

1 Dear Editor,

2 First, we thank the reviewers for their review of our manuscript. Their comments have helped us  
3 improve the quality of our manuscript. Please find attached our response to the reviewers'  
4 comments on our manuscript, 'A post-wildfire response in cave dripwater chemistry' by Nagra et al.

5 We have provided a detailed reply to all questions and comments raised by the three reviewers, as  
6 requested. We will provide a revised version should the decision be made to accept the manuscript.

7 If you have any further questions we will be happy to answer them. Thanks again!

8 Kind regards,

9 For the authors,

10 Gurinder Nagra

11

12

13 *Reviewer Comments*

14 Reviewer #1

15 **The authors present an excellent multi-year cave monitoring study that uniquely provides insight**  
16 **into the response of cave dripwater isotope and geochemical compositions to POST-fire**  
17 **vegetation dynamics.**

18 We thank the reviewer for these kind remarks.

19

20 **Overall, the manuscript does not present a rigorous, integrated argument. Many of the**  
21 **interpretations come across as speculative without rigorous constraint or consideration of**  
22 **alternative hypotheses.**

23 We appreciate the reviewers' comments, and agree to address them in our revised manuscript. We  
24 will go through the manuscript again to further integrate our argument and assure the manuscript  
25 reads more smoothly. And to apply more rigor to our argument, we will consider alternate  
26 hypotheses, and place more clear constraints on the claims we make. We constrain our claims by  
27 making a more quantitative comparison to differences in solute concentrations as a result of natural  
28 heterogeneity as suggested by Reviewer # 2.

29

30 **Furthermore, the use of groundwater and cave dripwater monitoring in another cave used as a**  
31 **pre-fire baseline is not compelling nor necessary, as the authors should be focusing on the**  
32 **response of dripwater to the recovery of the ecosystem following the fire, not to the fire itself.**  
33 **That is, the authors argue that a progressive decrease elemental concentrations reflects a gradual**  
34 **decrease in transpiration due to tree death - however, a fire would result in a dramatic,**

35 **instantaneous decrease in transpiration not a prolonged, multi-year response. It is recommended**  
36 **that the authors switch their perspective from "response to the fire" to "response to the recovery**  
37 **from the fire".**

38 We appreciate and acknowledge these comments. Regarding the use of data from another site as a  
39 pre-fire baseline, reviewer 2 and 3 recommend putting our site into context. We will adjust the  
40 wording in our revised text to reflect the 'response to the recovery from the fire'.

41 **L72 – is there a way to quantify “intense wildfire” and put it into context of the range of wild fires**  
42 **in the region or historical time interval? This seems pertinent to understanding the magnitude of**  
43 **the event when thinking about its implications for interpreting past events from cave deposits.**

44 The wildfire burnt 1200ha and was of a high enough intensity to calcine and cause fracturing of the  
45 limestone at the cave's entrance. We will add these details in the revised draft.

46 **L193 – What is 3 years based on? Another study? A best guess?**

47 The three years is an informed maximum water residence time based on the depth of the cave, the  
48 response time observed at a greater depth at Golgotha cave (Mahmud et al., 2016) and the isotope  
49 variability. We will make this clearer in the revised text.

50 **L201-205 – It seems that the authors tuned parameters of a forward model to best match modeled**  
51 **and observed drip hydrology. What observational window/interval was used? 2005-2011? I**  
52 **assume that the model was tuned to observations at Yonderup, not Golgotha, correct? Is it**  
53 **possible that multiple parameterization schemes could result in similar comparison between**  
54 **observations and data?**

55 We used the observational time window of 2005 to 2011. However, since our seepage reservoir  
56 requires a minimum 10-month residence time for year-round flow, we add monthly average rainfall  
57 data from 2003 to 2005 as a 'warm up' period for the model to avoid edge effects. But only the  
58 observation time window between 2005 and 2011 is shown. Yes, the model was tuned to  
59 observations at the studied cave site (Yonderup). We will make this clearer in the revised text.

60 The input parameters for the model are few (rainfall isotopic composition and the bedrock flow  
61 thresholds). And using numerous flow scenarios we show the full range of possibilities. We find that  
62 no hydro-climatic scenarios can explain our observed  $\delta^{18}\text{O}$ . Thus the model vs data offset suggests  
63 another factor is affecting  $\delta^{18}\text{O}$ . We attribute the higher  $\delta^{18}\text{O}$  values in our data to increased  
64 evaporation conditions post-fire. We will re-write this section in our revised manuscript in and add  
65 this detail to make this point clearer.

66

67 **L240 – Does a bivariate plot of  $\delta\text{H}$  vs  $\delta^{18}\text{O}$  support evaporative enrichment of these dripwaters?**  
68 **I.e., trend off the local meteoric water line with lower slope?**

69 Yes. Firstly, we would like to note evaporative enrichment that occurs in a high humidity  
70 environment (> 95% relative humidity) will occur with a similar slope to the local meteoric water line  
71 (LMWL), and this has been observed in cave environments (e.g. Cuthbert et al 2014). Cuthbert et al  
72 (2014) also characterized the scenarios which would result in offset from the LMWL and we have

73 based our interpretation on these. In our case, the bivariate plot (Fig.5) shows that the least squares  
74 regression (LSR) for cave dripwater falls within the standard error ( $\pm 0.45\%$ ) of the slope for the  
75 LMWL (weighted LSR), but with drip water isotopic composition shifted towards higher  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$   
76 on the LMWL. This falls under a *type 1* scenario suggested by Cuthbert et al., (2014). In the *type 1*  
77 scenario drip  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  do not deviate from the LMWL but are relatively enriched. The *type 1*  
78 *scenario* suggested by Cuthbert et al., supports our case for near-surface evaporation occurring in a  
79 humid near-surface environment (> 95% relative humidity).

80 **L270 – What does “it” refer to?**

81 Here ‘it’ refers to observed drip water composition. We have made the appropriate changes in text.

82 **L329-331, 338-340 – A fire is an abrupt event that likely decimated vegetation instantaneously.**  
83 **How can this be reconciled with a gradual trend of increasing dilution? If the fire resulted in a**  
84 **shut-down of transpiration, a flushing of transpiration-concentrated poor water might be**  
85 **expected followed by relatively dilute concentrations until vegetation reestablished and**  
86 **transpiration lead began to concentration waters again. This might explain a gradual increase in**  
87 **solute concentrations (as seen at Site 2a and noted in L340-342), but not decrease. However,**  
88 **increasing solute concentrations with increased transpiration assumes that vegetative nutrient**  
89 **uptake is reliable relative to concentration due to transpiration.**

90 Yes, a fire would lead to a dramatic reduction in transpiration above a site and we do agree that at  
91 Site 2a we see a vegetation recovery response post-fire. However, at site 1a a transpiration-  
92 reduction response is not necessarily reflected immediately after the fire. This can be due to a  
93 number of reasons; firstly, it simply takes time for water to reach the cave as the soil moisture deficit  
94 would have to be overcome. Second, many of the solutes would be released from the ash, so their  
95 concentrations depend on the dissolution of the elements from the ash over time. It is likely the  
96 decrease in concentrations are reflecting both a return of element concentrations to values without  
97 the influence of the tuart tree and the diminishing leaching of elements from the ash as suggested  
98 by reviewer 2. We will make the appropriate changes to shift our argument and add this explanation  
99 to our discussion in the revised draft.

100

101 **L332 – Is Golgotha Cave further inland, and therefore have less aerosol Cl deposition?**

102 Both caves are ~5 km from the coast and we have added this detail to the text. So, based on location  
103 alone, both are likely to have a similar amount of aerosol Cl deposition. However, variations in  
104 vegetation density between the coastline and the site could also influence Cl compositions to some  
105 extent. We add this to the revised text so the reader is aware.

106 **L346 – How would an increase in surface evaporation induce PCP?**

107 A higher rate of surface evaporation creates longer water-rock interaction times which is ideal for  
108 PCP. This is further detailed in Fairchild et al. 2000, which is referenced in the text. We will re-write  
109 this to further clarify the appropriate references in the text.

110 **L349 – Secession of microbial and root respiration should be abrupt and coincident with the fire,**  
111 **not a gradual signal.**

112 We agree and have deleted this.

113 **L364-386 – This discussion seems highly speculative with very limited constraints on the proposed**  
114 **interpretation.**

115 We argue that at Site 1a, elevated concentrations of  $\text{SO}_4$  and K are being maintained by the  
116 abundance of above-ground biomass ash – sourced from the tuart tree. While, Cl and other solutes  
117 such as Mg, Sr and Ca at Site 1a, are reflecting dilution or a decrease in leaching of these elements  
118 post-fire, as suggested by reviewer #2 (see comment L333 – 334). Grove et al., 1986 found post-fire  
119 soils in this southwest Australia region to contain 23% more S and 16% more K than pre-fire soils, up  
120 to 1 year after the fire, as a result of the ash deposition of the overlying biomass. Given the  
121 abundance of biomass above our site (the tuart tree), we attribute the high  $\text{SO}_4$  and K  
122 concentrations in post-fire soils from the burnt tree to be maintaining high  $\text{SO}_4$  and K concentrations  
123 in dripwater. However, we acknowledge the reviewer’s comments and will re-write this section and  
124 reduce the amount of speculation by quantifying the difference between  $\text{SO}_4$  and K concentrations  
125 in comparison to other solutes at our sites, as suggested by reviewer #2 (see comment Lines 430 –  
126 431).

127 **L397-398 – How does model dripwater  $\delta^{18}\text{O}$  agree with observed  $\delta^{18}\text{O}$ ? They look to be**  
128 **substantially offset. Additionally, it is not clear how simulation of dripwater  $\delta^{18}\text{O}$  supports the**  
129 **interpretation that tree death gradually reduced transpiration resulting in gradual decrease in**  
130 **solute concentrations.**

131 Yes, we agree with Reviewer 1 that the observed dripwater  $\delta^{18}\text{O}$  and modelled  $\delta^{18}\text{O}$  are distinctly  
132 offset after the fire. This offset is underpinning our argument: that there was increased evaporation  
133 after the fire due to the lack of shading (see reply to reviewer 1, regarding L201-205). We will make  
134 this clearer in the text at L258-269. We agree that the statement in L397-398 is inconsistent with this  
135 interpretation and thank the reviewer for pointing this out. We will delete this statement in a  
136 revised manuscript.

137 **L399 – A bit of context would be helpful – does this apply to a single tree? A stand of trees? At this**  
138 **(fire-affected) site? In this region?**

139 This is specific to a full grown tuart tree that is native to the coastal dune systems of southwest  
140 Australia. Drake et al., (2011) tested the transpiration potential of seedlings and full grown trees  
141 from Yalgorup National Park in different seasons to determine the pressure gradient created by their  
142 roots. We will add this detail in the revised text.

143 **L400 – it is not clear what “this potential” refers to, nor is it clear what gradient (i.e., from where**  
144 **to where) is being referred to. . . . How does a hydraulic gradient maintain high Cl concentrations?**

145 Our use of ‘this potential’ refers to the ‘potential energy’ or ability of tuart tree to extract water (by  
146 transpiration). This generates a pressure gradient between the soil and the tuart tree. The gradient  
147 transports both water and nutrients towards the tree, but not all are taken up and as a consequence  
148 are left behind in the soil solution, in particular salts like Cl. We will re-write and further clarify this in  
149 the revised manuscript.

150

151 **L408-410 – It is not clear how a fire in Feb 2005 results in a sharp increase in evaporation in the**  
152 **Mar 2007.**

153 As mentioned in the text, since the cumulative water balance (CWB) is positive we don't yet observe  
154 the effect of increased evaporation. Once CWB becomes negative ( $P < ET$ ) we see  $\delta^{18}O$  rise sharply.  
155 We believe the decrease in shading post-fire coupled with the  $P < ET$  condition was enough to drive  
156 this sharp increase in  $\delta^{18}O$  and eventually exhaust the reservoir feeding the drip.

157 **L419 –  $\delta^{18}O$  reflecting more evaporation but lower solute concentrations tracking less**  
158 **concentration (more dilution) due to less (evap)transpiration does not make the most compelling**  
159 **argument.**

160 Forward modelling of dripwater  $\delta^{18}O$  compared with the observations provides firm evidence that  
161 the relatively higher observed  $\delta^{18}O$  at both sites can only be explained by increased evaporation (see  
162 reply to reviewer 1, regarding L201-205). With this constraint, the simplest explanation for the  
163 declining trend in most of the solutes is a decrease in tree water use at Site 1a and removal from the  
164 surface and subsurface (leaching) post-fire (suggested by reviewer #2; see comment L333 – 334).  
165 The dominating effect of tree water-use on dripwater solute concentrations, post-fire is reported in  
166 Treble et al., (in press - accepted 8/4/16) which, using a mass-balance approach, quantifies the role  
167 transpiration from regrowth on solute concentrations at Golgotha Cave. We will rewrite this  
168 explanation in the revised text section to make it clearer for the reader.

169 **L243-276 – It is not clear how modeling  $\delta^{18}O$  contributes to this study. The results are not well**  
170 **integrated into the interpretations, and it is not clear why a model tuned to the drip hydrology**  
171 **does so poorly in accounting for observed dripwater  $\delta^{18}O$ . Is this meant to support the**  
172 **interpretation that evaporation of infiltrating water occurs somewhere between the surface and**  
173 **drip site, and that evaporation might also play a role in dictating solute compositions? If so, this**  
174 **does not come across clearly.**

175 **And**

176 **L453 – It is not clear how/why the modeled  $\delta^{18}O$  represents a control (no fire) scenario.**

177 By forward modelling  $\delta^{18}O$  we aim to predict the composition of  $\delta^{18}O$  in dripwater using rainfall  
178 isotopic composition and bedrock flow parameters. The model represents hydro-climatic influences  
179 only. It simulates the probable drip  $\delta^{18}O$  based on rainfall  $\delta^{18}O$  as an input and parameters  
180 representing the physical hydrology, as outlined in the m/s. It therefore acts as a control scenario in  
181 which only hydro-climatic influences are modifying drip  $\delta^{18}O$ . Hence, the fact that our drip  $\delta^{18}O$  are  
182 higher than that simulated, supports our argument that a post-fire increase in surface/near-surface  
183 evaporation is driving drip  $\delta^{18}O$ . See reply to Reviewer 1, L201-205. For further clarity, we will re-edit  
184 the modelling text with these comments in mind.

185 **L436 – use of groundwater and nearby cave dripwater as pre-fire conditions is not compelling, nor**  
186 **necessary. The argument that the recovery of an ecosystem after a disturbance is potentially**  
187 **reflected in cave dripwater is compelling on its own.**

188 Reviewer #2 suggests it is necessary to compare with regional sites. In our revisions we will try to  
189 reconcile comments by both reviewers.

190 **L441-444 – The case supporting these statements is not compelling. L446-447 – How has the**  
191 **vegetation forcing been delineated from the climate (i.e., CWB) signal?**

192 See reply above for L453 with regards to  $\delta^{18}\text{O}$ . Our interpretation that a response to post-fire  
193 conditions is the dominating response is further supported by our solute data. The fact that most of  
194 our solutes display contrasting behaviour at our two sites demonstrates that highly localised factors,  
195 not climate, are dominating. Cl is a conservative ion and hence is primarily driven by  
196 dilution/evaporation. We interpret Cl's decline post-fire at Site 1a to reflect dilution of the water  
197 store that the dead tree previously exploited, and a decrease in leaching of post-fire as suggested by  
198 reviewer #2 (see comment L333 – 334). In contrast, Cl increases at Site 2a, sympathetically with  
199  $\delta^{18}\text{O}$ , consistent with an increased evaporative demand on shallow water stores driven by post-fire  
200 reduction in shading and reduced albedo. We will clarify our concluding remarks to strengthen our  
201 arguments in a revised manuscript.

202 **L448 – Increasing K and SO<sub>4</sub> trends are not obvious from Fig. 3, and why would the degree of**  
203 **leaching increase with time from fire? It might be expected that there would be a pulse of K and**  
204 **SO<sub>4</sub> following the fire, then leaching would decline after the initial pulse.**

205 At Site 1a, K and SO<sub>4</sub> do not decrease like other solutes due to dilution and removal from the surface  
206 and subsurface through leaching, as above. The fact that K and SO<sub>4</sub> appear to be unaffected suggests  
207 that their flux has increased, counteracting the dilution and source decline, observed from other  
208 solutes. K and SO<sub>4</sub>, have been found to have higher concentrations, in post-fire soils (Grove et al.,  
209 1986), up to one-year post-fire, as result of over-lying biomass ash deposition. Thus our  
210 interpretation for the sustained, high concentrations of K and SO<sub>4</sub> is that ash from the burnt tuart  
211 tree has increased the K and SO<sub>4</sub> flux. Further the difference in dissolution rates could also explain  
212 the rate at which these solutes are leached into drip water in comparison to others, as mentioned in  
213 our reply to Reviewer #1 comment L329 – 331, 338 – 340.

214 At Site 2a, we see that SO<sub>4</sub> and K follow the concentration trend driven by the dominant forcing,  
215 evaporation, at this site (see previous comment) and do not show an ash signal. This due to the site  
216 having less biomass available to be converted to soluble ash, and thus it is a less dominant forcing.  
217 We will make sure that we quantify the differences between K and SO<sub>4</sub> vs. other solutes at our sites  
218 (see reviewer #2 comment L 430 – L431) and outline and possible scenarios for these differences in  
219 our revised manuscript.

220 **L449 – This is vague and not all that helpful of a conclusion.**

221 **And**

222 **L450 – No evidence to support this was presented in this manuscript so it is a bit odd to present as**  
223 **a conclusion.**

224 Thank you for pointing out the weaknesses in the conclusions. We will re-write the conclusion in a  
225 more quantitative fashion summarizing the findings of our study.

226

227 **Reviewer #2**

228 This manuscript presents a high quality cave monitoring study from 2005-2011. The authors  
229 collected a suite of analyses to better understand cave and climate processes, and ultimately how  
230 these signals are incorporated into speleothems.

231 Thank you for your kind comments.

232 **Plot the raw rainfall  $\delta^{18}\text{O}$  time series in Figure 3c along with the forward model dripwater  $\delta^{18}\text{O}$ . It**  
233 **will be interesting to see how the model alters the above-ground signal.**

234 We were very fortunate to be able to access these data for the purpose of performing the forward  
235 model calculations. These rainfall  $\delta^{18}\text{O}$  data are unpublished data created for the IAEA/GNIP  
236 program by ANSTO non-coauthors who are acknowledged in the Acknowledgements. As a separate  
237 paper on these data is currently in preparation, the owners of these data, understandably have  
238 declined to have these data published here in the time series format that is being requested.  
239 However, we do have permission to provide Reviewer 2 with a version of Figure 3 that does contain  
240 these data to satisfy their query.

241 **Why is the modelled dripwater  $\delta^{18}\text{O}$  so much smoother than the dripwater data? Is the temporal**  
242 **resolution of the rainfall collection too low? Or is the rainfall data being smoothed too much by**  
243 **the model?**

244 The rainfall data is monthly, as stated in the manuscript, so it is not the temporal resolution of the  
245 rainfall data. The model output had been smoothed to reflect a typical stalagmite sampling  
246 resolution. This smoothing has now been removed.

247 **Lines 262-264: The slope calculations are subject to serious edge effects. For example, the**  
248 **modelled dripwater  $\delta^{18}\text{O}$  has an inflection point early in the record in 2006. This is not observed in**  
249 **the dripwater data. Perhaps you could use a bootstrap to calculate the error on the slope, but**  
250 **given the high density of points in blue curve of Figure 3c leaving out one or two or even three**  
251 **points probably will not change the slope too much. But it is this inflection point early in the**  
252 **modelled dripwater record – that is not in the actual data that is causing the very different slopes.**  
253 **The trend over both look very similar from 2007-2011. Also, the slope at for site1a should be**  
254 **compared with the slope of the modelled  $\delta^{18}\text{O}$  over the same time period: ~2005-2007.**  
255 **Therefore, I'm not convinced that evaporation is driving such a large difference in the dripwater**  
256  **$\delta^{18}\text{O}$ .**

257 Given the slope calculation are prone to edge effects, we will remove the slope calculations. We  
258 believe that the 1-2‰ offset between modelled  $\delta^{18}\text{O}$  and observed  $\delta^{18}\text{O}$  is compelling enough to  
259 show  $\delta^{18}\text{O}$  is being driven by increased evaporation. This range in isotopic enrichment is consistent  
260 with isotopic enrichment due to evaporation in caves from semi-arid regions (Rutledge et al.,  
261 Markowska et al., 2014). Further the shallow depth of this cave (4 m) as well as the reduced shading  
262 and change in albedo following the fire, makes it even more prone to evaporative effects.

263

264 **Plot the Mg/Ca and Sr/Ca time series. I cannot do the calculation in my head using data from**  
265 **Figure 3e, f, and g. Do they co-vary in time? Or is the  $\ln(\text{Mg}/\text{Ca})$  vs.  $\ln(\text{Sr}/\text{Ca})$  relationship driven by**  
266 **changes that are not coeval?**

267 Yes, Mg/Ca and Sr/Ca time series co-vary in time we will add this as a figure in the revised  
268 manuscript. This coupled with the  $\ln(\text{Sr}/\text{Ca})$  vs  $\ln(\text{Mg}/\text{Ca})$  slopes (Fig. 6) and the diagnostic range (a  
269 slope  $\pm 0.88$ ) given by Sinclair et al., (2011) support our case for PCP. We will add these details in the  
270 revised manuscript.

271 **Mark on Figure 3 when the fire occurred**

272 We will add this in our next draft.

273 **Line 331: how do pre-fire Cl values compare between caves?**

274 Unfortunately, we do not have pre-fire Cl values for the studied cave site (Yonderup).

275 **Lines 333-334: without values from before the fire, it's hard to discern exactly what is the cave**  
276 **response to the fire. Could it not be that at Site 1a there was a spike of Cl, Mg, Ca, and Sr after the**  
277 **fire due to dissolving ash (Lines 357-358)? Then the downward trend would be the slow removal of**  
278 **those from the surface and sub-surface. Without data from before the fire to establish a baseline,**  
279 **most of the arguments about what the fire did are too speculative and unsupported.**

280 It is possible that at Site 1a there may have been a prior peak in Cl, Mg, Ca and Sr. However, any  
281 peak would have been a rapid increase prior to monitoring as they are highly soluble and CWB was  
282 positive. Solutes would also be affected by a direct concentration effect created as the soil and  
283 vadose zone water was heated by the fire. But as suggested by the reviewer, the trends we see here,  
284 post-fire, in these solutes, could be reflecting both the slow removal of these nutrients from the  
285 surface and subsurface, and reflecting a reduction in tree water-use. We will add this other possible  
286 cause for the decline of solutes at Site 1a to our argument in the revised text.

287

288 **Lines 430-431: The differences between Yonderup and Golgotha should be quantified. Listing**  
289 **many values in the table does not support the differences quoted in the text. There are 10**  
290 **differences to calculate 2 sites at one cave, 5 sites at the other. From the population of 10**  
291 **differences, one may then calculate the median, mean and standard deviation. Then it will be**  
292 **clear 1) how much the drip chemistry differs and 2) how much variability exists just heterogeneous**  
293 **environments, which means to say \*not\* fire-related.**

294 We will calculate the difference suggested to further quantify our interpretation and pre-empt  
295 speculation. We think a better way to describe the natural heterogeneity is by the observed  
296 differences between sites at each cave (e.g. Site 1A vs Site 2A). As this will show how each site  
297 responds over time.

298

299 **Reviewer #3**

300 **Monitoring studies such as this provide insight into karst and speleothem processes and as such**  
301 **are valuable data sets for the scientific community. Understanding the response of karst systems**  
302 **to fire is potentially a great asset for paleoclimate interpretations, as it may be a new way to track**



303 **past limitations in water availability. Application of a forward model to aid in the interpretation of**  
304 **the monitoring data is also a valuable aspect of this study.**

305 We thank the reviewer for these kind comments.

306 **However, the manuscript lacks rigor in the presentation and interpretation of the results as well as**  
307 **in the overall presentation. For example, the abstract provides no results that would support any**  
308 **of the conclusions.**

309 We will add more details from our finding in the abstract. This includes specifics on how we utilized  
310 the forward model to demonstrate that drip  $\delta^{18}\text{O}$  was 1-2 per mil higher than predicted by hydro-  
311 climatic processes supporting enhanced evaporation post-fire, further supported by the Cl data; that  
312 the distinct spatial and temporal differences between site  $\delta^{18}\text{O}$  and solutes rules out that both sites  
313 could be climatically controlled. We will also detail the potential role of the death of the tree in  
314 explaining the post-fire observations at dripwater site 1a. Further we will also detail the use of ash-  
315 derived  $\text{SO}_4$  and K that are leached in dripwater.

316 **The application to speleothem studies discussion in the Conclusions should be its own section and**  
317 **come before the Conclusions. To be useful to other researchers, it would be helpful to provide**  
318 **exposition of the subtleties that would be involved in such applications. For example, 1) how**  
319 **many different proxies would be needed to delineate fire influence, given that  $\delta^{18}\text{O}$  in**  
320 **speleothems is affected by many processes, including in-cave processes not related to climate or**  
321 **vegetation disturbance?; 2) How would a researcher delineate between drip sites impacted by fire**  
322 **but without a tree in the recharge zone for the drip, vs. climate processes, vs. in-cave processes?**

323 We thank the reviewer for this constructive comment and will add an ‘application to speleothems’  
324 section discussing the information that can be used out of our study to find a proxy for fire. We will  
325 bolster this section by discussing recently published studies from Golgotha Cave (Treble et al 2015  
326 and Treble et al GCA – accepted) that further strengthen the interpretation presented here.

327 We will specifically address the suggestions that the reviewer has made above. We believe a multi-  
328 proxy approach that uses a suite of soil and bedrock sourced elements and  $\delta^{18}\text{O}$  should be used as  
329 proxies to analyse a fire signal.

330 We recommend multivariate statistical analysis should be used to separate climate/seasonal forcing  
331 from a soil/ vegetative forcing. Here, we would expect the local soil/vegetative forcing to preserve  
332 the impact of a fire on stalagmite composition, in trace elements like S, K and P. All of which have  
333 been found to increase in abundance in post-fire soils, due to burnt biomass-ash. Further, given fire  
334 can increase discharge by reducing tree water-use, it is also worth looking at colloid associated  
335 metals, such as Al, Fe and Cu which could show immediate spikes after the timing of the fire.

336 We will also discuss the importance of site selection i.e. the depth of the cave, the overlying  
337 vegetation and the climate setting, as not all sites will be affected by fire. Our study helps to place  
338 constraints on the search for a paleo-fire signal in caves and also informs other monitoring studies of  
339 this nature.

340 **Line 28. How is the analysis unique? What specifically is unique about this analysis?**

341 **AND**

342 **75. see comment on abstract re 'unique'; 'analysis' is used twice in this sentence.**

343 The analysis is one of the first monitoring studies conducted in a post-fire regime that seeks to  
344 identify the nutrient dynamics and effects of wildfire on dripwater composition. We will make this  
345 point clearer in the revised copy.

346 **29-31. Run-on and awkward sentence.**

347 We will re-write this sentence.

348 **33. This is the most significant claim of the study. Explain how the  $\delta^{18}\text{O}$ , chemistry, indicate and**  
349 **support this claim. There is nothing in the abstract that provides any hint of what the results of the**  
350 **study are.**

351 **AND**

352 **35. How so? What are the results that indicate this?**

353 **AND**

354 **36. 'here we open a new avenue for speleothem science': Without answers to the above**  
355 **questions, this statement is not supported.**

356 These points will be covered by the modifications suggested in the first reply to Reviewer 3.

357 **46. 'local environmental factors' such as. . . ?**

358 Evaporation, transpiration and leaching from biomass-sourced ash. This will be inserted in this  
359 sentence for clarification in the revised copy.

360 **48-49. Monitoring studies have also focused on controls on calcite growth, the role of CO<sub>2</sub>,**  
361 **respiration, and other factors and processes (Wong et al. 2011, GCA; Breecker et al. 2012, GCA)**

362 We will add these references in the revised copy.

363 **51. AET – define acronyms upon first use. AET is not defined until figure 3**

364 **AND**

365 **137. ANSTO: define acronyms on first use.**

366 All acronyms will be explained upon first use in the revised manuscript.

367 **52. Is this 'the exception' or the rule? There are many more monitoring studies in water limited**  
368 **regions, including those cited above**

369 Will remove 'the exception'.

370 **62. Wong reference – this study is not an example of vegetation loss due to fire.**

371 We will more clearly state this study is an example of vegetation loss not vegetation loss due to fire.

372 **74. give cave name and location**

373 We will add the cave name (Golgotha Cave) and location 36.10 deg S, 115.05 deg E to the text and  
374 location figure of the revised manuscript.

375 **85-6. Give length of time comprised by temp record; give geologic age of Tamala Limestone. 113.**  
376 **'heterogeneous' in what way?**

377 We will add the length of time from which we obtained the averages for the temp and rainfall  
378 records. We will give the age (Quaternary) of the Tamala limestone. Further we will also add that the  
379 soil is heterogeneous in thickness and spatial coverage above our sites.

380 **114-5. make into two sentences**

381 We will separate this into two sentences.

382 **123-5. state the length of the collection interval.**

383 The length of the collection interval varied over time between 2 – 4 times a month we will detail this  
384 in the new revised draft. Bi-monthly is the best way to describe it.

385 **129, 132. State location of instruments. Is the ICP-AES a spectroscope or spectrometer? List it as**  
386 **such.**

387 Our cation concentrations were measured using a Thermo-Fischer inductively coupled plasma-  
388 atomic emission spectrometer (ICP-AES) at the Australian Nuclear Science and Technology  
389 Organization facility in Lucas Heights. We will add this to our revised draft.

390 **145. Was this drip size verified for the site? Why refer to an experimental study if you have both**  
391 **drip speed and volume/time?**

392 When drips were fast enough to be measured we recorded the time intervals between the drips.  
393 When the drip was too slow we left a bottle and measured the volume of the discharge over a  
394 particular period of time. Thus in order to represent the data in common units we needed to use this  
395 drip volume in order to convert all our discharge data into volume data. The calculations have been  
396 provided in the supplementary info excel sheet. We will clarify this in the new draft.

397 **151-4. What does 'this' refer to? RMC? P-AET? This is a run-on sentence and hard to follow.**

398 Here 'this' refers to the residual mass curve (RMC). We will detail this in the revised draft.

399 **157 what is 'FWE'? 160-161. define terms upon first use, which is much earlier than here**

400 FWE is the acronym for AET in the Australian Water Availability Project data sheets. We have  
401 clarified in this sentence that this is the parameter that we used since there are multiple parameters  
402 available to estimate evaporation terms from this dataset.

403 **170. How far away? Yanchep is not on the map FIG 1.**

404 Yonderup Cave which is shown on the map is in Yanchep NP. We will make this clearer in text.

405 **178-9. Only one year different from the study's fire. Was the burn less intense? Less destructive?**  
406 **How did it effect the lower and middle understory growth?**

407 The prescribed burn at Golgotha Cave was much less intense and much more spatially  
408 heterogeneous than the fire that is reported in our study. We will make this clearer in the text and  
409 refer to the manuscript currently under review for the impact of this fire on Golgotha Cave.

410 **Run-on sentence 188. the 'latter' what? Several different things are developed in the previous**  
411 **sentence.**

412 We use the term 'latter' to refer to seepage and fracture thresholds, we will make this clearer in the  
413 text.

414 **192-4. confusing sentence, try restating as: 'Storage time for water that enters the seepage**  
415 **reservoir is modeled as a Gaussian distribution. This time is set as a maximum age of 3 years to**  
416 **reflect the shallow depth of our cave system. The model allows for the mean and standard**  
417 **deviation to be specified for these functions.'** Furthermore, what is this based on? It's more the  
418 **nature of the flowpath through the vadose zone than it is the thickness of the vadose zone that**  
419 **will determine storage time. It seems that this value is adjusted later based on observations at the**  
420 **cave, but need to have a basis for this starting point.**

421 We will readjust the sentence as suggested. These sites have been found to be predominantly  
422 controlled by seepage or matrix flow due to the calcarenite type lithology at our sites (Mahmud et  
423 al., 2015). Further, capillary barriers have been found to affect hydrology at these sites which delay  
424 downward movement of water. With this knowledge and the shallow depth of the cave we set a 3-  
425 year maximum limit.

426 **198. What is 'a karst store'?**

427 By 'karst store' we were referring to water storage in the overlying bedrock. We will make this  
428 clearer in our next section

429 **217. Why 'soil water availability' here, and 'water availability' above?**

430 We will change 'soil water availability' to water availability to make it more consistent.

431 **224. 'increases' should be past tense**

432 We will change to 'increased'.

433 **236-9. Run-on sentence 239. Add (DJF) for northern hemisphere readers**

434 We will add 'DJF' after 'summer' and 'JJA' after 'winter' for northern hemisphere readers and divide  
435 the sentence with punctuation.

436 **240-1. when first defining 'thresholds' also define it in context of 'seepage thresholds'**

437 We will re-write our definition of thresholds by defining 'thresholds' in the context of seepage and  
438 fracture thresholds.

439 **243. Unclear. Try rewording: 'We attempted to model dripwater  $\delta^{18}\text{O}$  that matched the measured**  
440 **drip water values based on using the rainfall isotopic data set as our input?'**

441 We will reword this sentence, as suggested, to make it clearer.

442 **244. Is 30 mm rainfall? State this. Could other variables in the model cause the shutoff?**

443 Yes, 30 mm is the P – AET threshold. Given P – AET is the only input in the model and the only thing  
444 limiting the water from entering the seepage reservoir is the threshold. Thus it is unlikely that other  
445 variable in the model would cause the shutoff.

446 **247-251. Run-on. 253. is this the mean or the weighted mean?**

447 It is just the mean. We will make this clearer and fix the Run-on sentence.

448 **265. What makes a given model 'meaningful'?**

449 First, meaningful models referred to modeled scenarios that maintain full year round drip flow to  
450 match observed drip flow. This eliminated models with a seepage threshold of 40 mm or greater.  
451 And given not all rainfall enters the seepage reservoir, some threshold must exist, so we set a  
452 minimum seepage threshold of 10 mm. Second, models that reflect the minimum residence time (10  
453 months) to maintain full year round flow at our t our sites were chosen. Under these bedrock  
454 constraints we tested all possible scenarios from predominantly fracture flow (10 – 15 mm) to all  
455 seepage flow (10 – 1000 mm). We found that fracture dominated flow showed more variability than  
456 seepage dominated flow. However despite the variability, none of the models that we ran fit the  
457 observed  $\delta^{18}\text{O}$  at both sites. It is clear,  $\delta^{18}\text{O}$  at both sites is offset from modelled hydro-climatic  $\delta^{18}\text{O}$   
458 We will provide brief clarification on this in the revised manuscript.

459 **273-4. try rewording to: 'This is similar to a type 1 scenario defined by Cuthbert...'**

460 We will reword this sentence as suggested.

461 **288. Interpretation, not Results, belongs in Discussion 291-6. difficult to follow, rewrite. 300.**  
462 **Arguments should be in Discussion, not Results.**

463 We will move L288 and L297-301 to the Discussion.

464 **316. For sure evaporation will affect chloride in water in the same way dilution and mixing will,**  
465 **not potentially.**

466 We will change 'and potentially' to 'as well as'.

467 **348. This figure shows no diagnostic model, even though the caption states that both sites fall**  
468 **within the model, the figure doesn't show it.**

469 We will clarify this sentence by replacing 'shown by the diagnostic model' with 'as evidenced by the  
470 agreement of the  $\ln(\text{Sr}/\text{Ca})$  vs  $\ln(\text{Mg}/\text{Ca})$  slopes in our data with the diagnostic range (a correlation  
471 co-efficient  $\pm 0.88$ ) given by Sinclair et al., (2011)'.

472 **358. Unclear. If 3 increases Ca in dripwater, why would dripwater not reflect the increase? Is the**  
473 **signal from 1 and 2 so large as to make 3 background? Or is it simply sequence of events? Run-on**  
474 **sentence.**

475 Here we were aiming to put forward a possible number of scenarios as it is hard to constrain. It is  
476 likely 1 and 2 are more important than 3 which we will clarify in the revised manuscript.

477 **389. Reference figure earlier in text when describing site and processes.**

478 We will reference this figure earlier in the text as suggested.

479 **450-2, and 455. This is new information, more appropriate to include it prior to Conclusions**

480 We will move this information to the Discussion.

481 **456-7. Further, that a fire signal may be much more subtle in a speleothem if the fire impacted**  
482 **drip sites without trees above them.**

483 Thank you. We will add this point to our Discussion.

484 **458. Growth rate not covered in text, this is more new information that is more appropriate to**  
485 **include prior to conclusions. See general comment above about.**

486 We will move this information to the Discussion.

487 **Fig. 1. Where is Yanchep National Park located? Where does the inset sit on the map of Australia?**

488 Yonderup Cave is situated within Yanchep National Park. We will make this clearer in the figure in  
489 our revised manuscript

490 **Fig. 2. The same data are presented in Figs. 2 and 6. Only one of these is needed. If the authors are**  
491 **going to employ the Sinclair graphical model, then it would be Fig. 6. However, the discussion of**  
492 **Fig. 6 in the text and caption claim to show that the Sinclair PCP model holds and can account for**  
493 **the trends, yet there is no text or addition to the figure that supports this claim. In Fig. 2, state**  
494 **that the values plotted are for drip water. The interpretation given that both (each) site has an**  
495 **independent flow path is not explained. What specifically indicates this? Different starting points?**  
496 **Distinct slopes?**

497 We will replace Figure 2 with Figure 6 as it is diagnostic for PCP as suggested. We will quote the  
498 regression equations for both the normal and log-transformed data. We will clarify in the text that  
499 the difference between regressions calculated for each site (using either method) indicate  
500 independent drip paths as suggested by the reviewer. We will also add a  $\ln \text{Mg/Ca}$  and  $\ln \text{Sr/Ca}$  time  
501 series to make our case for PCP more clear.

502

503 **Fig. 3B. It's difficult to see Site 2a measured values. Since 'Est Dis', 'Meas Dis' used, also label sites**  
504 **as 'Calc' for clarity.**

505 We will make this adjustment to Fig 3.

506 **641. concentrations appear to be quadruple, not 'double' that of Site 2a.**

507 The concentrations are x4 for K and x2 to x3 for SO<sub>4</sub>. We will make this adjustment in the text.

508 **643-4. differentiation of temporal trends between 1a and 2a: I disagree with trying to make a**  
509 **difference here, as the data do not support this. Both sites show nearly the same slope of increase**  
510 **and the trends are obscured by gaps in the time series.**

511 We will delete this sentence.

512 **Fig. 4. Explain which thresholds. Why would the 10-75 mm threshold have a lower response than**  
513 **the 10-15 mm model? Why not present time series for the proposed fire-sensitive ions such as SO<sub>4</sub>**  
514 **and P? Show where the fire event occurred in relation to the time series.**

515 We thank the reviewer for picking up on this. We found that one of our input rainfall isotopic data  
516 points for the date for 07/2009 was + 5.16 instead of -5.16, this was offsetting the model for 15 –  
517 100, 10-15, 40 – 100 and 10 – 1000 mm thresholds which had the (+) instead of the (–) value while  
518 the 10 – 75mm had the correct negative value. We have also eliminated the smoothing in the model  
519 as suggested by reviewer #2. This figure is provided below. We will also add the time series of SO<sub>4</sub>  
520 and K along with an indicator for when the fire occurred to put the time series into context as  
521 requested.

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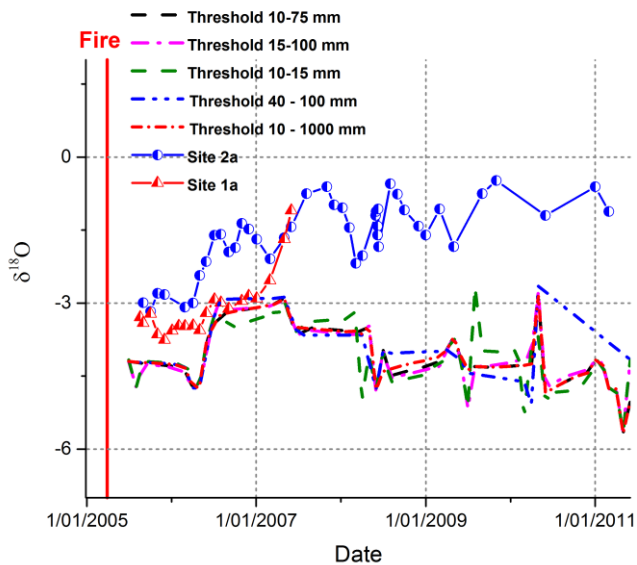
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536 **Fig. 5. Move Mean rainfall to below rainfall. Use color other than red since red is for cave drip info.**  
537 **Is the mean rainfall the mean or weighted mean?**

538 We will move the mean rainfall to below the rainfall and use a color other than red.

539 **Fig. 6. Place both sites on same plot with different symbols, in order to help the reader directly**  
540 **compare them. See comments above on Fig. 2. Presentation of time series for these element**  
541 **ratios would aid in their interpretation and how the processes proposed to account for the**  
542 **variation change with seasons, etc.**

543 We will add Mg/Ca and Sr/Ca time series to aid in the interpretation of processes affecting these  
544 trace elements such as PCP.

545



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579

580 **A post-wildfire response in cave dripwater chemistry**

581

582

583 **Gurinder Nagra<sup>1\*</sup>, Pauline C. Treble<sup>1, 2</sup>, Martin S. Andersen<sup>1</sup>, Ian J. Fairchild<sup>3</sup>,**

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585

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602 **Abstract**

603 Surface disturbances above a cave have the potential to impact cave dripwater discharge,  
604 isotopic composition and solute concentrations, which may subsequently be recorded in the  
605 stalagmites forming from these dripwaters. One such disturbance is wildfire, however the  
606 effects of wildfire on cave chemistry and hydrology remains poorly understood. Using  
607 dripwater data monitored at two sites in a shallow cave, beneath a forest, in southwest  
608 Australia, we provide one of the first cave monitoring studies, conducted in a post-fire  
609 regime, which seeks to identify the effects of wildfire, and post-fire vegetation dynamics, on  
610 dripwater  $\delta^{18}\text{O}$  composition and solute concentrations. We compare our post-wildfire  $\delta^{18}\text{O}$   
611 data with predicted dripwater  $\delta^{18}\text{O}$  using a forward model based on measured hydro-climatic  
612 influences alone. This will help to delineate hydro-climatic and fire-related influences on  
613  $\delta^{18}\text{O}$ . Further we also compare our data with both data from Golgotha Cave - which is in a  
614 similar environment but was not influenced by this particular fire - as well as regional  
615 groundwater chemistry, in an attempt to determine the extent to which wildfire affects  
616 dripwater chemistry. We find in our forested, shallow cave,  $\delta^{18}\text{O}$  is higher after the fire  
617 relative to modelled  $\delta^{18}\text{O}$ . We attribute this to increased evaporation due to reduced albedo  
618 and canopy cover. While the solute response post-fire varied between the two drip sites: at  
619 Site 1a, which had a large tree above it but was lost in the fire, we see a response reflecting  
620 both a reduction in tree-water use and a removal of nutrients (Cl, Mg, Sr and Ca) from the  
621 surface and subsurface. Solute such as  $\text{SO}_4$  and K maintain a high concentration, which we  
622 believe is due to the abundance of above ground ash. At Site 2a, which was covered by  
623 lower-middle storey vegetation, we see a solute response reflecting evaporative concentration  
624 of all studied ions (Cl, Ca, Mg, Sr,  $\text{SO}_4$  and K) similar to the trend in  $\delta^{18}\text{O}$  for this drip site.  
625 We open a new avenue for speleothem science in fire-prone regions, focusing on the  
626 geochemical records of speleothems as potential paleo-fire archives.

**Comment [GN1]:** An additional perspective as suggested by reviewer 2 (comment about incorporating vegetation recovery to the story)

**Comment [GN2]:** We detail how this study is unique in response to Reviewer 3's comment L28 and L75

**Comment [GN3]:** We add a clear description of the role of the forward model early in the paper (and throughout) for the reader as reviewer 1's comments suggested its role needed to be made clearer.

**Comment [GN4]:** Detailed outline of results added in the abstract as suggested by reviewer 3 (Comment L33 to L36). Further some changes in the writing of the manuscript have been made to improve the writing (suggested by Reviewers 1 and 3)

627 **1 Introduction**

628 Caves are observatories that preserve invaluable geochemical archives of past-climates; in  
629 the form of speleothems (stalagmites, stalactites and flowstones). The existing paradigm in  
630 speleothem science has largely focused on establishing paleoclimate proxies in stalagmites  
631 (e.g. McDermott et al., 2001; Treble et al., 2008; Woodhead et al., 2010). While these proxies  
632 are useful for reconstructing paleoclimates, their interpretations may hold a predisposed bias  
633 towards using these proxies as indicators of paleoclimate only.

634 To avoid this bias, we need to consider the sensitivity of these proxies to the effects of local  
635 environmental factors like in our case, fire. This is especially important as incorporating this  
636 perspective may not only be used to correct the climate proxy interpretation, but also yield  
637 novel information about paleo-environments. Paleo-environmental proxies are verified by  
638 conducting process-based in-cave monitoring studies. However, in-cave monitoring has  
639 predominantly focused on understanding the extent to which dripwater  $\delta^{18}\text{O}$  (Lachniet,  
640 2009), dripwater solute concentrations (Fairchild and Treble, 2009), speleothem calcite  
641 growth (Wong et al., 2011) and cave  $\text{CO}_2$  processes (Breecker et al., 2012), are affected by  
642 climate. Further, such studies have largely been restricted to mid to high latitude climate  
643 regions where precipitation (P) is larger than actual evapotranspiration (AET), and climate is  
644 likely to be a major control on dripwater composition.

645 In water-limited regions, dripwater chemistry is influenced to a greater extent by  
646 environmental factors such as evaporation (E) (Pape et al., 2010; Cuthbert et al., 2014;  
647 Rutledge et al., 2014) and transpiration (T), (Tremaine and Froelich, 2013; Treble et al.,  
648 2016). Wildfires, common in water-limited regions, are agents of change than can  
649 dramatically alter evaporation and transpiration rates by destroying vegetation. The potential  
650 impacts of vegetation loss from fire are both short-term and long-term. The short-term

**Comment [G5]:** Addition of an example of a local environmental factor. Response to reviewer 3 (comment L46)

**Comment [GN6]:** References added in response to reviewer 3 comment L 48 to 49)

**Comment [GN7]:** Added more references to demonstrate geographical evidence and broad range of literature that has been read. (Reviewer 3 comment L48-49)

**Comment [GN8]:** All acronyms have been stated on first use. In response to reviewer 3's comments

651 impacts could include: (1) an increase in evaporation rates due to changes in albedo and/or  
652 lack of shading (Silberstein et al., 2013); (2) a reduction in transpiration from reduced tree  
653 water use; (3) a reduction in soil microbial and root CO<sub>2</sub> production (Coleborn et al., 2016);  
654 (4) a decrease in cave CO<sub>2</sub> due to the destruction of vegetation (Wong et al., 2010), which can  
655 influence in-cave prior calcite precipitation (PCP); (5) the addition of plant ash to the soil  
656 profile, increasing concentrations of Ca, K, Mg, and S (Grove et al., 1986; Yusiharni and  
657 Gilkes, 2012a); and (6) altered infiltration patterns (González-Pelayo et al., 2010). While  
658 long-term impacts include: (1) the spatial redistribution of nutrients (Abbott and Burrows,  
659 2003); (2) regrowth impacts on water balance and nutrient flux (Treble et al., 2016); and (3) a  
660 reduction in total soil CO<sub>2</sub> due to the destruction of CO<sub>2</sub> sequestering microbial communities  
661 and plant roots, both significant sources of soil CO<sub>2</sub> (Coleborn et al., 2016). Despite the fact  
662 that wildfires regularly affect water-limited regions, their impacts on δ<sup>18</sup>O and solute  
663 concentrations in cave dripwater have not been reported.

**Comment [pT9]:** Added a more detailed description of the study (suggested by reviewers 3 L63).

664 We analyse the composition of cave dripwater over five years (August 2005 - March 2011) of  
665 cave monitoring in Yonderup Cave, a shallow cave system, in southwest Australia. Our  
666 monitoring followed an intense wildfire in February 2005 that burnt 1200 ha of Yanchep  
667 National Park. The fire was hot enough to calcine and fracture the limestone observed at the  
668 caves entrance (Supplementary Fig. 1). We compare our monitoring data to the regional  
669 groundwater geochemistry and published monitoring data (Treble et al. 2015) from Golgotha  
670 Cave in southwest Australia (lat. 36.10° S, long. 115.05° E). Our analysis provides one of the  
671 first analyses of the response of dripwater δ<sup>18</sup>O and solute concentrations to post-wildfire  
672 conditions in shallow caves located in the tree rooting zone.

**Comment [GN10]:** Cave name added in response to R3 comment L74

**Comment [GN11]:** A clearer description of the intensity of the wildfire has been added as suggested by Reviewer 1 (Comment L 72)

673

674 **2 Site description**

675 Our study was conducted in Yonderup Cave in Yanchep National Park (lat 31.5475° S, long  
676 115.6908° E), 20 km north of Perth, southwestern Australia (Fig. 1A). This region has a  
677 Mediterranean climate characterised by dry hot summers and cold wet winters with a 25-year  
678 (1990 to 2015) average annual surface temperature of 15.1°C and rainfall of 664 mm with  
679 85% of rainfall falling between May and October. Yonderup Cave is located in the young  
680 Quaternary Tamala Limestone Formation, a porous, partially lithified calcareous coastal dune  
681 sand. This karst process is said to be “syngenetic” with karstification occurring  
682 simultaneously with lithification of the host rock (Jennings, 1964; Fairchild and Baker,  
683 2012).

684 Yonderup Cave is situated in a tuart forest (*Eucalyptus gomphocephala*), with mature tuart  
685 trees 30 m high, and an understory of shrubs and trees standing 5-10 m high, Sheoak trees  
686 (*Allocasuarina fraseriana*) approx. 5-15 m high, and Balga trees (*Banksia attenuata*, *Banksia*  
687 *menziesii*, *Banksia grandis*, *Allocasuarina fraseriana*, *Xanthorrhoea pressii*). Tree roots are  
688 exposed in the cave, both in the roof (fine roots), and cave floor (thick tap roots). In February  
689 2005, the area above the cave was burnt in an intense wildfire (Fig. 1B, Department of Parks  
690 and Wildlife, pers. comm., 2015), substantially modifying vegetation above the cave  
691 including the death of mature trees and complete removal of canopy and understorey.

692 Over the period of August 2005 to March 2011, two drip sites in Yonderup Cave (Site 1a and  
693 Site 2a), were monitored for their chemical and hydrological variations. These two sites are  
694 22.8 m apart (Site 2a east of Site 1a, ~ 1 m slope towards the East), located at similar depths  
695 below the surface (~ 4 m) within the same chamber (~ 7 m height) and partially separated by  
696 a large boulder fall-in. We use an existing cave survey to determine the location of each cave  
697 drip site relative to the ground surface (Fig. 1C). A soil depth survey was conducted within 5  
698 m of each site (Supp. Table 3), along with visual vegetation/ground surface observations

Comment [G12]: Time of temp record given (R3 comment L85)

699 post-fire. Soil depths were measured every meter with a dynamic soil penetrometer in north,  
700 south, east and west directions and averaged soil depth above each site were calculated.

701 Site 1a, 30 m from the cave entrance, has a drip source within a large cluster of soda-straw  
702 stalactites known as the ‘Wheatfield’ (Supp. Figure 2B). This circular feature is  
703 approximately 1 m across and as it appears in an otherwise very sparsely decorated part of the  
704 ceiling, suggesting that it represents a focused flow path into the cave. The land surface  
705 above this site is flat with 70% coverage by shallow soil (average 124 mm thickness) and the  
706 remaining surface is exposed bedrock (approx. 30%). A tuart tree, located directly above Site  
707 1a, burnt and collapsed during the 2005 wildfire which resulted in the entire removal of  
708 canopy cover above Site 1a. No other trees are close enough to provide shade on the surface  
709 above Site 1a.

**Comment [GN13]:** ‘heterogenous’ removed in response to (R3 comment L113). We already demonstrate that the soil is heterogeneous by explaining that only 70% of the surface is covered by soil.

710 In contrast, Site 2a situated 50 m from the cave entrance is in a highly decorated part of the  
711 cave known as the ‘Cathedral’ characterised by large icicle shaped stalactites. Above Site 2a,  
712 the soil cover is thicker (200 mm) and more homogenous with no bedrock exposure, and no  
713 trees directly above, however there is a partial canopy cover from adjacent trees ~15 m away.

714

### 715 3 Data collection

716 Cave dripwater was collected from 1L high-density polyethylene (HDPE) collection vessels  
717 at the two sites Site 1a and Site 2a between August 2005 and March 2011 (~5.5 years) at  
718 approximately bi-monthly intervals. The water was separated into three aliquots: two aliquots  
719 were filtered with 0.45 µm mixed-cellulose filters into two 50 ml polypropylene bottles for  
720 major and minor ion determination; the third was stored with zero-headspace in a 12 ml  
721 amber glass bottle for stable isotopes. All aliquots were refrigerated below 5°C until analysis.  
722 Anion concentrations (Cl and SO<sub>4</sub>) were determined using a Dionex DX-600 ion

**Comment [GN14]:** Length of collection interval stated. (R3 comment L123-5)

723 chromatograph with self-regenerating suppressor on one aliquot. The second aliquot was  
724 acidified to 2% HNO<sub>3</sub> in the collection bottle and used for cation concentrations (Ca, K, Mg,  
725 Na, Si and Sr) using a Thermo Fisher inductively coupled plasma-atomic emission  
726 spectrometer (ICP-AES) ICAP7600 at the Australian Nuclear Science and Technology  
727 Organisation (ANSTO) facility. An internal standard with concentrations approximating the  
728 cave waters was included in each cation batch to check for between-run reproducibility.

**Comment [GN15]:** Spectrometer defined response to reviewer 3 L 129

**Comment [G16]:** ANSTO acronym defined (reviewer 3, Comment L145)

729 Dripwaters collected between August 2005 and May 2008 were analysed for δ<sup>18</sup>O using  
730 Isotope-Ratio Mass Spectrometry (IRMS) at the Australian National University (see Treble et  
731 al., 2013 for method). The remaining dripwaters were analysed for δ<sup>18</sup>O and δ<sup>2</sup>H at ANSTO  
732 using the Cavity Ring Down Spectroscopy (CRDS) method. Additionally, as there was  
733 sufficient remaining water in the stored aliquots analysed by IRMS for Site 2a, these were  
734 also re-analysed using CRDS to obtain a complete time series for δ<sup>2</sup>H. After Jan 2007  
735 dripwater volume at Site 1a became insufficient to collect all three aliquots. Collections of  
736 aliquots were prioritised in the following order: 1) stable isotopes; 2) cations and; 3) anions.

**Comment [GN17]:** Location of instrumental analysis has been specified in response to Reviewer 3 comment L129,132

737 At each cave visit for dripwater sampling, drip rates were manually recorded using a  
738 stopwatch and the level of water accumulated in the bottles was recorded to the nearest 100  
739 ml. Weekly discharge was estimated using a drip volume of 0.2 ml per drip (Collister and  
740 Mattey, 2008). When timing drip intervals became impractical, only the bottle level was  
741 recorded. Thus in order to represent the data in common units we needed to use the Collister  
742 and Mattey, 2008 drip volume in order to convert all our discharge data into volume data.  
743 We use both sets of measurements from the overlapping period to convert volume to  
744 discharge for when direct measurements for drip interval using the stop watch were lacking.

**Comment [GN18]:** Drip volume use justified in response to R3 comment L145

745 The calculations are provided in the supplementary info as excel sheets.



746 To distinguish dry and wet periods we applied a residual mass curve (RMC), (Hurst, 1951)  
747 to monthly P - AET data. The RMC is the cumulative sum of the monthly anomaly calculated  
748 from the 22 year mean and used to generate a time series of cumulative potential water  
749 surplus or deficit starting from Jan 2000, highlighting trends in above average or below  
750 average P – AET, we refer to this calculation as cumulative water balance (CWB) throughout  
751 the rest of this paper.

**Comment [GN19]:** Defined. In response to Reviewer 3 comment L151-4

752 Unpublished monthly  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  rainfall data (2005 – 2011) from Perth were obtained  
753 from ANSTO. We used modelled regional precipitation (P) and actual evapotranspiration  
754 ( $\text{AET}/\text{F}_{\text{WE}}$  is the sum of soil evaporation and transpiration by vegetation based on Priestly-  
755 Taylor equations) from the Australian Water Availability Project (AWAP) (Raupach et al.,  
756 2009; Raupach et al., 2011) with monthly parameters, to determine P – AET. AWAP  
757 precipitation (P), actual evapotranspiration ( $\text{AET}/\text{F}_{\text{WE}}$ ) and rainfall  $\delta^{18}\text{O}$  data were then used  
758 as input to the forward model (detailed in the next section) to predict cave dripwater  $\delta^{18}\text{O}$   
759 composition under various hydro-climatic scenarios. Predictions are based solely on P - AET  
760 data which are then compared to the dripwater observations.

**Comment [GN20]:** FWE defined, in response to reviewer 3 comment L157

**Comment [GN21]:** FWE, defined in response to reviewer 3 comment L157

761 Monthly rainfall  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  compositions were amount weighted and fitted with a linear  
762 regression (Hughes and Crawford, 2012) and compared to the long-term groundwater mean  
763 obtained from Turner and Thorpe (2001) and the cave dripwater to determine whether  
764 evaporation has affected cave dripwater isotopic composition (see section 4).

765 Post-fire solute and  $\delta^{18}\text{O}$  data from Yonderup Cave dripwater are also compared to other  
766 relevant published data. These include, long term Perth rainfall  $\delta^{18}\text{O}$  from Turner and Thorpe  
767 (2001), and local Yanchep rainfall solute data from Hingston and Gailitis, (1976), and  
768 published dripwater data from Golgotha Cave, located 300 km south of Yanchep. Golgotha  
769 Cave has been monitored since 2005 (Treble et al. 2013; 2015; 2016; Mahmud et al 2015).

770 The climate at Golgotha Cave is also Mediterranean, but receives annual mean rainfall of  
771 approx. 795 mm, which is 23% higher than Yanchep. Both caves are located within the  
772 Tamala Limestone Formation, however the caves vary in depths: Golgotha Cave is  
773 significantly deeper than Yonderup approx. 30 - 35 m. Golgotha Cave is covered by a more  
774 extensive forest of mixed marri/karri (*Eucalyptus calophylla* / *Eucalyptus diversicolor*) trees  
775 and this site has not experienced an intense wildfire since 1992 and no prescribed burns since  
776 2006. However, the prescribed burn at Golgotha Cave was much less intense and it was more  
777 controlled than the fire that is reported in our study.

**Comment [GN22]:** We detail that the Golgotha burn was much less intense than the wildfire noted in this study. Response to Reviewer 3 comment L178-179

#### 779 4 Forward model

780 We use the forward model employed by Baker et al., (2010). This model uses monthly  
781 rainfall  $\delta^{18}\text{O}$ , monthly (P – AET), from 2003 to 2011 (we use 2003 to 2005 data as a ‘warm  
782 up’ period to avoid edge effects), and adjustable bedrock flow thresholds for seepage flow  
783 and fracture flow to predict dripwater  $\delta^{18}\text{O}$  based on hydro-climatic influences. Seepage flow  
784 and fracture flow thresholds are hydrological P - AET thresholds that are required for  
785 infiltrating water to enter seepage or fracture reservoirs (for further details see Baker and  
786 Bradley, 2010). The Tamala Limestone retains high primary porosity thus seepage flow is  
787 likely to be dominant whilst fracture flow is less dominant and only likely to be activated  
788 during high infiltration (Treble et al., 2013; Mahmud et al., 2015).

**Comment [GN23]:** Addition of observational window in response to Reviewer 1 comment L201-L205.

Further we also re-wrote section 4 to make it clearer to the reader the role of the forward model in this study. (in response to reviewer 1 comment L453 and Reviewer 1 comment 243-272 and other general comments from the other reviewers and the editor.

789 Storage time for the water that enters the seepage reservoir is modelled as a Gaussian  
790 distribution. A maximum residence time of 3 years is set; this reflects the dominating  
791 seepage or matrix flow type at our site, the shallow depth (4 m) and the potential for capillary  
792 barrier effects to impact hydrology in this region (Mahmud et al., 2015). Further a minimum  
793 residence time of 10 months is required, to maintain the observed year round discharge at

**Comment [GN24]:** Changed – in response to Reviewer 1 comment L193

794 both sites. The model allows for the mean and standard deviation to be specified for these  
795 functions. Being conservative we specify the minimum residence time of  $10 \pm 2$  months. In  
796 contrast, the fracture-fed flow is instantaneously passed through the system (i.e. with a travel  
797 time of less than one month). In the model we can adjust the (P – AET) thresholds required  
798 for flow into the seepage reservoir and the threshold required for it to spill into the fracture  
799 flow. The seepage and fracture-fed components are mixed in the overlying bedrock reservoir,  
800 before predicting dripwater  $\delta^{18}\text{O}$  composition. By request the authors can supply the forward  
801 model as a spread sheet.

**Comment [GN25]:** Re-written in response to Reviewer 3 comment L192-4

**Comment [GN26]:** Changed from Karst store to overlying bedrock reservoir. Response to reviewer 3 comment L198

802 We tested a full range of seepage and fracture possibilities. This suite of model runs helps to  
803 place constraints on  $\delta^{18}\text{O}$  variability that can be explained by hydro-climatic variability alone.  
804 We compare these scenarios to the observed dripwater  $\delta^{18}\text{O}$  at our sites, to assist in our  
805 interpretation of the post-fire dripwater  $\delta^{18}\text{O}$  response.

## 806 5 Results

807 A time series of monthly P – AET, cumulative water balance (CWB), discharge, dripwater  
808  $\delta^{18}\text{O}$ , and ion concentrations for Sites 1a and 2a from August 2005 – March 2011 are shown  
809 in Figure 2.

810

### 811 5.1. Water balance

812 Firstly, we observe a distinct seasonality in the water availability (P – AET) (Fig. 2A), where  
813 winter months generate an excess (P > AET), while summer months generate a deficit (P <  
814 AET). Further, CWB shows three distinct trends throughout the monitoring period: 1) a  
815 decline over the period of January 2006 to June 2006, consistent with very low excess in P -  
816 AET; 2) an overall rise from June 2006 to February 2010; 3) a decrease in P - AET from  
817 February 2010 to September 2010. Site 1a and 2a display moderate and similar discharge  
818 rates, at the start of the monitoring period that continue until July 2006; Site 1a an average of  
819 90 ml  $\pm$  21 per week and Site 2a an average of 92 ml  $\pm$  23 per week. This coincides with  
820 infiltration indicated by positive CWB (Fig. 2B). In July 2006, Site 1a dramatically increased  
821 discharge five-fold to 468 ml/week on one cave visit, but had decreased to 55 ml  $\pm$  3 ml per  
822 week on the subsequent visit two weeks later and was completely dry, three months later.  
823 This site has not re-activated since (Department Parks and Wildlife, pers. comm.). Site 2a  
824 shows much less variation in discharge overall, but contains smooth long-term trends. Two  
825 periods of higher discharge are observed in August 2005 to May 2006 (average 92 ml  $\pm$  23  
826 ml per week) and April 2008 to February 2009 (average 93 ml  $\pm$  29 ml per week), both  
827 coinciding with positive trends in CWB.

828

829

**Comment [GN27]:** Changed to water availability to make the term consistent in response to Reviewer 3 comment L217.

**Comment [GN28]:** Changed from 'increases' to 'increased'. (reviewer 3 comment L224)

830 **5.2 Water isotopes**

831 Dripwater  $\delta^{18}\text{O}$  from Site 1a (Fig. 2C) shows no seasonal pattern but we see a steady increase  
832 of 1‰ to January 2007, then a further steeper rise of 1.5‰ in June 2007, after which the drip  
833 ceases. Dripwater  $\delta^{18}\text{O}$  from Site 2a presents an overall increasing trend rising from -3‰ to  
834 +0.7‰ over the monitoring period with a 6-month quasi-seasonal signal (approx. 2‰ range)  
835 that peaks in cooler months (June to October) generally coinciding with months when  
836 infiltration from rainfall occurs. We hypothesise that the  $P < \text{AET}$  environment in drier  
837 summer months isotopically enriches soil water, but this only arrives at the cave when  
838 seepage thresholds are exceeded in periods of  $P > \text{AET}$  (winter months).

**Comment [GN29]:** We have removed the slope calculations and p-tests in this section in response to reviewer 2's comment L262-264.

**Comment [GN30]:** Re-written and added month for Northern Hemisphere readers. Response to reviewer 3 comment 239.

839  
840 We forward modelled our rainfall isotopic data in order to predict drip-water  $\delta^{18}\text{O}$  under  
841 various hydro-climatic scenarios (Fig 3). Our sensitivity analyses of hydrological residence  
842 times and thresholds showed that seepage residence times, less than 10 months resulted in the  
843 seasonal cessation of dripwater, which is not observed at our sites. Therefore, a minimum  
844 seepage residence time is required to match our observations. Further, seepage threshold  
845 values greater than 40 mm ( $P - \text{AET}$ ) also resulted in the cessation of our drip site. Thus  
846 seepage threshold must be below 40 mm ( $P - \text{AET}$ ) to match our observation. Next we varied  
847 the fracture threshold between 15 mm and 1000 mm, the wide range reflecting our  
848 uncertainty over this parameter. However, we know that seepage flow is dominant at these  
849 sites (Mahmud et al., 2015). This suggests two things, first, the seepage threshold is low,  
850 second, the threshold required for water to 'overflow' from the seepage reservoir to fracture  
851 reservoir must be significantly higher than the seepage threshold. We note that scenarios with  
852 a lower fracture threshold (10 – 15 mm) show high variability in comparison to sites with a  
853 seepage dominated flow and no fracture flow (10 – 1000 mm). Based on the variable

**Comment [GN31]:** Re-written in continued response to reviewer 1 comment L243-L276 and Reviewer 3 comment L243

**Comment [GN32]:** We have defined the thresholds required for seepage and fracture flow ( $P - \text{AET}$ ), in response to Reviewer 3 comment L244).

854 morphology of stalactites and stalagmites at our sites we interpret discharge to be a  
855 combination of seepage and fracture flow, but with seepage clearly dominating. Hence we  
856 chose the 15 – 100 mm scenario to represent the hydrology at our cave site (Fig. 3). Our  
857 forward-modelled dripwater  $\delta^{18}\text{O}$  mean is -4.1‰, slightly less than the mean of Perth rainfall  
858 (-3.1‰). The time series of modelled dripwater  $\delta^{18}\text{O}$  (Fig. 2C) starts and remains at ~ -4.2‰  
859 until February 2006 where it dips slightly before rising sharply to -3‰ where it remains  
860 steady until February 2007. Here it begins a step-wise decline; declining from February to  
861 March 2007 by 0.5‰ and remaining stable again until February 2008. It then shows a further  
862 step-decline in March 2008 to -4.5‰, where it remains at approximately this value, albeit  
863 with a few small variations on timescales of months, until the end of the monitoring period.

864 In all meaningful modelled scenarios i.e. ones that have full year flow and test the full range  
865 of hydrological variability, estimated dripwater  $\delta^{18}\text{O}$  cannot replicate the higher observed  
866 dripwater  $\delta^{18}\text{O}$  which are +1‰ to +3‰ higher compared to modelled (Fig. 3). This clearly  
867 suggests another factor is affecting dripwater  $\delta^{18}\text{O}$  composition: likely near-surface  
868 evaporation.

**Comment [GN33]:** We have defined what comprises a 'meaningful model' in response to reviewer 3 comment L265.

870 To investigate an evaporation effect, we plot cave dripwater along the local meteoric water  
871 line (LMWL, weighted LSR) to test for isotopic enrichment (Fig. 4). Figure 4 shows that  
872 while the least squares regression (LSR) for cave dripwater falls within the standard error ( $\pm$   
873 0.45‰) of the slope for the local meteoric water line (LMWL, weighted LSR), drip water  
874 isotopic composition is concentrated towards heavier  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ . These results are  
875 consistent with evaporation in a high humidity environment as has been observed in semi-  
876 arid cave environments elsewhere (e.g. Cuthbert et al., 2014). Adopting Cuthbert's  
877 classification, our data falls under a type 1 scenario reported in Cuthbert et al., (2014). In the

**Comment [GN34]:** Re-worded as suggested by reviewer 3 comment L 273-274.

878 type 1 scenario,  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  do not deviate from the LMWL but are shifted along the  
879 LMWL towards higher values, as is the case with our data (Fig. 4). This means that our data  
880 are similarly impacted by evaporation occurring in a high humidity environment.

**Comment [GN35]:** Re-written to more clearly state that our bi-variate plot of  $\delta^{18}\text{O}$  vs  $\delta^{13}\text{C}$  supports our case for evaporation at our sites. Response to reviewer 1 comment L240

### 882 5.3 Water Solutes

883 There are significant differences in solute concentrations and trends between the two sites  
884 (Fig. 2E, 2F, 2G and 2H). Solute concentrations are typically higher at Site 1a versus Site 2a  
885 and they demonstrate opposite trends post-fire. At Site 1a, Cl, Ca, Mg and Sr decline overall,  
886 although this trend is step-wise for Ca, and reverses for Cl ~ 6 months before the drip ceases.

887 The trends in these solutes at Site 1a are inconsistent with the declining CWB during this  
888 period (Fig 2A), as we would expect a drying trend reflected through the evaporative  
889 concentration of solutes. In contrast at Site 2a, Cl and other solute concentrations show a  
890 direct relationship to CWB (i.e. increasing solute concentration with decreasing CWB from  
891 2006 until mid-2008 followed by decreasing solute concentrations with increasing CWB.

**Comment [G36]:** We have moved the interpretation of tree roots affecting the drip water chemistry for the discussion to Line 349-351. (in response to R3 comment 288)

892 Trends in  $\text{SO}_4$  and K are more subtle than for other solutes: at Site 1a, K shows a slight  
893 decline from the beginning of the monitoring until early 2007 and then has a small rise prior  
894 to drips ceasing. Although harder to judge in the shorter  $\text{SO}_4$  time series,  $\text{SO}_4$  also shows a  
895 small rise before drips cease, similar to K. Trends in K and  $\text{SO}_4$  for Site 2a are more subtle,  
896 although they both increase slightly over time. K and  $\text{SO}_4$  concentrations are, two to three  
897 times higher at Site 1a versus Site 2a and are considerably higher than those recorded at  
898 Golgotha Cave (Table 1). We also note that initial Cl and other solute concentrations at Site  
899 1a are twice that at Site 2a.

900

901 **6 Discussion**

902

903 **6.1 Post-fire hydrology**

904 Discharge at Site 1a is inconsistent with CWB: discharge rose as rainfall fell below the long-  
905 term mean ( $P < AET$ ) (Fig. 2A and 2B), suggesting that Site 1a received a localised increase  
906 in discharge despite the declining input from rainfall. In contrast, discharge at Site 2a is more  
907 closely related to the cumulative water balance (CWB), with higher discharge coinciding  
908 with periods of higher water surplus and lower discharge with lower water surplus.

909 Chloride is a chemically conservative and highly soluble solute (Graedel and Keene, 1996),  
910 and its concentrations in dripwater will therefore reflect concentration/dilution effects (Tooth  
911 and Fairchild, 2003; Tremaine and Froelich, 2013). Chloride concentrations at Site 2a  
912 increase during the period of declining CWB (2006 to mid-2008) suggesting that evaporation  
913 is concentrating Cl. Rising  $\delta^{18}O$  and other solutes over this period are also consistent with  
914 increased evaporation. From mid-2008 onwards, when CWB is positive ( $P > AET$ ), Cl  
915 decreases, consistent with an increase in infiltration and thus dilution (Fig. 2E).

**Comment [G37]:** Re-arranged sentences such that interpretation followed observations and observations made more conclusive/definitive in response to reviewer 3 comment L316

916 At Site 1a higher discharge also coincides with falling Cl concentrations also suggesting  
917 dilution (Fig. 2B, 2D). However, we note this coincides with a highly negative CWB i.e. drier  
918 than normal conditions. This suggests in this case, a non-climatic driver has influenced  
919 infiltration. We propose that a reduction in localised transpiration, following the 2005 fire,  
920 may be driving this. Deeply-rooted trees within the area have been reported to produce high  
921 Cl concentrations in the unsaturated zone (Turner et al., 1987). Site 1a had a tuart tree  
922 directly above it and tree roots are visible above Site 1a in the cave, but not at Site 2a. The  
923 proximity of the tree to Site 1a is the most likely explanation for the higher solute  
924 concentrations here (Treble et al., 2016). The death of the tree in the 2005 fires would remove  
925 the previous transpiration demand and hence result in effective dilution of the solutes during



926 infiltration, as observed. However, this reduction in transpiration would have been abrupt but  
927 we observe a response lasting 1.5 years after the fire. This could be due to a number of  
928 reasons; firstly, the minimum residence time is 10 months (for a year of continuous  
929 discharge) so a delay in the response is to be expected. Second, this occurred during a period  
930 in which the soil moisture deficit would have been larger than average, so a larger volume of  
931 cumulative infiltration would be needed to overcome this deficit- and move the more dilute  
932 solute into the cave.

933 It is also possible the decrease in concentrations reflect the diminishing element  
934 concentrations after an immediate flush of the more soluble ash-derived material (i.e. the tail  
935 of a solute pulse). However, post-fire, highly soluble solutes like Cl, will still reflect dilution  
936 due to increased discharge. So, it is likely that we are seeing a decline in these elements due  
937 to a combination of the removal of these nutrients from the surface and subsurface, and  
938 dilution.

939  
940 In the broader context we look at the differences in Cl at Yonderup Cave vs Golgotha Cave.  
941 Both caves are ~5 km from the coastline, so they likely have a similar amount of Cl aerosol  
942 deposition. Yet, post-fire Cl values at Yonderup Cave, are more than double the Cl  
943 concentrations from Golgotha Cave (Table 1). Further, within cave variability in Cl  
944 concentrations at Yonderup Cave are twice that at Golgotha Cave (Table 2). While variations  
945 in vegetation density (along with wildfire history) may have some role to play in the  
946 difference in mean dripwater Cl at each location, the higher within cave variability at  
947 Yonderup Cave (Table 2), suggests a post-fire setting increases variability in dripwater  
948 chemistry. The impact of this is discussed further and a conceptual model (Fig. 6) devised  
949 later in section 6.3.

**Comment [GN38]:** Paragraph added for further clarification of our interpretation in response to the comment by Reviewer 1 L329-331, 338-340.

**Comment [GN39]:** We add the other possible cause for the decline in solutes at Site 1a in our interpretation here and throughout the manuscript. The removal of solutes from the surface and subsurface post-fire. (response to reviewer 2 comment L333-334).

**Comment [G40]:** Figure 6 introduced earlier in response to Reviewer 3 L389

**Comment [GN41]:** Distance between Yonderup and the coast and Golgotha and their possible effects on Cl added in response to reviewer 1 comment L332.

950

## 951 6.2 Post-fire carbonate chemistry

952 Similar to Cl, concentrations of carbonate metals (Mg, Ca and Sr) at Site 1a also decrease;  
953 this reflects solutes being diluted due to reduced tree water-use. However, we note that at Site  
954 1a, Ca, for example, declines twice as much (in concentration, ~75 %) in comparison to Cl  
955 (~30%). Thus for Ca, another mechanism along with dilution is required to explain its non-  
956 linear step-like decline (Fig. 2G).

957 There are a number of mechanisms that could influence post-fire Ca concentrations. First, we  
958 consider increased near-surface evaporation inducing prior calcite precipitation (PCP).

959 Increased evaporation, can saturate solutes relative to calcite in karstic waters and promote  
960 degassing. Further, evaporation will slow the flow increase water-rock interaction times in  
961 the remaining water. Both of these conditions are ideal for PCP. Both our sites show evidence  
962 of PCP: the ln(Sr/Ca) vs ln(Mg/Ca) slopes in our data agree with the diagnostic range for  
963 PCP (a slope of + or - 0.88; cf. Fig. 5B) (Sinclair et al., 2011). Expressed as a time series  
964 (Fig. 5A), we see ln(Sr/Ca) and ln(Mg/Ca) increase simultaneously with  $\delta^{18}\text{O}$  and Cl (Fig. 2),  
965 suggesting that evaporation is indeed the common driving mechanism and is inducing PCP at  
966 Site 1a. For further information on PCP processes we recommend the reader to Fairchild et  
967 al., 2000; Sinclair 2011 and Treble et al., 2015.

968 A second mechanism influencing post-fire Ca concentrations may be the addition of plant ash  
969 (Yusiharni and Gilkes, 2012a) and highly soluble CaO (produced by the burning of exposed  
970 surface rock to the fire; Yusiharni and Gilkes, 2012b). Further, it is possible that the Ca  
971 decline may also reflect a decrease in Ca being leached post-fire that may have followed an  
972 earlier spike in Ca concentrations from the above. The extent to which each process is  
973 affecting the Ca concentration is difficult to assess in our data, especially since our

**Comment [GN42]:** Instead of saying that our figure provides a diagnostic model we have said that our figure shows our data 'falls within the diagnostic range' as was initially intended. Response to reviewer 3 comment L348. Further we have also re-written the Ca section in response to R3 comment 358

**Comment [GN43]:** We add further description on how increased surface evaporation could induce PCP, and direct the reader to further references in response to reviewer 1 comment L346.

**Comment [GN44]:** We have integrated a response to reviewer 3's Comment L358 and Reviewer 2's other possible explanation for the decline in solutes to help explain the decline in Ca.

974 monitoring did not commence until 6 months after the fire, and processes such as PCP and  
975 the addition of Ca from plant-ash are difficult to constrain.

Comment [G45]: Reworded

976 We now consider other carbonate metals: Mg and Sr, at Site 1a to further constrain our  
977 interpretation. We find Mg and Sr decline by ~30% (in terms of relative concentration). Thus  
978 it is likely that the same process affecting Cl is also affecting Mg and Sr, that is, dilution, a  
979 decline in leaching of biomass-sourced ash, or some combination of both. Ca concentrations  
980 decline by a relatively larger amount (75%) suggesting that additional processes are  
981 specifically affecting Ca. The rise in Mg/Ca at Site 1a strongly suggests that the remaining  
982 portion of the Ca decline may be attributed to PCP (Fig. 5A).

983 At Site 2a, a rising trend, reflects the concentration of solutes due to a rise in post-fire  
984 evapotranspiration, evidenced by increasing  $\delta^{18}\text{O}$  and Cl, evaporation of near-surface water  
985 stores (Fig. 2F, 2G and 2H respectively), and possibly to some extent, an increase in  
986 transpiration from vegetation recovery for Cl (Treble et al., 2016). Additionally, Ca  
987 concentrations also show a quasi-seasonal response, interpreted from the Mg/Ca time series  
988 to be driven by PCP, possibly due to seasonal P – AET (supported by similar seasonality in  
989 dripwater  $\delta^{18}\text{O}$ ; Fig. 2C) although in-cave PCP could also be contributing (Treble et al.,  
990 2015).

991 Sulphate and K post-fire at Site 1a are abnormally high in concentration, approximately three  
992 times higher in comparison to Site 2a (Figs. 2H, 2I; Tables 1, Table 2). This is  
993 counterintuitive to the initial dilution signal (owing to a decrease in tree water-use)  
994 interpreted for the other solutes. While at Site 2a,  $\text{SO}_4$  and K increase similar to other solutes,  
995 consistent with evaporative concentration (from post-fire conditions) and an increase in  
996 transpiration from (vegetation recovery). These observations suggest that there was an

Comment [GN46]: Table 2 (suggested by Reviewer 2) helps quantify the differences between  $\text{SO}_4$  and K between the two sites, pre-empting speculation. Of how the two Sites response post-fire. Helping satisfy Reviewer 1 comment L364-386. And Reviewer 2 comment 374-377

997 increase in the availability of SO<sub>4</sub> and K after the fire at Site 1a despite a decrease in tree-  
998 water use here (Fig. 2H and 2I).

**Comment [G47]:** Reduced speculation.  
Response to reviewer 1 comment L364 –  
L386

999

1000 We note the majority of aboveground SO<sub>4</sub> is predominantly stored within the lower to middle  
1001 storey of the forest (O'Connell and Grove, 1996), and post-fire soils contain 23% more S and  
1002 16% more K than pre-fire soils due to biomass-sourced ash deposition (Grove et al., 1986).  
1003 So SO<sub>4</sub> and K concentrations at each site may respond differently since the amount of  
1004 available SO<sub>4</sub> and K above each site is influenced by the amount of biomass burnt above the  
1005 site. Further, the dissolution rates of ash minerals containing these elements could also affect  
1006 the rate at which these nutrients are leached from the surface and subsurface and  
1007 subsequently their concentrations in dripwaters. We propose that the large amount of biomass  
1008 burnt above Site 1a – the tuart tree - is responsible for the much higher concentrations of SO<sub>4</sub>  
1009 and K at Site 1a dripwater relative to Site 2a. We also propose that the increase in near-  
1010 surface evaporation from 2007 onwards drives even higher concentrations of SO<sub>4</sub> and K (Fig  
1011 2H and 2I). Site 2a, which has much less pre-fire biomass, has much lower SO<sub>4</sub> and K  
1012 concentrations, consistent with our argument. Here, these solutes show a steady increasing  
1013 trend over the monitoring period. This is consistent with increased evapotranspiration post-  
1014 fire, which is also evident in the other solutes.

1015 Now, we compare our Yonderup Cave results to those of Golgotha Cave to put SO<sub>4</sub> and K  
1016 concentrations into context. Golgotha Cave, last experienced a wildfire in 1992 and a  
1017 controlled low-temperature prescribed burn in 2006, while Yonderup experienced a high  
1018 intensity burn in 2005. Firstly, we see five and ten times higher within cave differences in  
1019 SO<sub>4</sub> and K (respectively) at Yonderup Cave than we do at Golgotha Cave (Table 2). Also,  
1020 concentrations at Yonderup Cave are up to four and three times higher in SO<sub>4</sub> and K

1021 | (respectively) at Yonderup Cave than at Golgotha Cave (Table 1). These results coupled with  
1022 | the difference between Cl composition at Yonderup vs. Golgotha presents a clear case for  
1023 | more variability in burnt sites in comparison to unburnt sites.  
1024 |

### 1025 | **6.3 A multi-proxy fire signal in dripwater**

1026 | Here we propose post-fire scenarios for both sites at Yonderup Cave (refer to conceptual  
1027 | model Fig. 6) that account for the altered dripwater chemistry that is observed post-wildfire.  
1028 | A straight-forward relationship between cumulative water balance (CWB), discharge and Cl  
1029 | concentrations at Site 2a, suggests increased concentration of solutes (Ca, Mg, Sr, K and SO<sub>4</sub>)  
1030 | in response to an increase in near-surface evaporation (Fig. 2C and 2D).

1031 | In contrast solutes such as Cl, Mg, Sr, Ca at Site 1a show a declining trend. We have argued  
1032 | that this declining trend in these solutes is due to two underlying processes. First, a decrease  
1033 | in tree water-use (transpiration) due to the death of the tuart tree in the wildfire. Tuart trees  
1034 | are deeply-rooted, having adapted their root systems to access water from both the surface  
1035 | and subsurface. These roots have been found to generate potential energy between the tree  
1036 | and the soil to extract water and nutrients. Specifically, mature tuart trees generate a pressure  
1037 | gradient ranging from  $-0.86 \pm 0.11$  MPa (summer) to  $-0.35 \pm 0.02$  MPa (winter) (Drake et al.,  
1038 | 2011). So the death of the tuart tree in the wildfire had a significant local effect on hydrology  
1039 | at Site 1a resulting in an increase in discharge and dilution of solutes (Cl, Mg, Sr and Ca).  
1040 | Our study suggests that the consequent reduction in transpiration may not be immediately  
1041 | detected in dripwater, owing to transit time through the limestone and the requirement for  
1042 | overcoming a soil moisture deficit during drier than average climatic conditions.  
1043 | Hydrological effects such as ‘capillary barriers’ can also slow down vertical transport of  
1044 | infiltrating waters at our site (Mahmud et al., 2015).

**Comment [GN48]:** Context of Tuart tree transpiration added in response to reviewer 1 comment L399 and L400

1045 Second, the decline in solutes at Site 1a could be the result of a gradual return to pre-fire  
1046 concentrations following a pulse of increased input of solutes from ash (e.g. Site 1a scenario).  
1047 A contribution of SO<sub>4</sub> and K from burnt biomass may explain their relatively high  
1048 concentrations.

**Comment [G49]:** We have reduced speculation. This sentence along with adjustments in section 6.2 to answer the comment L448 by Reviewer 1.

1049 Further, isotopic composition at Site 1a show that δ<sup>18</sup>O is offset from modelled hydro-  
1050 climatic δ<sup>18</sup>O by ~+1‰ suggesting increased near-surface evaporation post-fire which we  
1051 attribute to the reduction of shading from the tuart tree post-fire - even in P > AET periods  
1052 (Fig. 2C). And post-2007, when P < AET conditions arrive, δ<sup>18</sup>O rises even higher and  
1053 discharge declines. Eventually, due to the persistent duration of P < AET conditions Site 1a  
1054 ceased dripping owing to eventual depletion of the near-surface reservoir feeding this drip.

**Comment [GN50]:** Added, in response to Reviewer 1 comment L397, showing that observed d18O and modelled do not agree. Further we demonstrate how the climate forcing (P > ET) inhibits a full evaporation post-fire to take effect.

1055 Site 2a also shows higher δ<sup>18</sup>O, during the post-fire period, which we attribute to an increase  
1056 in near-surface evaporation as a result of low albedo and reduced vegetation cover. The case  
1057 for evaporation at this site is supported by the rise in Cl and other solutes. The possibility of  
1058 an immediate spike in solutes from ash and a long-term decline from leaching, as was  
1059 discussed for Site 1a, is limited here, as there was less biomass available to burn above Site  
1060 2a. Therefore, we interpret increased evaporation and transpiration from regrowth post-fire to  
1061 be the dominant forcing at this site, similar to the findings of Treble et al. (2016).

**Comment [GN51]:** Re-structured sentence to respond to Reviewer 1 comment L408 to 410. Further with the restructuring of this section and appropriate clarification we believe we have a more compelling argument – answering Reviewer 1 comment L419

1062 We propose that differences in surface vegetation above sites can influence site specific drip  
1063 chemistry. For example, we interpret Site 1a was influenced by a reduction in transpiration  
1064 after the fire, due to the forcing biomass above the site; which may also have been a source of  
1065 post-fire ash at this site. At Site 2a the response to the fire was primarily an increase in near-  
1066 surface evaporation owing to changes in surface albedo. This variability within the cave  
1067 response at Yonderup Cave is significant, so too is the comparison between Yonderup Cave  
1068 drip chemistry and the drip chemistry at Golgotha Cave. The latter more generally highlights

1069 that an intense wildfire has variable, but multi-year effects on dripwater composition in  
1070 shallow caves.

**Comment [GN52]:** We have added a table of differences between Golgotha and Yonderup to quantify our interpretation and pre-empt speculation. As suggested by reviewer 2 comment (L430-431)

1071 From this we propose that post-fire condition persist up-to 5 – 10 years' post-fire, affecting  
1072 dripwater  $\delta^{18}\text{O}$  and solute concentrations. We would expect a full recovery, of  $\delta^{18}\text{O}$  and  
1073 solute concentrations back to pre-fire levels within 10 - 20 years as a result of revegetation  
1074 growth (Treble et al., 2016) and re-establishment of vegetation cover and pre-fire albedo.

**Comment [GN53]:** Introduced earlier in the text in response to reviewer 3 comment L450-4452 and 455

## 1076 **7 Application for a speleothem paleo-fire signal**

**Comment [GN54]:** We add this section in response to Reviewer 3's general comments and comments L450-458. This section helps to move hypothetical implication for future studies of this kind from the conclusion, addressing general comments from Reviewer 1 and 3 on the conclusion. And in the process adding my conclusiveness to our conclusion.

1077 Our post-wildfire dripwater response from  $\delta^{18}\text{O}$  was a 2‰ increase above that predicted by a  
1078 hydro-climatic model, and measured regional groundwater and Golgotha Cave  $\delta^{18}\text{O}$  data. If  
1079 this signal is preserved at equilibrium in speleothems this is equivalent to some of the largest  
1080 interpreted climatic changes seen in the Quaternary record. This highlights the significance  
1081 of the findings in our study, which suggests a fire signal could in fact be misinterpreted as  
1082 climate variability. Furthermore, the impact on Ca dissolution from the limestone bedrock  
1083 could have a significant effect by decreasing on speleothem growth rate.

**Comment [GN55]:** Possible effect of Ca on Growth rate introduced in text prior to conclusion as suggested by reviewer 3 comment L458.

1084 However, before attributing  $\delta^{18}\text{O}$  and growth rate abnormalities to fire, we must remember  
1085 there are a number of processes that effect speleothem  $\delta^{18}\text{O}$ . Thus it is important that a multi-  
1086 proxy approach, which uses isotopic composition as well as a suite of trace elements (sourced  
1087 from both soil and bedrock), is used to separate fire from other forcings such as climate and  
1088 other local factors. Further, our study was conducted in a shallow cave environment, where  
1089 perhaps the overlying vegetation can exert a more dominant forcing on dripwater hydrology  
1090 and chemistry relative to deeper caves. Deeper caves have more complex hydrology; this  
1091 involves mixing with other flow paths, which are possibly not fire affected. This may result  
1092 in the smoothing of the fire signal; making it harder to isolate (McDonald and Drysdale,

1093 2007). Further, a fire signal in cave dripwater and stalagmites, may be much subtler in  
1094 grassland environment in comparison to a forested environment as changes in the biomass  
1095 would be smaller and vegetation recovery presumable faster (Coleborn et al., 2016). We  
1096 recommend searching for fire signals in shallow cave environments in the tree rooting zone in  
1097 forested areas.

**Comment [GN56]:** Suggesting added, in response to reviewer 3, comment L456-457.

1098 One further approach that may help to differentiate fire and climate signals in a stalagmite  
1099 would be to use multivariate statistical techniques such as principal component analyses  
1100 (PCA). Using this technique, we would expect one component to reflect a bedrock/hydro-  
1101 climatic signal and another to preserve a local soil/vegetative forcing. The soil/vegetative  
1102 component could preserve the impact of a fire on stalagmite composition, in trace elements  
1103 like S, and K. Further, it is also possible an immediate spike in solutes (Mg, Ca, Cl, Sr, S, K  
1104 and P) from post-fire ash may be preserved in stalagmites and colloid associated metals, such  
1105 as Al, Fe and Cu from an increase in discharge post-fire. Future studies of this kind will open  
1106 a new avenue in speleothem research; speleothems as archives of paleo-fire.

1107

1108



## 1109 **8 Conclusions**

1110 We isolate a post-wildfire response by comparing a recently burnt cave monitoring site with  
1111 forward modelled  $\delta^{18}\text{O}$ , which predicts  $\delta^{18}\text{O}$  based on hydro-climatic factors, and nearby cave  
1112 monitoring and groundwater data. We provide a novel analysis of the multi-year impacts  
1113 wildfire has on cave dripwater. Our analysis shows a strong hydrologic relationship between  
1114 surface environments and shallow caves that are located within the tree rooting zone. This  
1115 finding is especially important in water-limited environments ( $P < ET$ ) as the overlying  
1116 vegetation can exert controls on the cave hydrogeochemical environment.

1117 A post-wildfire dripwater response is clearest in  $\delta^{18}\text{O}$  and Cl due to their sensitivity to  
1118 variation in near-surface evaporation (both  $\delta^{18}\text{O}$  and Cl) and transpiration (Cl). Cl is a  
1119 conservative ion and hence is driven mainly by dilution/evaporation. Cl declines post-fire  
1120 at Site 1a which we interpret as dilution of the water store that the dead tree previously  
1121 exploited. In contrast, Cl increases at Site 2a, sympathetically with  $\delta^{18}\text{O}$ , consistent with an  
1122 increased evaporative demand on shallow water stores driven by post-fire reduction in  
1123 shading and reduced albedo.  $\text{SO}_4$  and K are also important as at sites with abundant biomass  
1124 they can be leached at high concentrations as they are made more abundant in post-fire soils  
1125 due to the ash generated from a fire. Other solutes such as Mg, Sr and Ca support the  
1126 dominant local forcing at the site post-fire and can be extremely powerful when using a  
1127 multi-proxy approach.

1128 We propose a conceptual model for a multi-year post-wildfire cave dripwater response in  
1129 forested water-limited regions. This involves a 5 - 10 yr response of: 1) higher  $\delta^{18}\text{O}$  and Cl in  
1130 cave dripwater due to increased evaporation and decreased shading after the wildfire; 2)  
1131 increased K and  $\text{SO}_4$  due to the leaching of biomass-sourced ash, particularly in areas with  
1132 large biomass; and 3) increased variability in Mg, Sr and Ca due to changes in evaporation,

**Comment [GN57]:** The addition of section 7 and refinements in our conclusion help to make our conclusion more compelling and supportive. In response to reviewer 1 comment L449.

**Comment [GN58]:** By clarifying our model and rewriting our argument in response to all the reviewers comments we have a compelling case for our conclusion (response to Reviewer 1 comment L441-444 and L446-447)

1133 transpiration and water-rock interactions post-fire. We may expect a recovery within 10-20  
1134 years after the wildfire and a restore to pre-fire isotopic and trace element concentrations as a  
1135 result of increased bio-productivity from forest regrowth and a re-establishment of canopy  
1136 cover.

1137

1138 |

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**Tables and Figures**

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1279 Table 1. A summary table comparing the hydrogeochemistry of, (A) shallow Yonderup Cave drip sites burnt in a 2005 wildfire to; (B) deeper  
 1280 Golgotha Cave sites burnt in a 1992 wildfire and a 2006 prescribed burn; (C) unpublished Perth rainfall data during the monitoring period  
 1281 (ANSTO); (D) rainfall isotopic composition (Turner and Thorpe, 2001); (E) groundwater isotopic composition (Turner and Thorpe, 2001); (F)  
 1282 Yanchep rainfall solute composition (Hingston and Gailitis, 1976).

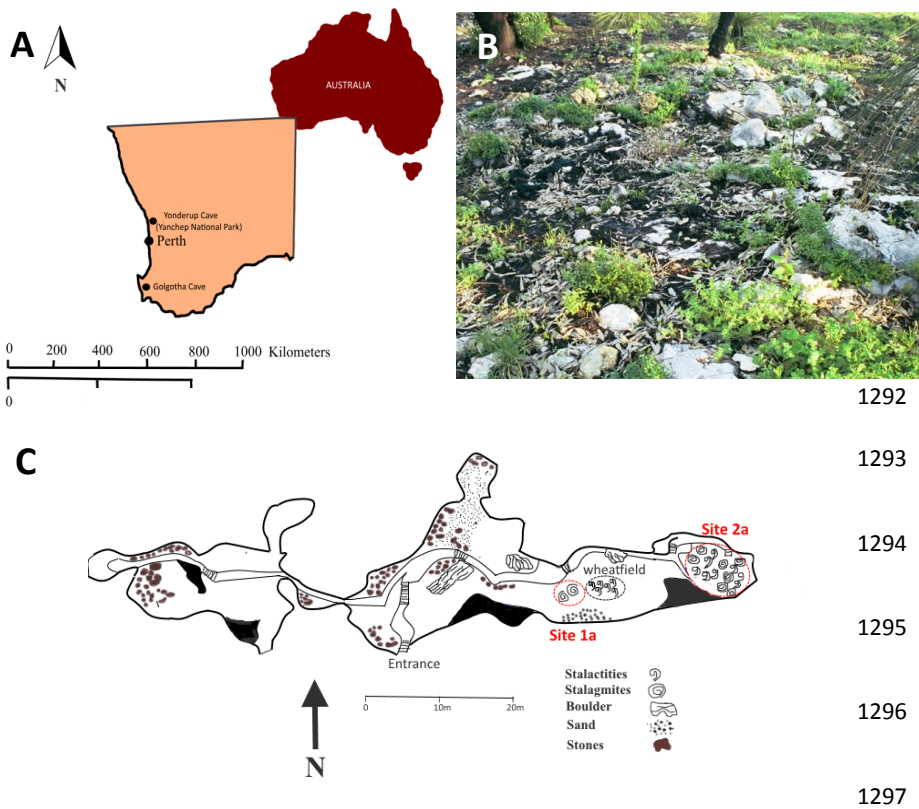
Location		Discharge (ml/day)	Ca (mmol/L)	Mg (mmol/L)	Sr (mmol/L)	Cl (mmol/L)	SO <sub>4</sub> <sup>2+</sup> (mmol/L)	K (mmol/L)	δ <sup>18</sup> O (per mil)	n
<b>(A) Yonderup</b>										
Site 1A	Median	12.2	1.2	0.46	0.01	8.1	0.25	0.15	-3.2	27
	SD	2.6	0.6	0.13	0.2 × 10 <sup>-2</sup>	0.6	0.02	0.03	1.23	
Site 2A	Median	6.1	1.42	0.37	0.01	5.91	0.14	0.05	-1.48	49
	SD	4.8	0.2	0.04	0.5 × 10 <sup>-3</sup>	1.75	0.05	0.8 × 10 <sup>-2</sup>	0.84	
<b>(B) Golgotha</b>										
Site 1A	Median	63	1.2	0.3	0.2 × 10 <sup>-2</sup>	3.35	8.0 × 10 <sup>-2</sup>	0.02	-4.1	85
	SD	4.7	0.22	0.03	0.183 × 10 <sup>-3</sup>	0.18	0.4 × 10 <sup>-2</sup>	0.3 × 10 <sup>-2</sup>	0.3	
Site 1B	Median	40	1.1	0.3	0.2 × 10 <sup>-2</sup>	3.41	8.0 × 10 <sup>-2</sup>	2.3 × 10 <sup>-2</sup>	-3.9	82
	SD	2	0.25	0.03	2.39 × 10 <sup>-4</sup>	0.25	0.5 × 10 <sup>-2</sup>	0.5 × 10 <sup>-2</sup>	0.1	
Site 2A	Median	47	1.1	0.32	0.2 × 10 <sup>-2</sup>	2.76	0.13	3.8 × 10 <sup>-2</sup>	-3.9	77
	SD	10	0.25	0.02	0.1 × 10 <sup>-3</sup>	0.22	0.8 × 10 <sup>-2</sup>	0.5 × 10 <sup>-2</sup>	0.1	
Site 2B	Median	67	1.22	0.33	0.2 × 10 <sup>-2</sup>	4.49	9.8 × 10 <sup>-2</sup>	3.8 × 10 <sup>-2</sup>		84
	SD	85	0.27	0.03	0.2 × 10 <sup>-3</sup>	0.24	0.6 × 10 <sup>-2</sup>	0.7 × 10 <sup>-2</sup>		
Site 2E	Median	524	1.9	0.31	0.1 × 10 <sup>-3</sup>	4.11	0.085	3.8 × 10 <sup>-2</sup>		51
	SD	45	0.2	0.02	0.2 × 10 <sup>-3</sup>	0.17	0.4 × 10 <sup>-2</sup>	0.4 × 10 <sup>-2</sup>		
<b>(C) ANSTO Rainfall</b>	<i>Mean</i>								-3.1	64
<b>(D) CSIRO long-term rainfall (Perth)</b>	<i>Mean</i>								-3.85	165
<b>(E) Regional groundwater (Perth)</b>	<i>Mean</i>								-4.68	43
<b>(F) Yanchep Rainfall Solutes</b>	<i>Mean</i>		0.008			0.01	0.1 × 10 <sup>-2</sup>	0.4 × 10 <sup>-2</sup>		

1283 Table 2. Summary of differences in mean concentration of solutes and isotopic composition of solutes among sites at Yonderup and Golgotha  
 1284 caves. We see that Cl, SO<sub>4</sub>, K and δ<sup>18</sup>O values, at both sites are distinctly different. Specifically, the solutes have higher concentrations and δ<sup>18</sup>O  
 1285 is higher at Yonderup Cave in comparison to Golgotha Cave.

**Comment [GN59]:** Table of differences added in response to reviewer 3 comment L430-431.

Site Differences								
	Discharge (ml/day)	Ca (mmol/L)	Mg (mmol/L)	Sr (mmol/L)	Cl (mmol/L)	SO <sub>4</sub> (mmol/L)	K (mmol/L)	δ <sup>18</sup> O (mmol/L)
<i>Yonderup site differences in mean</i>								
Site 1a and Site 2a	6.1	0.22	0.09	0.02	2.19	0.11	0.1	1.72
<i>Golgotha site differences in mean</i>								
Site 1a and Site 1b	23	0.100	0.000	0.000	0.060	0.000	0.003	0.200
Site 1a and Site 2a	16	0.100	0.020	0.000	0.290	0.050	0.018	0.200
Site 1a and Site 2b	22	0.020	0.030	0.000	1.140	0.018	0.018	--
Site 1a and Site 2e	461	0.700	0.010	0.019	0.760	0.005	0.018	--
Site 1b and Site 2a	7	0.000	0.020	0.000	0.650	0.050	0.013	0.000
Site 1b and Site 2b	27	0.120	0.030	0.000	1.080	0.018	0.013	--
Site 1b and Site 2e	484	0.800	0.010	0.019	0.700	0.050	0.013	--
Site 2a and Site 2b	20	0.120	0.010	0.000	1.730	0.032	0.000	--
Site 2a and Site 2e	477	0.800	0.020	0.019	1.350	0.045	0.000	--
Site 2b and Site 2e	457	0.680	0.010	0.019	0.380	0.013	0.000	--
Average Gol. difference	199.400	0.344	0.016	0.008	0.814	0.028	0.010	0.133
St. dev of Gol. difference	220.907	0.331	0.009	0.009	0.488	0.019	0.005	0.200

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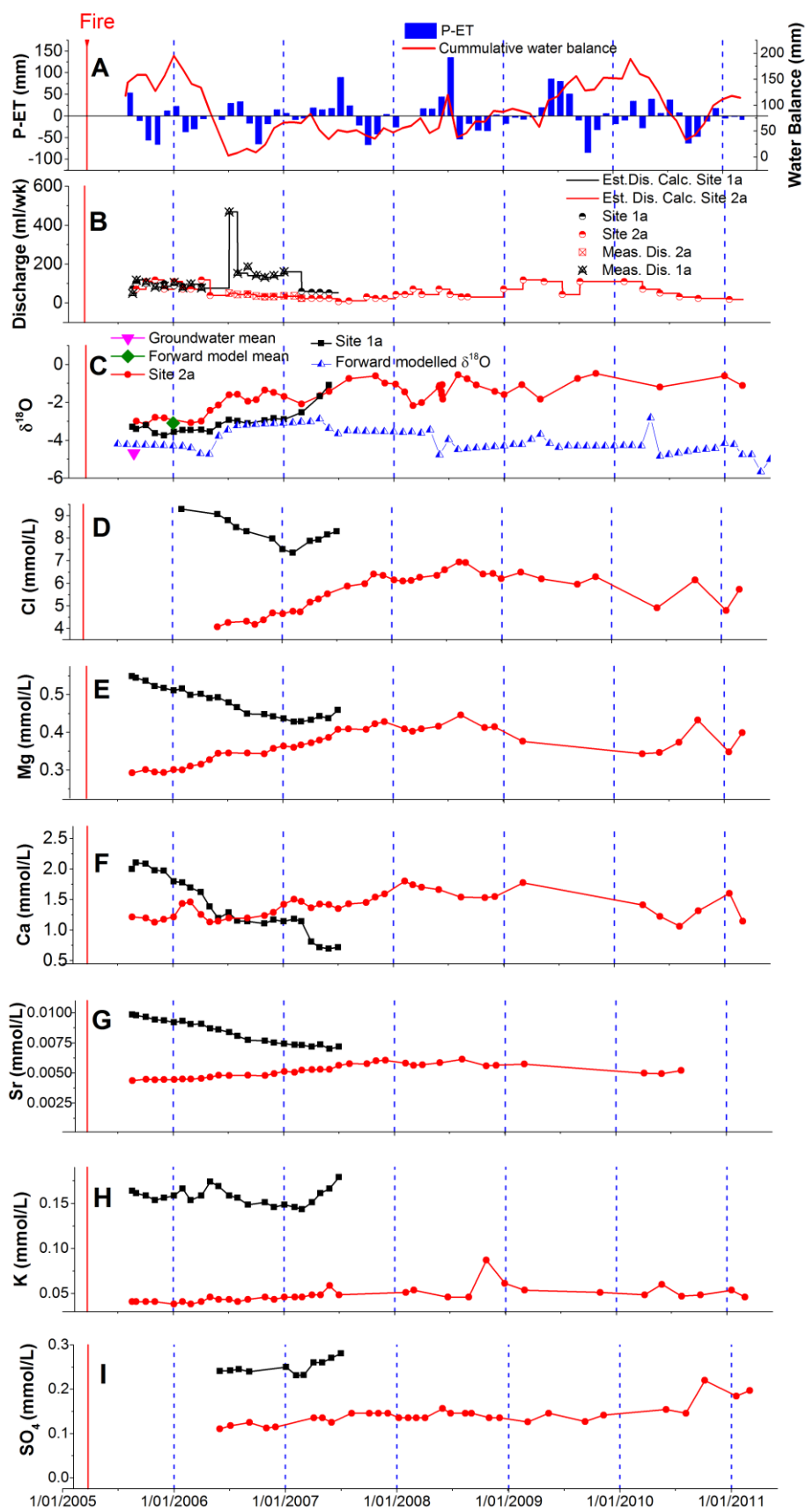
1298 Figure 1. Geographical location of our study site (A), a post fire photograph of the area (B)  
 1299 taken in August 2005 a photo of recovering shrubs and grass post wildfire, and (C) a map of  
 1300 Yonderup cave to scale originally surveyed by Watts and Henley, (1973).

**Comment [GN60]:** Yanchep national park noted on the map as suggested by Reviewer 3 comment Fig. 1

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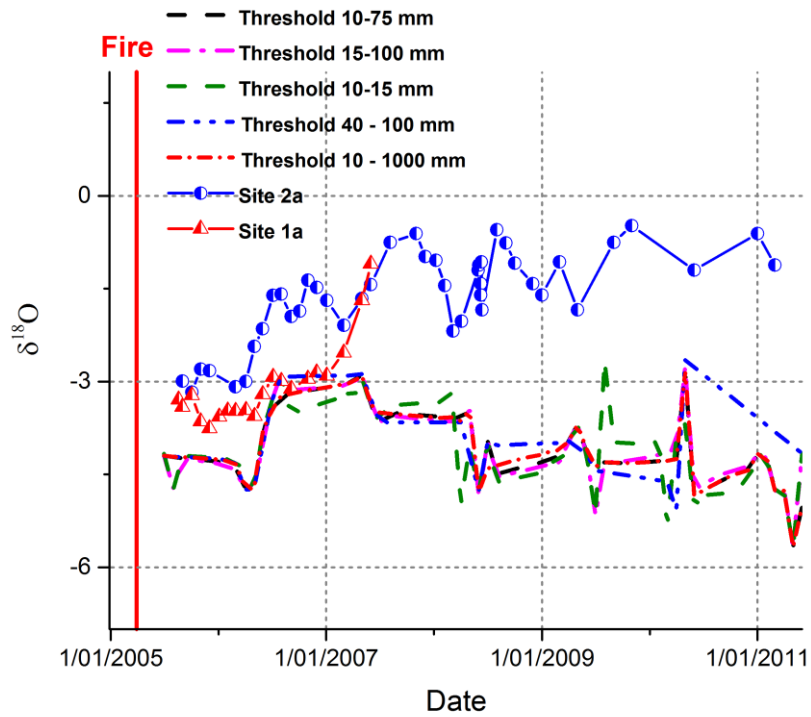
1322 Figure 2. Presents a post-fire time series of data from Site 1a and Site 2a. Note: Site 1a dries  
1323 up in June 2007. (A) Precipitation – actual evapotranspiration (P – AET), shows seasonal  
1324 variations of excess (above threshold) and deficiency (below threshold) on a monthly scale  
1325 overlaid with cumulative surface water balance. (B) Discharge is given in ml/week. Actual  
1326 measured discharge data is given in blue (Site 1a) and Black (Site 2a) while estimated data is  
1327 given in (red) this is then inferred to give measured discharge. Site 1a shows a spike in  
1328 measured discharge in August 2006 and a consequent decrease until the site is dry, while Site  
1329 2a shows little variability in discharge throughout the monitoring period. (C) Shows observed  
1330  $\delta^{18}\text{O}$  composition of cave dripwater from Site 1a and Site 2a with the forward modelled  $\delta^{18}\text{O}$   
1331 (red) and mean modelled  $\delta^{18}\text{O}$  (orange) and long-term groundwater  $\delta^{18}\text{O}$  mean (pink). (D) Cl  
1332 declines at Site 1a until Feb where it shows a slight increase until the drip becomes dry, while  
1333 Site 2a shows a steady increase until in July 2007 where it stabilizes for the remainder of the  
1334 monitoring period. (E) Post-fire response shows a decline in Mg at Site 1a until dry and a  
1335 steady increase at Site 2a until in Dec 2007 where it remains stable. (F) Site 1a shows step  
1336 wise decline in Ca at Site 1a until dry, while at Site 2a a very gradual increase until June 2007  
1337 is seen while the remainder of the monitoring period remains steady. (G) Response for Sr  
1338 shows Site 1a declining and Site 2a peaking in Dec 2007; an identical response to Mg and at  
1339 both sites. (H) K post-fire at Site 1a shows high concentrations, triple that of Site 2a but  
1340 stable, while at Site 2a shows a slight increase over time. (I)  $\text{SO}_4$  both Site 1a and Site 2a  
1341 show a slight increasing trend over time. But Site 1a has more than double the initial absolute  
1342 concentration in comparison to Site 2a, similar to other solutes.

**Comment [GN61]:** We have added the time of the fire to figure three. As suggested by Reviewer 2 and 3.

**Comment [GN62]:** Labels on figure adjusted in response to reviewer 3 comments figure 3B

**Comment [GN63]:** Changed to triple by Reviewer 3 comment L641

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1346 **Figure 3.** Modelled dripwater  $\delta^{18}\text{O}$  outputs under varying thresholds in our forward model

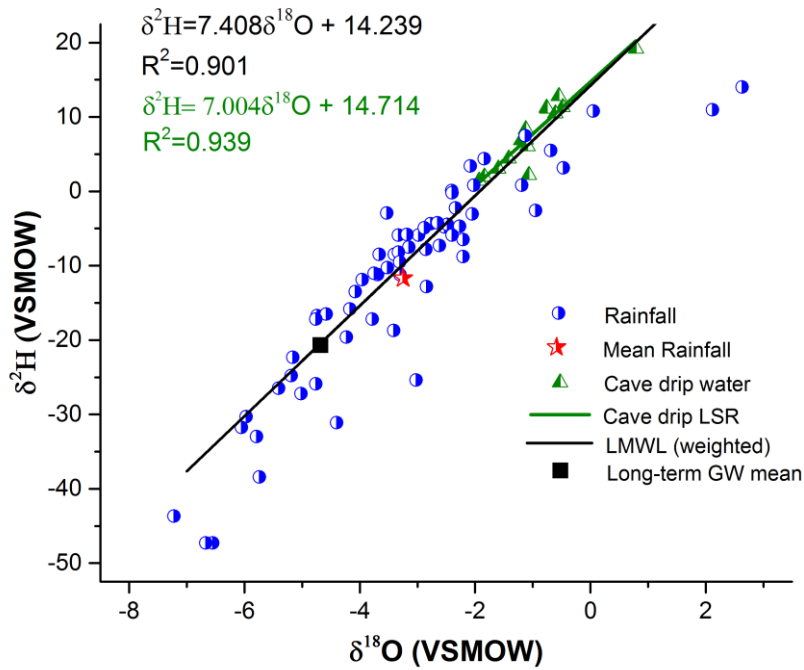
1347 (model from Baker et al., 2010) which accounts for climatic and various epikarst threshold

1348 values that control isotopic values. Given no output matches observed dripwater composition

1349 we can infer that a localised factor has influenced isotopic compositions

**Comment [G64]:** Updated Figure 3 as per response

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1352 Figure 4. Shows compositions of  $\delta^2\text{H}$  against  $\delta^{18}\text{O}$  of our rainfall data (blue) during the  
1353 monitoring period (ANSTO, unpublished), cave dripwater (red), long-term local groundwater  
1354 mean (black) (0 – 10 ka, n=43, Southern Perth Basin from Turner and Thorpe, (2001)), and  
1355 rainfall mean (red). A least squares regression (LSR) is plotted for cave dripwater (red line)  
1356 and falls close to the local meteoric water line (LMWL, black) which is calculated using a  
1357 weighted least squares regression (WLSR) using Hughes and Crawford, (2011).

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**Comment [GN65]:** Moved Mean rainfall to below rainfall and changed colour for cave drip data. In response to Reviewer 3



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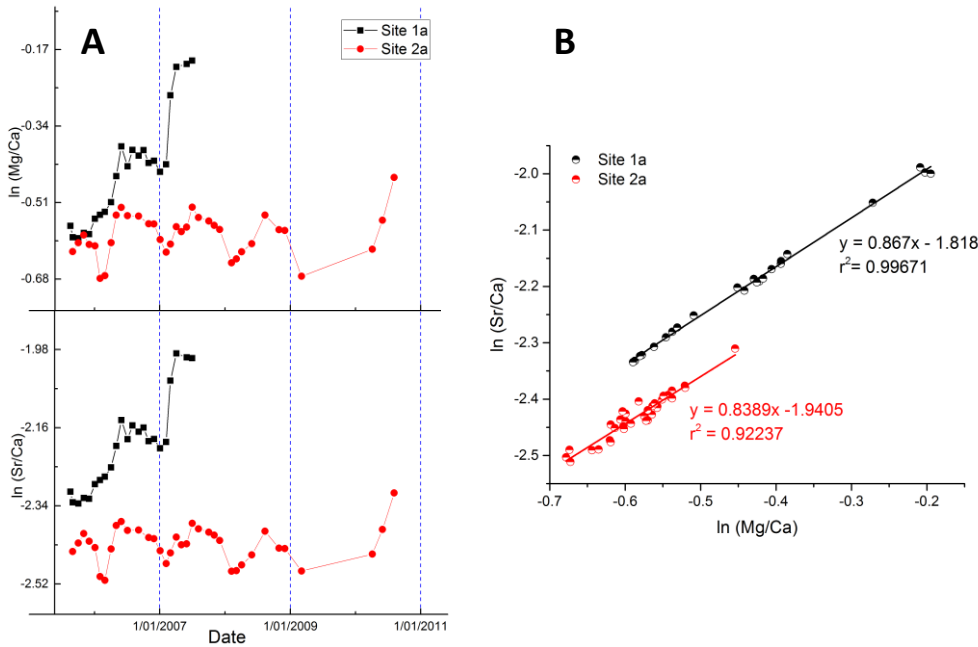
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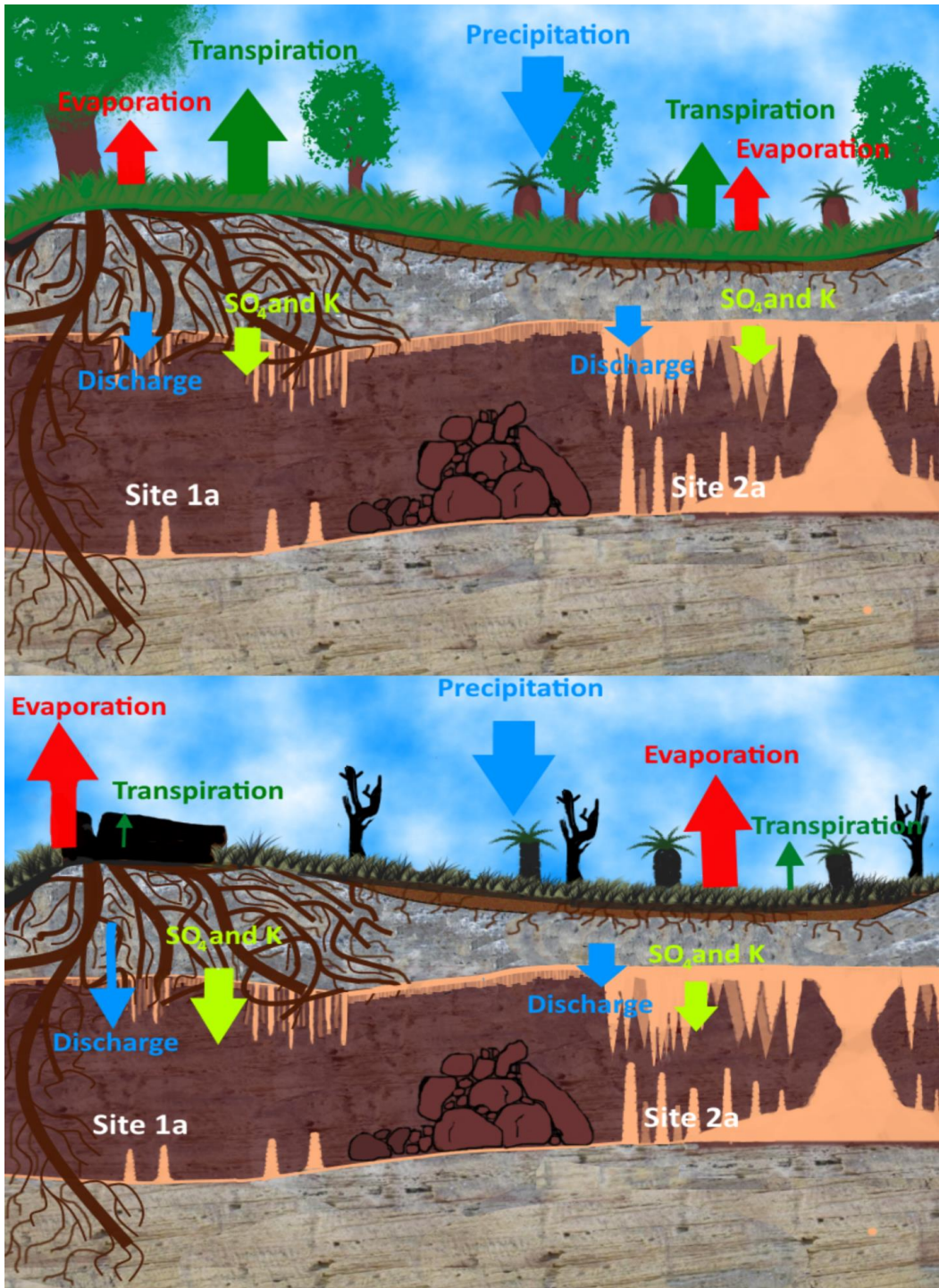
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1369 **Figure 5.** (A) A time series of ln(Sr/Ca) vs. ln(Mg/Ca). Site 1a shows a clear enrichment in  
1370 Mg/Ca and Sr/Ca, or an increase in PCP, post-2007, driven by evaporation. Site 2a on the  
1371 other hand shows quasi-seasonal variational in ln(Mg/Ca) and ln(Sr/Ca), suggesting here PCP  
1372 is likely dominated by seasonality. (B) Shows that at both sites, ln(Mg/Ca) vs. ln(Sr/Ca) falls  
1373 within the diagnostic range of PCP (a slope of + or - 0.88) suggesting PCP is occurring.

**Comment [GN66]:** We have added a Mg/Ca and Sr/Ca time series showing that they covary over time. And at site 1a it increased post-fire. In response to evaporation as suggested in Section 6.2 (in response to reviewer 2 comment Figure 5)



1375 Figure 6) Evaporation (red) increases post-fire at both sites due to a reduction in albedo and  
1376 vegetation cover while precipitation (blue) remains the same and initial transpiration (green)  
1377 decreases, but recovers over time. Site 2a shows higher  $\delta^{18}\text{O}$  and an increase in  
1378 concentrations of solutes including  $\text{SO}_4$  and K (lime) due to evaporation and slow increase in  
1379 transpiration due to vegetation recovery, with cumulative water balance (CWB) remaining  
1380 the same. While Site 1a shows, higher  $\delta^{18}\text{O}$  in response to increased evaporation and a  
1381 decline in solute concentrations in response to increased discharge and a decrease in  
1382 transpiration and removal of nutrients from the surface and subsurface. However, since  $\text{SO}_4$   
1383 and K are from biomass-sourced ash, the tuart tree above this site acts as a source of  
1384 increased  $\text{SO}_4$  and K. Discharge increased immediately (blue). But the drip became inactive  
1385 one year after the fire due to an increase in evaporation, which outweighed the reduction in  
1386 transpiration (green), leading to depletion of the near-surface reservoir feeding Site 1a and an  
1387 in active drip site.

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