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Using groundwater age to understand sources and dynamics of nutrient contamination through the catchment into Lake Rotorua, New Zealand

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Abstract

The water quality of Lake Rotorua has declined continuously over the past 50 yr despite mitigation efforts over recent decades. Delayed response of the groundwater discharges to historic land-use intensification 50 yr ago was the reason suggested by early tritium measurements, which indicated large transit times through the groundwater system. We use the isotopic and chemistry signature of the groundwater for detailed understanding of the origin, fate, flow pathways, lag times, and future loads of contaminants. A unique set of high-quality tritium data over more than four decades, encompassing the time when the tritium spike from nuclear weapons testing moved through the groundwater system, allows us to determine detailed age distribution parameters of the water discharging into Lake Rotorua.

The Rotorua volcanic groundwater system is complicated due to the highly complex geology that has evolved through volcanic activity. Vertical and steeply-inclined geological contacts preclude a simple flow model. The extent of the Lake Rotorua groundwater catchment is difficult to establish due to the deep water table in large areas, combined with inhomogeneous groundwater flow patterns.

Hierarchical cluster analysis of the water chemistry parameters provided evidence of the recharge source of the large springs near the lake shore, with discharge from the Mamaku ignimbrite through lake sediment layers. Groundwater chemistry and age data show clearly the source of nutrients that cause lake eutrophication, nitrate from agricultural activities and phosphate from geologic sources. With a naturally high phosphate load reaching the lake continuously via all streams, the only effective way to limit algae blooms and improve lake water quality in such environments is by limiting the nitrate load.

The groundwater in the Rotorua catchment, once it has passed through the soil zone, shows no further decrease in dissolved oxygen, indicating absence of electron donors in the aquifer that could facilitate microbial denitrification reactions. Nitrate from land-use activities that leaches out of the root zone of agricultural land into the deeper

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part of the groundwater system must be expected to travel with the groundwater to the lake.

The old age and the highly mixed nature of the water discharges imply a very slow and lagged response of the streams and the lake to anthropogenic contaminants in the catchment, such as nitrate. Using the age distribution as deduced from tritium time series data measured in the stream discharges into the lake allows prediction of future nutrient loads from historic land-use activities 50 yr ago. For Hamurana Stream, the largest stream to Lake Rotorua, it takes more than a hundred years for the groundwater-dominated stream discharge to adjust to changes in land-use activities. These time scales apply to activities that cause contamination, but also to remediation action.

1 Introduction

Detailed information on groundwater age distribution is required for the Lake Rotorua catchment to understand the agricultural contaminant loads that travel from land to the lake with the groundwater and discharge via springs and streams into the lake, with a large lag time. The water quality of Lake Rotorua has declined continuously over the past 50 yr, despite cessation of direct-to-lake sewage discharge in 1991 (Burger et al., 2011) and the fencing-off of streams in grazing land in parts of the lake catchment.

Land use in the catchment has intensified significantly over the past 60 yr and is now predominantly forest (39%), pasture (27%), and dairy (9%) (Burger et al., 2011; Rutherford et al., 2009). Increasing nitrate concentrations had been observed in virtually all of the major streams flowing into the lake during the period 1968–2003 (Hoare, 1987; Rutherford, 2003). Nitrate in groundwater under dairy farms and discharging from them measures $10 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$. In the absence of significant over-land run-off, nutrients from land use are transported with the water through the groundwater system to the lake. Early tritium measurements indicated large transit times through the groundwater system (the subject of this study). With a time lag $> 50 \text{ yr}$ in the groundwater

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system, nitrate loads to the lake may be expected to increase further in the future due to delayed arrival of nutrients from historic land use as they ultimately discharge from the groundwater system via the springs and streams into the lake. This trend will be exacerbated by any further intensification of land use within the catchment over recent decades, as this recently recharged water has largely not yet reached the streams (Morgenstern and Gordon, 2006).

Groundwater age is a crucial parameter for understanding the dynamics of the groundwater and the contaminants that travel with the water. Determining water age, and hence transit times, allows identification of delayed impacts of past and present land-use practices on water quality (Böhlke and Denver, 1995; Katz et al., 2001, 2004; McGuire et al., 2002; MacDonald et al., 2003; Broers, 2004; Moore et al., 2006; Sophocleous, 2012), and for identification of anthropogenic vs. geologic impacts on groundwater quality (Morgenstern and Daughney, 2012). Understanding the dynamics of groundwater is fundamental for most groundwater issues. Water age defines the transit time of water through catchments and hence is vital for conceptual understanding of catchment processes such as response to rainfall, stream flow generation, recharge source and rate (McGuire and McDonnell, 2006; Morgenstern et al., 2010, 2012; Stewart et al., 2010; Cartwright and Morgenstern, 2012). Water age, being directly related to fluid flux, is also very useful for calibrating numerical surface water and groundwater transport models (Goode, 1996; Burton et al., 2002; Molson and Frind, 2005; Bethke and Johnson, 2008). Water age provides important information on vulnerability to contamination and can therefore be used to assess the security of drinking water supplies, particularly from groundwater bores (Darling et al., 2005; Morris et al., 2005; New Zealand Ministry of Health, 2008). Water age measurements can also be used to quantify rates of hydrochemical evolution resulting from water–rock interaction (Katz et al., 1995; Burns et al., 2003; Glynn and Plummer, 2005; Beyer et al., 2014). These applications of water dating cover the spectrum from applied water resource management to fundamental scientific research.

In all of the above-mentioned applications it is important to constrain not only the mean age of water, but also the distribution of ages within a sample from the groundwater discharge. Transit time determinations in catchment hydrology typically identify a range of water ages contributing to stream flow, and the time- and location-dependent distribution of transit times provides insight into the processes that generate runoff (Maloszewski and Zuber, 1982; McGuire and McDonnell, 2006; Stewart et al., 2007; McDonnell et al., 2010). Use of water age determinations for calibration of numerical transport models must also account for the full distribution of age and its variation in space and time (Goode, 1996; Cornaton et al., 2011; Cornaton, 2012). Assessment of the security of drinking water from groundwater bores also requires an understanding of the water's age distribution (Eberts et al., 2012; Morgenstern, 2004). For example, New Zealand legislation states that a water supply bore is considered secure (unlikely to have a risk of contamination by pathogenic organisms) when less than 0.005 % of the water has been present in the aquifer for less than one year (New Zealand Ministry of Health, 2008).

For the Lake Rotorua catchment study, tritium is the tracer of choice. Tritium dating can be applied to both river/stream water and groundwater, whereas gas tracers are less suitable for surface waters that are in contact with air. Tritium ages, in contrast to gas tracer ages, include travel through the unsaturated zone (Zoellmann et al., 2001; Cook and Solomon, 1995); travel times can be up to > 40 yr through the thick unsaturated zones of the Rotorua catchment ignimbrite aquifers (Morgenstern et al., 2004). Tritium is not subject to transformation, degradation or retardation during water transport through the catchment. Tritium dating is applicable to water with mean residence times of up to about 200 yr (Cook and Solomon, 1997; Morgenstern and Daughney, 2012), as is typical of New Zealand's dynamic surface waters and shallow groundwaters. In addition, monitoring the movement of the pulse-shaped bomb-tritium through groundwater systems is an excellent opportunity to obtain information about the age distribution parameters of the groundwater. This is particularly useful in groundwater systems, such as the Rotorua system, that have high uncertainties within flow models

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due to a deep water table and preferential flow paths. Finally, tritium is a particularly sensitive marker for study of the timing of nitrate contamination in groundwater, because the main anthropogenic nitrate contamination of groundwater systems started coincidentally with the bomb-tritium peak from the atmospheric nuclear weapons testing after WWII, giving that water a very distinct age signal.

For the Rotorua catchment we have an extensive data set available over time and space. Tritium time series data for the main lake inflows cover more than four decades, and data covering the last decade are available with an extremely high spatial resolution of about 100 sites in the Lake Rotorua catchment. Tritium concentration can be measured at GNS Science with the required extremely high accuracy using 95-fold electrolytic enrichment prior to ultra-low level liquid scintillation spectrometry (Morgenstern and Taylor, 2009). Tritium is highly applicable for groundwater dating in the post-bomb low-tritium environment of the Southern Hemisphere, as bomb tritium from atmospheric thermonuclear weapons testing has now been washed out from the atmosphere for 20 yr, as is described in detail in Morgenstern and Daughney (2012).

The objective of this study is to understand the origin, fate, flow pathways, lag times, and future loads of contaminants that cause lake eutrophication in the Lake Rotorua catchment, central North Island, New Zealand. This will assist in mitigating the deterioration in lake water quality since the 1960s (Rutherford et al., 1989) that threatens the lake's significant cultural and tourist value. Environmental hydrochemistry tracers and age tracers are used to identify the recharge source of the main water discharges to Lake Rotorua, identify the source of the contaminants (anthropogenic vs. geologic), and to evaluate the water age distributions in order to understand groundwater processes, lag times, and the groundwater flow dynamics. The Rotorua groundwater system is complicated due to the catchment's highly complex geology, which has evolved through volcanic activity, and due to the deep water table of > 50 m in large areas, which prevents detailed groundwater studies and introduces uncertainty in catchment boundaries and flow patterns. The complex geology leads to inhomogeneous groundwater flow patterns, as indicated by large parts of the catchment having particularly large

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From 200 to 61 ka, volcanic activity in the vicinity of Lake Rotorua was relatively subdued. A number of eruptions from the Okataina Volcanic Centre (OVC), located to the east of the Rotorua caldera, produced widely dispersed but relatively thin airfall deposits. These pyroclastic materials caused periodic damming of drainage pathways and led to fluctuations in lake level that in turn resulted in widespread and variably thick sediments being deposited in the Rotorua Caldera. This period of relatively quiet volcanic activity ended with the Rotoiti and Earthquake Flat eruptions from the OVC, which produced widespread pyroclastic deposits, including the non-welded ignimbrites of the Rotoiti Formation and the Earthquake Flat Formation.

From 61 ka to present, numerous eruptions from the OVC (the most recent of which was in 1886) deposited airfall layers in the Lake Rotorua catchment area. Numerous rhyolite lava units were also emplaced during this period. The periodic deposition of pyroclastic materials, along with activity on faults of the Taupo Rift (Leonard et al., 2010), presumably caused fluctuations in the lake level, with current lake level being reached sometime within the last few thousand years (White et al., 2004). Due to the decline of lake level, Holocene alluvial sand and gravel deposits are found in stream channels and around the current lake shoreline.

The southern Rotorua basin hosts a vigorous geothermal system producing many hot water, hot mud, steam and geyser features, along with gas emission, between the southern edge of the Lake and about the southern edge of the caldera (Fig. 1). There is hot local groundwater flow in this area, generally flowing down hill northwards into the lake. Beyond this relatively confined area the groundwater system does not appear to interact with fluids from this geothermal system.

2.2 Hydrology

Lake Rotorua has a surface area of 79 km² and a mean depth of 10.8 m (Burger et al., 2011), with a total water volume of 0.85 km³. The assumed total catchment area is ca. 475 km² (White and Rutherford, 2009) (Fig. 1).

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Annual rainfall in the catchment is strongly affected by topography and varies from more than 2200 mm northwest of the lake to less than 1400 mm southeast of the lake (Hoare, 1980; White et al., 2007; Rutherford et al., 2008). Approximately 50 % of rainfall infiltrates into the groundwater system. This is based on two sources of information: (1) comparisons of rainfall and actual evapotranspiration that have been made for various parts of the catchment (Hoare, 1980; Dell, 1982; White et al., 2004, 2007; Rutherford et al., 2008); and (2) data from paired lysimeters, a standard rain gauge and a ground level rain gauge installed at Kaharoa (White et al., 2007) (Fig. 1). With 50 % of rainfall recharge, total infiltration into the groundwater system is estimated to be 14 500 L s⁻¹, based on the catchment shown in Fig. 1, excluding rainfall inputs direct to the lake, and assuming recharge is 50 % of rainfall. This rainfall recharge supports stream flow and potentially direct inputs of groundwater to the lake.

There are nine major streams (Fig. 1, for abbreviations refer to Table 1) and several minor streams that flow into the lake; the remainder of the inflows are provided from direct inputs of rainfall and lake-front features, and potentially from groundwater seepage through the lake bed. The major streams are baseflow-controlled and characterized by very constant water flow (Hoare, 1980) and temperature, and groundwater-derived baseflow accounts for approximately 90 % of the average flow in the typical Rotorua stream (Hoare, 1987). Baseflows in the nine major streams entering Lake Rotorua cumulatively amount to 11 800 L s⁻¹, and total inflows to the lake from minor streams and lake-front features amounts to 350 L s⁻¹ (Hoare, 1980; White et al., 2007) (Table 1).

With the lake water volume of 0.85 km³, the lake water turnover time via the groundwater-fed streams is 2.2 yr. The only surface outflow occurs through Ohau Channel via Lake Rotoiti (Fig. 1). Water balance calculations suggest that the total catchment area exceeds the surface water catchment area (White et al., 2007); in other words, groundwater from outside the surface water catchment is flowing through the aquifer system to Lake Rotorua (White and Rutherford, 2009).

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and Zuber, 1982, use the variable η ; $\eta = 1/f$). When $f = 0$ the model becomes equivalent to the piston flow model, and when $f = 1$ it becomes equivalent to the exponential model.

5 The two parameters of the response functions, the mean residence time (MRT) and the distribution of transit times (f), are determined by convoluting the input (tritium concentration in rainfall measured over time) to model water passage through the hydrological system in a way that matches the output (e.g., tritium concentrations measured in wells or springs). Because of its pulse-shaped input, tritium is a particularly sensitive tracer for identifying both of these two parameters, which can be deduced uniquely
10 by comparing the delay and the dispersion of the bomb-pulse tritium in the groundwater to that from tritium in the original rain input. This method is particularly useful for interpretation of ages of groundwater in the Lake Rotorua catchment because most of these groundwater discharges lack any information on mixing of groundwater with varying flow path lengths and of different age. The exponential piston flow model was
15 used because, with its age distribution, it has produced good matches to most (about a hundred) tritium time series data from springs and wells throughout New Zealand, including all hydrogeological situations (Morgenstern and Daughney, 2012).

For tracer age interpretation, the integral (Eq. 1) was used to convolute the historical rainfall tracer input to an output that reflects mixing in a groundwater system, with the
20 best match of the simulated output to the measured output time-series data (Fig. 3). The TracerLPM workbook (Jurgens et al., 2012) was used. The tritium input function is based on concentrations of tritium in rainfall measured monthly since the 1960s at Kaitoke, near Wellington, New Zealand (Morgenstern and Taylor, 2009). The Kaitoke rainfall input function is multiplied by a scaling factor of 0.87 to account for variation
25 in atmospheric tritium concentrations due to latitude and orographic factors. For the prevailing New Zealand climatic conditions there is no need for correction of the tritium input for seasonal infiltration (Morgenstern et al., 2010).

The problem of ambiguity in tritium dating over the last decades is demonstrated in Fig. 3. Hangarua Spring discharges very old water with a mean residence time of

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about 90 yr (see below), but during the late 1980s its tritium concentration was similar to that of very young water (rain curve in Fig. 3). At that time, the tritium concentration in Hangarua Spring would have been in agreement with both very young water and old
5 water with a mean residence time of 90 yr. Tritium data covering several decades, however, clearly distinguish this old water (low tritium concentration) from young rain water. Figure 3 also shows that due to the fading of the bomb-tritium in recent decades (tritium decay over four tritium half-lives since the bomb spike), in recent years the tritium concentration of old water is clearly distinguishable (lower) from that of young water, without ambiguity. The tritium time series data allow also for constraining groundwater
10 mixing models. Figure 3 shows the model output curves that match the measured tritium data. Given sufficient analytical accuracy, this is also possible for extremely low tritium concentrations; the data for Hamurana water intake spring (blue curve in Fig. 3) are all below 0.4 TU.

The application of mixing models is described in Morgenstern and Daughney (2012, Sect. 2.6). While throughout New Zealand generally we obtained good matches between tritium input and output data using the exponential piston flow model, it was not
15 possible to obtain adequate matches in the ignimbrite area of the Rotorua catchment using such a simple model. Using the dispersion model did not significantly improve the matches. However, using a binary mixing model of parallel contribution from two exponential piston flow models resulted in excellent matches. We justify this binary mixing
20 model by inferring two different flow contributions in the catchment to the stream and spring flow, mainly from deep old groundwater, and a contribution from younger groundwater from shallow aquifers, as indicated by very deep groundwater tables in the area (generally > 50 m) but at the same time also minor stream flows maintained by shallow
25 aquifers.

3.2 Sample collection and analysis

Samples were collected from 41 springs, from 31 groundwater-dominated stream flow sites, and from 26 groundwater wells. To obtain realistic residence times of the water

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passing through the whole groundwater system, sampling focused on the naturally flowing groundwater discharges, the springs and streams. Samples were collected at times of base flow conditions.

All nine major streams were sampled multiple times near the inflow into the lake. Most of these tritium time series go back to the early seventies and encompass the passage of the “bomb” tritium peak through the groundwater system, allowing determination of detailed age distribution parameters for these major inflows to the lake. These “historic” samples had been collected sporadically for various projects over the decades to study the transfer of the bomb-tritium through the hydrologic cycle. Over the recent decade, the streams have also been sampled for tritium at various points upstream, at various main confluences, or at main springs to obtain a detailed spatial distribution of water ages. Springs and wells were also once sampled for CFCs, SF₆, argon, and nitrogen, to obtain complementary age information.

Sampling locations are shown in Fig. 1. Many of the sites have no road access, with some of them in remote steep gullies. A portable sampling system was required for the gas samples to allow fresh water from the well or spring to be pumped into the sample bottles from below the water surface without air contact. We used a pneumatic Bennet pump, powered from a cylinder of compressed air at the remote locations, and from a compressor powered by the car battery at sites with car access. Sampling from streams (tritium only) involved simply dipping the bottle under the water surface and filling the bottle.

Sampling methods for hydrochemistry and nutrients were according to Daughney et al. (2007). Age tracer samples were collected without filtration or preservation. For tritium, a one-litre plastic bottle was filled to the top. For CFC samples, two 125 mL glass bottles with aluminium liner cap were filled, rigorously excluding air contact by filling from the bottom via a nylon tube and three times volume replacement below the surface of the overflowing sample water. One litre bottles were filled for SF₆.

Analytical details for hydrochemistry are described in Daughney et al. (2014). Details of the tritium analysis procedure are described in Morgenstern and Taylor (2009).

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While the early tritium measurements in the 1970s were performed with a detection limit of approximately 0.1 Tritium Units (TU), we now achieve significantly lower detection limit of 0.02 (TU) via tritium enrichment by a factor of 95 and reproducibility of tritium enrichment of 1 % via deuterium-calibration. Analysis procedures for CFC-11, CFC-12, and SF₆ are described in van der Raaij and Beyer (2014). Detection limits are $3 \times 10^{-15} \text{ mol kg}^{-1}$ for CFCs and $2 \times 10^{-17} \text{ mol kg}^{-1}$ for SF₆. Dissolved argon and nitrogen concentrations were measured for estimating the temperature at the time of recharge, and the excess air concentration, as described by Heaton and Vogel (1981), for calculation of the atmospheric partial pressure (ppt) of CFCs and SF₆ at the time of recharge.

4 Results and discussion

In the following section hydrochemistry cluster analysis and hydrochemistry evolution are discussed to assess the geographic sources of groundwater and groundwater processes in the aquifer. The nutrients nitrate, sulphate, potassium, and phosphate are discussed to evaluate their source (anthropogenic vs. geologic), lag time, fate, and impact on lake eutrophication. The age distributions of the groundwater discharges to Lake Rotorua are discussed to understand the conceptual groundwater flow pattern and the lag time in the groundwater system. The ultimate goal of this project is the use of the hydrochemistry and groundwater age parameters for calibration of a groundwater transport model for improved management of the nutrient loads to the lake – the subject of follow up papers.

4.1 Groundwater age interpretation

To obtain the unique solution for both parameters of the age distribution for a specific model, time series data are required (Sect. 3.1). Most of the large water inflows into Lake Rotorua have long time series data available (up to over four decades), allowing

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for well constrained age distribution parameters for both MRT and the fraction f between different flow models (or Peclet number for the dispersion model). The tritium time series data, together with the matching lumped parameter model simulations, are shown in Fig. 3 for two of the large springs, and in Fig. 4 for six of the large streams (covering 2/3 of stream baseflow to Lake Rotorua). For the sites with shorter time series data (sub-catchment stream discharges, groundwater wells), most of the sites have at least sufficient time series or multi tracer data for unambiguous robust age interpretations. If fraction f cannot be established uniquely from the tritium time series data, we applied mixing models that matched long tritium time series data from other sites with similar hydrogeologic settings to these sites. All 96 sites with tritium time series or tritium and complementary CFC and SF₆ data have unambiguous age interpretation. For the tritium time series data shown in Figs. 3 and 4, the lumped parameter models, with their respective age distribution parameters that match the measured data, are listed in Table 2.

Throughout New Zealand, and including all hydrogeologic situations (but mainly groundwater wells), the measured data of approximately a hundred long tritium time series over several decades can usually be matched by a simple lumped parameter model (e.g., Morgenstern and Daughney, 2012). The long-term tritium data from most of the large stream discharges shown in Fig. 4, however, cannot be matched by a simple model such as the exponential piston flow or dispersion model and require a more complex groundwater flow model combination. Using a binary mixing model, with parallel contributions from two exponential piston flow models, resulted in excellent matches. We justify this binary mixing model by inferring two different flow contributions in the catchment to stream and spring flow – from deep old groundwater, as indicated by very deep groundwater tables in the area (generally > 50 m), and from younger groundwater from shallow aquifers, as indicated by minor stream flows maintained by shallow aquifers. In Table 2 are also listed the average mean residence times between the two parallel models, weighted by their fraction within the total flow. For the MRTs, errors are typically ±1 yr for MRTs < 5 yr, ±2 yr for MRTs between 5 and 10 yr, ±3 yr for MRTs

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between 10 and 50 yr, ±5 yr for MRTs between 50 and 100 yr, and larger errors for older water towards the detection limit.

For convenience, the average MRTs are also listed in Fig. 4 next to their model output curve. It is obvious from Fig. 4 that all the main streams discharge very old water into Lake Rotorua. The tritium response of the streams is clearly distinguishable from that of young rain water (grey line). The youngest water discharge, Ngongotaha Stream (green), has an average MRT of 30 yr. All other main streams discharge significantly older water, up to MRT of 145 yr for Waingaehe Stream (dark blue). Note that even though bomb-tritium from atmospheric nuclear weapons testing in the 1960s has decayed enough to not cause ambiguous age interpretations anymore, it is still possible to detect the tail of the bomb-tritium for matching model parameters if tritium analyses have sufficient accuracy; all data for the Hamurana water intake spring (blue in Fig. 3) are below 0.4 TU.

Most of the streams and springs discharge very old groundwater into Lake Rotorua, with MRTs typically between 50–150 yr, indicating discharge from a large groundwater system with large water residence (turn-over) times. Only a few small sub-catchments with minor flow rates discharge young water (MRT < 20 yr), indicating local geologic units below the surface that don't allow water to infiltrate into and flow through larger, deeper groundwater systems.

The old age of the majority of the Lake Rotorua water inflows and the highly mixed nature of the water discharges (note the high fractions of exponential flow, up to 100 %, in Table 2) implies a very slow and lagged response of the streams and the lake to anthropogenic contaminants in the catchment, such as nitrate. The majority of the nitrate load currently discharging into the lake is thus from land-use activities 30 and more years ago.

About a hundred stream and groundwater well samples have been dated in the Lake Rotorua catchment. The groundwater age distributions are used in the following sections to identify hydrochemistry evolution, sources of contaminants, and to predict future nitrate loads that will enter Lake Rotorua from the large contaminated groundwater

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system. In a future paper, the conceptual groundwater flow model in the Lake Rotorua catchment will be inferred from the groundwater age distribution data. The data will subsequently be used for calibration of a groundwater transport model.

4.2 Hydrochemistry and recharge source

5 The hydrochemical composition of the groundwater and surface waters in the Rotorua catchment have been investigated by Morgenstern et al. (2004) and Donath et al. (2014); the following section summarises the results relevant to the present study.

Hydrochemistry is driven by interaction between water and the different major lithologies and can be used to track the origin of the groundwater. Hydrochemistry reflects the rhyolite ignimbrite and lava aquifer lithologies that dominate the Lake Rotorua catchment, with much lower concentrations of Ca, Mg and SO_4 and much higher concentrations of F, PO_4 and SiO_2 , compared to groundwater in other parts of New Zealand.

Several statistical and graphical techniques were applied to characterise the variations in hydrochemistry across parts of the catchment. Hierarchical Cluster Analysis (HCA) was shown to be a useful technique to identify samples with similar hydrochemical composition, and to relate the groundwater samples to their origin from one of the main aquifer lithologic units. HCA conducted with Ward's linkage rule allowed the samples to be partitioned into four hydrochemical clusters. Three of the clusters, accounting for the majority of the samples, were inferred to reflect water–rock interaction with the dominant lithologies in the catchment, namely Mamaku ignimbrite, lava, or paleo-lake sediments. Hydrochemistry inferred to indicate interaction between water and the Mamaku ignimbrite had Na and HCO_3 as the dominant cation and anion, respectively, and among the highest concentrations of Mg, PO_4 and SiO_2 and among the lowest concentrations of F, K and SO_4 observed. Hydrochemistry inferred to indicate interaction between water and rhyolite lava also had Na and HCO_3 as the dominant cation and anion, respectively, but had relatively low concentrations of PO_4 and among the highest concentrations of K. Hydrochemistry inferred to indicate interaction between water and sediments had Na-Ca- HCO_3 -Cl water type and relatively low concentrations

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of SiO_2 . The remaining cluster was inferred to represent geothermal influences on the hydrochemistry (e.g., elevated concentrations of Na, Cl, SO_4 , SiO_2 and NH_4).

Figure 5 shows the hydrochemical clusters of the water samples inferred to indicate interaction between water and Mamaku ignimbrite (light blue), lava (red), or lacustrine sediments (dark yellow). Note that samples assigned to the cluster inferred to indicate geothermal are not displayed in Fig. 5 or discussed further in the present study because geothermal influence is not the subject of this study.

Samples with hydrochemistry indicative of interaction with the Mamaku ignimbrite occur predominantly in the north and northwest portion of the catchment (Fig. 5, blue). All of the large springs discharging into Hamurana (Ham), Awahou (Awh), and Utuhina (Utu) streams have a Mamaku ignimbrite hydrochemistry signature (blue circles). The stream reaches in the Mamaku ignimbrite area upstream from these large springs are usually dry. This, together with the Mamaku ignimbrite hydrochemistry signature, implies that these large springs drain the large Mamaku ignimbrite areas upstream that have negligible surface runoff. In the north-west in the Hamurana and Awhou catchments, these springs emerge close to the lake shore within the sediment area (Fig. 5, yellow), indicating that close to the lake shore, due to the more impermeable nature of the intra-caldera sediment, the deeper groundwater flow from the Mamaku ignimbrite is forced to the surface. All of these large springs emerge within the slopes of the sediment formation where sediment layers are thinner compared to the level area closer to the present lake shore; the thin nature of the sediments on these slopes allows the water from the underlying ignimbrite to flow to the surface. No large spring occurs in the level area closer to the lake, where sediments are thicker. Also the large Utuhina Spring in the southeast emerges within the slopes of the sediment (Fig. 5), indicating the more impermeable nature of the sediment forcing the groundwater from the Mamaku ignimbrite to the surface in the area of thin sediment layers. The Utuhina Spring emerges below a small local lava dome feature, but the ignimbrite signature of the water indicates that this spring is the discharge from the large ignimbrite area southwest

from the Mamaku ignimbrite, the pH increases from just 6.3 to 6.6 over the age range from 0 to 100 yr, with a power law fit of $\ln(\text{pH}) = 0.025 \cdot \ln(\text{MRT}) + 1.76$, $R^2 = 0.47$. The groundwater from the sediment formation show no clear trend in pH with groundwater age, but displays higher pH for relatively young water, pH 6.5–7.2 for water with MRT between 1 and 60 yr.

As groundwater becomes more evolved over time due to water–rock interaction, concentrations of phosphorus, silica, bicarbonate, and fluoride typically increase due to dissolution of volcanic glass, silicate minerals, carbonates, and fluoride likely deposited from the volatile phases in magma exsolved during eruption (Morgenstern and Daughney, 2012). With increasing groundwater age, ion concentrations are expected to increase up to a maximum equilibrium concentration. In the following discussion, samples indicative of geothermal influence are excluded because they follow different thermodynamic equilibrium. Groundwater from the sediment formation often follows different or unclear trends compared to the rhyolite ignimbrite and lava formations.

Dissolved reactive phosphate ($\text{PO}_4\text{-P}$) in groundwater from all three formations, the rhyolite lava and ignimbrite aquifers, and the sediments originating from the same formations, shows excellent correlation with groundwater age (Fig. 7a, black curve), with $\ln(\text{PO}_4\text{-P}) = 0.458 \cdot \ln(\text{MRT}) - 4.72$, $R^2 = 0.94$.

Silica (SiO_2) also shows good correlation with groundwater age for the rhyolite ignimbrite and lava formations (Fig. 7b). The silica concentration of groundwater in lava formations (red circles) increases faster compared to ignimbrite (blue circles). The power fit to the lava data is $\ln(\text{SiO}_2) = 0.310 \cdot \ln(\text{MRT}) + 2.96$, $R^2 = 0.88$ (red curve), and to the ignimbrite data is $\ln(\text{SiO}_2) = 0.238 \cdot \ln(\text{MRT}) + 3.05$, $R^2 = 0.83$ (blue curve). The correlation between silica and groundwater age for the lacustrine sediment aquifers (yellow circles) is rather erratic; high silica concentration can also occur in very young groundwater.

For bicarbonate (HCO_3), only groundwater samples from the Mamaku ignimbrite show a reasonable correlation with groundwater age, with a power fit of $\ln(\text{HCO}_3) =$

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$0.206 \cdot \ln(\text{MRT}) + 2.58$, $R^2 = 0.71$ (Fig. 7c). The high data point at 58 mg L^{-1} was considered an outlier and not included in the fit.

Sodium (Na) in general also shows increasing concentration with groundwater age because it is part of common minerals and leached from these. But the correlation is poor (Fig. 7d). Note that elevated Na in young groundwater can also be caused by land-use impacts (Morgenstern and Daughney, 2012). Considering the high data point at 15.8 mg L^{-1} an outlier (it is from the same site considered as an outlier for HCO_3), the correlation for the data from all three formations is $\text{Na} = 0.017 \cdot \text{MRT} + 7.42$, $R^2 = 0.38$.

Fluoride concentrations (F) show good correlations with age for the rhyolite ignimbrite and lava formations (Fig. 7e), even though the trends are masked by the fact that the concentrations are close to the detection limit. Concentrations increase in lava formations significantly faster, with a power fit of $\ln(\text{F}) = 1.087 \cdot \ln(\text{MRT}) - 6.44$, $R^2 = 0.97$, compared to ignimbrite with $\ln(\text{F}) = 0.238 \cdot \ln(\text{MRT}) - 3.99$, $R^2 = 0.58$.

In groundwater of the Rotorua catchment (excluding groundwater from the eastern catchment indicating geothermal influence, which is not the subject of this study), the hydrochemistry parameters phosphate, silica, bicarbonate, sodium, and fluoride are purely of geologic origin, because they do not display elevated concentrations in young water that was recharged during the time of anthropogenic high intensity land-use activities. The groundwater samples show, for the rhyolite Mamaku ignimbrite and lava formations, excellent correlations across the western and northern Lake Rotorua catchment. All samples in each of these geologic units follow similar trends of hydrochemistry concentration vs. mean residence time, indicating the relatively homogeneous nature of these aquifers. In rhyolite lava formations, geochemical reactions lead to increased pH, Si, and F in groundwater significantly faster than in ignimbrite, indicating higher reaction rates for dissolution of these elements from lava formations. Rather erratic trends in water originating from the sediments suggest that these are not a homogeneous formation but rather finely layered lensoidal geologic deposits that vary spatially and support complex or fragmented groundwater systems. Good trends of hydrochemistry vs. groundwater age are an indication of robust age interpretations.

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4.4 Nutrients

Elevated nutrient levels in surface water cause poisonous algal blooms and lake eutrophication. Presence of both phosphate and nitrate, above a threshold concentration, triggers algae blooms in lakes. Limitation of one of these, P or N, can limit algae blooms. In New Zealand, increasing nutrient loads from high intensity animal farming and fertilisers have triggered lake eutrophication. In the absence of significant overland runoff, nitrate travels from the land to the lake via the groundwater, which eventually discharges into streams and lakes.

Nutrient concentrations in New Zealand groundwaters from agricultural sources have increased steadily after European settlement in the early 19th century and with development of the meat industry after 1880 (Morgenstern and Daughney, 2012). In a national context, for groundwater recharged before 1880 at pre-anthropogenic pristine conditions, low nutrient concentrations prevailed (e.g., nitrate $< 0.2 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$). In groundwater recharged between 1880 and 1955, nutrient concentrations are slightly elevated due to low intensity land use. In groundwater recharged after 1955 a sharp increase of nutrient concentrations is observed due to the impact of high intensity land use after World War II (Morgenstern and Daughney, 2012).

The main nutrients derived from land use in the Rotorua catchment, as indicated by elevated concentrations in young groundwater, are nitrate (NO_3), sulphate (SO_4), and potassium (K). These nutrient concentrations are shown in Figs. 8 and 9a and b vs. mean residence time, also correlated to recharge year (upper x-axes). The majority of the chemistry data of the Rotorua data set are from calendar year 2003, therefore mean residence time of about 50 and 125 yr correspond to mean groundwater recharge years 1955 and 1880, respectively. Homogeneous nitrate concentrations in discharges from within sub-catchments of typically $0.7 \pm 0.2 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$ indicate that nutrient inputs are derived from diffuse rather than a small number of point sources, pointing to agricultural sources.

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Nitrate concentrations (Fig. 8) in oxic groundwaters with MRT > 125 yr (recharged prior to 1880) in the Rotorua catchment are higher, with up to about $0.7 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$ (dotted line in Fig. 8) compared to other regions in New Zealand with $0.2 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$. The reason for elevated nitrate in water despite a high mean residence time is the high degree of mixing in the groundwater discharges from the highly porous unconfined Rotorua ignimbrite aquifers (see next section). In such aquifer conditions, groundwater from short and long flow paths converge at the groundwater discharges, causing a high degree of mixing of young and old water. For example, groundwater with a mean residence time (MRT) as high as 170 yr, using an exponential piston flow model with 95 % exponential flow volume within the total flow volume (see next section), contains over 20 % of water recharged after 1955. This post-1955 water can contribute significant amounts of nitrate from high-intensity land use, raising the nitrate concentration of the water mix considerably, despite such a long residence time.

A significant increase in nitrate occurred only recently (Fig. 8). Apart from one data point, an increase up to $1.5 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$ was observed only in water with MRTs of less than 75 yr, and a dramatic increase up to $14 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$ was observed in water with MRTs of less than 50 yr. Note the dramatic increase of nitrate in water with MRT < 20 yr, reflecting the increased conversions to dairy farming over the most recent decades before 2000. As the majority of the water discharges into Lake Rotorua are significantly older than a few decades, up to MRTs of 120 yr, the impact of the dairy conversions and their nitrate loads over the recent decades has to a large extent not yet reached the lake. Increased nitrate loads to the lake over the next decades must be expected as these nitrate loads work their way through the large groundwater system and eventually discharge into the streams and lake.

Sulphate and potassium are part of fertilizers and also show elevated concentrations in young groundwater (Fig. 9). Note that sulphate in groundwater in the eastern lake catchment has much elevated concentrations, up to $40 \text{ mg L}^{-1} \text{ SO}_4$, due to geothermal influence. Groundwater with indications of geothermal influence is not discussed in this study. Also note that sulphate can be biased due to anoxic SO_4 reduction. The

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data shown are, however, not from anoxic groundwater environments. Sulphate and potassium show slightly elevated concentrations, up to a factor of three, only in water with MRT < 50 yr, corresponding to water recharged after approximately 1950. Sulphate concentrations (in $\text{mg L}^{-1} \text{SO}_4$) in the Rotorua volcanic aquifers are significantly lower compared to a national survey: pre-anthropogenic concentration of 2 vs. 12, and high intensity land-use concentrations of up to 6 vs. 94 for the Rotorua volcanic aquifers and the national survey (Morgenstern and Daughney, 2012), respectively.

Phosphate, in conjunction with nitrate the cause for lake eutrophication, is not elevated in young groundwater (Fig. 7a) despite its frequent application as super-phosphate fertilisers. Absence of elevated PO_4 in young groundwater implies that fertiliser phosphate from non-point sources has not yet reached the saturated groundwater systems and is still retained in the soil. The presence of elevated PO_4 only in old groundwater indicates that its source is purely due to geological factors, because these waters were recharged before land-use intensification. PO_4 concentrations up to $0.1 \text{ mg L}^{-1} \text{PO}_4\text{-P}$ are observed, due to phosphate leaching from the rhyolite ignimbrite and lava formations. With most groundwater discharging into Lake Rotorua being very old (MRT > 50 yr), the water has naturally high PO_4 concentrations, well above the threshold for primary algae production of c. 0.03 mg L^{-1} total phosphate (Dodds, 2007).

The high phosphate load to the lake via groundwater is natural. As the turn-over time of the lake water is only 2.2 yr via the high PO_4 -bearing streams, there is a constantly high PO_4 load reaching the lake via all streams. Therefore, the only effective way to limit algae blooms and improve lake water quality in such environments is by limiting the nitrate load.

4.5 Prediction of future nitrate load

The water quality of Lake Rotorua has declined continuously over the past 60 yr, responding very slowly to historical agricultural and urban development in the catchment,

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and large amounts of groundwater have insidiously become contaminated over the last 60 yr because of the long travel times through the groundwater system of the Lake Rotorua catchment. The response time of the groundwater system to mitigation action will also be lengthy; it will take similar time frames until the contaminated water is flushed out of the aquifers. To improve lake water quality and define reduction targets for nutrients that affect lake water quality, prediction of future contaminant loads from current and historic activities in the Lake Rotorua catchment are required.

In the previous section we have shown that of the two main contaminants that together cause lake eutrophication, phosphorus is naturally present in the volcanic lake environment, but nitrate from anthropogenic sources has been leaching into the groundwater since the onset of industrial agriculture, delivering increasing nitrate loads to the lake. Figure 8 shows significantly elevated nitrate concentrations in groundwater recharged after 1955.

Due to the large lag time in the groundwater system, these younger groundwaters, with their higher nitrate load, have not yet worked their way fully through the groundwater system. Significant fractions of the groundwaters discharging to the lake are older (Figs. 3 and 4), with MRT > 50 yr, and were recharged before land-use intensification. Therefore the water discharges into the lake are currently still diluted by old pristine water. With the delayed arrival of nitrate from historic land use, which ultimately will discharge from the groundwater system via the springs and streams into the lake, nitrate loads to the lake from historic land-use activities must be expected to increase further in the future. No significant denitrification can be expected in the Rotorua groundwater system (Fig. 6a).

The age distributions functions derived from the tritium time series data in the stream discharges to Lake Rotorua (Table 2) can be used to project the future arrival to the lake of water that was recharged since land-use development in the catchment (Morgenstern and Gordon, 2006). The age distribution function for Hamurana Stream, the largest stream (Table 1), which discharges some of the oldest water to the lake (Fig. 2), is shown in Fig. 10.

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Figure 10 shows the two superimposed age distributions of exponential piston flow models: EPM2 with younger water of MRT = 12 yr and EPM1 with significantly older water of MRT = 185, together with the average MRT = 125 yr between the two models (blue). Only the water younger than 55 yr has been recharged after land-use intensification (red shaded) and contains elevated nitrate. The cumulative fraction of land-use impacted water is about 45 %, implying that more than half of the water is still pristine old water. After this old water is completely displaced by land-use impacted water, the nitrogen load of Hamurana Stream will approximately double. The projected increase in nitrogen load over time, as derived from the age distribution, is shown in Fig. 11.

Nitrate is clearly the major component of nitrogen in the Rotorua groundwater system (Morgenstern et al., 2004). Concentrations of nitrate in the catchment were low ($0.14 \text{ mg L}^{-1} \text{ NO}_3\text{-N}$) prior to catchment development, as determined from old groundwater (Morgenstern and Gordon, 2006). The prediction of nitrogen load increase is calculated by scaling the nitrogen load currently measured in the stream (full symbol, Fig. 11) according to its fraction of land-use impacted water to the various years over time, using the age distribution (Fig. 10).

Good agreement with average historic monitoring results of nitrogen loads (Rutherford, 2003, hollow symbols in Fig. 11) confirms that the assumptions regarding the baseline concentration and timing of nitrogen input in the catchment are reasonable. Using the age distributions derived for all stream discharges to the lake, we also projected the total nitrogen load increase to Lake Rotorua (Morgenstern and Gordon, 2006). In regards to phosphorus, there are no elevated phosphorus concentrations in young groundwater (Fig. 7a) and the phosphorus load to Lake Rotorua is projected to stay constant, as long as fertilizer phosphate does not break through the soil into the groundwater.

The time scales necessary for the Hamurana Stream to adjust to changes in land-use activities in the catchment are long. Due to the long residence time of the water in the large aquifer system, it takes more than a hundred years for the groundwater

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discharge to the lake to adjust to changes in land-use activities. These time scales apply to activities that cause contamination, but also to remediation action.

This projection of nitrogen load via the stream is based on actual nitrogen concentrations in the stream (combined with the age of the water) and accounts only for the nitrogen from land-use activities that leaches out of the root zone of agricultural land into the deeper part of the groundwater system. Any nitrogen uptake in the soil is already taken into account.

The above nitrogen prediction is based on constant nitrogen input since catchment development. This trend will, however, be exacerbated by any further intensification of land use within the catchment over recent decades, as this recently recharged water has largely not yet reached the streams.

5 Conclusions

This study shows how the isotopic and chemistry signature of groundwater can be used to help determine the sources and the dynamics of groundwater and contaminants that travel with it, in particular in complex groundwater systems that are difficult to characterise using conventional hydrogeologic methods, such as that of the Lake Rotorua catchment. The isotopic and chemistry signatures of the major groundwater-dominated stream discharges to the lake, after passing through the large aquifer system of the catchment, allow us to understand groundwater processes and lag time on a catchment scale.

Tritium time series data and complementary age tracers SF_6 and CFCs can be used to establish age distribution parameters, allowing for understanding of groundwater processes and dynamics, and the timing of groundwater contamination. This is particularly useful in catchments where little information is available on historic land-use activities.

After long-standing controversies about the recharge areas and hydraulic connections of the large springs near the northern shore of Lake Rotorua, hierarchical cluster analysis of the water chemistry parameters has provided evidence of their recharge

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areas, and the flow of Mamaku ignimbrite discharges through lake sediment layers in slope areas where the sediments are weaker.

Groundwater chemistry and age data show clearly the source of nutrients that discharge with the groundwater into the lake and cause lake eutrophication. Low nitrate concentration in old oxic groundwater and high nitrate concentration in young groundwater recharged after catchment development in the 1950s implies an anthropogenic source of nitrate from agricultural activities, while low phosphate (PO_4) concentrations in young groundwater but high PO_4 concentrations in old groundwater imply a geologic source. High PO_4 is a natural constituent of the groundwater that discharges via the streams into the lake, and with a turn-over time of the lake water of only 2.2 yr, there is a constantly high PO_4 load reaching the lake via all streams. Therefore, the only effective way to limit algae blooms and improve lake water quality in such environments is by limiting the nitrate load.

The groundwater in the Rotorua catchment, once it has passed through the soil zone, shows no further decrease in dissolved oxygen over the full range of residence time of the water in the aquifer, indicating absence of electron donors in the aquifer (e.g. organic matter) that could facilitate microbial denitrification reactions (Kendall and McDonnell, 1998; Tesoriero et al., 2007). Nitrate from land-use activities that leaches out of the root zone of agricultural land into the deeper part of the groundwater system is unlikely to undergo any significant degree of reduction through denitrification and must be expected to travel with the groundwater to the lake.

The old age of the water, with mean residence time of > 50 yr for most water discharges to the lake, implies that there is a large lag time for transmission of the nitrate through the groundwater system. Younger groundwaters, with their higher nitrate load, have not yet worked their way fully through the groundwater system. With increasing arrival of this nitrate from historic land uses, a further increase of the nitrate load to the lake must be expected in the future.

The old age and the highly mixed nature of the water discharges imply a very slow and lagged response of the streams and the lake to anthropogenic contaminants in

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the catchment, such as nitrate. Using the age distribution as deduced from tritium time series data measured in the stream discharges to the lake allows extrapolation of the nutrient load from historic land-use activities into the future. For Hamurana Stream, the largest stream to Lake Rotorua, it takes more than a hundred years for the groundwater-dominated stream discharge to adjust to changes in land-use activities. These time scales apply to activities that cause contamination, but also to remediation action.

Without age information on the groundwater-dominated streams, it would be difficult to obtain such an understanding of groundwater process, groundwater dynamics, and contaminant loads that travel with the groundwater.

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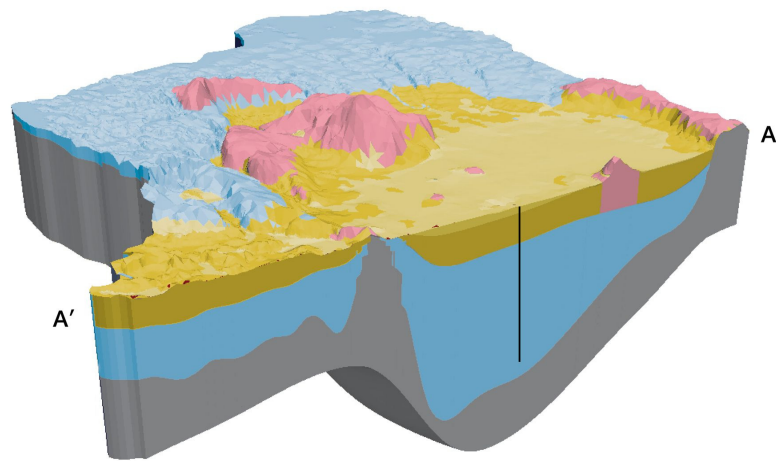


Figure 2. Three-dimensional geological model of the Lake Rotorua catchment area (from White et al., 2007). The location of the vertical cut-away face A'–A is shown in Fig. 1. The vertical exaggeration is 5×, with a 1 km vertical scale bar shown in the centre. The colour scheme is similar to that in Fig. 1 (bottom to top): Pre-Mamaku formations (grey), Mamaku Plateau Formation (light blue), sediment formation (dark yellow), lava domes (red) and Holocene alluvial deposits (light yellow).

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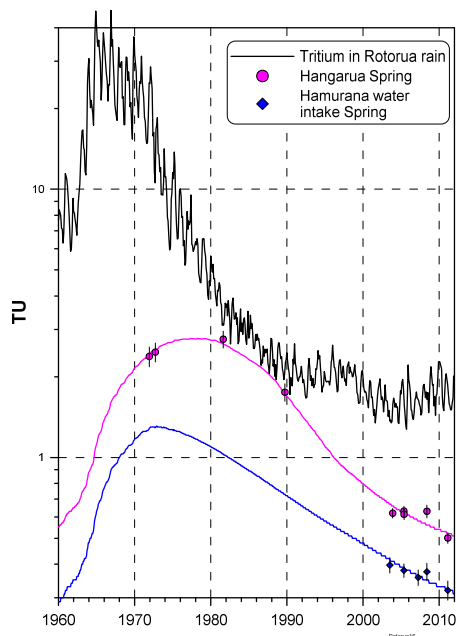


Figure 3. Tritium rain input for the Rotorua catchment, and measured tritium output at Hangarua Spring and at Hamurana water intake spring. The input curve is based on monthly measurements in Kaitoke near Wellington, New Zealand, scaled to the latitude of Rotorua with a factor 0.87, and smoothed by an exponential piston flow model with 0.3 yr mean residence time and 50 % exponential flow within the total flow volume. One TU = one tritium atom per 10^{18} hydrogen atoms. For the spring samples one-sigma measurement errors are shown. Note the logarithmic scale of the TU axis.

9952

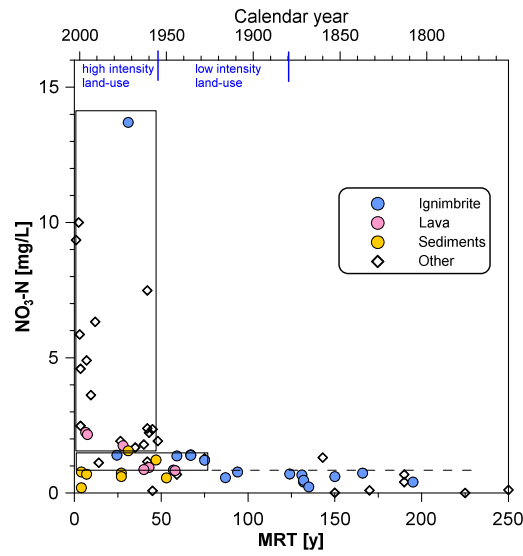


Figure 8. Nitrate (NO_3) vs. mean residence time (MRT). The sample colour code indicates water origin from the relevant geologic formation, as indicated by Hierarchical Cluster Analysis (HCA).

9957

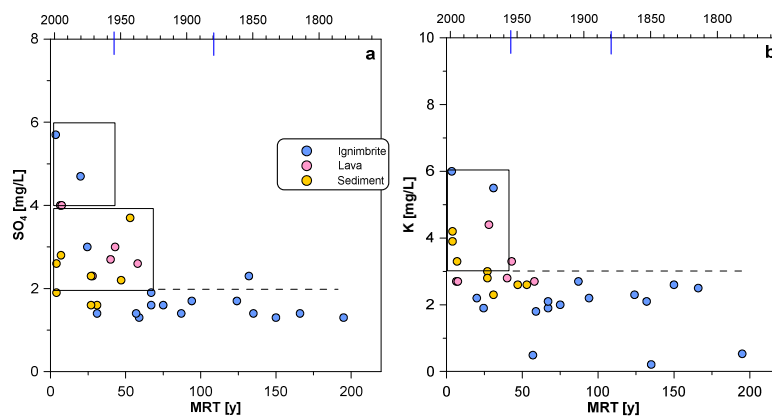


Figure 9. (a) Sulphate (SO_4) and (b) potassium (K) vs. mean residence time (MRT). The sample colour code indicates water origin from the relevant geologic formation, as indicated by Hierarchical Cluster Analysis (HCA). The upper axis indicates calendar year.

9958

