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**Quantifying  
heterogeneous  
transport in the field**

D. Schotanus et al.

# Quantifying heterogeneous transport of a tracer and a degradable contaminant in the field, under two infiltration rates

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Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



## Abstract

To examine the persistence of preferential flow paths in a field soil, and to compare the leaching of a degradable contaminant with the leaching of a non-degradable tracer, we did two field experiments, using a multicompartment sampler. The first experiment was done during the snowmelt period in early spring, characterized by high infiltration fluxes from snowmelt. The second experiment was done in early summer with irrigation to mimic homogeneous rainfall. In the second experiment, the soil was warmer and degradation of the degradable contaminant was observed. For both experiments, the highest tracer concentrations were found in the same area of the sampler, but the leached tracer masses of the individual locations were not highly correlated. Thus, the preferential flow paths were stable between seasons. With a lower infiltration rate, in the second experiment, more isolated peaks in the drainage and the leached masses were found than in the first experiment. Therefore it is concluded that the soil heterogeneity is mainly caused by local differences in the soil hydraulic properties, and not by macropores. With higher infiltration rates, the clustering of high and low leaching cells was higher. The leached masses of the degradable contaminant were lower than the leached masses of the non-degradable tracer, but the masses were highly correlated. The first-order degradation rate was  $0.02 \text{ d}^{-1}$ . The dispersivity varied between 1.9 and 7.1 cm. Soil heterogeneity is the main reason for the heterogeneous water flow and solute transport in this soil. Heterogeneous melting of snow does not influence the heterogeneous flow in the soil much at this scale.

## 1 Introduction

Preferential flow can lead to rapid transport of nutrients and chemicals through the unsaturated zone, which can result in groundwater contamination. Preferential flow processes are generally studied at multiple spatial scales, mostly microscale, core, profile and landscape scale (Allaire et al., 2009). Soil heterogeneity hinders the prediction of

HESSD

9, 4827–4868, 2012

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



---

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

---

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



5 the movement of mass in field situations, because it often results in faster movement of water and solutes than would be expected from the soil matrix properties. When average parameters are used for soil and climatic properties, contaminant leaching may be underestimated by a model (Allaire et al., 2009). Preferential flow can be included

10 in models, for which parameters are needed to quantify the importance of preferential flow in a field soil (Feyen et al., 1998). When one wants to model solute transport, it is important that the main process causing preferential flow is identified. As the soil is a complex system, field experiments can give better insight in which processes are important for the occurrence of preferential flow. With this knowledge, a better prediction

15 can be made of the potential risk of groundwater contamination for a particular soil and contaminant.

A common conceptual approach is to distinguish two flow domains in the soil: macropore flow and matrix flow (Feyen et al., 1998). Macropore flow can transport infiltrating water and contaminant fastly, compared to matrix flow. As a result, when the main part

20 of the water and solutes is transported by macropores, the breakthrough curve (BTC) is skewed, the peak concentration is reached rapidly, and tailing is long (Boll et al., 1997). When matrix flow dominates, the BTC is less skewed, and the peak concentration is reached later than with macropore flow.

25 The partitioning of water between macropores and matrix depends on the flow rate. Quisenberry et al. (1994) found that macropore flow occurred with high flow rates. With lower flow rates, macropore flow diminished, but the general spatial pattern was the same as for higher flow rates. At high flow rates, the soil moisture content is generally higher, resulting in more filled pores and more connections between these pores (Williams et al., 2003). Besides in soils with macropores, channelling can also occur in soils where the saturated conductivity varies in space. When the flow rate is high compared to the saturated hydraulic conductivity of the soil, most of the soil is saturated. Under these conditions, coarse textured regions are better conducting than fine textured regions (Roth, 1995). Therefore, coarser textured regions are likely to develop flow channels with a vertical orientation under a high infiltration rate. With a water flux

that is much lower than the saturated hydraulic conductivity, flow channels will also develop, but then in the fine textured regions instead of the coarse textured. This was experimentally supported by Lennartz et al. (2008), who found that the hydraulic conductivity was higher in fine textured regions, with a low flow rate compared to the saturated hydraulic conductivity. Quisenberry et al. (1994) and Williams et al. (2003) do not report the saturated hydraulic conductivity of their soils, which therefore cannot be related with the flow rate.

Preferential flowpaths may change in time due to bioactivity, or due to complete saturation or drying of the soil. Preferential flowpaths may originate at locations where the soil is wetter than the surrounding soil, at a certain moment. When the soil is completely saturated or dried, this type of preferential flowpaths may originate at another location later on (Öhrström et al., 2004). Buchter et al. (1995) performed an experiment twice on the same soil column, with complete saturation and drying in between. The velocities and the dispersion coefficients were highly correlated between both experiments. Thus, under the same conditions, the flow pattern was an intrinsic soil property. Also Williams et al. (2003) and Lennartz et al. (2008) found that preferential flow paths were stable.

The travel time distribution in the soil quantifies preferential solute transport, and is especially relevant for degradable solutes (Jury and Gruber, 1989). This travel time distribution can be used to estimate the degree of degradation and thus leaching of degradable contaminants. The rapid transport in macropores, compared to the matrix, leads to a higher leaching of a degradable contaminant when macropores exist than when only matrix flow occurs (Pot et al., 2005). Most studies investigated the leaching of tracers, while the effect of the travel time on the biodegradation of degradable contaminants is highly relevant for the potential contamination risk for natural systems.

When biodegradation of contaminants is studied, the experimental conditions should resemble the natural field conditions as closely as possible, because the degradation rate caused by the micro-organisms depends on temperature, and soil moisture content, amongst others (Stotzky, 1997). Furthermore, in a structured soil, biodegradation

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



---

**Quantifying  
heterogeneous  
transport in the field**

---

D. Schotanus et al.

---

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

may be lower than in a mixed sample, because micro-organisms or the substrate cannot enter all aggregates (Chenu and Stotzky, 2002). Also when soil heterogeneity is quantified, the soil structure should be intact. Most studies have been done in the laboratory, under different conditions than in the field (Quisenberry et al., 1994; Buchter et al., 1995; Pot et al., 2005) and hence may under- or overestimate the soil heterogeneity. Estimations of the effect of soil heterogeneity on contaminant leaching may be more realistic in the field, using wick samplers (Holder et al., 1991; Boll et al., 1997). A disadvantage of wick samplers is that the pressure that is applied to the soil is constant. Furthermore, the spatial resolution may be too coarse (around  $100\text{ cm}^2$ ) to capture all small scale flow processes (Poletika and Jury, 1994; Buchter et al., 1995). Bloem et al. (2010) developed a multicompartment sampler (MCS), in which the pressure is adjusted according to the pressure in the surrounding soil. The spatial resolution of the MCS is high: fluxes through  $3.15 \times 3.15\text{ cm}^2$  can be measured. This instrument was used in a field experiment (Bloem et al., 2009), studying the leaching of a tracer. The spatial pattern in leaching of a degrading contaminant might be different than a non-degrading tracer as a result of differences in travel times.

Examples of degradable contaminants for which a travel time distribution in the unsaturated zone may help to estimate the leaching to the groundwater are easily degradable contaminants like pesticides. Another easily degradable contaminant is propylene glycol (PG) which is used in de-icing chemicals at airports (Jaesche et al., 2006; French et al., 2001). At Oslo airport in Norway de-icing chemicals are used to remove snow and ice from airplanes before departure during winter time. Although this is done on a platform to collect the de-icing chemicals, a fraction of these de-icing chemicals can contaminate the snow surrounding the runway. During snowmelt, the infiltration rates are high, and the de-icing chemicals can be transported rapidly through the coarse textured soil on which the airport is located (French and Van der Zee, 1999). Groundwater contamination with the de-icing chemical should be avoided, as the airport is located on a large unconfined aquifer (French et al., 2001). Propylene glycol is degradable by micro-organisms, and non-adsorbing to soil particles (French et al., 2001).

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**Quantifying heterogeneous transport in the field**D. Schotanus et al.

---

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Snowmelt may add extra heterogeneity to the water flow, as a snow cover acts as a porous medium, in which meltwater can flow in preferential paths (Waldner et al., 2004; Marsh, 1991). Furthermore, the rate of snowmelt can differ locally which introduces more heterogeneous flow at the plot scale, with horizontal differences in infiltration (French and Van der Zee, 1999).

It has not been investigated often which particular process is responsible for the preferential leaching of a tracer in a field soil. Furthermore, experiments on preferential leaching usually focus on non-degradable tracers, while the preferential leaching of a degradable contaminant might be different. We did two field experiments, using the MCS, to quantify the spatial variability in solute leaching, to quantify possible seasonal effects, and to compare the leaching of a non-degradable tracer with degradable PG. The first experiment was done during the snowmelt period in early spring. This experiment was characterized by high infiltration fluxes from the meltwater and a frozen soil. The second experiment was done in late spring/early summer with irrigation to mimic homogeneous rainfall with known infiltration amounts. In the second experiment, the soil was warmer and degradation of the degradable contaminant was observed.

With the two experiments we answer the following research questions: (1) How persistent is preferential flow in the field? (2) How different is the spatial variability in leaching for a tracer and a degradable solute? (3) What determines the heterogeneous flow in this soil: soil heterogeneity, the infiltration rate, or snowmelt heterogeneity?

## 2 Materials and methods

### 2.1 Experiment

To measure spatial and temporal variability in solute leaching and drainage a multicompartiment sampler (MCS) (Bloem et al., 2010) was installed at the field station Morepen, near Oslo Airport, Norway. The MCS has a size of  $31.5 \times 31.5 \text{ cm}^2$ , and consists of 100 separate drainage collectors. The surface of the MCS consists of porous metal

plates, to which pressure can be applied. From a trench (French et al., 1994) a horizontal tunnel was dug, leaving the soil above this tunnel undisturbed. The tunnel was secured with a wooden frame, to prevent the surrounding soil from collapsing. At 68 cm distance from the trench wall the MCS was installed at 51 cm below soil surface (Fig. 1).

A 2 mm thick layer of wetted soil from the tunnel was applied to the surface of the MCS, to ensure a good contact between the MCS and the soil above it. The wooden frame was removed, and the tunnel was backfilled, to avoid boundary effects. Four tensiometers were installed near the trench wall, at 51 cm depth. The average pressure head measured by the four tensiometers was applied to the MCS, plus 15 cm extra pressure head to compensate for a pressure head drop in the porous metal plates (Bloem et al., 2009). The pressure in the MCs was variable. The cell numbering of the MCS is given in Fig. 2. The field station is located in a flat area with coarse glaciofluvial sediments (sand and gravel) (French and Van der Zee, 1999). The soil is an Entic Haplorthods (French et al., 2001), soil properties are given in Table 1. The soil surface was covered with short grass.

Two experiments were done: one during snowmelt (26 March–23 May 2010) and one with irrigation (23 May–4 July 2010). For the snowmelt experiment  $1092 \text{ g m}^{-2}$  propylene glycol (PG) and  $10 \text{ g m}^{-2}$  bromide was diluted in  $2 \text{ l m}^{-2}$ , and sprayed uniformly on top of an undisturbed snow cover (26 March 2010). De-icing fluid (Kilfrost, 2012) containing PG was diluted to reach this applied mass. The application area was large enough to ensure that boundary effects can be neglected. In the snowmelt experiment the infiltration originated from snowmelt or rainfall. Snowmelt was measured daily by measuring the depth and equivalent water depth of the snow cover at three locations at the field station. At these locations, no de-icing chemical was applied. The snow in the application area melted faster than at these locations, due to the de-icing chemical. At 12 April the snow was patchy above the MCS, and at 13 April all snow above the MCS was melted. From the daily difference of the equivalent water depth of the snow cover, and the rainfall, the infiltration per day is calculated (Fig. 3). The infiltration originating from the snowmelt occurred 1 or 2 days earlier in the area where PG was applied, than

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



in the area where the snowmelt was measured. We will ignore this difference, as it is not important for the current data analysis. No ice layer was found below the snow cover, at the soil surface. This means that the possible heterogeneous infiltration of meltwater was not attributable to ice.

5 For the irrigation experiment  $1103 \text{ gm}^{-2}$  PG and  $10 \text{ gm}^{-2}$  bromide was diluted in  $51 \text{ m}^{-2}$ , and sprayed uniformly on the soil surface at 23 May 2010. Also in the irrigation experiment, the application area was large enough to ensure that boundary effects can be neglected. The irrigation scheme and the daily mean air temperature are given in Fig. 4. Initially, the irrigation rate was based on the measurements from the snowmelt  
10 experiment, such that the irrigation experiment could be finished in the limited time that was available for it. However, during the irrigation experiment, the transport of bromide was found to be too slow. Therefore, the irrigation rate was increased from 2 June onward. In between the irrigations, the soil surface was covered with plastic, to minimize evapotranspiration and to avoid the infiltration of rainwater. Using a nozzle, the soil surface was irrigated with tapwater, except on 12 June. This was a rainy day and 7.8 mm  
15 of rainwater infiltrated, while the plastic was removed. On 13 June, 7.5 mm water was irrigated. The irrigation rate was approximately  $6.6 \text{ mm h}^{-1}$ . Neither surface runoff, nor ponding was observed. Between 31 May and 2 June, evaporation was measured in a pan under the plastic and was  $1 \text{ mm d}^{-1}$ . These were warm days, the average evapotranspiration rate during the entire irrigation experiment probably was lower.

20 Drainage from the MCS was collected in the trench, with a frequency that depended on the amount of drainage. For the snowmelt experiment, samples were taken daily from 1 April until 15 April. Then, samples were taken at 17 and 21 April, and 9 and 23 May. During the irrigation experiment, samples were taken every second day, on the same day as, and prior to, the irrigations. After collecting, the samples were weighted  
25 to determine the volume of drainage. If a sample was smaller than 4 ml, it was stored cool. In the next sampling round, the sample was added to this stored small sample. In samples larger than 4 ml, the bromide concentration was measured with an Orion 9635BNWP Bromide Ion Selective Electrode. Besides, in the irrigation experiment, the

---

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

---

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)



PG concentration was measured with a GC analyser. We assumed that degradation of PG during the snowmelt experiment was low, due to the low temperature and high PG concentration in the applied solution (Jaesche et al., 2006). Therefore, the PG concentration was not measured in the snowmelt samples. At the end of the experiment, all samples that still were smaller than 4 ml after the final sampling round were mixed, and Br and PG were measured in this mixed sample.

## 2.2 Data analysis

The experiment lead to a data set with volumes of drainage, and leaching concentrations for each sampling round and each cell. With drainage, we refer to water volumes that were collected, while we use leaching to refer to solute masses. To quantify how the drainage, or the leaching in a particular cell is related to the drainage, or leaching in a neighbouring cell, we calculate the spatial autocorrelation using Moran's  $I$  (Strock et al., 2001):

$$I = \frac{n}{S_0} \frac{\sum_i \sum_j W_{ij} (X_i - \bar{X}) (X_j - \bar{X})}{\sum_i (X_i - \bar{X})^2}, \quad S_0 = \sum_i \sum_{j \neq i} W_{ij} \quad (1)$$

where  $n$  is the number of cells,  $W_{ij}$  is a measure of the neighbouring of the cells  $i$  and  $j$ , and  $X_i$  is the measured total drainage, or solute leaching in cell  $i$ . Cells are neighbours when they share one side. In  $I$  both the values of drainage and leaching, and the spatial information, with the weighing  $W_{ij}$ , are included.  $I$  varies between  $-1$  and  $1$ , where  $1$  means that the values of neighbouring cells are perfectly positively autocorrelated. When there is no autocorrelation, the expected value of  $I$  is  $-1/(n-1)$ , which is  $-0.010$  with 100 cells.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



## 2.3 Results and discussion

### 2.3.1 Stability of the spatial patterns of drainage and concentration in time

We will start with describing the spatial patterns of drainage and solute concentrations to determine the stability of these patterns during and between the snowmelt and the irrigation experiment.

Figure 5 shows the spatial distribution of the volume of drainage, and of the bromide concentration during the snowmelt experiment at five selected cumulative drainage depths. The cumulative drainage depth is used as a time axis instead of the day (Wierenga, 1977), to facilitate a comparison with the irrigation experiment, which had a different infiltration rate. The bromide concentration is scaled with the applied mass of bromide per  $m^2$ , to facilitate a comparison with propylene glycol (PG) during the irrigation experiment later on. The initial bromide concentration was not used for scaling, because the applied bromide was diluted with meltwater from the snow.

At 13 mm after solute application, drainage started in the left middle part of the sampler. At 66 mm, most cells drained water, but the highest water fluxes were still found in the left middle part. Throughout the snowmelt experiment, the highest drainage volumes were found in this area. The bromide concentrations were low in the first drainage at 13 mm. The highest bromide concentrations could be found in the lower left part of the sampler at 44 mm of drainage. At 66 mm, the maximum bromide concentrations were lower than at 44 mm. At 66 mm, bromide leached through a larger part of the sampler and the concentrations were more equally spread in space than at 44 mm. At 103 mm of drainage, the concentrations decreased further, and the highest concentrations were now found on the right side. At 121 mm of drainage, the drainage did not contain bromide anymore.

The drainage patterns showed more isolated peaks than the bromide concentration. The cells with the highest drainage did not necessarily leach the highest concentration of bromide. The spatial pattern of the bromide concentration had a larger variation in

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



time than the pattern of drainage. From Fig. 5 it is clear that both water flow and solute transport in this soil are heterogeneous.

Figure 6 shows the spatial distributions of the volume of drainage, of the bromide concentration, and of the PG concentration during the irrigation experiment at five selected cumulative drainage depths. During the irrigation experiment, water drained over the whole MCS, with a few isolated peaks that were stable throughout the irrigation experiment. Bromide leaching started at 70 mm, the highest bromide concentrations were found in the lower left area of the sampler. At 101 mm, the bromide concentrations peaked in the lower left, the middle right area, and in the upper left cell. Around those peaks, cells also started leaching bromide, but generally with lower concentrations. At 124 mm, most cells leached bromide and the differences between the concentrations of the cells were small. At 124 mm, the concentrations decreased relatively to the concentrations at 101 mm. In the irrigation experiment, bromide leaching started after a larger amount of drainage than during the snowmelt experiment. Thus, at the start of the irrigation experiment, more water was stored between the soil surface and the MCS than at the start of the snowmelt experiment.

Like the bromide concentration, the PG concentration firstly increased in the lower left part (at 41 and 70 mm). At 101 mm, the concentrations were lower than at 70 mm, the highest concentrations were found on the right side of the sampler. At 124 mm, the drainage did not contain PG anymore. The leaching of PG started earlier than the leaching of bromide. This was also observed by French et al. (2001), and may be caused by density driven flow. The density of pure de-icing fluid is 1.043 times the density of pure water (Kilfrost, 2012). After dilution the density of the applied solution was approximately 1.005 times the density of water. The PG concentrations decreased both earlier and faster than the bromide concentration. The faster decrease in the concentration can be attributed to degradation of PG by micro-organisms. French et al. (2001) found that the first order degradation constant for PG in a field soil was between 0.015 and  $0.04 \text{ d}^{-1}$ . Fitted from the breakthrough curves of bromide and PG for the

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



entire MCS, we found a first order degradation constant of  $0.02 \text{ d}^{-1}$  for PG during the irrigation experiment.

From the measured pressure heads, the average soil moisture content was calculated, using the van Genuchten-Mualem model (van Genuchten, 1980), and the soil hydraulic parameters in Table 1 (Fig. 7). Figure 7 shows that the soil moisture content was generally higher during the snowmelt period than during the irrigation experiment. Peaks in the soil moisture content during the snowmelt resulted from infiltrating melt-water during warmer days. The maximum soil moisture contents from the snowmelt and the irrigation experiment were similar, but during the snowmelt experiment, the soil dried less than during the irrigation experiment. The average soil moisture content was 0.10 during the snowmelt experiment (26 March–23 May), 0.14 during the snowmelt period (26 March–12 April), and 0.10 during the irrigation experiment (23 May–4 July). During the snowmelt period, when high infiltration rates occur, the soil is wetter and therefore a larger part of the coarse textured soil is highly conductive than under drier conditions as in the irrigation experiment. In the irrigation experiment, the drainage showed more isolated peaks than for the snowmelt experiment. Williams et al. (2003) showed that the degree of preferential flow resulting from macropore flow would increase with increasing flow rate. In the snowmelt experiment with higher infiltration rates, however, the number of isolated peaks in the drainage decreased (comparing the drainage in Figs. 5 and 6). The heterogeneous flow in the soil, therefore, is probably caused by small changes in the soil hydraulic properties, and not by macropores.

Figure 8 shows the areas of the MCS in which the highest concentrations in Figs. 5 and 6 were measured. For both the snowmelt and the irrigation experiment, the areas with the highest concentrations are similar. In the snowmelt experiment, the differences in concentrations between all cells seem to be larger than during the irrigation experiment. The high concentrations in a small area of the sampler can be caused by snowmelt. With a high infiltration rate and a wet soil, a small part of the soil can transport solutes rapidly, while little dilution occurs due to a low exchange with the surrounding soil. This leads to a larger spatial variability in concentrations than for the

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



lower infiltration rates during the irrigation experiment. The differences in the concentrations were not caused by heterogeneous infiltration of the meltwater from the snow. If this were the case, also the drainage should be heterogeneous, with the same spatial patterns as the concentration, which was not observed (Fig. 5). Based on this, it is concluded that the infiltration of the meltwater was homogeneous, compared to the soil heterogeneity.

Based on the snowmelt and irrigation experiment we conclude that generally, the spatial patterns of the concentration are similar with high or low infiltration rates, but the differences between the concentrations are larger with a high infiltration rate. Öhrström et al. (2004) identified the soil moisture content at tracer application as an important driver for the formation of flow fingers. Areas that had a higher soil moisture content than surrounding areas at the start of an experiment appeared to be areas where preferential flowpaths were formed. On the other hand, Van Ommen et al. (1989) found that the soil moisture content in and outside preferential flow paths were similar. In our experiment, the soil moisture content in and outside flowpaths was unknown. However, the locations where the solute concentrations were highest, were similar for the two infiltration rates. Therefore, we conclude that the soil moisture content probably does not determine which part of the soil is highly conductive for this soil.

During the irrigation experiment, the highest PG concentrations, and the highest bromide concentrations can be found in the same areas, but at different time steps. Thus, the spatial patterns in the concentrations of a tracer and a degradable solute are similar.

### 2.3.2 Effect of the travel time distribution on BTCs

Figures 5 and 6 reveal that the soil consisted of fast and slow responding areas. Buchter et al. (1995) made a division of the soil in fast and slow cells, based on the height of the peak concentration of the BTC, and the moment that this peak concentration was reached. For each cell, we determined the number of days from solute application until the peak concentration was reached. The resulting histograms are

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



shown in Fig. 9. It must be mentioned that the irrigation experiment was ended before the BTCs were complete for all cells. Cells in which the concentration was still increasing at that time, mostly have their highest concentration at day 40 or 41, which then is called the peak concentration in Fig. 9. Furthermore, cells that did not leach during the irrigation experiment are included (bar at day number 2). For the snowmelt experiment, the histogram has a bell-shaped distribution, and shows that most cells reached the peak concentration at 15 days after solute application. For the irrigation experiment, the histogram is not bell-shaped.

Based on the histograms, the cells are divided in three groups: fast, average, and slow cells. Fast cells are the cells that leach before the mode (before day 15 in the snowmelt experiment, and before day 30 in the irrigation experiment). Average cells have their peak concentration when most cells have (at day 15 and 16 in the snowmelt experiment, and at day 30 in the irrigation experiment). Slow cells have their peak concentration later than the average cells (after day 16 in the snowmelt experiment, and after day 30 in the irrigation experiment). Cells can belong to different groups for the snowmelt and the irrigation experiment. The average concentrations of bromide and PG were calculated for the groups of cells. The BTCs of the fast, average, and slow cells for bromide during the snowmelt and the irrigation experiment are given in Fig. 10.

During the snowmelt experiment, the bromide concentration increased rapidly in the fast cells, as a result of the high infiltration rate caused by snowmelt. At 66 mm of drainage, only patchy snow was left above the MCS. The bromide concentration decreased rapidly in the fast cells when the snow had melted (79 mm drainage). In the snowmelt experiment, the height of the concentration peak was higher than in the irrigation experiment, probably as a result of the higher infiltration rate. The bromide concentration also increased earlier in the snowmelt experiment than in the irrigation experiment. The value of the peak concentration decreased from fast to slow cells in both the snowmelt and the irrigation experiment.

**Quantifying  
heterogeneous  
transport in the field**

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



For the BTCs in Fig. 10, the transport parameters from the convection-dispersion equation were fitted using CXTFIT (Toride et al., 1995). For the fitting, the cumulative drainage of the entire sampler was used as a time axis, to obtain the required steady state conditions (Wierenga, 1977). This gives the velocity  $v$  a unit of  $\text{cm mm}^{-1}$ , and the dispersion coefficient  $D$  a unit of  $\text{cm}^2 \text{mm}^{-1}$ . Using the average drainage per day during either the snowmelt or the irrigation experiment, the parameters were converted to units of  $\text{cm d}^{-1}$ , and  $\text{cm}^2 \text{d}^{-1}$  (Table 2). From the fitted parameters the dispersivity  $\alpha$  was calculated. Both  $v$  and  $D$  decrease from fast to slow cells for the snowmelt and the irrigation experiment. For the fast and average cells, the velocities are higher during the snowmelt than the irrigation experiment. This is caused by the higher infiltration rate, resulting from the snowmelt. For the snowmelt experiment, the dispersivity of the slow cells is lower than the dispersivity of the fast and average cells. For the irrigation experiment, the dispersivity decreases from fast to slow cells. The dispersivities were lower during the irrigation experiment than during the snowmelt experiment. This shows that the dispersivity depends on the flow velocity. The dispersivities of the fast, average, slow cells, and of the total sampler are similar. The dispersivities are in the same order of magnitude of the values that are given in the reviews of Beven et al. (1993) and Vanderborght and Vereecken (2007).

We hypothesize that in fast cells, the values of the peak concentration (relative to the applied concentrations) of PG and bromide would be similar, because PG degradation can only occur during a short period. In slow cells, the travel time is longer, and therefore more PG would be degraded than in the fast cells. As a result, in slow cells, the PG concentration would be lower than the bromide concentration, and this difference would increase with increasing travel time.

Figure 11 shows the BTCs for PG and bromide in the irrigation experiment, for the fast, average, and slow cells. For the fast cells, the PG concentration increased earlier than the bromide concentration. This was also observed by French et al. (2001), and can be caused by density-driven flow. The values of the peak concentration and the recession thereafter were similar for bromide and PG. The average cells show

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**Quantifying  
heterogeneous  
transport in the field**

---

D. Schotanus et al.

---

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[⏪](#)[⏩](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

comparable features, but with lower concentrations than for the fast cells. Also, the PG concentration decreased earlier than the bromide concentration in the average cells, which possibly is attributable to degradation.

Initially, the PG concentrations in the fast and average cells might have been too high for the micro-organisms, and therefore degradation might have been inhibited. PG degradation can be inhibited if the PG concentration is high, and the inhibition limit depends on the type of micro-organisms, and temperature, amongst others (Bausmith and Neufeld, 1999; Jaesche et al., 2006). The first order degradation rate of PG increases with a decreasing PG concentration (Bausmith and Neufeld, 1999). Once the concentration was below the inhibition level, as was probably the case in the average cells, the concentration decreased rapidly due to degradation. In the slow cells, PG was degraded before bromide reached the peak concentration. Thus, in the slow cells the PG concentration appears to decrease faster than in the fast cells, the decrease in the bromide concentration appears to be similar in the fast, average, and slow cells. This means that the degree of degradation of PG would increase from fast to slow cells. The degradation constant was only fitted for the entire sampler ( $k = 0.02\text{d}^{-1}$ ), and not for the fast, average, and slow cells separately, because the BTCs of bromide are not complete. The incomplete BTCs hinder the estimation of the degradation constants, mainly in the slow cells.

We conclude that the travel time distribution influences the concentrations of degradable solutes, and the differences between the concentrations of degradable and non-degradable solutes increase from fast to slow transporting areas. Therefore, in parts of the soil that transport water and solutes rapidly, the leaching of degradable contaminants will be higher than in slow transporting areas.

### 2.3.3 Spatial patterns of total drainage and leaching

From Fig. 11 was concluded that the degradation of PG appears to increase from fast to slow cells. To further examine this conclusion, we will discuss the spatial pattern of the

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





total leached masses of bromide for both the snowmelt and the irrigation experiment, and PG for the irrigation experiment.

Figure 12 shows the total leached mass of bromide for the snowmelt experiment, for the irrigation experiment, and the total leached mass of PG during the irrigation experiment, together with the relative differences between the leached PG and bromide (Fig. 12d). The leached masses of PG and bromide should be compared with some caution, because the bromide BTCs were not complete for all cells before the experiment was stopped. Therefore, the total bromide leaching was underestimated, especially for slow cells.

From Fig. 12a and b follow that the highest bromide leaching occurred in the same area for the snowmelt and the irrigation experiment. In the snowmelt experiment, most leaching occurred in the middle left part of the sampler, while in the irrigation experiment, most leaching occurred in the areas that are marked in Fig. 8. In the low leaching areas, the leaching was higher during the snowmelt experiment than during the irrigation experiment. Apparently, with high infiltration rates, as in the snowmelt experiment, the leaching was more homogeneous. More homogeneous leaching was caused by more homogeneity in drainage, because the spatial variability in the concentration was larger in the snowmelt experiment than in the irrigation experiment (Fig. 5). The more homogeneous drainage can be caused by the higher infiltration rate, which results in a wetter soil. In a wetter soil, a larger part of the matrix drains water than in a dry soil, because the hydraulic conductivity is higher. Furthermore, Fig. 12a and b show that in the irrigation experiment, with low infiltration rates, there were more isolated peaks in the leached masses than in the snowmelt experiment.

The spatial patterns in the bromide and PG leaching during the irrigation experiment were similar (Fig. 12b, c). Figure 12d shows that the PG leaching was generally lower than the bromide leaching (i.e. less than 100%). In the upper right area of the MCS, the leaching of PG was smallest compared to bromide, but since also little bromide leached that was not well visible in Fig. 12b and c. Thus, the spatial distributions of the leached masses are similar, because the leaching of the non-degradable solute in

**Quantifying heterogeneous transport in the field**

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



this area was low as well. The highest PG concentrations, and the highest bromide concentrations can be found in the same areas, but at different times (Fig. 6), and the cumulative leached masses were similar. This means that the spatial pattern of the drainage was stable in time during the irrigation experiment.

During the snowmelt experiment, the bromide recovery was 43%. During the irrigation experiment, the bromide recovery was 42%, and the PG recovery was 32%. The low bromide recovery suggests flow bypassing the sampler. However, in a closed lysimeter experiment from the same area, the bromide recovery was 42% (Lißner et al., 2011). Thus, the extent of flow bypassing the sampler is probably much lower than would be expected based on the bromide recovery. On average for all cells and for the entire irrigation experiment, the leaching of PG is 75% from the bromide leaching, which suggests that 25% of the PG was degraded during this experiment. As mentioned before, the BTCs of bromide were not complete before the irrigation experiment was stopped, and thus the bromide leaching was underestimated. This implies that the PG degradation was also underestimated.

### 2.3.4 Quantifying preferential flow

Figures 5, 6, and 12 give a visual representation of the results. In the following section we will quantify differences in the spatial distribution of the leaching of bromide and PG, and the drainage.

For this purpose, the cells are sorted with decreasing drained volume or leached mass, and then cumulated (Quisenberry et al., 1994; Strock et al., 2001; De Rooij and Stagnitti, 2002). Doing so, removed all spatial information. The cumulative drainage and leaching are plotted as a function of the cumulative sampler area in Fig. 13. The cumulative sampler area is defined as the surface area of the number of cells that corresponds to the number of the sorted drained volume or leached mass. In Fig. 13, the cells are sorted separately for each experiment and solute. Uniform leaching would give a straight line. Figure 13 shows that 50% of the drainage and leaching occurred in 12–20% of the sampler area.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



---

**Quantifying heterogeneous transport in the field**

---

D. Schotanus et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

The lines for the drainage lie below the lines of the solutes for both experiments. This can be caused by the longer time period of sampling for the drainage. The solutes were applied as a pulse, and therefore, the sampling period was generally slightly shorter than for the drainage. When the peak concentration of bromide already passed, drainage water was still sampled. It is more correct to use the leaching instead of the drainage, as the leaching is labelled using a pulse, and the drainage volume itself is not labelled. However, we did include the drainage in Fig. 13, because the sampling period of drainage is not much larger than of bromide, and the drained volumes and leached masses are distributed differently in space (Figs. 5 and 6).

The drainage during the snowmelt experiment lies below the drainage during the irrigation experiment, which implies a larger homogeneity, and strengthens the conclusion based on Fig. 12.

For bromide mass, the line of the irrigation experiment lies above the line of the snowmelt experiment. This means that the bromide leaching was more heterogeneous during the irrigation experiment than during the snowmelt experiment. The distances between the lines of the bromide leaching and of the drainage are different for both experiments in Fig. 13. Therefore, we conclude that the heterogeneous solute leaching is both caused by differences in the drainage and the spatial distribution of the solute concentrations. The leaching of bromide and PG during the irrigation experiment was similarly heterogeneous. This was also concluded from Fig. 12.

We studied the stability of the preferential flow paths by comparing the drainage and the bromide leaching in each cell in both experiments (Fig. 14a, b). Lennartz et al. (2008) found that preferential flow paths were stable in a sandy soil in the laboratory. The drainage during the snowmelt experiment and the irrigation experiment had a low linear correlation coefficient ( $R^2 = 0.43$ ). Also the bromide leaching during the snowmelt experiment and the irrigation experiment were not well correlated ( $R^2 = 0.56$ ). With the high resolution MCS we used, we found that the bromide leaching may be different in the exact locations (Fig. 14), even though the area of the MCS

with the highest bromide leaching is similar for both experiments (Fig. 12). Also, the fraction of the solute that leached through a cell was different for each experiment.

Comparing the total bromide leaching with the total PG leaching in each cell in the irrigation experiment (Fig. 14c) shows the leached masses per cell were strongly correlated ( $R^2 = 0.95$ ). Thus, with the same infiltration rate, the spatial patterns of the leached masses are similar for a degradable solute and a non-degradable tracer. In Fig. 11, the BTCs for bromide appeared to be different than the BTCs for PG, depending on the division in the fast, average, and slow cells. However, this did not result in a different total PG leaching, relative to the bromide leaching, for the fast, average, and slow cells. Apparently, for solutes with this degradation rate and under this infiltration rate, the travel time distribution does not influence the overall leaching much.

In Figs. 13 and 14, the cells were approximated as individual stream tubes, while ignoring the possibility of cell clustering. The soil above neighbouring cells could exchange water and solutes, which may have a distinct influence when using a high resolution sampler. We calculated the spatial autocorrelation coefficient of the total drainage and the total leached masses, to determine how the drainage, or the leaching in a particular cell is related to the drainage, or leaching in neighbouring cells. Table 3 shows the values for the spatial autocorrelation coefficient  $r$ , calculated with Eq. (1). With a high spatial autocorrelation coefficient, the cross-sectional area of the preferential flowpaths is large, while with a low spatial autocorrelation the preferential flowpaths mainly consist of isolated peaks. The spatial autocorrelation was higher for bromide than for the drainage, for both the snowmelt and the irrigation experiment. As with Fig. 13, this may be a result of the slightly shorter sampling time for leaching than for drainage. During the snowmelt experiment, the spatial autocorrelation was higher than during the irrigation experiment. Thus, with high infiltration rates there is more clustering of high leaching cells. This is consistent with Figs. 12 and 13 which show that with a higher water flux the differences in leaching between the cells are smaller. The reason for this might be that with a higher water flux, a larger part of this coarse

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



textured soil is highly conductive. As expected from Fig. 14c, the spatial autocorrelation of PG leaching was similar as of bromide leaching.

The spatial patterns of the leached masses of bromide and the spatial distribution of the highest bromide concentrations were rather similar for the snowmelt and the irrigation experiment. Therefore, we conclude that soil heterogeneity is the main reason for the heterogeneous water flow and solute transport in this soil. At this scale, the heterogeneous melting of snow does not influence the heterogeneous flow in the soil much. The applied infiltration rate, and the corresponding soil moisture content, influence the differences between the concentrations of the cells. As a result the areas with the highest leaching were similar for both experiments, but the leached masses in the cells were not highly correlated.

### 3 Conclusions

The leaching of a non-degradable and a degradable solute was measured in two field experiments, with different infiltration rates. The spatial and temporal variability of leaching, and the seasonal effects on leaching were quantified, and additionally the leaching of the non-degradable and the degradable solute was compared. One experiment was done during snowmelt, and was characterized by a high infiltration rate from the meltwater. In the other experiment, the soil was irrigated, to mimic homogeneous rainfall with lower infiltration rates.

Regarding the persistence of preferential flow, the highest bromide concentrations were found in the same area for both experiments, and thus were independent of the flow rate. In the snowmelt experiment the spatial differences in the concentrations appeared to be larger than in the irrigation experiment, possibly due to a lower exchange with the surrounding soil (and thus less dilution) caused by the higher infiltration rate.

The highest bromide leaching occurred in the same area for the snowmelt and the irrigation experiments, but not in exactly the same cells. Furthermore, with low infiltration rates, as in the irrigation experiment, there are more isolated peaks in the bromide

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



leaching than in the snowmelt experiment. The heterogeneous flow in the soil, therefore, is probably caused by small changes in the soil hydraulic properties, and not by macropores.

The spatial autocorrelation of the leaching of cells is higher with high infiltration rates.

The reason for this might be that with a higher water flux, a larger part of the coarse textured soil is highly conductive.

The highest PG concentrations, and the highest bromide concentrations were found in the same areas in the irrigation experiment. The patterns in the concentrations of a tracer and a degradable solute were similar in space, but not in time, as the leaching of PG started earlier than the leaching of bromide. The leached masses were highly correlated. The leaching of PG was lower than the leaching of bromide, due to degradation.

Soil heterogeneity is the main reason for the heterogeneous water flow and solute transport in this soil. Heterogeneous melting of snow did not influence the heterogeneous flow in the soil much at the scale of the MCS. The applied infiltration rate, and the corresponding soil moisture content, influenced the differences between the concentrations of the cells. As a result the areas with the highest leaching were similar for both experiments, but the leached masses in the cells were not highly correlated.

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## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



## References

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## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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- 30



## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page	
Abstract	Introduction
Conclusions	References
Tables	Figures
◀	▶
◀	▶
Back	Close
Full Screen / Esc	
Printer-friendly Version	
Interactive Discussion	

**Table 1.** Soil properties of the field site, at 50 cm depth. The soil hydraulic parameters are from the textural class Sand (Carsel and Parrish, 1988).

% fine sand	% medium and coarse sand	% gravel	$K_{sat}$ (ms <sup>-1</sup> )	$\theta_{res}$ (-)	$\theta_{sat}$ (-)	$\alpha$ (cm <sup>-1</sup> )	$n$ (-)
15 <sup>a</sup>	75 <sup>a</sup>	10 <sup>a</sup>	$6.65 \cdot 10^{-4,b}$	0.045 <sup>c</sup>	0.43 <sup>c</sup>	0.145 <sup>c</sup>	2.68 <sup>c</sup>

<sup>a</sup> French et al. (1994).  
<sup>b</sup> French et al. (2001).  
<sup>c</sup> Carsel and Parrish (1988).



## Quantifying heterogeneous transport in the field

D. Schotanus et al.

**Table 2.** Transport parameters velocity  $v$ , dispersion coefficient  $D$ , and dispersivity  $\alpha$  ( $= \frac{D}{v}$ ) for fast, average, and slow cells, and the total sampler, for bromide.

Experiment	Group	$v$ (cm d <sup>-1</sup> )	$D$ (cm <sup>2</sup> d <sup>-1</sup> )	$\alpha$ (cm)
Snowmelt	Fast	1.8	12	6.6
	Average	1.5	11	7.1
	Slow	1.3	3.2	2.4
	Total	1.7	9.2	5.6
Irrigation	Fast	1.5	6.1	4.1
	Average	1.4	5.2	3.8
	Slow	1.3	2.6	2.0
	Total	1.6	3.1	1.9

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



## Quantifying heterogeneous transport in the field

D. Schotanus et al.

**Table 3.** Spatial autocorrelation according to Moran's  $I$ . A value of 1 means that the spatial autocorrelation is perfect, and a value of  $-0.01$  means that there is no spatial autocorrelation.

Experiment and drainage/solute	$I$
Snowmelt drainage (mm)	0.46
Snowmelt bromide (mg)	0.61
Irrigation drainage (mm)	0.14
Irrigation bromide (mg)	0.26
Irrigation PG (mg)	0.30

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 1.** Experimental setup. Installation of the multicompartment sampler (MCS), shown from the trench **(a)**. The location of the MCS shown from above **(b)**. The marked square indicates the location of the MCS, 51 cm below soil surface. The roof of the trench is visible in the background. The width of the MCS is 31.5 cm.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



1	11	21	31	41	51	61	71	81	91
2									
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**Fig. 2.** Cell number configuration of the MCS. Only cell numbers 1–10, and 11, 21, ..., 91 are given.

**Quantifying heterogeneous transport in the field**

D. Schotanus et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

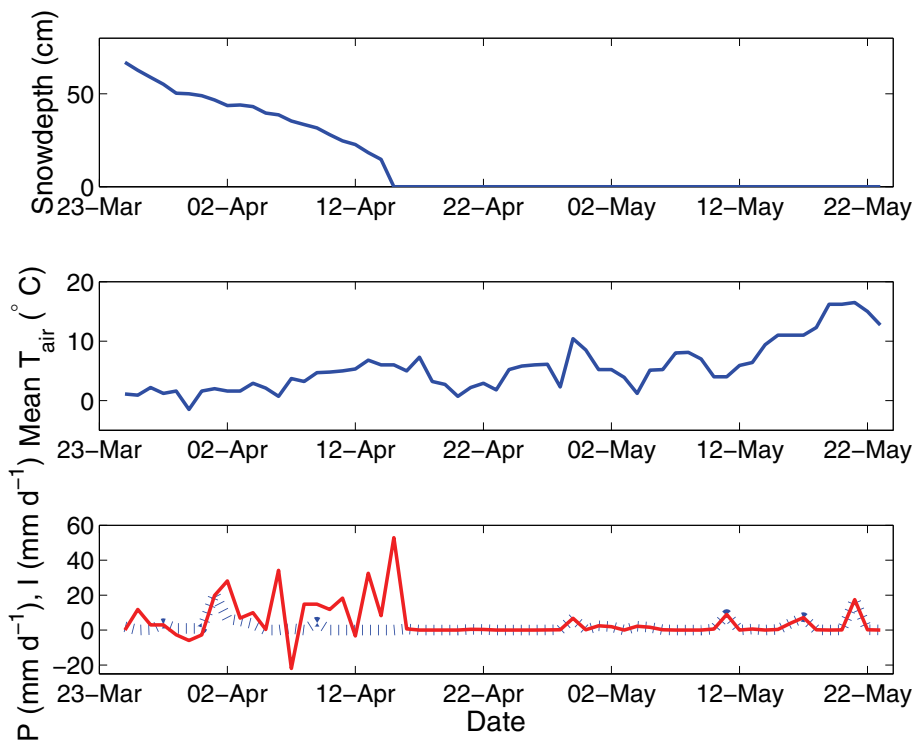
Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 3.** Weather series for the snowmelt experiment (in 2010), mean  $T_{\text{air}}$  is the daily mean temperature,  $P$  is the daily precipitation (dotted blue line),  $I$  is the infiltration (red line). The infiltration is calculated from the daily difference in the equivalent water depth of the snow cover, plus the precipitation.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

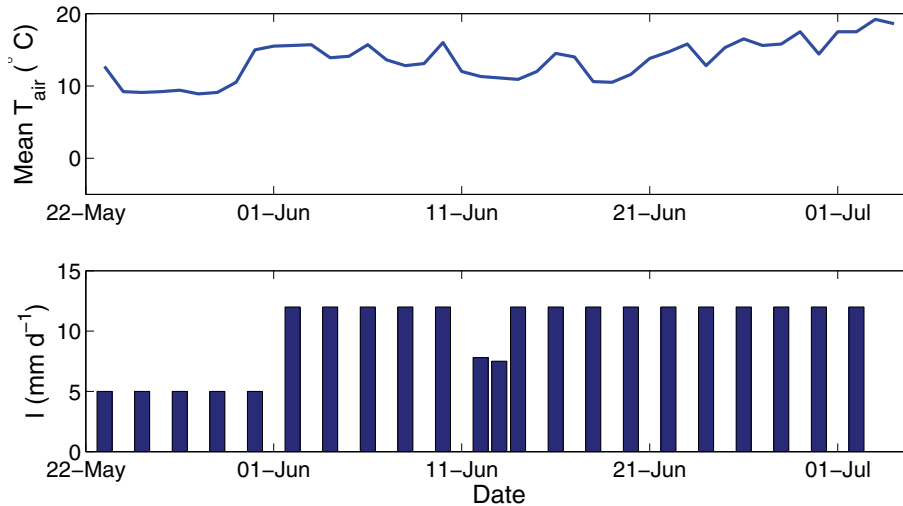
Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 4.** Weather series for the irrigation experiment (in 2010), same remarks as Fig. 3.

**Quantifying heterogeneous transport in the field**

D. Schotanus et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

Back Close

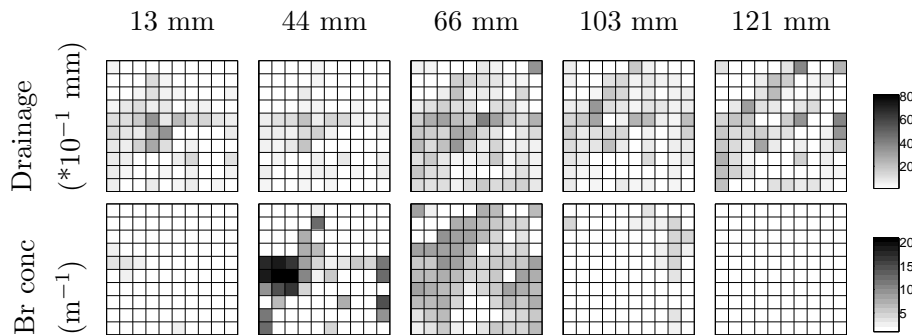
Full Screen / Esc

Printer-friendly Version

Interactive Discussion







**Fig. 5.** Spatial distributions of the drainage, and the bromide concentration during the snowmelt experiment over the multicompartament sampler consisting of  $10 \times 10$  cells of  $3.15 \times 3.15 \text{ cm}^2$  each. The bromide concentration is scaled with the applied mass per  $\text{m}^2$ . The titles refer to the total cumulative drainage since solute application.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

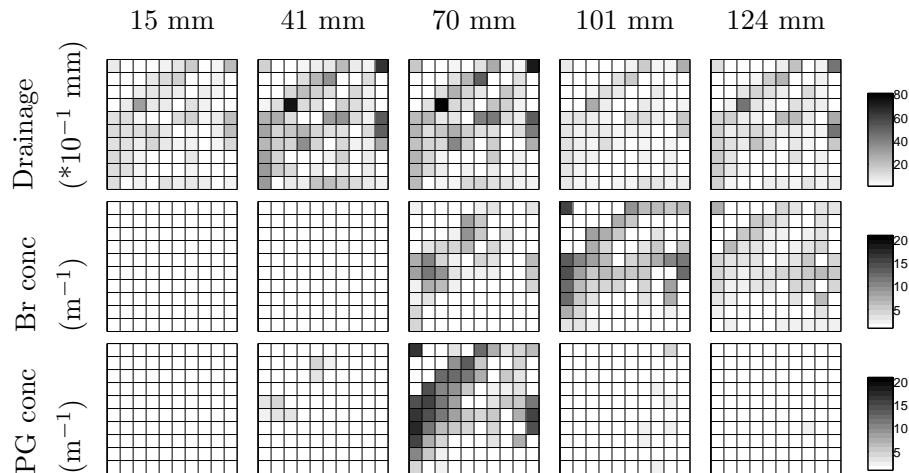
Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 6.** Spatial distributions of the drainage, the bromide concentration, and the PG concentration during the irrigation experiment over the multicompart sampler consisting of  $10 \times 10$  cells of  $3.15 \times 3.15 \text{ cm}^2$  each. The concentrations of bromide and PG are scaled with the applied mass per  $\text{m}^2$ . The titles refer to the total cumulative drainage since solute application.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

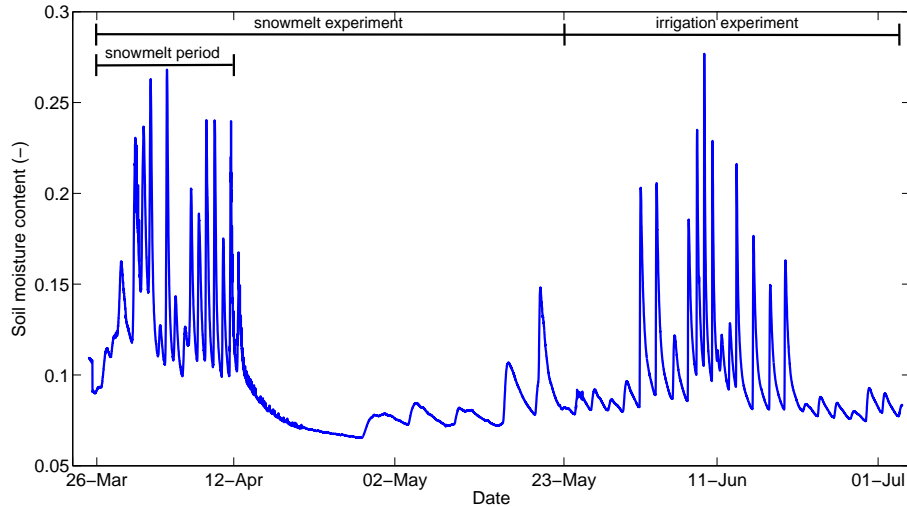
Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 7.** Average soil moisture content (–) at 51 cm depth during the snowmelt and the irrigation experiment (in 2010), calculated from pressure heads measured by 4 tensiometers.

**Quantifying heterogeneous transport in the field**

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

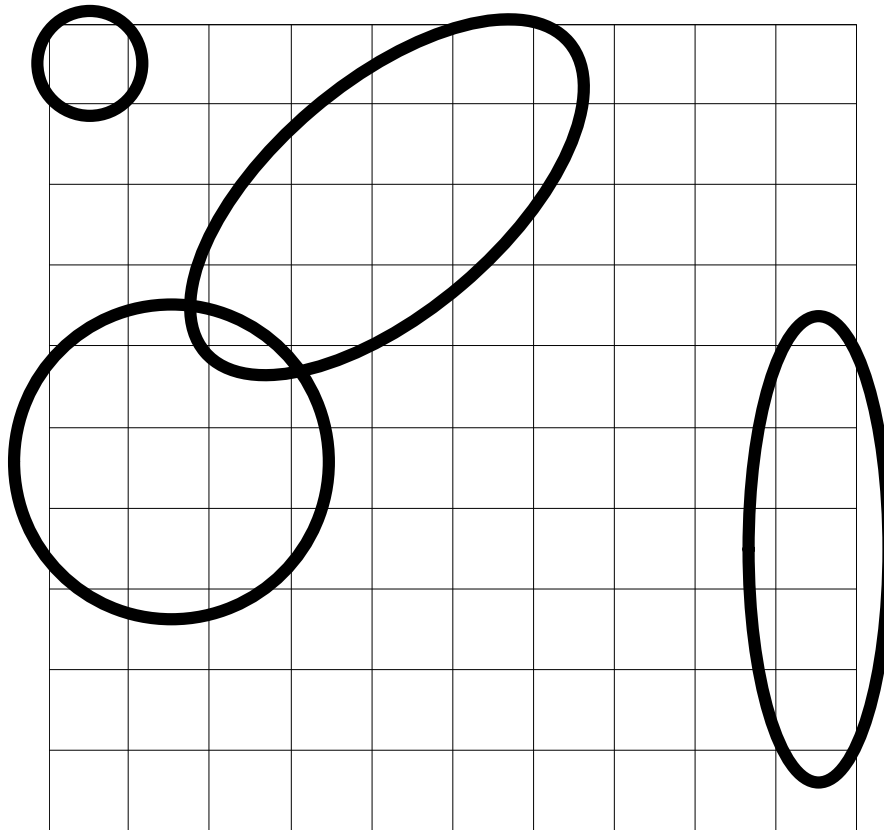
Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 8.** The areas from the multicompartament sampler where the highest leaching of bromide and PG occurred (based on Figs. 5 and 6), marked with ellipses.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

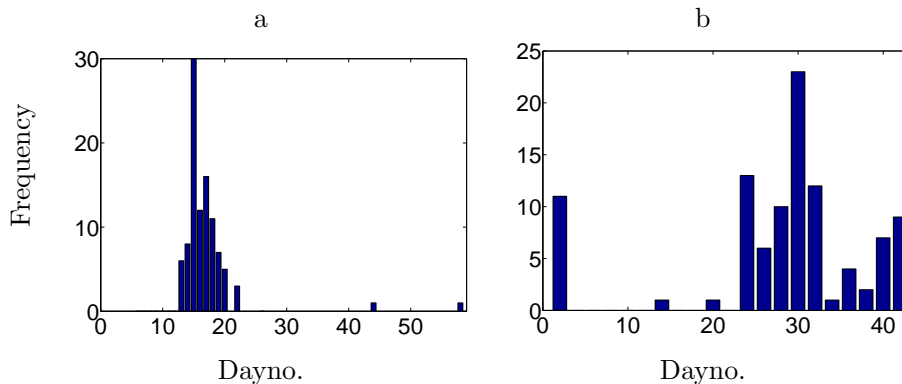
Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion





**Fig. 9.** Histograms of the daynumber on which the peak bromide concentration occurred, for the snowmelt experiment **(a)**, and the irrigation experiment **(b)**. Dayno. 0 is 26 March 2010 **(a)**, and 23 May 2010 **(b)**.

**Quantifying heterogeneous transport in the field**

D. Schotanus et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

Back Close

Full Screen / Esc

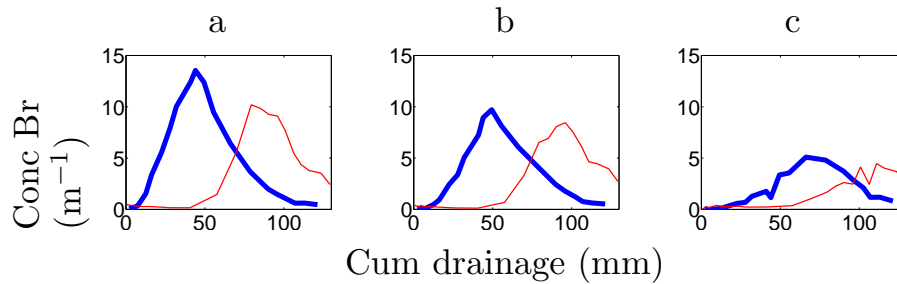
Printer-friendly Version

Interactive Discussion



Quantifying heterogeneous transport in the field

D. Schotanus et al.



**Fig. 10.** Breakthrough curves of bromide for the snowmelt experiment (thick blue line) and the irrigation experiment (thin red line), for fast cells (a), average cells (b), and slow cells (c).

Title Page

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

Back Close

Full Screen / Esc

Printer-friendly Version

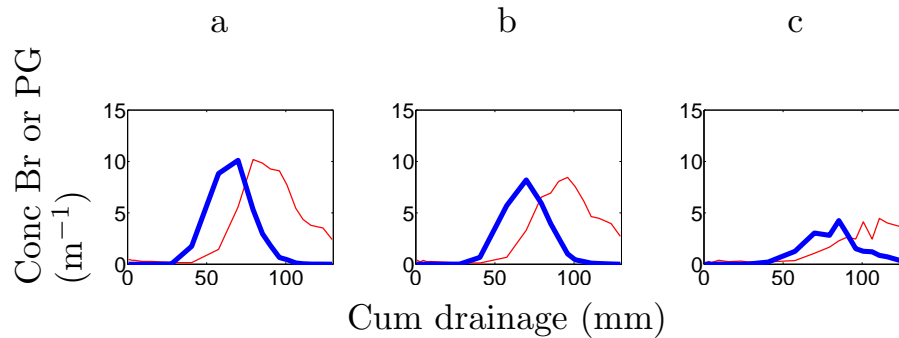
Interactive Discussion



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**Quantifying  
heterogeneous  
transport in the field**D. Schotanus et al.

---



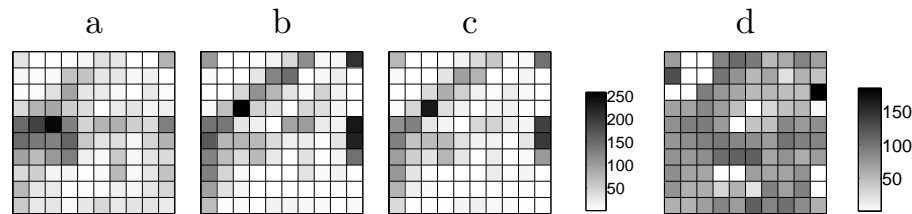
**Fig. 11.** Breakthrough curves of bromide (thin red line) and PG (thick blue line) for the irrigation experiment, for fast cells **(a)**, average cells **(b)**, and slow cells **(c)**.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

---

**Quantifying  
heterogeneous  
transport in the field**D. Schotanus et al.

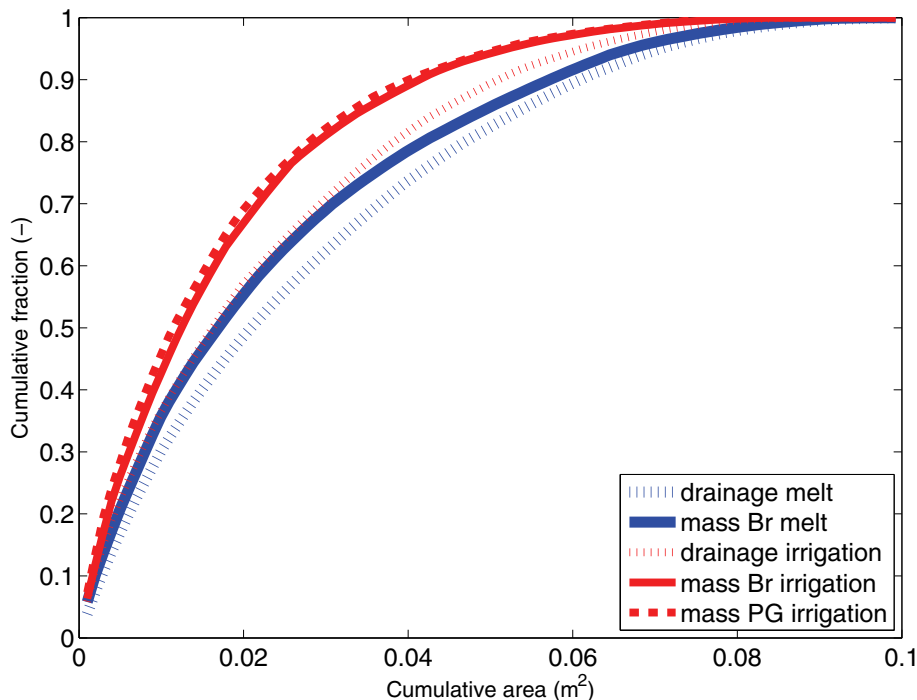
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**Fig. 12.** Spatial distributions of the total bromide leaching during snowmelt, total bromide leaching during irrigation, and total PG leaching during irrigation over the multicompart sampler consisting of  $10 \times 10$  cells of  $3.15 \times 3.15 \text{ cm}^2$  each. Bromide leaching during the snowmelt experiment **(a)**, bromide leaching during the irrigation experiment **(b)**, PG leaching during the irrigation experiment **(c)**, leaching is given in % from what would have leached in each cell with perfect parallel flow. PG leaching divided by bromide leaching during the irrigation experiment **(d)**, both scaled with the applied solute mass per  $\text{m}^2$ , given in %.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)





**Fig. 13.** Cumulative drainage and leaching of bromide during the snowmelt and irrigation experiments, and PG during the irrigation experiment (–). The drainage and the leaching per cell were scaled with the total drainage or the total leached mass for the sampler during either the snowmelt experiment or the irrigation experiment. The cells are sorted with decreasing amount of drainage, or bromide or PG leaching.

## Quantifying heterogeneous transport in the field

D. Schotanus et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

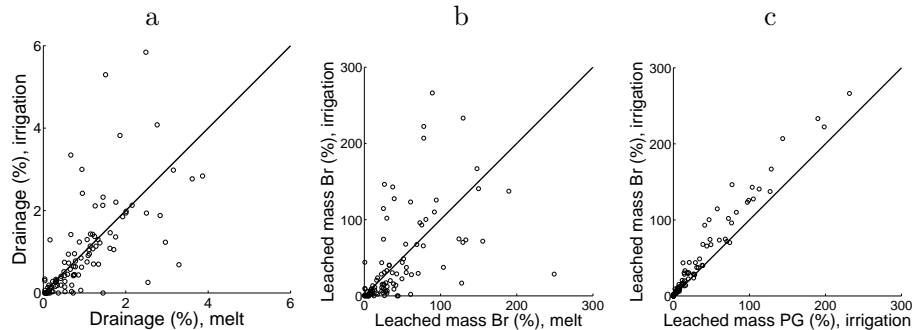
Interactive Discussion



---

**Quantifying heterogeneous transport in the field**D. Schotanus et al.

---



**Fig. 14.** Comparison of total leaching per cell: drainage of the snowmelt and irrigation experiment **(a)**, bromide leaching of the snowmelt and irrigation experiment **(b)**, and bromide leaching and PG leaching of the irrigation experiment **(c)**. The drainage was scaled with the drainage of the entire sampler of either the snowmelt or irrigation experiment. Solute leaching is given in % from what would have leached in each cell with perfect parallel flow. The 1:1 lines are given as well.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)