

Can mussels be used as sentinel organisms for characterisation of pollution in urban water systems?

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Abstract

Urbanisation strongly impacts aquatic ecosystems by decreasing water quality and altering water cycles. Today, much effort is put towards the restoration and conservation of urban waterbodies to enhance ecosystem service provision leading to liveable and sustainable cities. To enable a sustainable management of waterbodies, the quantification of the temporal and spatial variability of pollution levels and biogeochemical processes is essential. Stable isotopes have widely been used to identify sources of pollution in ecosystems. For example, increased nitrogen levels in waterbodies are often accompanied with a higher nitrogen stable isotope signature ($\delta^{15}\text{N}$), which can then be detected in higher trophic levels such as mussels. The main aim of this study was to assess the suitability of nitrogen stable isotope as measured in mussels (*Mytilus edulis*), as an indicator able to resolve spatial and temporal variability of nitrogen pollution in an urban, tidally influenced estuary (Swan River estuary; Western Australia). Nitrogen concentrations were generally low and nitrogen stable isotope values of nitrate throughout the estuary were well within natural values of uncontaminated groundwater, organic nitrate from soils or marine derived sources, indicating groundwater inflow rather than pollution by human activity was responsible for differences between sites. The $\delta^{15}\text{N}$ signature in mussels was very stable over time within each site which indicated that mussels can be used as time-integrated sentinel organisms in urban systems. In addition, our study shows that the nature of the relationship between $\delta^{15}\text{N}$ in the mussels and the nitrate in the water can

1 provide insights into site specific biogeochemical transformation of nutrients. We suggest
2 that mussels and other sentinel organisms can become a robust tool for the detection and
3 characterization of the dynamics of a number of emerging anthropogenic pollutants of
4 concern in urban water systems.

6 1 Introduction

7 Humans exert a growing impact on the environment supporting them. Today, more than
8 50% of the world's population is living in cities and this percentage is projected to further
9 increase to up to 80% by 2050 (Pickett et al., 2011; United Nations, 2013). Impervious
10 surfaces in cities lead to less rainfall infiltrating the soil. Instead, stormwater runoff is
11 directly transported to waterbodies, polluting them with nutrients, heavy metals, and
12 bacteria (Makepeace et al., 1995; Brezonik and Stadelmann, 2002). Urbanisation has
13 resulted in increased eutrophication of waterbodies leading to deteriorated ecosystems
14 worldwide, reducing natural biodiversity and ecosystem services (Heathwaite, 2010).
15 Environmental management is often hampered by a limited understanding of the temporal
16 and spatial variability of pollution levels, the sources of contamination and the processes
17 within systems that affect the recovery of a system (Kooistra et al., 2001; Scheffer et al.,
18 2001; Lahr and Kooistra, 2010). In addition, the traditional hierarchical water management
19 practices that are still in use around the world have been criticised as being ineffective and
20 leaving little scope for adaptation to changes (Pahl-Wostl, 2007; van de Meene et al.,
21 2011). The current trend to decentralise such urban water management might allow for
22 more local management of water resources, indicating the need for improving our
23 understanding of the variability of pollution levels in a range of urban waterbodies with
24 greater emphasis on local processes.

25 Many urban estuaries are highly impacted by human activity due to direct input of
26 pollutants, such as nitrogen from urban, agriculture and industry areas (e.g., Oczkowski et
27 al., 2008) which can lead to eutrophication. In urban estuaries, tributaries often transport
28 high amounts of nitrogen from the watershed into the estuary, causing water quality
29 problems including toxic bloom development (Hamilton, 2000; Atkins et al., 2001).
30 Nitrogen concentration gradients might develop with higher upstream and lower
31 downstream values, where nutrients are diluted by seawater (Dähnke et al., 2010; Fry et
32 al., 2011). This can lead to a spatial variability of nitrogen concentration within estuaries.

1 Nitrogen pollution can also be highly variable in time with higher nitrogen concentrations
2 in estuaries found during times of high water input by tributaries. Smaller scale variability
3 in temporal and spatial nitrogen concentrations can additionally stem from local
4 differences in hydrological processes (Linderfelt and Turner, 2001), variations in fertilizer
5 use in agricultural areas or temporal failure of septic tank systems leading to leakage of
6 sewage, leading to localised places of concern for water management.

7 Anthropogenic nitrogen and organic pollution of water systems, including the interaction
8 between surface and groundwater, have been successfully investigated using a range of
9 stable isotopes (Sikdar and Sahu, 2009; Yang et al., 2012; Lutz et al., 2013). In addition,
10 stable isotopes have been widely used in purely hydrological studies focused on flow
11 paths, hydraulic residence time and other hydrological dynamics (Clay et al., 2004;
12 Rodgers et al., 2005; Volkmann and Weiler, 2014). Stable isotopes of nitrogen (N), carbon
13 (C), sulfur (S) and oxygen (O) in water and biota have also been applied as an integrated
14 measure of ecosystem processes (Robinson, 2001; Chaves et al., 2003; Pace et al., 2004).
15 Furthermore, the analysis of the nitrogen signature has proven to be an especially
16 powerful tool as an indicator of anthropogenic contamination (Lake et al., 2001;
17 McKinney et al., 2002; Fry and Allen, 2003; Xu and Zhang, 2012) and landuse
18 (Harrington et al., 1998; Broderius, 2013; Carvalho et al., 2015), bearing on the fact that
19 the sources of contamination such as animal manure, sewage, septic waste, some fertilizers
20 carry higher nitrogen signatures values and consequently a higher $\delta^{15}\text{N}$ (Heaton, 1986;
21 Cabana and Rasmussen, 1996; Kellman, 2005; Choi et al., 2007). This signal is then
22 passed on to higher trophic levels up the food chain (Cabana and Rasmussen, 1994;
23 Carvalho et al., 2015): Elevated $\delta^{15}\text{N}$ signals in nitrate have been shown to lead to elevated
24 $\delta^{15}\text{N}$ signals in organisms that directly take up nitrate from the water, such as
25 phytoplankton and microbes (Harrington et al., 1998). These organisms form an important
26 part of particulate organic matter (POM), which serves as food for filter feeders (e.g.,
27 mussels). Mussels that ingest POM with elevated $\delta^{15}\text{N}$ signal will then also show a higher
28 $\delta^{15}\text{N}$ signal.

29 Assessing anthropogenic pollution of a system by directly measuring the isotopic signature
30 of nitrogen containing nutrients (e.g., nitrate, ammonium) or of aquatic short-lived
31 organisms with fast tissue turnover times, such as phytoplankton, may significantly under-
32 or overestimate the average level of pollution, as the result strongly depends on the time of
33 measurement. Bivalves on the other hand, which include the blue mussel are primary

1 consumers with limited movement, and have been suggested as suitable site-specific
2 bioindicators of time-averaged persistence of nutrient pollutants, because their isotopic
3 signature fluctuates less than that of their food sources due to longer tissue turnover rates
4 (Raikow and Hamilton, 2001; Post, 2002; Fukumori et al., 2008; Fertig et al., 2010; Wang
5 et al., 2013). The blue mussel, *Mytilus edulis*, is a common sessile bivalve in estuarine and
6 marine environments that is able to adapt to a wide range of environmental conditions,
7 such as food concentration, temperature and salinity (e.g., Thompson and Bayne, 1974;
8 Widdows et al., 1979; Zandee et al., 1980; Almadavillela, 1984), and that shows low
9 sensitivity to anthropogenic pressures (Mainwaring et al., 2014). As such, this species is
10 able to thrive at different pollution levels and has therefore been used as an indicator
11 species for pollution (Phillips, 1976) and as a model organism for physiological, genetic
12 and toxicological studies (Luedeking and Koehler, 2004) for some time. Earlier studies in
13 polluted freshwater and marine systems found positive relationships between the
14 concentration of nitrogen and the isotopic signature of nitrogen in mussels, and between
15 the isotopic signature of nitrate-N and that of mussels (Cabana and Rasmussen, 1996;
16 McClelland et al., 1997; Costanzo et al., 2001; Anderson and Cabana, 2005; Gustafson et
17 al., 2007; Wen et al., 2010), suggesting that bivalves are suitable indicators of changes in
18 nutrient pollution load from agriculture and wastewater to waterbodies. However, very
19 little information exists on the use of these stable isotopic signatures in urban systems.

20 The main aim of this study was to identify the variability of nitrogen concentration in an
21 urban estuary over time and space and to ascertain the suitability of the isotopic signature
22 ($\delta^{15}\text{N}$) of blue mussel (*Mytilus edulis*) tissue as an indicator of nitrogen pollution in urban
23 water systems. Specifically, we anticipated that (1) a higher input of nitrogen-rich waters
24 upstream would lead to a higher isotopic signatures of nitrate, (2) spatial differences in the
25 level of nitrates in the water would lead to spatial differences in mussel isotopic signature,
26 and (3) the increased distance from the estuary mouth would lead to elevated ^{15}N values in
27 mussels due to elevated ^{15}N inputs from nitrogen-rich waters upstream.

28 29 **2 Materials and Methods**

30 **2.1 Study sites**

31 The study was performed in the lower reaches of the heavily urbanised Swan River estuary
32 that flows through Perth, Western Australia (Fig. 1) (Atkins and Klemm, 1987). The

1 catchment of this estuary is approximately 121,000 km² (Peters and Donohue, 2001) and
2 encompasses urban, rural, agricultural and forested areas. In the urban area, drains contain
3 sewerage and unsewered areas (Peters and Donohue, 2001). The Swan River estuary
4 experienced a major toxic cyanobacterial bloom in 2000, when a large rainfall event
5 increased nutrient concentrations and decreased salinity within the estuary (Hamilton,
6 2000; Atkins et al., 2001), indicating that this estuary is prone to nutrient pollution from
7 the watershed. The Swan River estuary is influenced by mostly diurnal tides with a mean
8 tidal range at the mouth of the estuary of 0.8 m. At the same time, the estuary is seasonally
9 forced with a large discharge of freshwater from the tributaries during the wetter winter
10 months (May to September), and little freshwater discharge during dry summers. This
11 leads to fresh to brackish water in parts of the estuary in winter with a freshwater lens
12 overlying saltwater, and an inland progression of the saltwater wedge, making the estuary
13 a saltwater habitat during drier months (Stephens and Imberger, 1996). The Swan River
14 estuary is permanently open to the ocean and has two major freshwater tributaries, the
15 Swan River and the Canning River (Fig. 1). While there are also several short stormwater
16 drains leading into the lower Swan River estuary that could potentially provide nutrient
17 input into the Swan River estuary from the adjacent land, these drains did not flow during
18 the study.

19 Seven sites within the Lower Swan River estuary were sampled 6 times for blue mussels
20 and 9 times for nutrients, chlorophyll-*a*, temperature, salinity, pH and oxygen during the
21 wetter season (April - November 2010). The sites were jetties at Point Walter (WP) (32° 0'
22 39.23" S, 115° 47' 15.11" E), Minim Cove Park (MC) (32° 1' 21.23" S, 115° 45' 57.38" E),
23 Swan River Canoe Club (SCC) (32° 0' 27.31" S, 115° 46' 18.73" E), Claremont (Cl) (31°
24 59' 23.80" S, 115° 46' 52.97" E), Broadway (BRD) (31° 59' 25.55" S, 115° 49' 5.49" E),
25 Applecross (AC) (32° 0' 17.59" S, 115° 49' 58.29" E), Como Beach (CB) (31° 59' 37.46"
26 S, 115° 51' 10.33" E) (Fig. 1). While MC and SCC are situated at the deeper part of the
27 estuary (depth < 17 m), all other sites are located in the shallower part (depth < 10 m)
28 (Stephens and Imberger, 1996). The jetty at Cl is situated in a shallow bay (depth
29 approximately 2 m) with established seagrass meadows and abundant macroalgae and
30 macrophytes (Department of Water, 2010). Additionally, a one-time marine reference
31 measurement was performed towards the end of the study outside the estuary at Woodman
32 Point Jetty (WO; 32° 7' 26.97" S, 115° 45' 32.10" E) (Fig. 1).

33

1 2.2 Sampling and analyses

2 On each date, sampling was performed 0.5 to 1 h prior to high and low tide at each site,
3 respectively. While mussels were sampled only once per day, all other parameters were
4 sampled at high and low tide. Salinity, pH, water temperature and oxygen were measured
5 at 20 cm depth with hand-held probes (WP-81; TPS-DO₂). At each site, one water sample
6 for quantification of nutrient concentration (TP = total phosphorous, NO_x = nitrate (NO₃) +
7 nitrite (NO₂), NH₄⁺ = ammonium), phytoplankton biomass (as chlorophyll-*a*), and stable
8 isotope analysis of NO₃ (δ¹⁵N, δ¹⁸O) and particulate organic matter (POM; δ¹⁵N) were
9 taken from 10 to 20 cm below the surface and brought back to the laboratory in glass
10 bottles that were stored on ice. Nine blue mussels per site were randomly taken from the
11 pylons of the jetties at each site from between 20 and 40 cm depth and brought into the
12 laboratory on ice in bags containing water from the respective site. There were no mussels
13 at WP in November.

14 In the laboratory, total phytoplankton concentration at each site was measured with a
15 bench top version of the FluoroProbe (bbe Moldaenke, Germany) as μg chl-*a* L⁻¹ (Beutler
16 et al., 2002; Ghadouani and Smith, 2005). Water for quantification of NO_x (LOQ = 0.14
17 μM) and NH₄⁺ (LOQ = 0.21 μM) concentrations was filtered through 0.45 μm syringe
18 filters (Ht Tuffryn, Pall, Australia) and kept frozen until analysis at the Marine and
19 Freshwater Research Laboratory (Murdoch University, Western Australia) using a Lachat
20 Quikchem Flow Injection Analyser. Water for analysis of nitrate δ¹⁵N and δ¹⁸O was
21 filtered through 0.2 μm syringe filters (Ht Tuffryn, Pall, Australia) and kept frozen until
22 analysis at the UC Davis Stable Isotope Facility (Davis, California, USA) using a
23 ThermoFinnigan GasBench plus PreCon trace gas concentration system interfaced to a
24 ThermoScientific Delta V Plus isotope-ratio mass spectrometer (Bremen, Germany), with
25 the bacteria denitrification method (Sigman et al., 2001). All values are reported in per mill
26 (‰) with respect to the international standards (δ¹⁵N: air; δ¹⁸O: Vienna Standard Mean
27 Ocean Water, VSMOW). The limit of quantification for this analysis was 0.71 μM NO₃-N
28 and the external errors of analysis were 0.4 ‰ for nitrate δ¹⁵N and 0.8 ‰ for nitrate δ¹⁸O.
29 Raw water was used for quantification of TP with the ascorbic acid method (APHA,
30 1998).

31 To determine the isotopic composition of nitrogen in particulate organic matter (POM),
32 which is the food source for mussels that presents the direct link between nitrate and the
33 mussels, 0.7 - 2.5 L of water was filtered onto pre-combusted 25 mm GF/C filters

1 (Whatman), which were then dried for 24 h at 60°C and stored in a desiccator until
2 analysis. Harvested mussels were measured and dissected to obtain the foot tissue for
3 stable isotope analysis. The feet of three individuals per site were combined, dried at 60°C
4 for at least 24 h, fully homogenized with mortar/pestle, and stored in a desiccator until a
5 subsample was analysed for mussel $\delta^{15}\text{N}$ and C:N ratio. As 9 mussels per site were
6 collected, this resulted in three replicates for stable isotope analysis per site, with each
7 replicate comprised of the feet of three mussels. This method was adopted from Lancaster
8 and Waldron (2001) as the minimum detectable difference between two populations was
9 negatively associated with the number of replicate samples and the number of individual
10 animals combined in each replicate. Therefore, this method is preferred, when only small
11 differences in the stable isotope signatures are expected. We used foot tissue for the
12 analysis, because it is easy to identify and obtain, and because its $\delta^{15}\text{N}$ value presents a
13 time-averaged value of $\delta^{15}\text{N}$ of the food source. Stable isotope analysis of mussel feet
14 tissue and POM was performed at the West Australian Biogeochemistry Centre
15 (University of Western Australia, Australia) with a continuous flow Delta V Plus mass
16 spectrometer (connected with a Thermo Flush 1112 via Conflo IV) (Thermo-Finnigan,
17 Germany). All values are reported in per mill (‰) with respect to the international
18 standard (air). The external errors of analysis were 0.10 ‰ for $\delta^{15}\text{N}$. To check whether the
19 size of mussels was correlated with their $\delta^{15}\text{N}$, 13 mussels with shell lengths between 30
20 and 54 mm were sampled from MC in July.

21

22 **2.3 Data processing and statistical analyses**

23 Relationships between parameters (i.e. nutrient concentrations, physical parameters, chl-*a*,
24 stable isotope values) and distance to the estuary mouth were analysed with linear
25 regressions. Differences between sites were analysed with one-way ANOVA or Kruskal
26 Wallis one-way ANOVA, in cases where the normality test failed (Sokal and Rohlf, 1995).
27 If significant, the parametric Tukey (equal variances) or the non-parametric Games Howell
28 (non-equal variances) post hoc tests were used to identify which sites were different. The
29 Mann-Whitney U test was used to compare chl-*a* concentrations between high and low
30 tide. All analyses were done with IBM® SPSS® Statistics 20 or Sigma Plot® Statistics 11.0,
31 and significance level was set to $P < 0.05$ unless stated otherwise.

32

1 3 Results

2 3.1 Physicochemical parameters

3 Rainfall was below average in 2010 with 421 mm for the entire sampling period, while the
4 average for this period was 690 mm in the previous 17 years (1993-2009; Bureau of
5 Meteorology, 2016). This resulted in a lower than usual discharge from the tributaries into
6 the estuary with a mean discharge from the Swan River of $7.5 \times 10^5 \text{ m}^3 \text{ d}^{-1}$ in 2010
7 compared to an average discharge of $8.4 \times 10^6 \text{ m}^3 \text{ d}^{-1}$ for the period of 1993-2009 for the
8 same season (min. – max: $1.99 \times 10^6 \text{ m}^3 \text{ d}^{-1}$ (2002) – $2.21 \times 10^7 \text{ m}^3 \text{ d}^{-1}$ (1996) (Department
9 of Water, 2016). This might have contributed to higher salinities throughout the entire
10 estuary during this study than previously reported (Stephens and Imberger, 1997) and no
11 relationship between salinity and distance to the estuary mouth was detected. During high
12 tide, the salinity at all sites was between 24.2 and 32.4 and there was no difference in
13 salinity between sites which can be considered brackish to saline (salinity of seawater is
14 35). Although salinity was not different between sites at low tide either, sites further away
15 from the ocean (AC, CB, BRD) were entirely freshwater between April and June, while
16 saline (mean \pm SE; 27.4 ± 0.4) conditions prevailed at all sites between July and
17 November. There were no differences between sites in temperature (temporal range 12.5 –
18 23°C; Kruskal-Wallis $H = 0.584$, $df = 6$), dissolved oxygen (temporal range 6.4 - 11.6 mg
19 L^{-1} , one-way ANOVA $F_{6,84} = 0.764$; 63 – 124 % *sat.*, one-way ANOVA $F_{6,84} = 0.515$), and
20 pH (temporal range 6.7 - 8.4; one-way ANOVA $F_{6,112} = 0.163$). Total chl-*a* concentration
21 was between 1.4 and 9.5 $\mu\text{g L}^{-1}$ with a mean of 3.9 $\mu\text{g L}^{-1}$ (coefficient of variation = 0.18).
22 Total chl-*a* concentration was similar between sites (ANOVA; $F_{6,70} = 1.45$), and did not
23 differ between low and high tide at any site (Mann-Whitney U Test).

24

25 3.2 Nutrient concentrations

26 Overall, NO_x and NH_4^+ concentrations were low in the Swan River estuary. The
27 concentration of NO_x ranged between below quantifiable limits (LOQ = 0.14 μM) and
28 15.0 μM (median 0.29; mean \pm SD 0.72 ± 1.7), and differed significantly between sites
29 (Kruskal Wallis One way ANOVA, $H = 50.03$, $df = 6$) (Fig. 2). The concentration of NH_4^+
30 ranged between the limit of quantification (LOQ = 0.21 μM) and 2.6 μM (median 0.78;
31 mean \pm SD = 0.85 ± 0.58) and did not differ between sites (Kruskal Wallis One way
32 ANOVA, $H = 7.9$, $df = 6$). On average, NO_x was the dominant N source at MC, SCC and

1 WO, while nitrogen from NH_4^+ was greater at all other sites (Fig. 2) (Kruskal Wallis one-
2 way ANOVA, $H = 59.0$, $df = 6$). Total phosphorous was below or just above the limit of
3 quantification ($\text{LOQ} = 0.32 \mu\text{M}$) throughout the study and did not show any spatial or
4 temporal trend.

5 The concentrations of total dissolved inorganic nitrogen ($\text{TDIN} = \text{NO}_x + \text{NH}_4$) (μM) and
6 NO_x (μM) were higher towards the estuary mouth (Fig. 2), although these relationships
7 were weak (TDIN : $r^2 = 0.113$, $y = -0.186x + 3.69$, $F_{1,117} = 14.86$; NO_x : $r^2 = 0.153$, $y = -$
8 $0.196x + 2.98$, $F_{1,117} = 21.16$) and were driven by site MC only. Ammonium concentrations
9 were not correlated with the distance from the estuary mouth ($F_{1,117} = 0.41$).

10 The TN:TP ratio (weight) of particulate organic matter was between 0 and 6.5 with 84% of
11 the samples being below 2.2 in our study, indicating a high possibility of nitrogen
12 limitation in this system (Redfield, 1958; Geider and La Roche, 2002).

14 3.3 Stable isotope values of NO_3

15 Analysis of the stable isotope composition of NO_3 was limited to a total of 25 samples that
16 fulfilled nutrient concentration requirements for the analysis ($0.71 \mu\text{M NO}_3\text{-N}$). Of these, 9
17 were from MC, 10 from SCC, 2 from AC, 3 from CB, and 1 from WP. Nitrate $\delta^{15}\text{N}$ values
18 varied between -1.3 and 10.4 ‰, while nitrate $\delta^{18}\text{O}$ values ranged between 18.4 and
19 72.9 ‰. Nitrate $\delta^{15}\text{N}$ increased exponentially with increasing NO_x concentration ($F_{1,23} =$
20 10.50) (Fig. 3) and differed between sites (one-way ANOVA; $F_{4,25} = 5.94$). A post-hoc test
21 (Games Howell) indicated that nitrate at MC was ^{15}N enriched (mean \pm SD; 7.92 ‰ \pm
22 2.55 ; $n = 12$) compared to SCC (2.71 ‰ \pm 1.02 ; $n = 10$) and AC (-0.19 ‰ \pm 1.51 ; $n = 2$).
23 There was no temporal trend in nitrate $\delta^{15}\text{N}$ at sites MC and SCC, respectively, which
24 were the only two sites for which sufficient data for such an analysis were available.
25 Nitrate $\delta^{18}\text{O}$ was not significantly different between sites ($F_{4,25} = 0.059$).

27 3.4 Particulate organic matter (POM) $\delta^{15}\text{N}$ values

28 POM $\delta^{15}\text{N}$ values were between 6.2 and 9.9 ‰ with no significant difference between sites
29 ($F_{6,25} = 1.327$). A weak but significant negative relationship between POM $\delta^{15}\text{N}$ values and
30 TDIN concentration was detected ($r^2 = 0.163$, $y = -0.044x + 9.37$, $F_{1,28} = 5.44$), while a
31 significant positive relationship between nitrogen stable isotope signatures of POM and

1 mussels was found ($r^2 = 0.303$, $y = 0.20x + 7.40$, $F_{1,14} = 6.08$) (Fig. 4). The relationship
2 between $\delta^{15}\text{N}$ of POM and nitrate was not significant; however as this calculation was
3 based on only five data points where simultaneous measurements of the two $\delta^{15}\text{N}$ values
4 were available, the value of this result is uncertain.

6 3.5 Mussel $\delta^{15}\text{N}$ values

7 Values of $\delta^{15}\text{N}$ of mussels varied between 6.8 and 10.3 ‰ and the range was therefore
8 smaller than the range seen in nitrate $\delta^{15}\text{N}$ (-1.3 and 10.4 ‰). No significant relationship
9 between mussel length and mussel $\delta^{15}\text{N}$ (linear regression; $F_{1,13} = 2.235$) and no temporal
10 trend in mussel $\delta^{15}\text{N}$ was detected (Fig. 5). Mussel $\delta^{15}\text{N}$ was significantly different
11 between sites (one-way ANOVA; $\delta^{15}\text{N}$: $F_{6,98} = 42.53$) and was negatively correlated with
12 the concentration of total dissolved inorganic nitrogen ($r^2 = 0.486$, $F_{1,5} = 4.73$, $P < 0.1$)
13 (Fig. 6). When site CI was omitted, the strength of the relationship increased ($r^2 = 0.838$,
14 $F_{1,4} = 20.69$, $P < 0.05$), while the relationship was not significant with an r^2 of 0.009 only
15 when site MC was omitted (Fig. 6). Mussel $\delta^{15}\text{N}$ increased significantly with distance from
16 the estuary mouth ($r^2 = 0.563$, $y = 0.12x + 7.74$, $F_{1,110} = 141.65$) (Fig. 7) and showed a
17 significant negative relationship between the $\delta^{15}\text{N}$ values of mussel and nitrate ($r^2 = 0.711$,
18 $F_{2,10} = 24.65$) (Fig. 8).

20 4 Discussion

21 Urban development poses a major threat to aquatic ecosystems, resulting in a range of
22 systems with different impact levels. The management of these waterbodies, whether they
23 are historical, hybrid or novel (Hobbs et al., 2014), requires a detailed knowledge on the
24 complex interactions of processes in these systems. The limited understanding of spatial
25 and temporal variabilities of pollutants is often the major limitation to successful and long-
26 lasting restoration and protection efforts (Kooistra et al., 2001; Lahr and Kooistra, 2010).
27 As such it is essential to develop in-depth knowledge of local processes and pollution
28 levels that will allow a decentralised management approach adapted to local issues (van de
29 Meene et al., 2011).

30 Our study supports this notion by showing that the concentration of nitrates and the
31 nitrogen stable isotope signatures of nitrate and of mussels were different between sites in
32 the Swan River estuary. Site-specific differences in nutrient concentrations can be caused

1 by local input of nutrients or by **spatial** differences in nutrient cycling caused by
2 physicochemical conditions or biological factors (Michener and Lajtha, 2007).
3 Additionally, nutrient input from the watershed often leads to higher nutrient
4 concentrations upstream. During our study, freshwater input into the estuary was weak,
5 leading to the estuary being mainly influenced by ocean water. This might have been the
6 reason that nutrient concentrations did not increase upstream in our study and that nitrogen
7 concentrations were in general low, leading to the conclusion that the Swan River estuary
8 does not represent a highly impacted urban estuary. However, differences in NO_x and
9 TDIN concentrations between sites suggested a significant site-specific input of nutrients
10 into the Swan River estuary. This is supported by the fact that mean nitrogen
11 concentrations at the site closest to the ocean (MC) were higher than the concentrations in
12 the ocean (WO) pointing towards a local input of non-marine NO_x at MC.

13 Earlier studies indicated that the nitrogen stable isotope ratio of dissolved inorganic
14 nitrogen was often higher at sites with high anthropogenic nitrogen pollution (Heaton,
15 1986; Cabana and Rasmussen, 1996). **In the Swan River estuary, NO_3 was enriched and
16 there was a positive relationship between nitrate $\delta^{15}\text{N}$ and the concentration of NO_x
17 throughout the estuary, although this was strongly driven by site MC. Because the isotopic
18 signatures of nitrates were well in the range of values reported for surface water (c. -4 - +9
19 ‰; Xue et al., 2009), uncontaminated groundwater (c. -1 - +8 ‰; Xue et al., 2009),
20 organic nitrate from soils (0 - +10 ‰; Heaton, 1986), pristine streams (+1.8- +2.2 ‰;
21 Harrington et al., 1998), or naturally available marine-derived dissolved inorganic nitrogen
22 (c. 6-8 ‰; Dudley and Shima, 2010), our study does not suggest differences in the level of
23 human impact between sites. Additionally, nitrate $\delta^{18}\text{O}$ values in our study are similar to
24 values indicative of the atmospheric source (+20 - +80 ‰; Kendall, 1998; Xue et al.,
25 2009), suggesting that the higher concentration and enriched signature of NO_x at site MC
26 is unlikely to result from anthropogenic pollution, but might rather be due to addition of
27 NO_x by groundwater inflow, potentially in combination with different productivity or
28 biochemical processes at this site compared to any of the other sites. Overall, results from
29 the stable isotope analysis in combination with nitrogen concentrations indicate that
30 anthropogenic nutrient pollution in the Swan River estuary is low.**

31 **The fraction of NO_x of the TDIN pool (%) was significantly different between sites (data
32 not shown; $y = 0.15x - 6.9$, $r^2 = 0.215$, $F_{1,23} = 6.30$, $P < 0.05$), with site MC having a higher
33 mean fraction (mean = 62.5%) compared to all other sites, except for SCC. An earlier**

1 study by Sugimoto et al. (2009) also found a positive relationship between nitrate $\delta^{15}\text{N}$
2 values and the nitrate fraction in TDIN which they explained by *in situ* isotopic effects
3 during nitrification. However whether higher $\delta^{15}\text{N}$ values of nitrate at MC are related to
4 site specific nitrification rates in our estuary needs further investigation as the $\delta^{18}\text{O}$ and
5 $\delta^{15}\text{N}$ values of nitrate are rather representative of atmospheric NO_3 deposition values
6 (Durka et al., 1994; Fang et al., 2011) and nitrification is likely to play a minor role at
7 ammonium concentrations $<5 \mu\text{M}$ (Day et al., 1989) that prevail in the Swan River
8 estuary.

9 Earlier studies found that nitrogen $\delta^{15}\text{N}$ values are reflected in higher trophic levels in a
10 predictable way with a positive relationship between $\delta^{15}\text{N}$ of nitrate, primary producer and
11 primary consumer (e.g., mussels) (Cabana et al., 1994; Cabana and Rasmussen, 1996;
12 Harrington et al., 1998; Oczkowski et al., 2008; Carvalho et al., 2015). In addition and
13 identical to our study, the range of $\delta^{15}\text{N}$ values for nitrate and POM has been shown to be
14 wider than the range for primary producers, indicating a time-averaging effect in mussels
15 (Gustafson et al., 2007; Wang et al., 2013). Previous studies reported mussel $\delta^{15}\text{N}$ values
16 between +6.6 and +16.7 ‰ in densely populated areas (Cabana and Rasmussen, 1996),
17 polluted inland waterbodies (Wen et al., 2010; Wang et al., 2013) and a eutrophic estuary
18 (Fry et al., 2011). Our values are at the lower end of this range, with mussel $\delta^{15}\text{N}$ values in
19 our study being between 6.8 - 10.3 ‰, indicating the estuary is not highly polluted by
20 wastewater, agriculture or fertilizers. We also found a positive relationship between food
21 (POM) and mussel $\delta^{15}\text{N}$, but a negative relationship between nitrate $\delta^{15}\text{N}$ and consumers
22 (mussels), which was strongly affected by site MC. Such negative relationships were
23 previously found in systems with very high nitrogen concentrations ($\text{DIN} > 40 \mu\text{M}$)
24 (Oczkowski et al., 2008), because in these systems primary producers can be choosy and
25 will preferentially uptake lighter NO_x , leading to a higher fractionation at higher
26 concentrations (Lake et al., 2001; Oczkowski et al., 2008). Therefore, the residual NO_x in
27 those waters retains more ^{15}N -enriched material, leading to a positive relationship between
28 nitrogen concentration and nitrate $\delta^{15}\text{N}$, while consumers which incorporate primary
29 producers will have a lighter signature. Because such fractionation is unlikely at TDIN
30 concentrations below $1 \mu\text{M}$ (Oczkowski et al., 2008), this mechanism is unlikely for most
31 of our sites where mean TDIN concentration was $< 1.5 \mu\text{M}$. This is also supported by the
32 lack of relationship between mussel $\delta^{15}\text{N}$ and TDIN concentration when omitting MC.
33 However, we cannot rule out that this process contributed partially to the low mussel $\delta^{15}\text{N}$

1 values detected at MC as TDIN concentrations were higher at this site with a mean of
2 3.6 μM . An alternative explanation would be that POM could originate upstream where
3 nitrate might have had higher $\delta^{15}\text{N}$ values (not quantified in this study). Upon entering the
4 estuary, POM mixes with estuarine POM, uncoupling the within-estuary $\delta^{15}\text{N}$ nitrate and
5 POM $\delta^{15}\text{N}$ values. This could also explain the strong relationship between $\delta^{15}\text{N}$ in mussels
6 and the distance from the estuary mouth found in our study. Such a strong relationship can
7 be expected in estuaries with low pollution levels due to the aforementioned mixing, while
8 little spatial variability in $\delta^{15}\text{N}$ values of primary consumers can be expected in heavily
9 polluted estuaries due to the dominance of upstream POM, as was shown by Oczkowski et
10 al. (2008).

11 The relationship between mussel $\delta^{15}\text{N}$ and TDIN concentration within the estuary was
12 much stronger when omitting site Cl and not significant when omitting site MC. Site Cl
13 was the shallowest site with a high density of macroalgae and seagrass. These benthic
14 primary producers are known to incorporate nutrients from the groundwater and pore
15 water (Penniford and Davis, 2001). As pore water in the Swan River estuary contains a
16 high concentration of ammonium (Linderfelt and Turner, 2001), this is taken up by the
17 benthic primary producers, and, when recycled, nitrogen with a different $\delta^{15}\text{N}$ value is
18 released into the water column. Therefore, nitrogen $\delta^{15}\text{N}$ in the water column at this site is
19 likely to differ from that of all other sites, which could explain why mussel $\delta^{15}\text{N}$ values at
20 Cl do not fit the general negative relationship. Due to constantly low nitrate concentration
21 at this site, the stable isotope composition of nitrate could not be tested in our study. Site
22 MC was closest to the ocean, was one of the deepest sites and had a higher TDIN
23 concentration compared to all other sites, which in turn did not show differences in TDIN
24 concentrations between them. This emphasises that the differences in mussel $\delta^{15}\text{N}$ between
25 sites detected in our estuary might rather reflect site-specific nutrient cycling processes
26 than nitrogen pollution itself.

27 Fluctuation of mussel $\delta^{15}\text{N}$ at each site over time was low compared to the differences
28 between sites, indicating that observed differences between sites prevailed and were not
29 obscured by time effects. This is important for assessing site-specific processes and source
30 inputs and highlights that mussels can be used as time-integrated sentinel organisms in
31 urban systems. The limited temporal variation likely reflected the physiochemical state of
32 the system during the study period; in our study, the estuary was dominated by marine
33 influences due to reduced river discharge. This might have further resulted in a dampening

1 effect of possible fluctuations of the nitrate $\delta^{15}\text{N}$ value caused by changes in watershed
2 input. Our results therefore highlight that while high seasonal variations of stable isotope
3 signature in mussels can be connected to seasonal changes in watershed input and
4 chemistry in large rivers (Fry and Allen, 2003), this is less pronounced in tidally
5 influenced estuaries or during drier conditions with low freshwater input.

6 7 **5 Conclusion**

8 The findings of our study corroborate that stable isotope analysis is a valuable tool for
9 identifying spatial variability of nutrient pollution and local processes in an urban, tidally
10 influenced estuary. As such, stable isotope analysis of a model organism, such as the blue
11 mussel can deliver essential information for future decentralised water management
12 practices that are focused on local process understanding. We propose to further
13 investigate its use for assessing the pollution by co-occurring non-nutrient pollutants, such
14 as oils and heavy metals, which are entering waterbodies simultaneously with nutrients
15 during stormwater events and which are critical in urban systems.

16 Based on nutrient concentrations and stable isotope analysis, our data provide detailed
17 evidence that the lower Swan River estuary did not present a highly impacted urban
18 estuary during our study. The nitrate stable isotope signature in the water suggested that
19 the higher concentration of nitrate at two sites (MC, SCC) were due to a natural input of
20 nitrate uncontaminated groundwater (Xue et al., 2009) rather than human pollution. The
21 stable spatial differences in mussel $\delta^{15}\text{N}$ values over time highlight the value of this
22 organism as a bioindicator of spatial water quality assessment. The negative trends
23 between mussel $\delta^{15}\text{N}$ values and nitrate concentration or nitrate $\delta^{15}\text{N}$ values emphasize that
24 mussels might not be good indicators for NO_3 sources in systems with low pollution
25 levels. Instead, the small differences in mussel stable isotope signatures might reflect
26 differences in site specific nutrient cycling caused by physicochemical conditions or
27 biological factors rather than nitrogen pollution. This is important information for local
28 management, but would have gone undetected at high pollution levels as the larger
29 deviations of nitrogen stable isotope values would have made such small differences in
30 mussel values invisible. In addition, we advocate future studies in similarly (low) polluted
31 systems that include stable isotope analysis of other food web end-members and nutrients
32 of the groundwater, to develop baselines of spatial natural isotopic variability in urban

1 aquatic systems which will help identifying the importance of local biogeochemical
2 processes for pollution control.

3 In conclusion, this work shows the value of using stable isotope analysis as an integrative
4 tool to establish an understanding on local processes and pollution levels in aquatic
5 systems. With an increasing importance of managing urban aquatic systems sustainably,
6 our work presents an important proof-of concept study in this context. In addition, we
7 propose that it could help to define divisions in tidal estuaries based on natural
8 characteristics and the human dimension that are meaningful for monitoring and
9 management and for which reference conditions have to be identified (Ferreira et al.,
10 2006).

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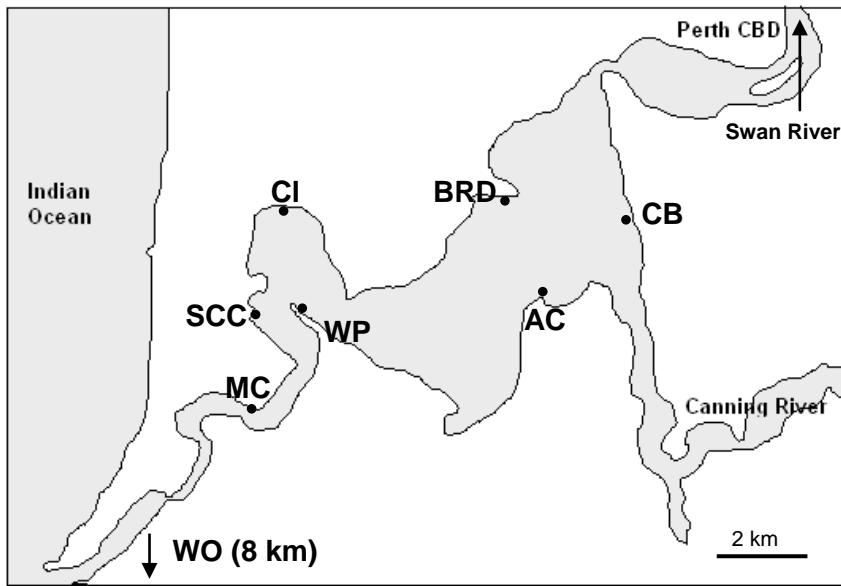
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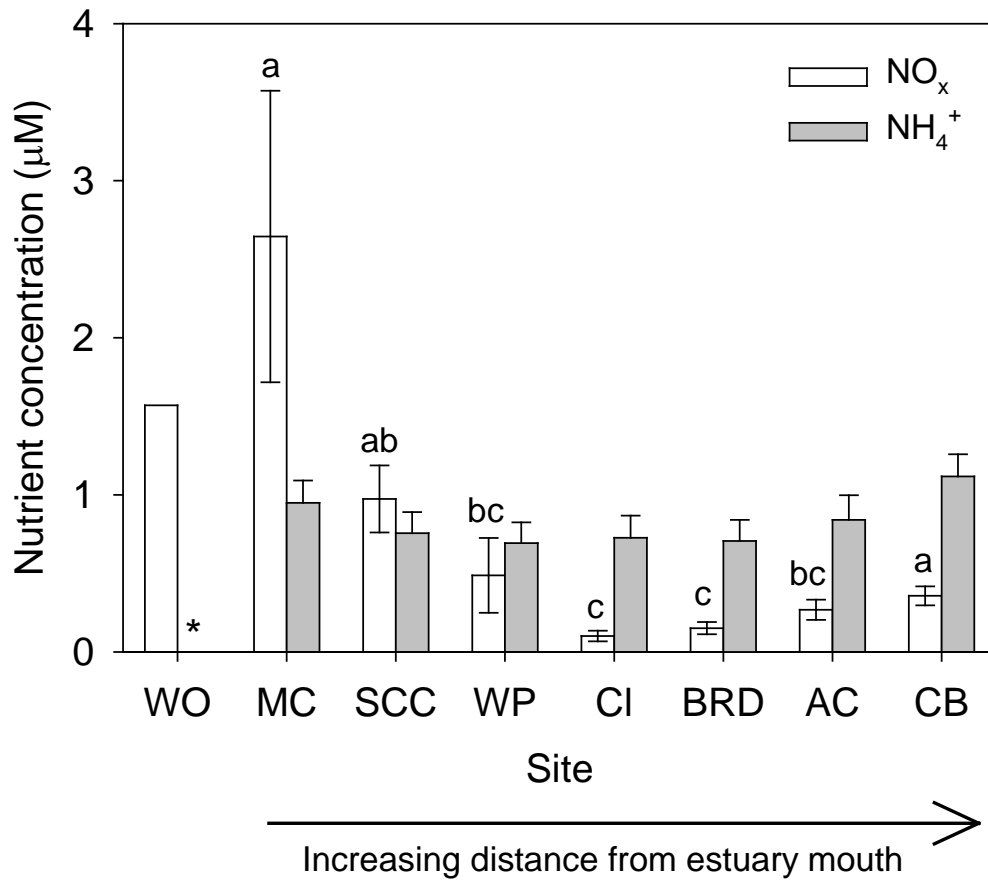
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4 Figure 1. Map indicating the 7 sampling sites (jetties) within the Lower Swan River
5 estuary, Perth, Western Australia. AC = Applecross, BRD = Broadway, CB = Como
6 Beach, CI = Claremont (Freshwater Bay), MC = Minim Cove, SCC = Swan River Canoe
7 Club, WP = Point Walter; the ocean reference site was located 8 km south of the estuary
8 mouth (WO = Woodman Jetty).

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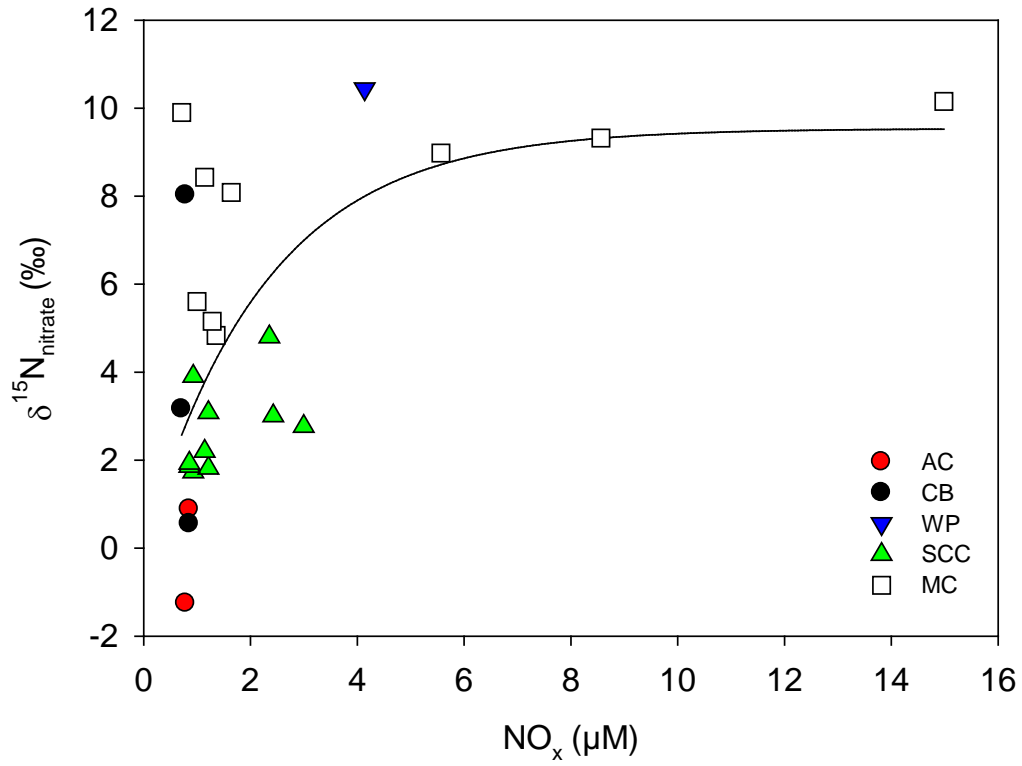


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2 Figure 2. Mean concentration of NO_x and NH₄⁺ (µM) at each site. Letters indicate
 3 differences between sites for NO_x concentrations, with sites sharing the same letter being
 4 not significantly different. Error bars represent one standard error (N = 17). Asterisk at
 5 WO indicates that mean value of NH₄ was below the limit of quantification.

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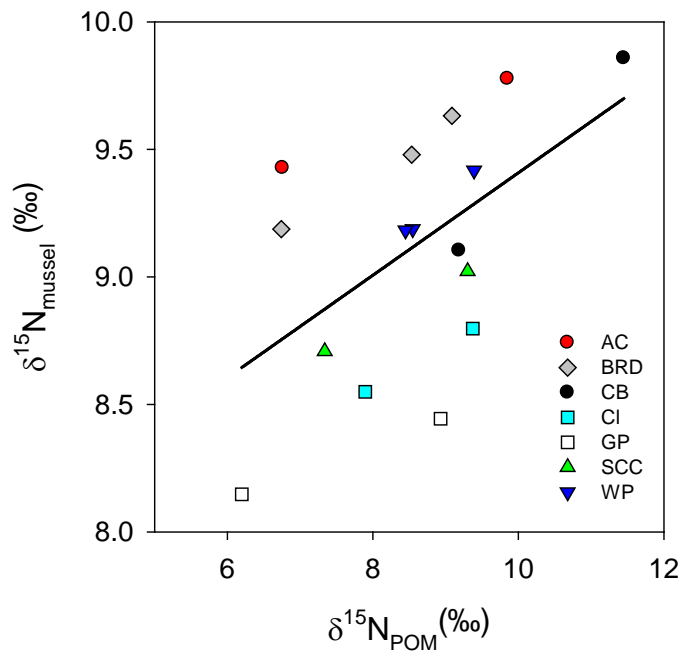
3 Figure 3. Relationship between nitrate $\delta^{15}\text{N}$ (‰) and the concentration of NO_x (μM)

4 ($r^2 = 0.313$, $y = 9.54(1 - e^{-0.44x})$, $P < 0.05$).

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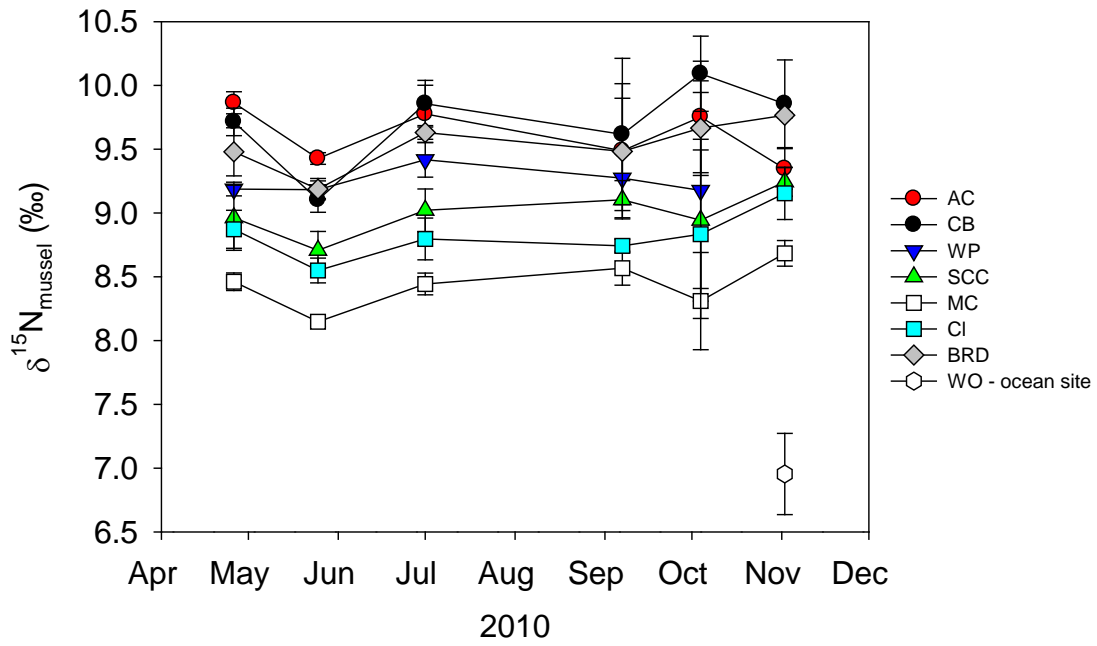
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2 Figure 4. Relationship between mean $\delta^{15}\text{N}$ of POM and mussel (‰). Linear regression is
 3 calculated using all data points ($r^2=0.303$, $y = 0.20x + 7.40$, $F_{1,14} = 6.08$, $P < 0.05$).

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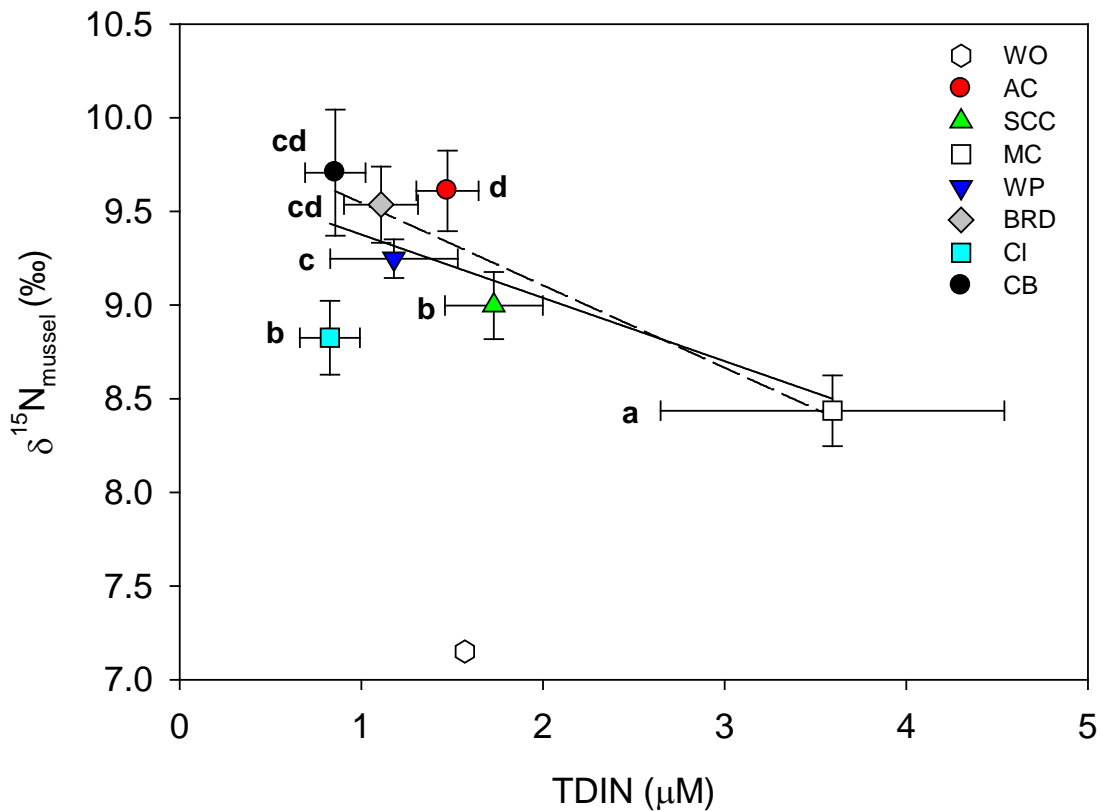


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2 Figure 5. Mean $\delta^{15}\text{N}$ mussel signature (‰) at each site over time. Error bars represent
 3 standard deviations of $N = 3$ for April to July and WO, and $N = 2$ for September to
 4 November 2010.

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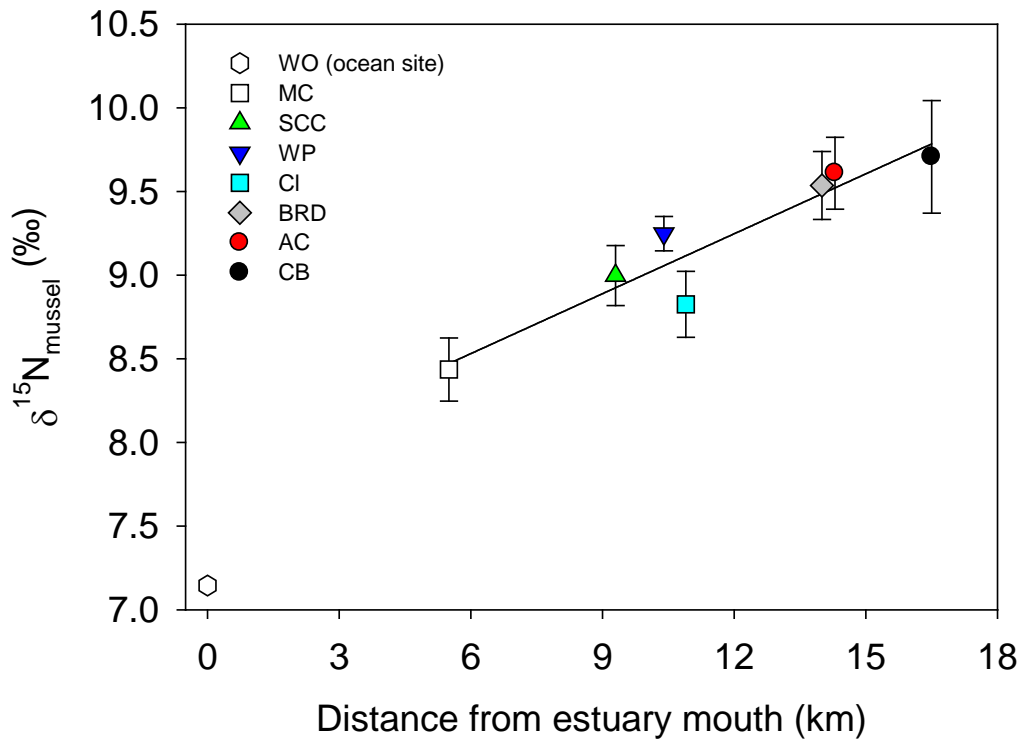


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2 Figure 6. Relationship between mean mussel $\delta^{15}\text{N}$ (‰) and total dissolved inorganic
 3 nitrogen (TDIN) (μM). Error bars represent standard deviation for mussels ($N = 6$ for all
 4 sites except for WP where $N = 5$) and standard error of for TDIN ($n = 17$). The solid line
 5 represents the relationship calculated for all sites ($r^2=0.486$, $y=-0.338x+9.71$), the broken
 6 line when site CI is omitted ($r^2=0.838$, $y=-0.440x+9.98$). Letters indicate differences in
 7 $\delta^{15}\text{N}_{\text{mussels}}$ (ANOVA with Games Howell post hoc test), with sites sharing the same letter
 8 being not significantly different. **WO was not included in the regressions.**

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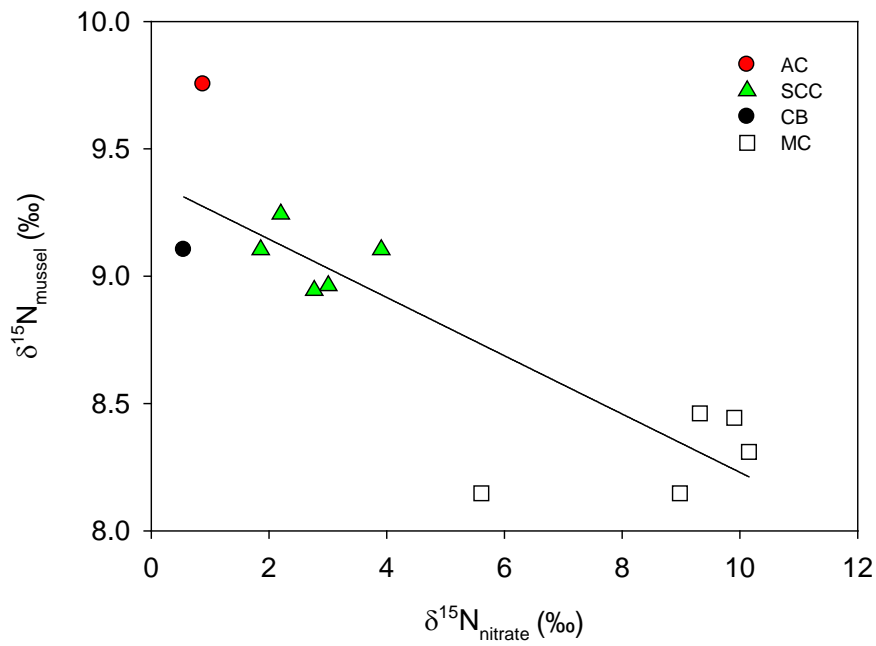


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2 Figure 7. Relationship between mean $\delta^{15}\text{N}$ of mussels (‰) and distance of sites from
 3 estuary mouth ($r^2 = 0.563$, $y = 0.12x + 7.74$). Error bars represent standard deviation of
 4 $N = 6$ for all sites except for WP where $N = 5$.

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2 Figure 8. Relationship between nitrogen stable isotope signature of mussel and nitrate in
 3 the water ($r^2 = 0.711$, $y = -0.114x + 9.37$).

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