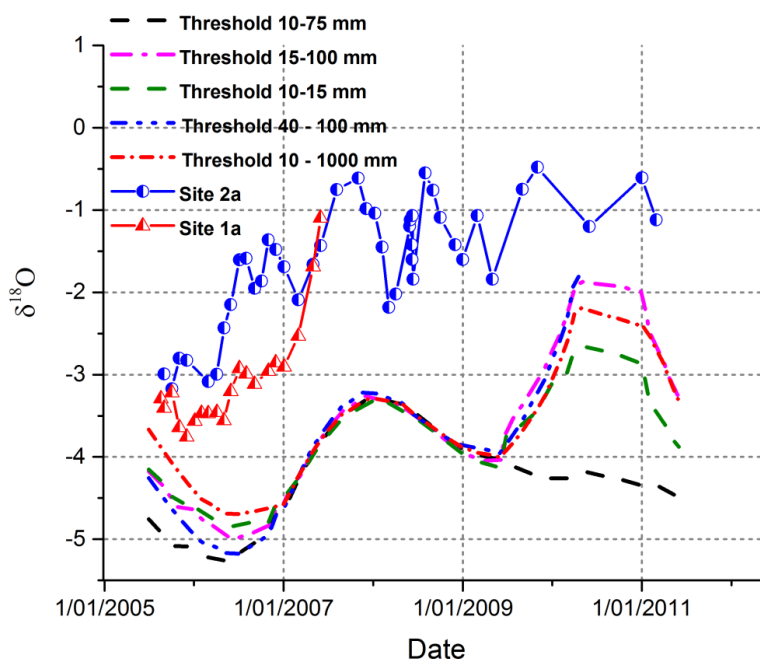


623 Figure 3. Presents a post-fire time series analysis of Site 1a and Site 2a. Note: Site 1a dries up  
624 in June 2007. (A) Precipitation – actual evapotranspiration (P – AET), shows seasonal  
625 variations of excess (above threshold) and deficiency (below threshold) on a monthly scale  
626 overlaid with cumulative surface water balance. (B) Discharge is given in ml/week actual  
627 measured discharge data is given as Meas. Dis. 1a and 2a, while calculated data is given as  
628 Site 1a and Site 2a this is then inferred to give our estimated measured discharge (Est. Dis.  
629 Site 1a and Site 2a). Site 1a shows a spike in measured discharge in August 2006 and a  
630 consequent decrease until the site is dry, while Site 2a shows little variability in discharge  
631 throughout the monitoring period. (C) Shows observed  $\delta^{18}\text{O}$  composition of cave dripwater  
632 from Site 1a and Site 2a with the forward modelled  $\delta^{18}\text{O}$  (15 – 100 mm; blue) and mean  
633 modelled  $\delta^{18}\text{O}$  (green) and long-term groundwater  $\delta^{18}\text{O}$  mean (pink). (D) Cl declines at Site  
634 1a until Feb where it shows a slight increase until the drip becomes dry, while Site 2a shows  
635 a steady increase until in July 2007 where it stabilizes for the remainder of the monitoring  
636 period. (E) Post-fire response shows a decline in Mg at Site 1a until dry and a steady increase  
637 at Site 2a until in Dec 2007 where it remains stable. (F) Site 1a shows step wise decline in Ca  
638 at Site 1a until dry, while at Site 2a a very gradual increase until June 2007 is seen while the  
639 remainder of the monitoring period remains steady. (G) Response for Sr shows Site 1a  
640 declining and Site 2a peaking in Dec 2007; an identical response to Mg and at both sites. (H)  
641 K post-fire at Site 1a shows high concentrations, double that of Site 2a but stable, while at  
642 Site 2a shows a very gradual increase but is relatively stable over time. (I)  $\text{SO}_4$  at Site 1a  
643 remains stable until 2007 where it increases until dry and at Site 2a it shows a steady  
644 increasing trend. Site 1a has double the initial absolute concentration in comparison to Site  
645 2a, similar to other solutes.

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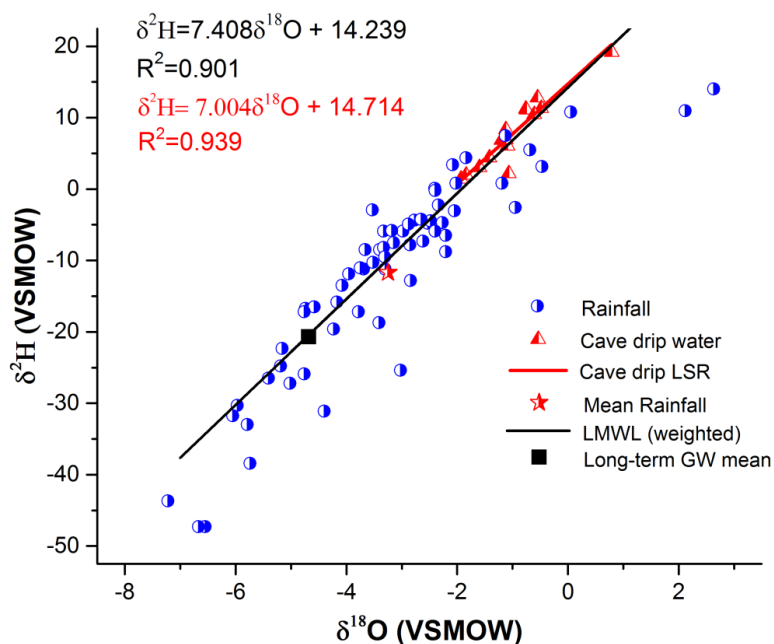


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649 Figure 4. Shows outputs under varying thresholds in our forward model (model from Baker et  
650 al., 2010) which accounts for climatic and various epikarst threshold values that control  
651 isotopic values. Given no output matches observed dripwater composition we can infer that a  
652 localised factor has influenced isotopic compositions



653



654

655 Figure 5. Shows compositions of  $\delta^2\text{H}$  against  $\delta^{18}\text{O}$  of our rainfall data (blue) during the  
656 monitoring period (ANSTO, unpublished), cave dripwater (red), long-term local groundwater  
657 mean (black) (0 – 10 ka, n=43, Southern Perth Basin from Turner and Thorpe, (2001)), and  
658 rainfall mean (red). A least squares regression (LSR) is plotted for cave dripwater (red line)  
659 and falls close to the local meteoric water line (LMWL, black) which is calculated using a  
660 weighted least squares regression (WLSR) using Hughes and Crawford, (2011).

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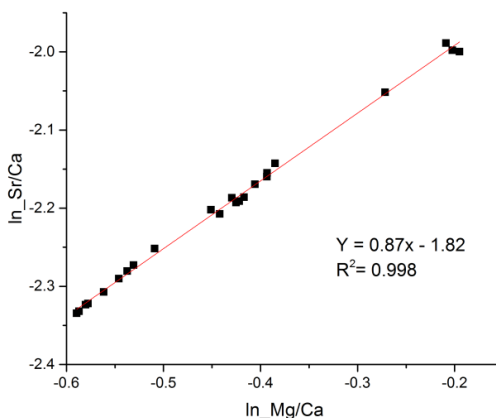
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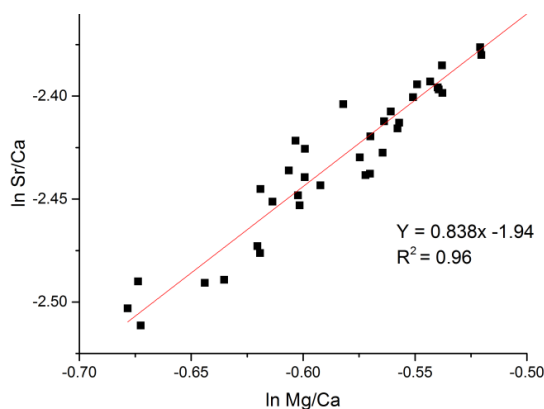
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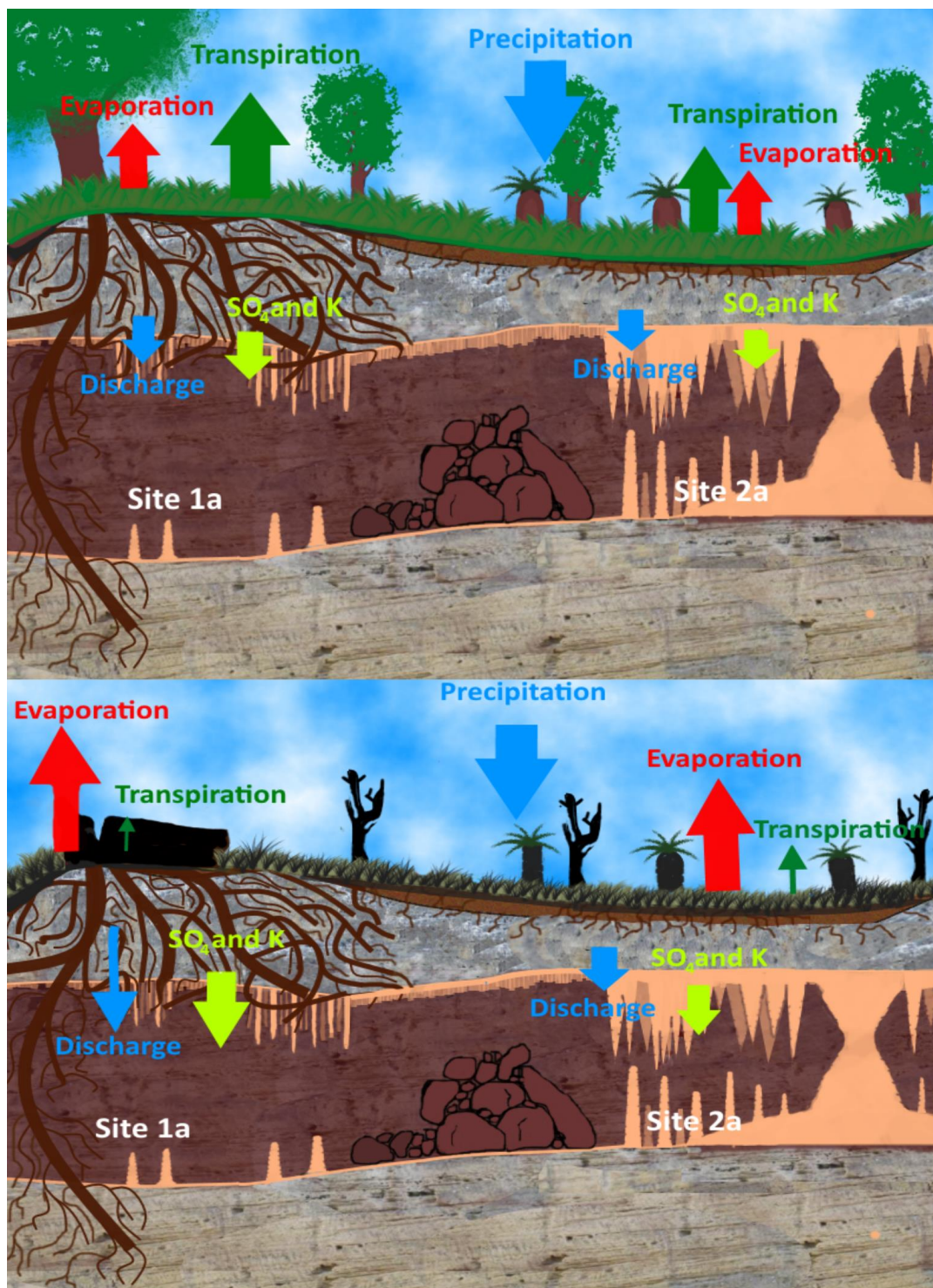
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674 Figure 6. Shows  $\ln_{Sr/Ca}$  vs.  $\ln_{Mg/Ca}$  using the method described in Sinclair, (2011) at Site 1a  
675 (A) and Site 2a (B) to determine if PCP is occurring. Both sites fall within the diagnostic model  
676 for water-rock interactions that include PCP by Sinclair (2011) and Treble et al. (2015) who found  
677 PCP to be a dominating in-cave process in Golgotha Cave 300 km, south of Yonderup Cave. Thus  
678 a number of processes may be responsible for causing heterogeneity in Ca concentrations.

679





681 Figure 7) Evaporation (red) increases post-fire at both sites due to a reduction in albedo and  
682 vegetation cover while precipitation (blue) remains the same and initial transpiration (green)  
683 decreases, but recovers over time. Site 2a shows higher  $\delta^{18}\text{O}$  and an increase in concentrations of  
684 solutes including  $\text{SO}_4$  and K (lime) due to evaporation and slow increase in transpiration due to  
685 vegetation recovery, with cumulative water balance (CWB) remaining the same. While Site 1a  
686 shows higher  $\delta^{18}\text{O}$  in response to evaporation and a decline in solute concentrations in response to  
687 increased discharge and a decrease in transpiration. However since  $\text{SO}_4$  and K are from biomass-  
688 sourced ash increased discharge post fire at site 1a increases  $\text{SO}_4$  and K in dripwater. Discharge  
689 increased immediately (blue) at this site but only lasted a few months post-fire. The drip became  
690 inactive one year after the fire an increase evaporation outweighed the reduction in transpiration  
691 (green), leading to depletion of the near-surface reservoir feeding Site 1a and an in active drip site.