## Comment of reviewer #2 concerning author's reply dated 2015/12/2

Paper: HESS-2015-365 MS

Once more a comment concerning equations (4) - (6):

The authors claim that they use the equations according to McIlveen correctly.

But this is not the case.

McIlveen defines q as the specific mass (mass of water per total mass) and r as mass of water per mass of dry air.

Using this definitions **equation (4) has to be formulated with q** (according to McIlveen as well as according to my comments before).

Equation (5) corrected by the authors after my first comment (q = 0.622 pw /p) is only valid for low moisture content (see McIlveen).

Using the notations r and q of McIlveen equation (6) is indeed correct.

Nevertheless, the exact equations for the mutual conversion of r (=X) and q (=w) given in my last comment (see below) are in my opinion more suitable to describe atmospheric processes with moist air.

 $R_{air}$ ,  $R_w$  = gas constants of air and water

w = mass fraction of water (mass of water per mass of mixture)

X = mass of water per mass of dry air.

p<sub>w</sub>, p<sub>air</sub> = partial pressures of water and air, respectively.

 $p = p_{air} + p_w$  (total pressure)

Quotient of molar masses (water to air) 18/29 = 0.622.

Taking into account the definitions above the following correct equations (4) - (6) can be formulated:

$R = w R_w + (1-w) R_{air}$	(4)

 $X = 0.622 p_w / (p - p_w)$ (5)

$$w = X / (1 + X)$$
 (6)