



# Supplement of

# Stream restoration and sanitary infrastructure alter sources and fluxes of water, carbon, and nutrients in urban watersheds

M. J. Pennino et al.

Correspondence to: M. J. Pennino (mpennino@princeton.edu)

The copyright of individual parts of the supplement might differ from the CC-BY 3.0 licence.

1	Supporting	Information
-		

22

Stream restoration and sanitary infrastructure alter sources and fluxes of water, carbon,
and nutrients in urban watersheds

6 Michael J. Pennino, Sujay S. Kaushal, Paul M. Mayer, Ryan M. Utz, Curtis A. Cooper

## 7 1 Methods

### 8 **1.1 Site Description**

9 Red Run (RRN), Powder Mill Run (PMR), and Dead Run (DRN) are all part of 10 the Gwynns Falls watershed, which is monitored by Baltimore Ecosystem Study's Long 11 Term Ecological Research (www.beslter.org). Previous information on runoff and 12 nutrient export in these watersheds can be found elsewhere (Groffman et al., 2004; 13 Kaushal et al., 2008a; Kaushal et al., 2011; Duan et al., 2012). Specific land use and 14 other site characteristics of each watershed are found in Figure S1 and listed in Table S1. 15 Minebank Run is an urban headwater stream located in Towson, MD. The 16 headwater reach of Minebank Run was restored in 1999 and the lower reach in 2005 17 (Doheny et al., 2006). The stream restorations at MBR consisted of bank stabilization, 18 the addition of riffles and meander features with step-pool habitat, a hydrologically 19 connecting streambed and floodplain wetland, and the planting of trees in the riparian 20 zone (Kaushal et al., 2008b). 21 Red Run, located in Owings Mills, MD, has been gauged by UMBC Center for

Urban Environmental Research and Education (CUERE) to measure stream flow since

1	2005. RRN is characterized by newer development and relatively high-density
2	stormwater management (SWM) facilities in both residential and commercial
3	development areas within the watershed. The newer residential and commercial
4	developments in RRN watershed have implemented stormwater detention facilities,
5	draining areas of impervious surfaces, such as detention basins, wet ponds, stormwater
6	bioretention facilities, and sand filters. This watershed also has extensive riparian
7	buffers. The outflows of SWM facilities are either connected to a stream or flow into a
8	forested buffer. Red Run is missing discharge data from July 2011 through September
9	2011, due to construction under the bridge where the gauge is located. For daily flow
10	data, we were able to estimate the missing mean daily flow by creating a linear
11	relationship with nearby USGS gauged streams of similar size and discharge ( $R^2 > 0.95$ ).
12	For the real-time data Flashiness estimates, the missing discharge was left out of the
13	calculations. Also, the Red Run flow data has not been through USGS QA/QC and is
14	based on a provisional rating curve.
15	Powder Mill Run and Dead Run are also subwatersheds of the BES LTER site.
16	Dead Run has 6 USGS stream gauging stations along its stream network, and has had
17	weekly stream chemistry sampling since 1998 (Groffman et al. 2004; Kaushal et al.
18	2008a). The Dead Run watershed has been the focus of numerous eco-hydrological
19	investigations and there exists a rich dataset on nitrogen dynamics (e.g. Groffman et al.,
20	2004; Kaushal et al., 2008a; Kaushal et al., 2011; Newcomer et al., 2012), hydrologic
21	characteristics, and water mass balances (Klocker et al., 2009; Ryan et al., 2010; Ryan et
22	al., 2011; Sivirichi et al., 2011).
22	

#### 1 **1.2** Stream Chemistry Analysis

2 During each stream chemistry sampling, measurements of dissolved oxygen 3 (DO), pH, and temperature were recorded. DO and pH meters were calibrated each day 4 before use in the field. All stream chemistry samples were analyzed using standard EPA 5 methods at the National Risk Management Research Lab (NRMRL) in Ada, OK, USA. 6 Water samples requiring filtration were filtered in the lab within 24 hours using a 47mm 7 (0.45 µm) glass fiber filter, then preserved until analysis. All nitrogen species, except 8 samples for stable isotope analysis, were preserved by acidifying to pH 2 with sulfuric 9 acid and stored frozen in HDPE bottles until analysis. Samples for TOC and DOC were 10 stored in 40 ml glass VOA bottles, preserved by acidifying to pH 2, using phosphoric 11 acid and refrigerated until analysis.  $NO_3^-$  isotope samples were filtered and preserved 12 frozen in HDPE bottles. Samples for H<sub>2</sub>O isotope analysis were unfiltered and collected 13 for only 1 year during 2010-2011. H<sub>2</sub>O isotope samples were unfiltered and collected in 14 a 30 ml glass Qorpak bottle, that was filled and capped under water in the field so that 15 there are no air bubbles or headspace, and then sealed with a Teflon-lined cap. The EPA NRMRL Isotope Ration Mass Spectrometer (IRMS) lab analyzes the  $\delta^2$ H and  $\delta^{18}$ O of 16 water samples following RSKSOP 296. Samples for carbon quality analysis were filtered 17 18 and saved in amber glass Qorpak vials and kept refrigerated (see further methods below). 19 The rest of the samples were preserved frozen in HDPE bottles until analysis. 20 Fluorescence samples were analyzed within 1-3 weeks. All other samples, besides  $NO_3^{-1}$ 21 isotopes samples, were analyzed within 1-2 months. TKN,  $NH_4^+$ ,  $NO_3^- + NO_2^-$ , ortho-P and Br<sup>-</sup> were measured using Lachat flow 22 23 injection analyses (Lachat Instruments, Loveland, CO USA). Total organic nitrogen

1	(TON) was calculated as TKN of unfiltered water minus ammonium and dissolved
2	organic nitrogen (DON) was calculated as TKN of filtered water minus ammonium (with
3	TKN being the sum of organic nitrogen plus ammonia/ammonium). Total nitrogen (TN)
4	was calculated as TKN plus $NO_3^- + NO_2^-$ . $NH_4^+$ was measured using EPA FIA Method
5	10-107-06-1-A. Br and Iodide were analyzed by EPA FIA Method 10-135-21-2-A. $SO_4$
6	was analyzed by FIA Method 10-116-10-1-C. Chloride was analyzed by FIA Method 10-
7	117-07-1-B. H <sub>2</sub> O isotopes were measured using a high temperature conversion elemental
8	analyzer (tc/ea), a continuous flow unit, and an isotope ratio mass spectrometer (IRMS)
9	(EPA's RSKSOP 296v0). Total organic carbon (TOC) and dissolved organic carbon
10	(DOC) were analyzed using a total organic C analyzer with high-temperature Pt-
11	catalyzed combustion and NDIR detection (Shimadzu TOC-VCPH, Columbia, MD
12	USA). Samples for carbon quality analysis were analyzed on a Fluoromax-4
13	spectrofluorometer (Jobin Yvon Horiba) at the University of Maryland Biogeochemistry
14	Lab.
15	

#### **1.3 Fluorescence Analysis** 16

On the day of analysis, samples were allowed to warm to room temperature and 17 18 placed in a 1 cm quartz cuvette for analysis. Each day that samples were run on the 19 spectrofluoromometer, a blank sample was also run at the beginning of the day using 20 milli-Q water. Between samples, the cuvette was rinsed three times with milli-Q water 21 and then sample rinsed with the next sample, two times. A fluorescence EEMs was collected for each sample and also collected for Milli-Q water as a blank. A full EEMs 22 23 was also collected for a quinine sulfate standard (10 ppb in 0.1 N H<sub>2</sub>SO<sub>4</sub>). This quinine

1	sulfate standard can be used when comparing samples run on different instruments and
2	used to monitor instrument drift over time (Del Vecchio and Blough, 2004; Fellman et
3	al., 2009). A scan rate of 4nm/sec was used and the excitation and emission slit widths
4	were 5 nm. The EEMs were collected under the S1/R1 mode to correct for instrument
5	drift. The standard inner-filter corrections (IFC) were not done on most samples because
6	absorbance measurements were not attained for most samples. However, for the samples
7	where absorbance was collected (using scanning spectrophotometer), the IFC was done
8	and it was found that there is $< 5\%$ difference between EEMs with and without IFC.
9	We analyzed fluorescence EEMs for the following indices: the fluorescence
10	index, FI (McKnight et al., 2001) the humification index, HIX (Zsolnay et al., 1999;
11	Huguet et al., 2009), the biological freshness index, BIX (Huguet et al., 2009), and the
12	protein to humic fluorescence intensities ratio (Coble, 1996; Stolpe et al., 2010). FI
13	values are ~ 1.9 for microbial derived fulvic acids and ~1.4 for terrestrial derived fulvic
14	acids sources (McKnight et al., 2001; Cory et al., 2010). BIX values of $< 0.7, 0.8 - 1.0$ ,
15	or $> 1.0$ are associated with terrestrial sourced DOM, algal sourced DOM, or aquatic
16	bacterial sources, respectively (Huguet et al., 2009). HIX was used to distinguish the
17	humic or autochthonous nature of the organic matter in the sample (Zsolnay et al., 1999;
18	Ohno, 2002). Higher HIX values suggest DOM of strong humic character and terrestrial
19	origin, while lower values indicate weaker humic character and higher autochthonous
20	sourced DOM. Further details are described in Pennino et al. (2014). Fluorescence EEM
21	peak intensities at specific excitation and emission wavelengths were also used to
22	determine the relative contribution of protein (at excitation 275 nm and emission 340
23	nm), fulvic (at excitation 310 nm and emission 420 nm) and humic (at excitation 350 nm

and emission 480 nm) dissolved organic matter (Coble, 1996; Stolpe et al., 2010) and
then used to calculate the protein to humic (P/H) organic matter ratio in each sample.
The higher the peak intensity for each of these organic matter types the greater the
proportion of that organic matter in the sample

#### 1 References

- 2 Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation
- 3 emission matrix spectroscopy, Marine Chemistry, 51, 325-346, 10.1016/0304-
- 4 4203(95)00062-3, 1996.
- 5 Cory, R. M., Miller, M. P., McKnight, D. M., Guerard, J. J., and Miller, P. L.: Effect of
- 6 instrument-specific response on the analysis of fulvic acid fluorescence spectra,
- 7 Limnology and Oceanography-Methods, 8, 67-78, 2010.
- 8 Craig, H.: Isotopic variations in meteoric waters, Science, 133, 1702-1703,
- 9 10.1126/science.133.3465.1702, 1961.
- 10 Del Vecchio, R., and Blough, N. V.: On the origin of the optical properties of humic
- substances, Environmental Science & Technology, 38, 3885-3891, 10.1021/es049912h,
- 12 2004.
- 13 Doheny, E. J., Starsoneck, R. J., Striz, E. A., and Mayer., P. M.: Watershed
- 14 characteristics and pre-restoration surface-water hydrology of Minebank Run, Baltimore
- 15 County, Maryland, water years 2002–04. USGS Scien- tific Investigations Rep. 2006–
- 16 5179. USGS, Reston, VA, 2006.
- 17 Duan, S. W., Kaushal, S. S., Groffman, P. M., Band, L. E., and Belt, K. T.: Phosphorus
- 18 export across an urban to rural gradient in the Chesapeake Bay watershed, Journal of
- 19 Geophysical Research-Biogeosciences, 117, 10.1029/2011jg001782, 2012.
- 20 Fellman, J. B., Miller, M. P., Cory, R. M., D'Amore, D. V., and White, D.: Characterizing
- 21 Dissolved Organic Matter Using PARAFAC Modeling of Fluorescence Spectroscopy: A
- 22 Comparison of Two Models, Environmental Science & Technology, 43, 6228-6234,
- 23 10.1021/es900143g, 2009.

1	Groffman, P. M., Law, N. L., Belt, K. T., Band, L. E., and Fisher, G. T.: Nitrogen fluxes
2	and retention in urban watershed ecosystems, Ecosystems, 7, 393-403, 10.1007/s10021-
3	003-0039-x, 2004.
4	Huguet, A., Vacher, L., Relexans, S., Saubusse, S., Froidefond, J. M., and Parlanti, E.:
5	Properties of fluorescent dissolved organic matter in the Gironde Estuary, Organic
6	Geochemistry, 40, 706-719, 10.1016/j.orggeochem.2009.03.002, 2009.
7	Kaushal, S. S., Groffman, P. M., Band, L. E., Shields, C. A., Morgan, R. P., Palmer, M.
8	A., Belt, K. T., Swan, C. M., Findlay, S. E. G., and Fisher, G. T.: Interaction between
9	urbanization and climate variability amplifies watershed nitrate export in Maryland,
10	Environmental Science & Technology, 42, 5872-5878, 10.1021/es800264f, 2008a.
11	Kaushal, S. S., Groffman, P. M., Mayer, P. M., Striz, E., and Gold, A. J.: Effects of
12	stream restoration on denitrification in an urbanizing watershed, Ecological Applications,
13	18, 789-804, 10.1890/07-1159.1, 2008b.
14	Kaushal, S. S., Groffman, P. M., Band, L. E., Elliott, E. M., Shields, C. A., and Kendall,
15	C.: Tracking Nonpoint Source Nitrogen Pollution in Human-Impacted Watersheds,
16	Environmental Science & Technology, 45, 8225-8232, 10.1021/es200779e, 2011.
17	Kendall, C., and Coplen, T. B.: Distribution of oxygen-18 and deuterium in river waters
18	across the United States, Hydrological Processes, 15, 1363-1393, 10.1002/hyp.217, 2001.
19	Klocker, C. A., Kaushal, S. S., Groffman, P. M., Mayer, P. M., and Morgan, R. P.:
20	Nitrogen uptake and denitrification in restored and unrestored streams in urban Maryland,
21	USA, Aquatic Sciences, 71, 411-424, 10.1007/s00027-009-0118-y, 2009.
22	McKnight, D. M., Boyer, E. W., Westerhoff, P. K., Doran, P. T., Kulbe, T., and
23	Andersen, D. T.: Spectrofluorometric characterization of dissolved organic matter for

1 indication of precursor organic material and aromaticity, Limnology and Oceanography,

2 46, 38-48, 2001.

- 3 Newcomer, T. A., Kaushal, S. S., Mayer, P. M., Shields, A. R., Canuel, E. A., Groffman,
- 4 P. M., and Gold, A. J.: Influence of natural and novel organic carbon sources on
- 5 denitrification in forest, degraded urban, and restored streams, Ecological Monographs,
- 6 82, 449-466, 10.5061/dryad.4gk00, 2012.
- 7 Ohno, T.: Fluorescence inner-filtering correction for determining the humification index
- 8 of dissolved organic matter, Environmental Science & Technology, 36, 742-746,
- 9 10.1021/es0155276, 2002.
- 10 Pennino, M. J., Kaushal, S. S., Beaulieu, J. J., Mayer, P. M., and Arango, C. P.: Effects of
- 11 urban stream burial on nitrogen uptake and ecosystem metabolism: implications for
- 12 watershed nitrogen and carbon fluxes, Biogeochemistry, 121, 247-269, 10.1007/s10533-

13 014-9958-1, 2014.

- 14 Ryan, R. J., Welty, C., and Larson, P. C.: Variation in surface water-groundwater
- 15 exchange with land use in an urban stream, Journal of Hydrology, 392, 1-11,
- 16 10.1016/j.jhydrol.2010.06.004, 2010.
- 17 Ryan, R. J., Larson, P. C., and Welty, C.: The logistics and mechanics of conducting
- 18 tracer injection experiments in urban streams, Urban Ecosystems, 14, 87-117,
- 19 10.1007/s11252-010-0144-5, 2011.
- 20 Sivirichi, G. M., Kaushal, S. S., Mayer, P. M., Welty, C., Belt, K. T., Newcomer, T. A.,
- 21 Newcomb, K. D., and Grese, M. M.: Longitudinal variability in streamwater chemistry
- 22 and carbon and nitrogen fluxes in restored and degraded urban stream networks, Journal
- 23 of Environmental Monitoring, 13, 288-303, 10.1039/c0em00055h, 2011.

1	Stolpe, B., Guo, L., Shiller, A. M., and Hassellov, M.: Size and composition of colloidal
2	organic matter and trace elements in the Mississippi River, Pearl River and the northern
3	Gulf of Mexico, as characterized by flow field-flow fractionation, Marine Chemistry,
4	118, 119-128, 10.1016/j.marchem.2009.11.007, 2010.
5	Zsolnay, A., Baigar, E., Jimenez, M., Steinweg, B., and Saccomandi, F.: Differentiating
6	with fluorescence spectroscopy the sources of dissolved organic matter in soils subjected
7	to drying, Chemosphere, 38, 45-50, 10.1016/s0045-6535(98)00166-0, 1999.
8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 4 35 36 37 38 39	

1	Table ST. Cor	nparison of carb	on and nutrient col	ncentrations (mean	± S.E.).
		MBR	RRN	PMR	DRN
	DOC (mg/L)	$1.6 \pm 0.1^{a}$	$1.7 \pm 0.1^{a}$	$2.5 \pm 0.3^{b}$	$3.7 \pm 0.2^{\circ}$
	TOC (mg/L)	$1.6 \pm 0.1^{a}$	$1.9 \pm 0.2^{a}$	$2.9 \pm 0.4^{b}$	$4.0 \pm 0.2^{c}$
	$NO_3^{-}$ (mg/L)	$1.4 \pm 0.0^{a}$	$1.3 \pm 0.05^{b}$	$1.7 \pm 0.1^{\circ}$	$1.0 \pm 0.05^{d}$
	TN (mg/L)	$1.7 \pm 0.1^{a}$	$1.5 \pm 0.05^{b}$	$1.9 \pm 0.1^{\circ}$	$1.3 \pm 0.1^{d}$
	$PO_4^{-3}$ (µg/L)	$17 \pm 0.6^{a}$	$17 \pm 0.7^{a}$	$27 \pm 3.6^{b}$	$20 \pm 0.8^{b}$
	TP ( $\mu$ g/L)	$45 \pm 1.9^{a}$	$51 \pm 2.6^{ab}$	$59 \pm 3.4^{bc}$	$65 \pm 4.2^{\circ}$
	$\Gamma(\mu g/L)$	$9.1 \pm 0.3^{a}$	$8.3 \pm 0.2^{ab}$	$7.6 \pm 0.3^{b}$	$15 \pm 0.7^{\circ}$
	$F^{-}(mg/L)$	$0.10 \pm 0.01^{a}$	$0.10 \pm 0.01^{a}$	$0.20 \pm 0.02^{b}$	$0.17 \pm 0.02^{ab}$
r				wder Mill Run, DRN	
2 3				Serences ( $p < 0.05$ ), b	
4			itinely sampled dat		ased on pan wise
4 5	comparisons of	t three years of for	annery sampled dat	1.	
6					
7					
8					
9					
10					
11					
12					
13					
14					
15					
16					
17					
18					
19					
20					
21					
22					
23					
24					
25					
23 26					
20 27					
27					
29					
30					
31					
32					
33					
34					
35					
36					
37					

1 Table S1. Comparison of carbon and nutrient concentrations (mean  $\pm$  S.E.).

Table S2. Flashiness metrics for mean daily carbon, nitrogen, and phosphorus loads 1

2  $(\text{mean} \pm \text{S.E.}).$ 

			D 10	Powder Mill	
DOC		Minebank Run	Red Run	Run	Dead Run
DOC	Daily Load (g/day)	$19 \pm 2^a$	$35 \pm 11^{a}$	$53 \pm 8^{b}$	$107 \pm 20^{c}$
	CV	$1.85 \pm 0.14^{a}$	$1.64 \pm 0.15^{a}$	$2.29 \pm 0.13^{b}$	$2.71 \pm 0.14^{b}$
	Mean diff. (g/day)	$0.02 \pm 0.00^{a}$	$0.03 \pm 0.01^{b}$	$0.07 \pm 0.01^{\circ}$	$0.15 \pm 0.03^{d}$
	Flashiness Index	$1.00 \pm 0.08^{a}$	$0.73 \pm 0.06^{b}$	$1.20 \pm 0.07^{ac}$	$1.38 \pm 0.08^{\circ}$
NO <sub>3</sub> <sup>-</sup>	Daily Load (g/day)	$10.6 \pm 0.85^{a}$	$10.4 \pm 0.83^{a}$	$19.0 \pm 1.74^{b}$	$11.8 \pm 1.26^{a}$
	CV	$1.23 \pm 0.09^{a}$	$0.68 \pm 0.07^{b}$	$1.61 \pm 0.11^{\circ}$	$1.88 \pm 0.10^{d}$
	Mean diff. (g/day)	$0.007 \pm 0.001^{a}$	$0.004 \pm 0.001^{b}$	$0.018 \pm 0.002^{c}$	$0.013 \pm 0.002^{\circ}$
	Flashiness Index	$0.74\pm0.06^a$	$0.33\pm0.03^b$	$0.88\pm0.06^{ac}$	$1.02 \pm 0.05^{c}$
TN	Daily Load (g/day)	$12.4 \pm 0.99^{a}$	$13.1 \pm 1.23^{a}$	$27.1 \pm 2.99^{b}$	$19.4 \pm 2.33^{ab}$
	CV	$1.24 \pm 0.09^{a}$	$0.82\pm0.08^b$	$1.90 \pm 0.13^{c}$	$2.07 \pm 0.11^{c}$
	Mean diff. (g/day)	$0.01 \pm 0.00^{a}$	$0.01 \pm 0.00^{b}$	$0.03 \pm 0.00^{c}$	$0.02 \pm 0.00^{c}$
	Flashiness Index	$0.74 \pm 0.06^{a}$	$0.39\pm0.03^b$	$1.01 \pm 0.06^{\rm c}$	$1.11 \pm 0.05^{c}$
PO <sub>4</sub> <sup>-3</sup>	Daily Load (g/day)	$0.17 \pm 0.02^{a}$	$0.22 \pm 0.04^{a}$	$0.40 \pm 0.05^{b}$	$0.50 \pm 0.08^{b}$
	CV	$1.65 \pm 0.13^{ab}$	$0.22 \pm 0.01$ $1.39 \pm 0.12^{a}$	$1.96 \pm 0.13^{b}$	$2.57 \pm 0.13^{\circ}$
	CV	$0.0002 \pm$	$0.0002 \pm$	$0.0005 \pm$	$0.0007 \pm$
	Mean diff. (g/day)	$0.00002^{a}$	$0.00002^{a}$	0.0001 <sup>b</sup>	0.0001 <sup>b</sup>
	Flashiness Index	$0.91 \pm 0.07^{a}$	$0.64 \pm 0.05^{b}$	$1.03 \pm 0.07^{a}$	$1.32 \pm 0.07^{\rm c}$
ТР	Daily Load (g/day)	$0.4 \pm 0.05^{a}$	$0.8 \pm 0.21^{ab}$	$0.9 \pm 0.11^{b}$	$1.0 \pm 0.13^{b}$
	CV	$1.61 \pm 0.13^{ab}$	$1.41 \pm 0.13^{a}$	$1.95 \pm 0.13^{bc}$	$2.08 \pm 0.11^{\circ}$
	0.1	$0.0004 \pm$	$0.0007 \pm$	$0.0010 \pm$	$0.0012 \pm$
	Mean diff. (g/day)	0.0001 <sup>a</sup>	0.0003 <sup>a</sup>	0.0001 <sup>b</sup>	$0.0002^{b}$
	Flashiness Index	$0.89 \pm 0.07^{a}$	$0.64 \pm 0.05^{b}$	$1.03 \pm 0.07^{\rm ac}$	$1.11 \pm 0.05^{\circ}$
I.	Daily Load (g/day)	$0.06 \pm 0.003^{a}$	$0.08 \pm 0.01^{a}$	$0.07 \pm 0.01^{a}$	$0.17 \pm 0.02^{b}$
	CV	$0.94 \pm 0.06^{a}$	$1.00 \pm 0.10^{a}$	$1.35 \pm 0.07^{b}$	$1.79 \pm 0.09^{c}$
	Mean diff. (g/day)	$0.00 \pm 0.00^{a}$	$0.00 \pm 0.00^{a}$	$0.00 \pm 0.00^{\rm b}$	$0.00 \pm 0.00^{\circ}$
	Flashiness Index	$0.60 \pm 0.05^{a}$	$0.47 \pm 0.04^{a}$	$0.81 \pm 0.05^{b}$	$0.98 \pm 0.05^{\circ}$
F <sup>-</sup>	Daily Load (g/day)	$0.5 \pm 0.05^{a}$	NA	$41.3 \pm 10.09^{b}$	$1.5 \pm 0.17^{c}$
	CV	$0.76 \pm 0.06^{a}$	NA	$3.68 \pm 0.23^{b}$	$1.63 \pm 0.09^{\circ}$
	Mean diff. (g/day)	$0.70 \pm 0.000$ $0.000 \pm 0.000^{a}$	NA	$0.07 \pm 0.02^{b}$	$0.001 \pm 0.000$
	Flashiness Index	$0.000 \pm 0.000$ $0.53 \pm 0.05^{a}$	NA	$0.07 \pm 0.02$ $2.22 \pm 0.64^{b}$	$0.001 \pm 0.000$ $0.91 \pm 0.04^{\circ}$
D:00-	ent letters (a, b, c, or d) in				

mean absolute value of the difference between daily loads; Flashiness Index = absolute value of mean

3 4 5 6 difference in load between days, divided by the mean load.

7

Different letters (a, b, c, or d) indicate significant differences (p < 0.05), based on daily loads. Values are based on mean monthly data from 2010-2012. CV = Coefficient of Variation; Mean diff. =

		Minebank		Powder Mill	
		Run	Red Run	Run	Dead Run
DOC	Mean (mg/L)	$1.6 \pm 0.09^{a}$	$1.8 \pm 0.10^{a}$	$2.5 \pm 0.24^{b}$	$3.6 \pm 0.19^{c}$
	CV	$0.36 \pm 0.06^{a}$	$0.25 \pm 0.06^{a}$	$0.40 \pm 0.08^{a}$	$0.31 \pm 0.04^{a}$
	Mean diff. (mg/L)	$0.68 \pm 0.09^{a}$	$0.67 \pm 0.12^{a}$	$1.40 \pm 0.27^{ab}$	$1.26 \pm 0.16^{b}$
	Flashiness Index	$0.40 \pm 0.05^{a}$	$0.38 \pm 0.06^{a}$	$0.48 \pm 0.06^{a}$	$0.34 \pm 0.04^{a}$
NO <sub>3</sub> -	Mean (mg/L)	$1.4 \pm 0.05^{a}$	$1.2 \pm 0.06^{b}$	$1.7 \pm 0.07^{c}$	$1.0 \pm 0.06^{d}$
	CV	$0.11 \pm 0.02^{a}$	$0.15 \pm 0.04^{a}$	$0.25 \pm 0.04^{b}$	$0.28 \pm 0.06^{ab}$
	Mean diff. (mg/L)	$0.28\pm0.03^a$	$0.35 \pm 0.05^{ab}$	$0.44 \pm 0.05^{b}$	$0.34 \pm 0.06^{ab}$
	Flashiness Index	$0.20\pm0.02^a$	$0.30\pm0.05^a$	$0.30\pm0.06^a$	$0.36\pm0.06^a$
TN	Mean (mg/L)	$1.7 \pm 0.07^{a}$	$1.4 \pm 0.06^{b}$	$1.9 \pm 0.08^{c}$	$1.4 \pm 0.06^{b}$
	CV	$0.09\pm0.02^a$	$0.11\pm0.02^{ab}$	$0.18\pm0.03^{ab}$	$0.19 \pm 0.04^{b}$
	Mean diff. (mg/L)	$0.38\pm0.08^a$	$0.34 \pm 0.05^{a}$	$0.56 \pm 0.06^{b}$	$0.34 \pm 0.06^{a}$
	Flashiness Index	$0.22\pm0.03^a$	$0.24\pm0.04^a$	$0.44 \pm 0.16^{a}$	$0.26\pm0.04^a$
$PO_4^{-3}$	Mean (µg/L)	$17 \pm 0.8^{a}$	$17 \pm 0.9^{a}$	$27 \pm 3.0^{b}$	$20 \pm 1.0^{b}$
	CV	$0.14 \pm 0.04^{a}$	$0.12 \pm 0.03^{a}$	$0.31 \pm 0.08^{a}$	$0.16 \pm 0.04^{a}$
	Mean diff. (µg/L)	$3.6 \pm 1.0^{a}$	$2.9 \pm 0.7^a$	$14 \pm 5.1^{b}$	$4.0 \pm 0.6^{b}$
	Flashiness Index	$0.29\pm0.11^a$	$0.29\pm0.14^a$	$0.44\pm0.14^a$	$0.25\pm0.09^a$
ТР	Mean (µg/L)	$47 \pm 2^{a}$	$51 \pm 3^{ab}$	$60 \pm 4^{bc}$	$66 \pm 4^{c}$
	CV	$0.18\pm0.03^a$	$0.22 \pm 0.06^{a}$	$0.23 \pm 0.06^{a}$	$0.31 \pm 0.05^{a}$
	Mean diff. (µg/L)	$10.5 \pm 1.1^{a}$	$15.3 \pm 2.5^{a}$	$18.9 \pm 3.6^{ab}$	$26.1 \pm 4.2^{b}$
	Flashiness Index	$0.24\pm0.03^a$	$0.31\pm0.05^a$	$0.33\pm0.06^a$	$0.40\pm0.05^a$
ľ	Mean (µg/L)	$9.1 \pm 0.36^{a}$	$8.0\pm0.28^{b}$	$7.5 \pm 0.34^{b}$	$14.4 \pm 0.69^{c}$
	CV	$0.18\pm0.03^{ab}$	$0.11 \pm 0.02^{a}$	$0.20\pm0.04^{ab}$	$0.31 \pm 0.05^{b}$
	Mean diff. (µg/L)	$1.99\pm0.28^a$	$1.17 \pm 0.20^{b}$	$1.71 \pm 0.20^{a}$	$5.25 \pm 0.51^{c}$
	Flashiness Index	$0.22\pm0.03^a$	$0.15 \pm 0.03^{b}$	$0.25\pm0.04^a$	$0.42\pm0.06^c$
F <sup>-</sup>	Mean (mg/L)	$0.11 \pm 0.02^{a}$	$0.10\pm0.02^{ab}$	$0.18 \pm 0.02^{b}$	$0.17\pm0.04^{ab}$
	CV	$0.51 \pm 0.12^{a}$	$0.31\pm0.23^a$	$0.39\pm0.07^a$	$0.42 \pm 0.10^{a}$
	Mean diff. (µg/L)	$0.08\pm0.02^a$	$0.04 \pm \mathrm{NA}^{\mathrm{a}}$	$0.08 \pm 0.01^{a}$	$0.17 \pm 0.05^{a}$
	Flashiness Index	$0.72 \pm 0.14^{a}$	$0.31 \pm NA^{a}$	$0.41 \pm 0.06^{b}$	$0.95 \pm 0.24^{ab}$

Different letters (a, b, c, or d) indicate significant differences (p < 0.05). CV = coefficient of variation; Mean diff. = mean absolute value of the difference between routine sample concentrations; Flashiness Index = mean change in concentration / mean concentration. 3 4

				Powder Mill	
		Minebank Run	Red Run	Run	Dead Run
$\delta^2$ H-H <sub>2</sub> O	Mean (‰)	$-43.3 \pm 1.74^{a}$	$-44.9 \pm 2.35^{a}$	$-43.0 \pm 2.19^{a}$	$-43.8 \pm 3.22^{a}$
	CV	$-0.13 \pm 0.06^{a}$	$-0.14 \pm 0.05^{a}$	$-0.24 \pm 0.06^{a}$	$-0.20 \pm 0.06^{a}$
	Mean Diff (‰)	$5.55 \pm 1.57^{a}$	$9.21 \pm 2.06^{a}$	$10.83 \pm 2.09^{a}$	$10.29 \pm 2.27^{a}$
	Flash Index	$-0.12 \pm 0.03^{a}$	$-0.19 \pm 0.03^{ab}$	$-0.24 \pm 0.04^{b}$	$-0.23 \pm 0.05^{ab}$
δ <sup>18</sup> O-H <sub>2</sub> O		$-6.7 \pm 0.20^{a}$	$-7.0 \pm 0.26^{a}$	$-6.6 \pm 0.27^{a}$	$-6.7 \pm 0.40^{a}$
0 О-п <sub>2</sub> О	Mean (‰)	$-0.11 \pm 0.03^{a}$	$-0.12 \pm 0.03^{a}$	$-0.19 \pm 0.04^{a}$	$-0.16 \pm 0.04^{a}$
	CV		$-0.12 \pm 0.03$ $1.1 \pm 0.23^{a}$	$-0.19 \pm 0.04$ $1.3 \pm 0.23^{a}$	$-0.16 \pm 0.04$ $1.3 \pm 0.27^{a}$
	Mean Diff (‰)	$0.82 \pm 0.15^{a}$			
	Flash Index	$-0.12 \pm 0.02^{a}$	$-0.15 \pm 0.03^{a}$	$-0.20 \pm 0.03^{a}$	$-0.19 \pm 0.04^{a}$
$\delta^{15}$ N-NO <sub>3</sub> <sup>-</sup>	Mean (‰)	$7.0 \pm 0.25^{ac}$	$6.2 \pm 0.25^{a}$	$8.1 \pm 0.13^{b}$	$7.4 \pm 0.27^{bc}$
5	CV	$0.11 \pm 0.03^{a}$	$0.16 \pm 0.06^{a}$	$0.15 \pm 0.03^{a}$	$0.20 \pm 0.04^{a}$
	Mean Diff (‰)	$0.87 \pm 0.13^{a}$	$1.0 \pm 0.17^{a}$	$1.13 \pm 0.13^{ab}$	$1.49 \pm 0.16^{b}$
	Flash Index	$0.13 \pm 0.02^{a}$	$0.19\pm0.05^{ab}$	$0.14\pm0.02^{ab}$	$0.22 \pm 0.03^{b}$
\$180 MG -		to to a zzab	4.1. × 0.22 <sup>a</sup>	z o . o z bc	<b>7</b> 0 + 0 0 c <sup>C</sup>
$\delta^{18}$ O-NO <sub>3</sub> <sup>-</sup>	Mean (‰)	$4.9 \pm 0.37^{ab}$	$4.1 \pm 0.32^{a}$	$5.8 \pm 0.51^{bc}$	$7.8 \pm 0.86^{\circ}$
	CV	$0.28 \pm 0.06^{a}$	$0.28 \pm 0.07^{a}$	$0.34 \pm 0.09^{a}$	$0.45 \pm 0.07^{a}$
	Mean Diff (‰)	$2.04 \pm 0.37^{a}$	$1.65 \pm 0.31^{a}$	$2.69 \pm 0.59^{a}$	$4.91 \pm 0.81^{b}$
	Flash Index	$0.40 \pm 0.07^{a}$	$0.37 \pm 0.05^{a}$	$0.39 \pm 0.07^{a}$	$0.58\pm0.08^a$

1	Table S4	Flashiness	metrics fo	or water and	I nitrate sources	$(mean \pm SE)$

3 4 5 Different letters (a, b, c, or d) indicate significant differences (p < 0.05). CV = coefficient of variation; Mean Diff = mean absolute value of the difference between biweekly isotope values; Flashiness Index = mean change in isotope value / mean isotope value.

		Minebank Run	Red Run	Powder Mill Run	Dead Run
HIX	Mean	$0.88\pm0.01^a$	$0.79\pm0.05^{ab}$	$0.80 \pm 0.02^{b}$	$0.83 \pm 0.02^{ab}$
	CV	$0.06 \pm 0.02^{a}$	$0.02 \pm 0.00^{a}$	$0.04 \pm 0.01^{a}$	$0.05 \pm 0.03^{a}$
	Mean Diff	$0.07 \pm 0.02^{a}$	$0.09\pm0.05^a$	$0.07\pm0.01^a$	$0.06 \pm 0.02^{a}$
	Flash Index	$0.08 \pm 0.02^{a}$	$0.16 \pm 0.11^{a}$	$0.09\pm0.02^a$	$0.07\pm0.02^a$
BIX	Mean	$0.69 \pm 0.08^{a}$	$0.61 \pm 0.06^{b}$	$0.71 \pm 0.07^{a}$	$0.75\pm0.08^a$
	CV	$0.03 \pm 0.01^{a}$	$0.06 \pm 0.02^{a}$	$0.03 \pm 0.01^{a}$	$0.09 \pm 0.03^{a}$
	Mean Diff	$0.14 \pm 0.07^{a}$	$0.12 \pm 0.06^{a}$	$0.14 \pm 0.07^{a}$	$0.19\pm0.07^a$
	Flash Index	$0.94\pm0.83^a$	$0.56 \pm 0.46^{a}$	$0.64 \pm 0.54^{a}$	$0.75\pm0.60^a$
FI	Mean	$1.15 \pm 0.10^{a}$	$1.10 \pm 0.10^{a}$	$1.12 \pm 0.10^{a}$	$1.22 \pm 0.07^{a}$
	CV	$0.03 \pm 0.02^{a}$	$0.03 \pm 0.01^{a}$	$0.03 \pm 0.00^{a}$	$0.10 \pm 0.03^{a}$
	Mean Diff	$0.19 \pm 0.10^{a}$	$0.18 \pm 0.10^{a}$	$0.18 \pm 0.10^{a}$	$0.22 \pm 0.07^{a}$
	Flash Index	$0.36 \pm 0.29^{a}$	$0.38 \pm 0.31^{a}$	$0.35 \pm 0.27^{a}$	$0.20 \pm 0.07^{a}$
P/H ratio	Mean	$0.80 \pm 0.06^{a}$	$0.73 \pm 0.03^{a}$	$1.16 \pm 0.08^{b}$	$0.92 \pm 0.07^{ab}$
	CV	$0.25 \pm 0.08^{a}$	$0.30 \pm 0.04^{a}$	$0.23 \pm 0.07^{a}$	$0.25 \pm 0.09^{a}$
	Mean Diff	$0.27 \pm 0.05^{ab}$	$0.17 \pm 0.04^{a}$	$0.23 \pm 0.07$ $0.41 \pm 0.06^{b}$	$0.28 \pm 0.10^{ab}$
	Flash Index	$0.34 \pm 0.07^{a}$	$0.24 \pm 0.05^{a}$	$0.36 \pm 0.06^{a}$	$0.28 \pm 0.08^{a}$
Protein	Mean	$0.09 \pm 0.03^{a}$	$0.11 \pm 0.04^{a}$	$0.25 \pm 0.12^{a}$	$0.26 \pm 0.08^{a}$
	CV	$0.49 \pm 0.17^{a}$	$0.50 \pm 0.10^{a}$	$0.53 \pm 0.10^{a}$	$0.54 \pm 0.08^{a}$
	Mean Diff	$0.03 \pm 0.01^{a}$	$0.06 \pm 0.02^{a}$	$0.18 \pm 0.11^{a}$	$0.01 \pm 0.00^{a}$
	Flash Index	$0.05 \pm 0.01^{a}$ $0.46 \pm 0.13^{a}$	$0.58 \pm 0.13^{a}$	$0.74 \pm 0.28^{a}$	$0.45 \pm 0.10^{a}$
Fulvic	Mean	$0.14 \pm 0.04^{a}$	$0.17 \pm 0.06^{a}$	$0.38 \pm 0.24^{a}$	$0.40 \pm 0.14^{a}$
	CV	$0.44 \pm 0.13^{a}$	$0.48 \pm 0.15^{a}$	$0.57 \pm 0.12^{a}$	$0.46 \pm 0.09^{a}$
	Mean Diff	$0.09 \pm 0.03^{a}$	$0.13 \pm 0.05^{a}$	$0.52 \pm 0.29^{a}$	$0.32 \pm 0.13^{a}$
	Flash Index	$2.15 \pm 1.68^{a}$	$2.98 \pm 2.44^{a}$	$10.08 \pm 9.41^{a}$	$3.75 \pm 3.31^{a}$
Humic	Mean	$0.11 \pm 0.03^{a}$	$0.13 \pm 0.04^{a}$	$0.23 \pm 0.14^{a}$	$0.27 \pm 0.09^{a}$
	CV	$0.43 \pm 0.14^{a}$	$0.19 \pm 0.01$ $0.50 \pm 0.18^{a}$	$0.23 \pm 0.11^{a}$ $0.59 \pm 0.13^{a}$	$0.27 \pm 0.09^{a}$ $0.48 \pm 0.09^{a}$
	C v Mean Diff	$0.45 \pm 0.14$ $0.06 \pm 0.02^{a}$	$0.09 \pm 0.03^{a}$	$0.30 \pm 0.16^{a}$	$0.48 \pm 0.09^{a}$ $0.19 \pm 0.07^{a}$
		$1.05 \pm 0.59^{a}$	$1.28 \pm 0.76^{a}$	$4.01 \pm 3.33^{a}$	$0.19 \pm 0.07$ $1.49 \pm 1.06^{a}$

1 Table S5. Flashiness metrics for carbon source metrics (mean  $\pm$  S.E.).

Different letters (a, b, c, or d) indicate significant differences (p < 0.05).

CV = coefficient of variation; Mean Diff = mean absolute value of the difference between biweekly sample indices; Flashiness Index = mean change in index value / mean index value.

HIX = humification index; BIX = biological freshness index; FI = fluorescence index; P/H ratio = protein-

234 567 to-humic ratio; Protein = protein-like organic matter fluorescence intensity index; Fulvic = fulvic-like organic matter fluorescence intensity index; Humic = humic-like organic matter fluorescence intensity index.

8

9

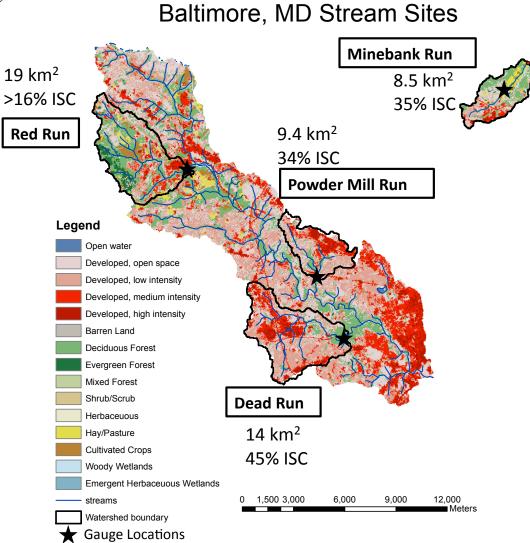
10

11

1 Figure Captions

Figure S1. Site map of study streams in the Baltimore Metropolitan Region showing land cover characteristics. Figure S2. Flow duration curves for each site: (a) Minebank Run, (b) Red Run (c), Powder Mill Run, and (c) Dead Run, showing the range of flows and also at what flows the 3 years of regular sampling took place. Figure S3. Comparison of (a) DOC, (b)  $NO_3^{-3}$  and (c)  $PO_4^{-3}$  concentration vs. time, from routine samples collected at each of the four stream gage sites. Figure S4. Water isotope comparison:  $\delta^2$ H-H<sub>2</sub>O vs.  $\delta^{18}$ O-H<sub>2</sub>O for (a) all samples from the gauge locations and (b) samples taken from the mouth of each stream during the winter 2010. GMWL = Global Meteoric Water Line, LMWL = Local Meteoric Water Line (Craig, 1961; Kendall and Coplen, 2001) (Craig 1961; Kendall & Coplen 2001). Figure S5. Seasonal relationship between  $\delta^{15}$ N-NO<sub>3</sub><sup>-</sup> vs.  $\delta^{18}$ O-NO<sub>3</sub><sup>-</sup> for (a) spring 2010, (b) spring 2011, (c), summer 2010, and (d) summer 2011, for routine samples at each of the four stream sites. 

Figure S1. 



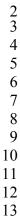
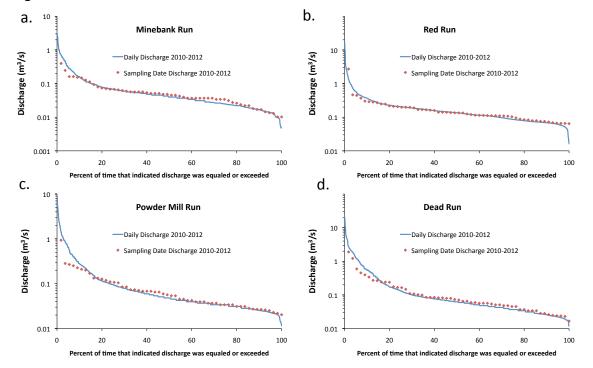
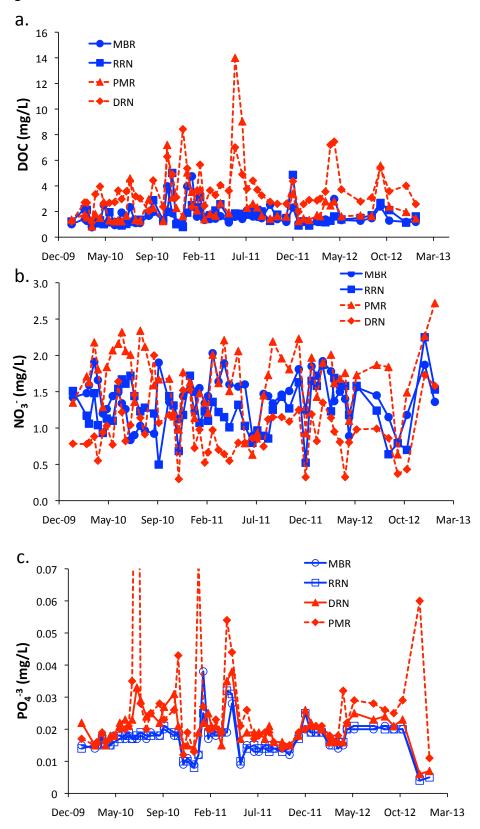


Figure S2. 



1 Figure S3.



1 Figure S4.

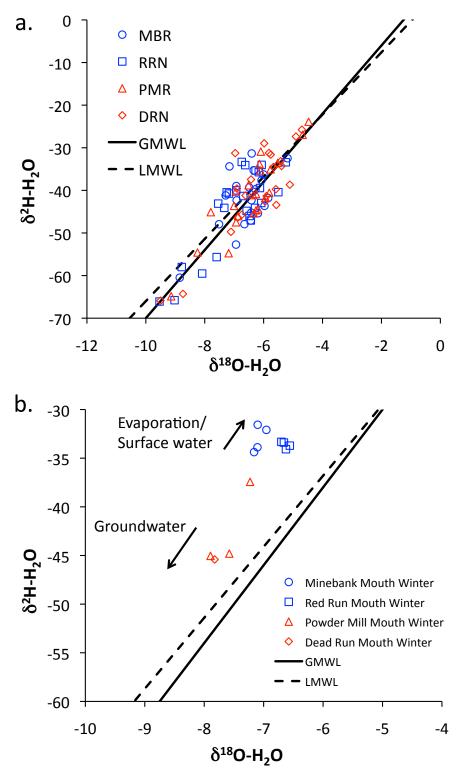
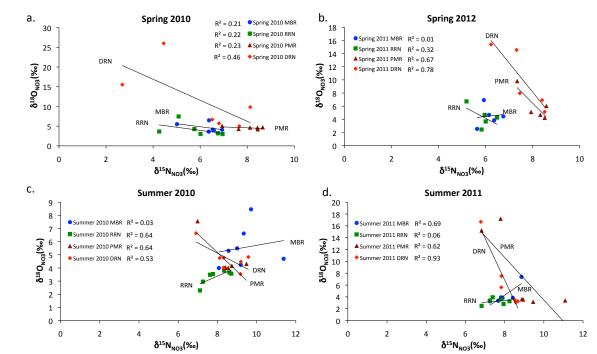


Figure S5. 





- 4 5