

# Using groundwater age and hydrochemistry to understand sources and dynamics of nutrient contamination through the catchment into Lake Rotorua, New Zealand

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Keywords:

Catchment hydrology, groundwater source, nutrients, nitrate, geochemical reaction rates, hierarchical cluster analysis, lake eutrophication, surface water, groundwater dating, age distribution, tritium

## Abstract

The water quality of Lake Rotorua has steadily declined over the past 50 years despite mitigation efforts over recent decades. Delayed response of the groundwater discharges to historic land-use intensification 50 years ago was the reason suggested by early tritium measurements, which indicated large transit times through the groundwater system. We use the isotopic and chemistry signature of the groundwater for detailed understanding of the origin, fate, flow pathways, lag times, and future loads of contaminants. A unique set of high-quality tritium data over more than four decades, encompassing the time when the tritium spike from nuclear weapons testing moved through the groundwater system, allows us to determine detailed age distribution parameters of the water discharging into Lake Rotorua.

The Rotorua volcanic groundwater system is complicated due to the highly complex geology that has evolved through volcanic activity. Vertical and steeply-inclined geological contacts preclude a simple flow model. The extent of the Lake Rotorua groundwater catchment is difficult to establish due to the deep water table in large areas, combined with inhomogeneous groundwater flow patterns.

Hierarchical cluster analysis of the water chemistry parameters provided evidence of the recharge source of the large springs near the lake shore, with discharge from the Mamaku ignimbrite through lake sediment layers. Groundwater chemistry and age data show clearly the source of nutrients that cause lake eutrophication, nitrate from agricultural activities and phosphate from geologic sources. With a naturally high phosphate load reaching the lake continuously via all streams, the only effective way to limit algae blooms and improve lake water quality in such environments is by limiting the nitrate load.

The groundwater in the Rotorua catchment, once it has passed through the soil zone, shows no further decrease in dissolved oxygen, indicating absence of bioavailable electron donors along flow paths that could facilitate microbial denitrification reactions. Nitrate from land-use activities that leaches out of the root zone of agricultural land into the deeper part of the groundwater system must be expected to travel with the groundwater to the lake.

The old age and the highly mixed nature of the water discharges imply a very slow and lagged response of the streams and the lake to anthropogenic contaminants in the catchment, such as nitrate. Using the age distribution as deduced from tritium time series data measured in the stream discharges into the lake allows prediction of future nutrient loads from historic land-use activities 50 years ago. For Hamurana Stream, the largest stream to Lake Rotorua, it takes more than a hundred years for the

53 groundwater-dominated stream discharge to adjust to changes in land-use activities. About half of the  
54 currently discharging water is still pristine old water, and after this old water is completely displaced by  
55 water affected by land use, the nitrogen load of Hamurana Stream will approximately double. These  
56 time scales apply to activities that cause contamination, but also to remediation action.

57

# 1 Introduction

Detailed information on groundwater age distribution is required for the Lake Rotorua catchment to understand the agricultural contaminant loads that travel from land to the lake with the groundwater and discharge via springs and streams into the lake, with a large lag time. The water quality of Lake Rotorua has declined continuously over the past 50 years, despite cessation of direct-to-lake sewage discharge in 1991 (Burger et al., 2011) and the fencing-off of streams in grazing land in parts of the lake catchment.

Land use in the catchment has intensified significantly over the past 60 years and is now predominantly forest (39%), pasture (27%), and dairy (9%) (Burger et al., 2011; Rutherford et al., 2009). Increasing nitrate concentrations had been observed in virtually all of the major streams flowing into the lake during the period 1968-2003 (Hoare, 1987; Rutherford, 2003). We measured nitrate concentrations of 6-10 mg/L  $\text{NO}_3\text{-N}$  in three young groundwater samples under dairy farms in the SE catchment. In the absence of significant over-land run-off, nutrients from land use are transported with the water through the groundwater system to the lake. Early tritium measurements indicated large transit times through the groundwater system (the subject of this study). With a time lag > 50 years in the groundwater system, nitrate loads to the lake may be expected to increase further in the future due to delayed arrival of nutrients from historic land use as they ultimately discharge from the groundwater system via the springs and streams into the lake. This trend will be exacerbated by any further intensification of land use within the catchment over recent decades, as this recently recharged water has largely not yet reached the streams (Morgenstern and Gordon, 2006).

Groundwater age is a crucial parameter for understanding the dynamics of the groundwater and the contaminants that travel with the water. Determining water age, and hence transit times, allows identification of delayed impacts of past and present land-use practices on water quality (Böhlke and Denver, 1995; Katz et al., 2001; McGuire et al., 2002; MacDonald et al., 2003; Broers, 2004; Katz et al., 2004; Moore et al., 2006), and for identification of anthropogenic versus geologic impacts on groundwater quality (Morgenstern and Daughney, 2012). Understanding the dynamics of groundwater is fundamental for most groundwater issues. Water age is defined by the transit time of water through catchments and hence is vital for conceptual understanding of catchment processes such as response to rainfall, stream flow generation, recharge source and rate (McGuire and McDonnell, 2006; Morgenstern et al., 2010; Stewart et al., 2010, Morgenstern et al., 2012; Cartwright and Morgenstern, 2012). Water age, being directly related to fluid flux, is also very useful for calibrating numerical surface water and groundwater transport models (Goode, 1996; Burton et al., 2002; Molson and Frind, 2005; Bethke and Johnson, 2008). Water age provides important information on vulnerability to contamination and can therefore be used to assess the security of drinking water supplies, particularly from groundwater bores (Darling et al., 2005; Morris et al., 2005; New Zealand Ministry of Health, 2008). Water age measurements can also be used to quantify rates of hydrochemical

102 evolution resulting from water-rock interaction (Katz et al., 1995; Burns et al., 2003;  
103 Glynn and Plummer, 2005, Beyer et al., under review). These applications of water  
104 dating cover the spectrum from applied water resource management to fundamental  
105 scientific research.

106 In all of the above-mentioned applications it is important to constrain not only  
107 the mean age of water, but also the distribution of ages within a sample from the  
108 groundwater discharge. Transit time determinations in catchment hydrology typically  
109 identify a range of water ages contributing to stream flow, and the time- and location-  
110 dependent distribution of transit times provides insight into the processes that  
111 generate runoff (Maloszewski and Zuber, 1982; McGuire and McDonnell, 2006;  
112 Stewart et al., 2007; McDonnell et al., 2010). Use of water age determinations for  
113 calibration of numerical transport models must also account for the full distribution of  
114 age and its variation in space and time (Goode, 1996; Cornaton et al., 2011; Cornaton,  
115 2012). Assessment of the security of drinking water from groundwater bores also  
116 requires an understanding of the water's age distribution (Eberts et al., 2012,  
117 Morgenstern, 2004). For example, New Zealand legislation states that a water supply  
118 bore is considered secure (unlikely to have a risk of contamination by pathogenic  
119 organisms) when less than 0.005% of the water has been present in the aquifer for less  
120 than one year (New Zealand Ministry of Health, 2008).

121 For the Lake Rotorua catchment study, tritium is the tracer of choice. Tritium  
122 dating can be applied to both river/stream water and groundwater, whereas gas tracers  
123 are less suitable for surface waters that are in contact with air. Tritium ages, in  
124 contrast to gas tracer ages, include travel through the unsaturated zone (Zoellmann et  
125 al., 2001; Cook and Solomon, 1995); travel times can be > 40 years through the thick  
126 unsaturated zones of the Rotorua catchment ignimbrite aquifers (Morgenstern et al.,  
127 2004). Tritium is not subject to transformation, degradation or retardation during  
128 water transport through the catchment. Tritium dating is applicable to water with  
129 mean residence times of up to about 200 years (Cook and Solomon, 1997;  
130 Morgenstern and Daughney, 2012), as is typical of New Zealand's dynamic surface  
131 waters and shallow groundwaters. In addition, monitoring the movement of the pulse-  
132 shaped bomb-tritium through groundwater systems is an excellent opportunity to  
133 obtain information about the age distribution parameters of the groundwater. This is  
134 particularly useful in groundwater systems, such as the Rotorua system, that have high  
135 uncertainties within flow models due to a deep water table and preferential flow paths.  
136 Finally, tritium is a particularly sensitive marker for study of the timing of nitrate  
137 contamination in groundwater, because the main anthropogenic nitrate contamination  
138 of groundwater systems started coincidentally with the bomb-tritium peak from the  
139 atmospheric nuclear weapons testing after WWII; water recharged before this post-  
140 war upsurge in intensive agriculture has low tritium and low nitrate concentrations.

141 For the Rotorua catchment we have an extensive data set available over time  
142 and space. Tritium time series data for the main lake inflows cover more than four  
143 decades, and data covering the last decade are available with an extremely high spatial  
144 resolution of about 100 sites in the Lake Rotorua catchment. Tritium concentration  
145 can be measured at GNS Science with the required extremely high accuracy using 95-

146 fold electrolytic enrichment prior to ultra-low level liquid scintillation spectrometry  
147 (Morgenstern and Taylor, 2009). Tritium is highly applicable for groundwater dating  
148 in the post-bomb low-tritium environment of the Southern Hemisphere, as bomb  
149 tritium from atmospheric thermonuclear weapons testing has now been washed out  
150 from the atmosphere for 20 years, as is described in detail in Morgenstern and  
151 Daughney (2012).

152 The objective of this study is to understand the origin, fate, flow pathways, lag  
153 times, and future loads of contaminants that cause lake eutrophication in the Lake  
154 Rotorua catchment, central North Island, New Zealand. This will assist in mitigating  
155 the deterioration in lake water quality since the 1960s (Rutherford et al., 1989) that  
156 threatens the lake's significant cultural and tourist value. Environmental  
157 hydrochemistry tracers and age tracers are used to identify the recharge source of the  
158 main water discharges to Lake Rotorua, identify the source of the contaminants  
159 (anthropogenic versus geologic), and to evaluate the water age distributions in order  
160 to understand groundwater processes, lag times, and the groundwater flow dynamics.  
161 The Rotorua groundwater system is complicated due to the catchment's highly  
162 complex geology, which has evolved through volcanic activity, and due to the deep  
163 water table of >50 m in large areas, which prevents detailed groundwater studies and  
164 introduces uncertainty in catchment boundaries and flow patterns. The complex  
165 geology leads to inhomogeneous groundwater flow patterns, as indicated by large  
166 parts of the catchment having particularly large positive or negative specific water  
167 yields (White et al., 2004). The groundwater discharge in the northern catchment is  
168 unusually large for the size of the surface water catchment, probably due to  
169 preferential flow paths that route groundwater towards the north across the surface  
170 slope in this part of the catchment.

171 The tritium and age distribution data is currently being used to calibrate a  
172 numeric groundwater transport model. The use of this rich data set for groundwater  
173 transport model calibration is part of a larger investigation that evaluates the  
174 calibration of hydrological and hydrogeological models using hydrochemical data,  
175 including tracers of water age, with the aim of using tracer-calibrated groundwater  
176 models for nutrient transport and economic modelling (e.g., Lock and Kerr, 2008;  
177 Rutherford et al., 2009), ultimately supporting optimal and sustainable land and water  
178 management in catchments. The broader findings from the Rotorua investigation will  
179 be applied to the many other New Zealand catchments for which time series age tracer  
180 data are available (Stewart and Morgenstern, 2001; Morgenstern, 2004; Stewart and  
181 Thomas, 2008; Gusyev et al., 2014).

182

183

## 184 **2. Hydrogeological setting**

185

### 186 **2.1 Geology**

187

188 Lake Rotorua is located in a roughly circular caldera basin in the central North  
189 Island, New Zealand (Fig. 1), situated in the Taupo Volcanic Zone (TVZ), an area of  
190 silicic volcanism with NW-SE extension and geothermal activity roughly 60 km wide  
191 by 300 km long that is related to subduction of the Pacific plate beneath the  
192 Australian plate off New Zealand's east coast (Wilson et al., 1995; Spinks et al.,  
193 2005). Figure 1 shows the surficial geology. Mesozoic greywacke, which outcrops to  
194 the east and west of the TVZ, forms the basement rocks in the area. Younger  
195 formations are predominantly rhyolite ignimbrites, rhyolite and dacite lava domes,  
196 and lacustrine and alluvial sediments derived from these volcanic lithologies. Note  
197 that 'rhyolite' is sometimes used colloquially to refer to rhyolite lava, but the use of  
198 'rhyolite' is here applied only as a formal compositional definition (volcanic rocks  
199  $>69\%$  SiO<sub>2</sub>). Deposits of rhyolite composition may be pyroclastic (explosively  
200 formed, including airfall deposits and the pyroclastic flow deposit termed  
201 'ignimbrite') or they may be lavas (effusively erupted without explosion). Rhyolite  
202 lavas are viscous and often push up into high lava domes over the eruption vent. The  
203 geological formations and processes of greatest relevance to the hydrology and  
204 hydrogeology of the Lake Rotorua catchment area are shown in the three-dimensional  
205 geological model of White et al. (2004) (Fig. 2) and summarised in the following  
206 paragraphs.

207

208 Fig 1

209

210 From 2 million to 240 thousand years ago (ka), a number of rhyolite lava  
211 domes were emplaced and volcanic activity from TVZ calderas resulted in pyroclastic  
212 deposits across the area, including the highly welded Waiotapu ignimbrite (ca. 710 ka;  
213 'older ignimbrite' in Fig. 1) and a range of variably welded, variably altered,  
214 sometimes-jointed ignimbrites of which the Matahina, Chimp and Pokai formations  
215 (ca. 320-270 ka) are most significant and mapped within 'undifferentiated rhyolite  
216 pyroclastics' in Fig. 1. These ignimbrites are expected to be the main basal units for  
217 groundwater aquifers in the study area (White et al., 2004).

218 The period from 240 to 200 ka is defined by the eruption that deposited the  
219 Mamaku Plateau Formation (240 ka) and formed the Rotorua Caldera. The caldera  
220 collapse down-faulted parts of older lava domes positioned across the western and  
221 northern edge of the caldera. The Mamaku Plateau Formation is predominantly  
222 composed of ignimbrite (hereafter 'Mamaku ignimbrite'), which is variably welded,  
223 variably jointed and very permeable. Several rhyolite lava domes (mainly Ngongotaha  
224 and its neighbours) began to develop soon after the caldera collapse. Also, soon after  
225 the eruption, a lake began to form in the collapsed depression, leading to the  
226 deposition of lacustrine fine ash and pumice, commonly referred to as lacustrine  
227 sediments (Leonard et al., 2010); note that these sediments have sometimes been  
228 referred to as Huka or Huka Group sediments throughout the TVZ, but this definition  
229 formally refers only to specific units near Taupo City and the term Huka is avoided  
230 here.

231

232 Fig. 2

233  
234 From 200 to 61 ka, volcanic activity in the vicinity of Lake Rotorua was  
235 relatively subdued. A number of eruptions from the Okataina Volcanic Centre (OVC),  
236 located to the east of the Rotorua caldera, produced widely dispersed but relatively  
237 thin airfall deposits. These pyroclastic materials caused periodic damming of drainage  
238 pathways and led to fluctuations in lake level that in turn resulted in widespread and  
239 variably thick sediments being deposited in the Rotorua Caldera. This period of  
240 relatively quiet volcanic activity ended with the Rotoiti and Earthquake Flat eruptions  
241 from the OVC, which produced widespread pyroclastic deposits, including the non-  
242 welded ignimbrites of the Rotoiti Formation and the Earthquake Flat Formation.

243 From 61 ka to present, numerous eruptions from the OVC (the most recent of  
244 which was in 1886) deposited airfall layers in the Lake Rotorua catchment area.  
245 Numerous rhyolite lava units were also emplaced during this period. The periodic  
246 deposition of pyroclastic materials, along with activity on faults of the Taupo Rift  
247 (Leonard et al., 2010), presumably caused fluctuations in the lake level, with current  
248 lake level being reached sometime within the last few thousand years (White et al.,  
249 2004). Due to the decline of lake level, Holocene alluvial sand and gravel deposits are  
250 found in stream channels and around the current lake shoreline.

251 The southern Rotorua basin hosts a vigorous geothermal system producing  
252 many hot water, hot mud, steam and **geyser** features, along with gas emission,  
253 between the southern edge of the Lake and about the southern edge of the caldera  
254 (Fig. 1). There is hot local groundwater flow in this area, generally flowing down hill  
255 northwards into the lake. Beyond this relatively confined area the groundwater system  
256 does not appear to interact with fluids from this geothermal system.

## 257 258 **2.2 Hydrology**

259  
260 Lake Rotorua has a surface area of 79 km<sup>2</sup> and a mean depth of 10.8 m  
261 (Burger et al., 2011), with a total water volume of 0.85 km<sup>3</sup>. The assumed total  
262 catchment area is ca. 475 km<sup>2</sup> (White and Rutherford, 2009) (Fig. 1).

263 Annual rainfall in the catchment is strongly affected by topography and varies  
264 from more than 2200 mm northwest of the lake to less than 1400 mm southeast of the  
265 lake (Hoare, 1980; White et al., 2007; Rutherford et al., 2008). Approximately 50% of  
266 rainfall infiltrates into the groundwater system. This is based on two sources of  
267 information: 1) comparisons of rainfall and actual evapotranspiration that have been  
268 made for various parts of the catchment (Hoare, 1980; Dell, 1982; White et al., 2004;  
269 White et al., 2007; Rutherford et al., 2008); and 2) data from paired lysimeters, a  
270 standard rain gauge and a ground level rain gauge installed at Kaharoa (White et al.,  
271 2007) (Fig. 1). With 50% of rainfall recharge, total infiltration into the groundwater  
272 system is estimated to be 14,500 l/s, based on the catchment shown in Fig. 1,  
273 excluding rainfall inputs direct to the lake, and assuming recharge is 50% of rainfall.  
274 This rainfall recharge supports stream flow and potentially direct inputs of  
275 groundwater to the lake.

276 There are nine major streams (Fig. 1, for abbreviations refer to Table 1) and  
277 several minor streams that flow into the lake; the remainder of the inflows are  
278 provided from direct inputs of rainfall and lake-front features, and potentially from  
279 groundwater seepage through the lake bed. The major streams are baseflow-controlled  
280 and characterized by very constant water flow (Hoare, 1980) and temperature, and  
281 groundwater-derived baseflow accounts for approximately 90% of the average flow in  
282 the typical Rotorua stream (Hoare, 1987). Baseflows in the nine major streams  
283 entering Lake Rotorua cumulatively amount to 11,800 l/s, and total inflows to the lake  
284 from minor streams and lake-front features amounts to 350 l/s (Hoare, 1980; White et  
285 al., 2007) (Table 1).

286

287 Tab.1

288

289 With the lake water volume of  $0.85 \text{ km}^3$ , the lake water turnover time via the  
290 groundwater-fed streams is 2.2 years. The only surface outflow occurs through Ohau  
291 Channel via Lake Rotoiti (Fig. 1). Water balance calculations suggest that the total  
292 catchment area exceeds the surface water catchment area (White et al., 2007); in other  
293 words, groundwater from outside the surface water catchment is flowing through the  
294 aquifer system to Lake Rotorua (White and Rutherford, 2009).

295

## 296 **2.3 Hydrogeology**

297

298 Groundwater flow in the Lake Rotorua catchment is influenced by several  
299 fundamental geological characteristics. First, the Mamaku ignimbrite, the dominant  
300 hydrogeological feature in the catchment, is assumed to be up to 1 km thick in the  
301 centre of the caldera depression, and from about 200 to tens of metres thick outside of  
302 the caldera, decreasing generally with distance from the caldera (Fig. 2). Ignimbrite  
303 tends to fill in pre-existing valleys and landforms, so its thickness can be quite  
304 variable over horizontal distances of as little as hundreds of metres. Transit times of  
305 groundwater through such a thick aquifer may be lengthy compared to times in the  
306 shallow alluvial aquifers used for water supply in many other parts of the world and  
307 New Zealand. Second, the ignimbrites in the Lake Rotorua catchment are known to be  
308 variably welded, altered and jointed, with the potential for preferential groundwater  
309 flow **paths**. Groundwater may be routed from its recharge area along lengthy  
310 preferential flow paths and discharge in neighbouring surface catchments, leading to  
311 water ages that vary substantially even within a localised area (the presence of such  
312 preferential flow paths can therefore be demonstrated with age tracers). Third, the  
313 broadly circular collapse faults of the caldera constitute a major structural feature that  
314 may influence the flow of groundwater within the catchment.

315

316 The major water contribution to Lake Rotorua is from the western catchment  
317 that drains the eastern flanks of the Mamaku Plateau (Fig. 1). The Mamaku ignimbrite  
318 formation serves as a major source of groundwater in the area (Gordon, 2001). A  
319 large area ( $\sim 250 \text{ km}^2$ ) is drained by several major springs ( $>1,000 \text{ l/s}$ ) emerging from  
the ignimbrite on the western side of Lake Rotorua. Given the large extent and



320 thickness of the ignimbrite aquifer, a large groundwater reservoir exists, with long  
321 water residence times expected in the aquifer. Taylor and Stewart (1987) estimated  
322 the mean residence time of the water of some of the springs as 50-100 years.

323 The post-240 ka ignimbrites in this area (and some lava domes) are extremely  
324 porous; they sustain hardly any overland water flow (Dell, 1982), with most of the  
325 stream beds dry throughout most of the year except during heavy rain, and they allow  
326 the infiltrated water to percolate down to an extremely deep groundwater table >50 m  
327 (the ignimbrite formations around Ngongotaha Dome are an exception).

328 Little is known about the hydrogeology of the groundwater system; borehole  
329 data collected by drillers is often not of sufficient quality to identify and correlate  
330 aquifer units. Rosen et al. (1998) developed a schematic model for the Mamaku  
331 ignimbrite, with a lower and upper ignimbrite aquifer sheet considered permeable,  
332 and a middle sheet considered impermeable but fractured and not acting as an  
333 aquitard for the other two sheets. This is probably an over-simplification in many  
334 areas (see Milner et al., 2003) but does point at horizontally-planar discontinuities  
335 within the formation that appear to influence groundwater flow. The water, after easy  
336 passage through the large aquifer, is forced to the surface only 1–2 km before the lake  
337 shore via large groundwater springs, feeding large streams that drain into Lake  
338 Rotorua.

339 The water-bearing lava dome formations that predate the Mamaku ignimbrite  
340 are likely to have fracture flow, based on spring discharge permeability analysis,  
341 varying depending on fracture sizes and linkages. Faulting associated with the  
342 Rotorua caldera has offset several of the rhyolite domes and groundwater may flow  
343 through these faults.

344 The paleo-lake sediments that post-date the Mamaku ignimbrite and rhyolite  
345 lava formations comprise silt, sand, and gravel (ignimbrite, obsidian and rhyolite  
346 pumice) and are considered permeable, with lenses of low permeability that can also  
347 act as confining layers.

348 Overall, understanding of the Rotorua groundwater system is complicated due  
349 to the highly complex geology that has evolved through volcanic activity. Vertical  
350 and steeply-inclined geological contacts are common, precluding a simple horizontal-  
351 layer-based succession model throughout the catchment **usually applicable in**  
352 **sedimentary basins**. Aquifers have not been well determined due to insufficient bore  
353 log data, and also the extent of the Lake Rotorua groundwater catchment is difficult to  
354 establish, due to the deep water table of > 50 m in large areas at the catchment  
355 boundaries, combined with an inhomogeneous groundwater flow pattern, as indicated  
356 by the groundwater discharge in the northern catchment being too large compared to  
357 the size of the surface water catchment.

358

### 359 **3. Methods**

360

#### 361 **3.1 Determination of water age**

362

363 The age of the groundwater at the discharge point characterises the transit time  
 364 of the water through a groundwater system. For groundwater dating, we use tritium  
 365 time series (repeated sampling after several years), and the complementary tracers  
 366 tritium, CFCs, and SF<sub>6</sub> together where possible. The method of dating young  
 367 groundwater with mean ages of less than 200 years for current New Zealand Southern  
 368 Hemispheric conditions is described in detail in Morgenstern and Daughney (2012,  
 369 sections 2.3 and 2.4). In short: tritium dating, in previous decades problematic due to  
 370 interference from the artificial tritium produced by atmospheric nuclear weapons  
 371 testing in the early sixties, has now become very efficient and accurate due to the  
 372 fading of the bomb-tritium.

373 For groundwater dating, one or more tracer substances are measured that have  
 374 a time-dependent input into the groundwater system or a well-defined decay-term  
 375 (e.g., radioactive decay). The tracer concentration data is then fitted using a lumped-  
 376 parameter model (Maloszewski and Zuber, 1982; Zuber et al., 2005). For dating  
 377 young groundwater—i.e., water less than about 100 years, the most commonly used  
 378 tracers are tritium, chlorofluorocarbons (CFCs) and sulphur hexafluoride (SF<sub>6</sub>) (Cook  
 379 and Solomon, 1997; Edmunds and Smedley, 2000; Stewart and Morgenstern, 2001;  
 380 Morgenstern et al., 2010). The measured output tracer concentration in the  
 381 groundwater ( $C_{out}$ ) is then compared to its tracer concentration at the time of rainfall  
 382 input ( $C_{in}$ ) using the convolution integral:

383  
 384

$$385 \quad C_{out}(t) = \int_0^{\infty} C_{in}(t - \tau) e^{-\lambda\tau} g(\tau) d\tau \quad (1)$$

386 where  $t$  is the time of observation,  $\tau$  is the transit time (age),  $e^{-\lambda\tau}$  is the decay term with  
 387  $\lambda = \ln(2)/T_{1/2}$  (e.g., radioactive decay of tritium with a half-life  $T_{1/2}$  of 12.32 years) and  
 388  $g(\tau)$  is the system response function (Cook and Herczeg, 1999; Zuber et al., 2005).

389 The system response function accounts for the distribution of ages within the  
 390 water sample, for example from mixing of groundwater of different ages within the  
 391 aquifer, or at the well (Maloszewski and Zuber, 1982, 1991; Goode, 1996; Weissman  
 392 et al., 2002; Zuber et al., 2005). The two response functions most  
 393 commonly used are the exponential piston flow model and the dispersion model  
 394 (Zuber et al., 2005). The exponential piston flow model combines the piston flow  
 395 model, assuming piston flow within a single flow tube in which there is minimal  
 396 mixing of water from different flow lines at the discharge point (e.g., confined  
 397 aquifer), and the exponential model, assuming full mixing of water from different  
 398 flow paths with transit times at the groundwater discharge point that are exponentially  
 399 distributed (e.g., mixing of stratified groundwater at an open well in an unconfined  
 400 aquifer). The response functions of the various models are described in Maloszewski  
 401 and Zuber (1982) and Cook and Herczeg (1999). To interpret the ages of the Lake  
 402 Rotorua catchment data set, the exponential piston flow model was used, given by:

403

$$404 \quad g = 0 \quad \text{for } \tau < T(1-f) \quad (2)$$

405 
$$g = \frac{1}{Tf} e^{(-\frac{\tau}{Tf} + \frac{1}{f} - 1)} \quad \text{for } \tau \geq T(1-f) \quad (3)$$

406  
 407 where T is the Mean Residence Time (MRT), f is the ratio of the volume of  
 408 exponential flow to the total flow volume at the groundwater discharge point, and T(1  
 409 - f) is the time that takes the water to flow through the piston flow section of the  
 410 aquifer (N.B. Maloszewski and Zuber (1982) use the variable  $\eta$ ;  $\eta = 1/f$ ). When f = 0  
 411 the model becomes equivalent to the piston flow model, and when f = 1 it becomes  
 412 equivalent to the exponential model.

413 The two parameters of the response functions, the mean residence time (MRT)  
 414 and the distribution of transit times (f), are determined by convoluting the input  
 415 (tritium concentration in rainfall measured over time) to model water passage through  
 416 the hydrological system in a way that matches the output (e.g., tritium concentrations  
 417 measured in wells or springs). Because of its pulse-shaped input, tritium is a  
 418 particularly sensitive tracer for identifying both of these two parameters, which can be  
 419 deduced uniquely by comparing the delay and the dispersion of the bomb-pulse  
 420 tritium in the groundwater to that from tritium in the original rain input. This method  
 421 is particularly useful for interpretation of ages of groundwater in the Lake Rotorua  
 422 catchment, where most of the groundwater discharges lack any other information on  
 423 mixing of groundwater with varying flow path lengths and of different age, such as  
 424 ratio of confined to unconfined flow volume, or screen depth for wells.

425 For tracer age interpretation, the integral (Eq. 1) was used to convolute the  
 426 historical rainfall tracer input to an output that reflects mixing in a groundwater  
 427 system, with the best match of the simulated output to the measured output time-series  
 428 data (Fig. 3). The TracerLPM workbook (Jurgens et al. 2012) was used. The tritium  
 429 input function is based on concentrations of tritium in rainfall measured monthly since  
 430 the 1960s at Kaitoke, near Wellington, New Zealand (Morgenstern and Taylor, 2009).  
 431 The Kaitoke rainfall input function is multiplied by a scaling factor of 0.87 to account  
 432 for variation in atmospheric tritium concentrations due to latitude and orographic  
 433 factors, as deduced from measurements from rain at various locations in New Zealand  
 434 (e.g. Morgenstern et al., 2010). For the prevailing New Zealand climatic conditions  
 435 there is no need for correction of the tritium input for seasonal infiltration  
 436 (Morgenstern et al., 2010).

437  
 438 Fig. 3

439  
 440 The problem of ambiguity in tritium dating over the last decades is  
 441 demonstrated in Fig. 3. Hangarua Spring discharges old water with a mean residence  
 442 time of about 90 years (see below), but during the late 1980s its tritium concentration  
 443 was similar to that of very young water (rain curve in Fig. 3). At that time, the tritium  
 444 concentration in Hangarua Spring would have been in agreement with both very  
 445 young water and old water with a mean residence time of 90 years. Tritium data  
 446 covering several decades, however, clearly distinguish this old water (low tritium

447 concentration) from young rain water. Figure 3 also shows that due to the fading of  
448 the bomb-tritium in recent decades (tritium decay over four tritium half-lives since the  
449 bomb spike), in recent years the tritium concentration of old water is clearly  
450 distinguishable (lower) from that of young water, without ambiguity. The tritium time  
451 series data allow also for constraining groundwater mixing models. Figure 3 shows  
452 the model output curves that match the measured tritium data. Given sufficient  
453 analytical accuracy, this is also possible for extremely low tritium concentrations; the  
454 data for Hamurana water intake spring (blue in Fig. 3) are all below 0.4 TU, which is  
455 below the detection limit of many tritium laboratories ([http://www-](http://www-naweb.iaea.org/napc/ih/IHS_programme_ihl_tric.html)  
456 [naweb.iaea.org/napc/ih/IHS\\_programme\\_ihl\\_tric.html](http://www-naweb.iaea.org/napc/ih/IHS_programme_ihl_tric.html)).

457 The application of mixing models is described in Morgenstern and Daughney  
458 (2012, section 2.6). Throughout New Zealand, for springs and wells in almost all  
459 hydrogeological situations, the exponential piston flow model, with its age  
460 distribution, has produced good matches to most (about a hundred) tritium time series  
461 data. It was not, however, possible to obtain adequate matches in the ignimbrite area  
462 of the Rotorua catchment using such a simple exponential piston flow model.  
463 Alternatively, using the dispersion model did not improve the matches. The complex  
464 volcanic aquifers of the Lake Rotorua catchment, which have evolved through  
465 volcanic activity, require a more complex system response function. A combination of  
466 two exponential piston flow models was used.

467  
468

### 469 **3.2 Sample collection and analysis**

470

471 Samples were collected from 41 springs, from 31 groundwater-dominated  
472 stream flow sites, and from 26 groundwater wells. To obtain the residence times of  
473 the water discharging into the lake after passage through the entire groundwater  
474 system, sampling focused on the naturally flowing groundwater discharges, the  
475 springs and streams. Samples were collected at times of base flow conditions.

476 All nine major streams were sampled multiple times near the inflow into the  
477 lake, typically 3-4 times (Figs. 3 and 4). Most of these tritium time series go back to  
478 the early seventies and encompass the passage of the ‘bomb’ tritium peak through the  
479 groundwater system, allowing determination of detailed age distribution parameters  
480 for these major inflows to the lake. These ‘historic’ samples had been collected  
481 sporadically for various projects over the decades to study the transfer of the bomb-  
482 tritium through the hydrologic cycle. Over the recent decade, the streams have also  
483 been sampled for tritium at various points upstream, at various main confluences, or  
484 at main springs to obtain a detailed spatial distribution of water ages. Springs and  
485 wells were also once sampled for CFCs, SF<sub>6</sub>, argon, and nitrogen, to obtain  
486 complementary age information.

487 Sampling locations are shown in Fig. 1. Many of the sites have no road access,  
488 with some of them in remote steep gullies. A portable sampling system was required  
489 for the gas samples to allow fresh water from the well or spring to be pumped into the

490 sample bottles from below the water surface without air contact. We used a pneumatic  
491 Bennet pump, powered from a cylinder of compressed air at the remote locations, and  
492 from a compressor powered by the car battery at sites with car access. Sampling from  
493 streams (tritium only) involved simply dipping the bottle under the water surface and  
494 filling the bottle.

495 Sampling methods for hydrochemistry and nutrients were according to  
496 Daughney et al. (2007). Age tracer samples were collected without filtration or  
497 preservation. For tritium, a one-litre plastic bottle was filled to the top. For CFC  
498 samples, two 125 mL glass bottles with aluminium liner cap were filled, rigorously  
499 excluding air contact by filling from the bottom via a nylon tube and three times  
500 volume replacement below the surface of the overflowing sample water. One litre  
501 bottles were filled for SF<sub>6</sub>.

502 Analytical details for hydrochemistry are described in Daugney et al. (under  
503 review). Details of the tritium analysis procedure are described in Morgenstern and  
504 Taylor (2009). While the early tritium measurements in the 1970s were performed  
505 with a detection limit of approximately 0.1 Tritium Units (TU), we now achieve  
506 significantly lower detection limit of 0.02 (TU) via tritium enrichment by a factor of  
507 95 and reproducibility of tritium enrichment of 1% via deuterium-calibration.  
508 Analysis procedures for CFC-11, CFC-12, and SF<sub>6</sub> are described in van der Raaij and  
509 Beyer (under review). Detection limits are  $3 \times 10^{-15}$  mol kg<sup>-1</sup> for CFCs and  $2 \times 10^{-17}$  mol  
510 kg<sup>-1</sup> for SF<sub>6</sub>. Dissolved argon and nitrogen concentrations were measured for  
511 estimating the temperature at the time of recharge, and the excess air concentration, as  
512 described by Heaton and Vogel (1981), for calculation of the atmospheric partial  
513 pressure (ppt) of CFCs and SF<sub>6</sub> at the time of recharge.

514

## 515 **4 Results and discussion**

516

517 In the following section hydrochemistry cluster analysis and hydrochemistry  
518 evolution are discussed to assess the geographic sources of groundwater and  
519 groundwater processes in the aquifer. The nutrients nitrate, sulphate, potassium, and  
520 phosphate are discussed to evaluate their source (anthropogenic versus geologic), lag  
521 time, fate, and impact on lake eutrophication. The age distributions of the  
522 groundwater discharges to Lake Rotorua are discussed to understand the conceptual  
523 groundwater flow pattern and the lag time in the groundwater system. The ultimate  
524 goal of this project is the use of the hydrochemistry and groundwater age parameters  
525 for calibration of a groundwater transport model for improved management of the  
526 nutrient loads to the lake – the subject of follow up papers.

527

### 528 **4.1 Groundwater age interpretation**

529

530 To obtain the unique solution for both parameters of the age distribution for a specific  
531 model, time series data are required (Section 3.1). Most of the large water inflows into

532 Lake Rotorua have long time series data available (up to over four decades), allowing  
533 for well constrained age distribution parameters for both MRT and the fraction  $f$   
534 between different flow models (or pecelet number for the dispersion model). The  
535 tritium time series data, together with the matching lumped parameter model  
536 simulations, are shown in Fig. 3 for two of the large springs, and in Fig. 4 for six of  
537 the large streams (covering 2/3 of stream baseflow to Lake Rotorua). For the sites  
538 with shorter time series data (sub-catchment stream discharges, groundwater wells),  
539 most of the sites have at least sufficient time series or multi tracer data for  
540 unambiguous robust age interpretations. If fraction  $f$  cannot be established uniquely  
541 from the tritium time series data, we applied mixing models that matched long tritium  
542 time series data from other sites with similar hydrogeologic settings to these sites. All  
543 96 sites with tritium time series or tritium and complementary CFC and SF<sub>6</sub> data have  
544 unambiguous age interpretation. For the tritium time series data shown in Figs. 3 and  
545 4, the lumped parameter models, with their respective age distribution parameters that  
546 match the measured data, are listed in Table 2.

547

548 Fig. 4

549 Tab. 2

550

551 Throughout New Zealand, and including all hydrogeologic situations (but  
552 mainly groundwater wells), we have measured approximately a hundred long tritium  
553 time series covering several decades. A simple lumped parameter model, the  
554 exponential piston flow model, usually can match these time series data well (e.g.,  
555 Morgenstern and Daughney, 2012). The long-term tritium data from most of the large  
556 stream discharges shown in Fig. 4, however, cannot be matched by a simple model  
557 such as the exponential piston flow or dispersion model and require a more complex  
558 groundwater flow model combination. Using a binary mixing model, with parallel  
559 contributions from two exponential piston flow models, resulted in excellent matches.  
560 We justify this binary mixing model by inferring two different flow contributions in  
561 the catchment to stream and spring flow—from deep old groundwater, as indicated by  
562 very deep groundwater tables in the area (generally > 50 m), and from younger  
563 groundwater from shallow aquifers, as indicated by minor stream flows maintained by  
564 shallow aquifers. In Table 2 are also listed the average mean residence times between  
565 the two parallel models, weighted by their fraction within the total flow. For the  
566 MRTs, errors caused by our tritium measurement error and uncertainty in tritium  
567 input are typically  $\pm 1$  y for MRTs < 5 y,  $\pm 2$  y for MRTs between 5 and 10 y,  $\pm 3$  y  
568 for MRTs between 10 and 50 y,  $\pm 5$  y for MRTs between 50 and 100 y, and larger  
569 errors for older water towards the detection limit.

570 For convenience, the average MRTs are also listed in Fig. 4 next to their  
571 model output curve. It is obvious from Fig. 4 that all the main streams discharge very  
572 old water into Lake Rotorua. The tritium response of the streams is clearly  
573 distinguishable from that of young rain water (grey line). The youngest water  
574 discharge, Ngongotaha Stream (green), has an average MRT of 30 years. All other  
575 main streams discharge significantly older water, up to MRT of 145 years for

576 Waingaehe Stream (dark blue). Note that even though bomb-tritium from atmospheric  
577 nuclear weapons testing in the 1960s has decayed enough to not cause ambiguous age  
578 interpretations anymore, it is still possible to detect the tail of the bomb-tritium for  
579 matching model parameters if tritium analyses have sufficient accuracy; all data for  
580 the Hamurana water intake spring (blue in Fig. 3) are below 0.4 TU.

581 Most of the streams and springs discharge very old groundwater into Lake  
582 Rotorua, with MRTs typically between 50–150 years, indicating discharge from a  
583 large groundwater system with large water residence (turn-over) times. Only a few  
584 small sub-catchments with minor flow rates discharge young water (MRT < 20 years),  
585 indicating local geologic units below the surface that don't allow water to infiltrate  
586 into and flow through larger, deeper groundwater systems.

587 Substantial fractions of that long residence time in the groundwater system  
588 may occur during passage through the thick unsaturated zones (50 – 100 m) as  
589 indicated by CFC and SF<sub>6</sub> results measured at groundwater wells and springs  
590 (Morgenstern et al., 2004). CFCs and SF<sub>6</sub> in groundwater are still exchanged with the  
591 atmosphere during passage through the unsaturated zone, therefore CFC and SF<sub>6</sub> ages  
592 represent travel time through the saturated zone only. Large observed differences  
593 between CFC and SF<sub>6</sub> ages, compared to tritium ages of up to 40 years and greater for  
594 the older waters, therefore indicate travel time of the groundwater through the  
595 unsaturated zone of >40 years for the older groundwater discharges.

596 The old age of the majority of the Lake Rotorua water inflows and the highly  
597 mixed nature of the water discharges (note the high fractions of exponential flow, up  
598 to 100%, in Table 2) implies a very slow and lagged response of the streams and the  
599 lake to anthropogenic contaminants in the catchment, such as nitrate. The majority of  
600 the nitrate load currently discharging into the lake is thus from land-use activities 30  
601 and more years ago.

602 About a hundred stream and groundwater well samples have been dated in the Lake  
603 Rotorua catchment. The groundwater age distributions are used in the following  
604 sections to identify hydrochemistry evolution, sources of contaminants, and to predict  
605 future nitrate loads that will enter Lake Rotorua from the large contaminated  
606 groundwater system. In a future paper, the conceptual groundwater flow model in the  
607 Lake Rotorua catchment will be inferred from the groundwater age distribution data.  
608 The data will subsequently be used for calibration of a groundwater transport model.

609

610

## 611 **4.2 Hydrochemistry and recharge source**

612

613 The hydrochemical composition of the groundwater and surface waters in the  
614 Rotorua catchment have been investigated by Morgenstern et al. (2004) and Donath et  
615 al. (under review); the following section summarises the results relevant to the present  
616 study.

617 Hydrochemistry is driven by interaction between water and the different major  
618 lithologies and can be used to track the origin of the groundwater. Hydrochemistry  
619 reflects the rhyolite ignimbrite and lava aquifer lithologies that dominate the Lake  
620 Rotorua catchment, with much lower concentrations of Ca, Mg and SO<sub>4</sub> and much  
621 higher concentrations of F, PO<sub>4</sub> and SiO<sub>2</sub>, compared to groundwater in other parts of  
622 New Zealand.

623 Several statistical and graphical techniques were applied to characterise the  
624 variations in hydrochemistry across parts of the catchment. Hierarchical Cluster  
625 Analysis (HCA) was shown to be a useful technique to identify samples with similar  
626 hydrochemical composition, and to relate the groundwater samples to their origin  
627 from one of the main aquifer lithologic units. HCA conducted with Ward's linkage  
628 rule allowed the samples to be partitioned into four hydrochemical clusters. Three of  
629 the clusters, accounting for the majority of the samples, were inferred to reflect water-  
630 rock interaction with the dominant lithologies in the catchment, namely Mamaku  
631 ignimbrite, lava, or paleo-lake sediments. Hydrochemistry inferred to indicate  
632 interaction between water and the Mamaku ignimbrite had Na and HCO<sub>3</sub> as the  
633 dominant cation and anion, respectively, and among the highest concentrations of Mg,  
634 PO<sub>4</sub> and SiO<sub>2</sub> and among the lowest concentrations of F, K and SO<sub>4</sub> observed.  
635 Hydrochemistry inferred to indicate interaction between water and rhyolite lava also  
636 had Na and HCO<sub>3</sub> as the dominant cation and anion, respectively, but had relatively  
637 low concentrations of PO<sub>4</sub> and among the highest concentrations of K.  
638 Hydrochemistry inferred to indicate interaction between water and sediments had Na-  
639 Ca-HCO<sub>3</sub>-Cl water type and relatively low concentrations of SiO<sub>2</sub>. The remaining  
640 cluster was inferred to represent geothermal influences on the hydrochemistry (e.g.,  
641 elevated concentrations of Na, Cl, SO<sub>4</sub>, SiO<sub>2</sub> and NH<sub>4</sub>).

642 Figure 5 shows the hydrochemical clusters of the water samples inferred to  
643 indicate interaction between water and Mamaku ignimbrite (light blue), lava (red), or  
644 lacustrine sediments (dark yellow). Note that samples assigned to the cluster inferred  
645 to indicate geothermal are not displayed in Fig. 5 or discussed further in the present  
646 study because geothermal influence is not the subject of this study.

647 Samples with hydrochemistry indicative of interaction with the Mamaku  
648 ignimbrite occur predominantly in the north and northwest portion of the catchment  
649 (Fig. 5, blue). All of the large springs discharging into Hamurana (Ham), Awahou  
650 (Awh), and Utuhina (Utu) streams have a Mamaku ignimbrite hydrochemistry  
651 signature (blue circles). The stream reaches in the Mamaku ignimbrite area upstream  
652 from these large springs are usually dry. This, together with the Mamaku ignimbrite  
653 hydrochemistry signature, implies that these large springs drain the large Mamaku  
654 ignimbrite areas upstream that have negligible surface runoff. In the north-west in the  
655 Hamurana and Awahou catchments, these springs emerge close to the lake shore  
656 within the sediment area (Fig. 5, yellow), indicating that close to the lake shore, due  
657 to the more impermeable nature of the intra-caldera sediment, the deeper groundwater  
658 flow from the Mamaku ignimbrite is forced to the surface. All of these large springs  
659 emerge within the slopes of the sediment formation where sediment layers are thinner  
660 and weaker compared to the level area closer to the present lake shore; we infer that



661 the thin nature of the sediments on these slopes allows the water from the underlying  
662 ignimbrite to flow to the surface. No large spring occurs in the level area closer to the  
663 lake, where sediments are thicker. Also the large Utuhina Spring in the southeast  
664 emerges within the slopes of the sediment (Fig. 5), indicating the more impermeable  
665 nature of the sediment forcing the groundwater from the Mamaku ignimbrite to the  
666 surface in the area of thin sediment layers. The Utuhina Spring emerges below a small  
667 local lava dome feature, but the ignimbrite signature of the water indicates that this  
668 spring is the discharge from the large ignimbrite area southwest of the lava dome  
669 feature. **The small lava feature may be fractured, discontinuous, or act as a water**  
670 **conduit, allowing water discharge from the ignimbrite behind.**

671 Shallow wells and streams that gain most of their recharge and flow within the  
672 lacustrine sediments display a characteristic hydrochemical signature (Fig. 5, yellow  
673 circles). Such samples originate from the downstream parts of Waiteti (Wtt) and  
674 Ngongotaha (Ngo) Streams. The study by Donath et al. (2014) also detected this  
675 characteristic hydrochemistry in samples collected with higher spatial resolution in  
676 parts of the Ngongotaha subcatchment that are not discussed here.

677 Hydrochemistry of the water draining the Ngongotaha lava dome west of the  
678 lake (Fig. 5, red circles) is inferred to indicate interaction with lava formations. The  
679 Ngongotaha dome, similar to the Mamaku ignimbrite, has no drainage via surface  
680 flow (stream beds are dry), indicating a highly porous nature, likely due to fractures  
681 and pumiceous zones within the dome. Only where the rhyolite dome intercepts the  
682 paleo-lake sediments is the groundwater flow from the lava forced to the surface due  
683 to the low permeability of the sediments.

684 The investigated water discharges from the eastern catchment of Lake Rotorua  
685 entirely show geothermal influence in their hydrochemistry composition (not the  
686 subject of this study).

687

688 Fig. 5

689

690 The above HCA results give, for the first time, consistent evidence of the link  
691 between the main recharge areas in the Mamaku ignimbrite, and the main  
692 groundwater discharges into the lake.

693

### 694 **4.3 Hydrochemistry evolution**

695

696 In the following two sections hydrochemistry data versus groundwater age is  
697 discussed for a better understanding of groundwater processes and geologic versus  
698 anthropogenic origin of contaminants.

699 The groundwater of the Rotorua rhyolite ignimbrite and lava dome aquifers  
700 (Fig. 6a) displays high dissolved oxygen (DO), between 5 and 11 mg/L (50-100% of  
701 equilibrium with air). There is no trend of decreasing DO with increasing age,  
702 indicating **microbial reduction reactions are insignificant** in this volcanic aquifer  
703 within time scales of the water residence time in the aquifer. Microbial reduction

704 reactions, facilitated by the presence of organic matter or other electron donors (e.g.,  
705 pyrite), would usually consume the dissolved oxygen in the groundwater. Reduction  
706 of oxygen is energetically the most favourable reaction that micro-organisms use in a  
707 series of reactions, with the result that other reduction reactions (e.g., denitrification)  
708 typically do not occur until most of the dissolved oxygen has been consumed. These  
709 reduction reactions take time, and if these reactions are supported by the presence of  
710 electron donors in the geologic formation, it is expected that old waters become  
711 increasingly anoxic (e.g., Böhlke et al., 2002, Tesoriero et al., 2011). No trend of  
712 decreasing DO with increasing groundwater age was observed, suggesting absence of  
713 significant amounts of electron donors such as organic matter or pyrite in this  
714 ignimbrite formation. This is supported by its depositional history as a single large  
715 ignimbrite formation without any organic matter involved. Absence of oxygen  
716 reduction indicates that there is no potential for significant denitrification reactions in  
717 this aquifer system.

718 Absence of a trend of decreasing DO with increasing groundwater age, but  
719 rather constant DO in very young and old groundwater of between 50 and 100%  
720 suggests that the partial oxygen reduction that is observed has occurred in the soil  
721 zone, which does contain organic matter, and that once the water has passed the soil  
722 zone, no further oxygen reduction processes occur. Only one groundwater sample in  
723 the Rotorua catchment was depleted in oxygen (below 1 mg/L). This is related to the  
724 paleo-lake sediments, suggesting localised deposits of reactive organic matter in these  
725 sediments, as would be expected in lake sediments.

726

727

Fig. 6

728

729 The pH of groundwater usually increases over time due to ongoing  
730 hydrogeochemical reactions, resulting in an increasing pH of groundwater with age.  
731 In New Zealand we observed in groundwater an increase from about pH 6 for very  
732 young groundwater (<1 year) to about pH 8 in very old groundwater (> 10,000 years)  
733 (Morgenstern and Daughney, 2012). For the Rotorua catchment, the groundwater pH  
734 data from lava formations (Fig. 6b) show a sharp increase from pH 5.8 to 6.7 over the  
735 age range from 5 to 50 years, with a power law fit of  $\ln(\text{pH}) = 0.073 * \ln(\text{MRT}) +$   
736  $1.62$ ,  $R^2 = 0.98$ . For the groundwater from the Mamaku ignimbrite, the pH increases  
737 from just 6.3 to 6.6 over the age range from 0 to 100 years, with a power law fit of  
738  $\ln(\text{pH}) = 0.025 * \ln(\text{MRT}) + 1.76$ ,  $R^2 = 0.47$ . The groundwater from the sediment  
739 formation show no clear trend in pH with groundwater age, but displays higher pH for  
740 relatively young water, pH 6.5–7.2 for water with MRT between 1 and 60 years.

741 As groundwater becomes more evolved over time due to water-rock  
742 interaction, concentrations of phosphorus, silica, bicarbonate, and fluoride typically  
743 increase due to dissolution of volcanic glass, silicate minerals, carbonates, and  
744 fluoride likely deposited from the volatile phases in magma exsolved during eruption  
745 (Morgenstern and Daughney, 2012). With increasing groundwater age, ion  
746 concentrations are expected to increase up to a maximum equilibrium concentration.  
747 In the following discussion, samples indicative of geothermal influence are excluded

748 ~~because they follow different thermodynamic equilibrium.~~ Groundwater from the  
749 sediment formation often follows different or unclear trends compared to the rhyolite  
750 ignimbrite and lava formations.

751 Dissolved reactive phosphate ( $\text{PO}_4\text{-P}$ ) in groundwater from all three  
752 formations, the rhyolite lava and ignimbrite aquifers, and the sediments originating  
753 from the same formations, shows excellent correlation with groundwater age (Fig. 7a,  
754 black curve), with  $\ln(\text{PO}_4\text{-P}) = 0.458 * \ln(\text{MRT}) - 4.72$ ,  $R^2 = 0.94$ .

755 Silica ( $\text{SiO}_2$ ) also shows good correlation with groundwater age for the  
756 rhyolite ignimbrite and lava formations (Fig. 7b). The silica concentration of  
757 groundwater in lava formations (red circles) increases faster compared to ignimbrite  
758 (blue circles). The power fit to the lava data is  $\ln(\text{SiO}_2) = 0.310 * \ln(\text{MRT}) + 2.96$ ,  $R^2$   
759  $= 0.88$  (red curve), and to the ignimbrite data is  $\ln(\text{SiO}_2) = 0.238 * \ln(\text{MRT}) + 3.05$ ,  
760  $R^2 = 0.83$  (blue curve). The correlation between silica and groundwater age for the  
761 lacustrine sediment aquifers (yellow circles) is rather erratic; high silica concentration  
762 can also occur in very young groundwater.

763

764 Fig. 7

765

766 For bicarbonate ( $\text{HCO}_3$ ), only groundwater samples from the Mamaku  
767 ignimbrite show a reasonable correlation with groundwater age, with a power fit of  
768  $\ln(\text{HCO}_3) = 0.206 * \ln(\text{MRT}) + 2.58$ ,  $R^2 = 0.71$  (Fig. 7c). The high data point at 58  
769 mg/L was considered an outlier and not included in the fit.

770 Sodium (Na) in general also shows increasing concentration with groundwater  
771 age because it is part of common minerals and leached from these. But the correlation  
772 is poor (Fig. 7d). Note that elevated Na in young groundwater can also be caused by  
773 land-use impacts, ~~as observed in other parts of New Zealand~~ (Morgenstern and  
774 Daughney, 2012). Considering the high data point at 15.8 mg/L an outlier (it is from  
775 the same site considered as an outlier for  $\text{HCO}_3$ ), the correlation for the data from all  
776 three formations is  $\ln(\text{Na}) = 0.07 * \ln(\text{MRT}) + 1.9$ ,  $R^2 = 0.27$ .

777 Fluoride concentrations (F) show good correlations with age for the rhyolite  
778 ignimbrite and lava formations (Fig. 7e), even though the trends are masked by the  
779 fact that the concentrations are close to the detection limit. Concentrations increase in  
780 lava formations significantly faster, with a power fit of  $\ln(\text{F}) = 1.087 * \ln(\text{MRT}) -$   
781  $6.44$ ,  $R^2 = 0.97$ , compared to ignimbrite with  $\ln(\text{F}) = 0.238 * \ln(\text{MRT}) - 3.99$ ,  $R^2 =$   
782  $0.58$ .

783 In groundwater of the Rotorua catchment (excluding groundwater from the  
784 eastern catchment indicating geothermal influence, which is not the subject of this  
785 study), the hydrochemistry parameters phosphate, silica, bicarbonate, sodium, and  
786 fluoride are purely of geologic origin, because they do not display elevated  
787 concentrations in young water that was recharged during the time of anthropogenic  
788 high intensity land-use activities. The groundwater samples show, for the rhyolite  
789 Mamaku ignimbrite and lava formations, excellent correlations across the western and  
790 northern Lake Rotorua catchment. ~~The~~ samples in each of these geologic units follow  
791 similar trends of hydrochemistry concentration versus mean residence time, indicating

792 the relatively homogeneous nature of these aquifers. Rather erratic trends in water  
793 originating from the sediments suggest that these are not a homogeneous formation  
794 but rather finely layered lensoidal geologic deposits that vary spatially and support  
795 complex or fragmented groundwater systems. Good trends of hydrochemistry versus  
796 groundwater age **may be an indirect indication** of robust age interpretations.

797 **In rhyolite lava formations, geochemical reactions lead to increased pH, Si,**  
798 **and F in groundwater significantly faster than in ignimbrite, indicating higher reaction**  
799 **rates for dissolution of these elements from lava formations. While this is important**  
800 **for understanding water-rock interaction, we do not yet have sufficient information on**  
801 **the litho-geochemistry to develop a mechanistic understanding of the reaction**  
802 **processes.**

803

#### 804 **4.4 Nutrients**

805

806 Elevated nutrient levels in surface water cause poisonous algal blooms and  
807 lake eutrophication. Presence of both phosphate and nitrate, above a threshold  
808 concentration, triggers algae blooms in lakes. Limitation of one of these, P or N, can  
809 limit algae blooms. In New Zealand, increasing nutrient loads from high intensity  
810 animal farming and fertilisers have triggered lake eutrophication. In the absence of  
811 significant overland runoff, nitrate travels from the land to the lake via the  
812 groundwater, which eventually discharges into streams and lakes.

813 Nutrient concentrations in New Zealand groundwaters from agricultural  
814 sources have increased steadily after European settlement in the early 19<sup>th</sup> century and  
815 with development of the meat industry after 1880 (Morgenstern and Daughney,  
816 2012). In a national context, for groundwater recharged before 1880 at pre-  
817 anthropogenic pristine conditions, low nutrient concentrations prevailed (e.g., nitrate  
818 < 0.2 mg/L NO<sub>3</sub>-N). In groundwater recharged between 1880 and 1955, nutrient  
819 concentrations are slightly elevated due to low intensity land use. In groundwater  
820 recharged after 1955 a sharp increase of nutrient concentrations is observed due to the  
821 impact of high intensity land use after World War II (Morgenstern and Daughney,  
822 2012).

823 The main nutrients derived from land use in the Rotorua catchment, as  
824 indicated by elevated concentrations in young groundwater, are nitrate (NO<sub>3</sub>),  
825 sulphate (SO<sub>4</sub>), and potassium (K). These nutrient concentrations are shown in Fig. 8  
826 and Fig. 9 a and b versus mean residence time, also correlated to recharge year (upper  
827 x-axes). The majority of the chemistry data of the Rotorua data set are from calendar  
828 year 2003, therefore mean residence time of about 50 and 125 years correspond to  
829 mean groundwater recharge years 1955 and 1880, respectively. Homogeneous nitrate  
830 concentrations in discharges from within sub-catchments of typically  $0.7 \pm 0.2$  mg/L  
831 NO<sub>3</sub>-N indicate that nutrient inputs are derived from diffuse rather than a small  
832 number of point sources, pointing to agricultural sources.

833

834 Fig. 8

835 Fig. 9

836

837 Fig. 8 also includes data (labelled 'other') from the sites in the Lake Rotorua  
838 catchment that could not be assigned to one of the HCA clusters because these sites  
839 had not been analysed for the full suite of hydrochemical parameters required for  
840 input into HCA. In several surveys only nitrate was measured to obtain a higher  
841 spatial resolution of the nitrate distribution. The analysis of all hydrochemical  
842 parameters, as required for HCA, was mainly undertaken at the large discharges into  
843 the lake that contain old water, and only few of these sites contain water young  
844 enough to show the impact of recent land use intensification. Therefore the 'other'  
845 samples were added to Fig. 8 to better represent younger waters. In addition, samples  
846 from the eastern catchment having a geothermal signature are also included in the  
847 cluster 'other'. The geothermal influence is minimal and does not affect the nitrate  
848 signature, and hence does not bias the display of results in Fig. 8.

849 Nitrate concentrations (Fig. 8) in oxic groundwaters with MRT > 125 years  
850 (recharged prior to 1880) in the Rotorua catchment are higher, with up to about 0.7  
851 mg/L NO<sub>3</sub>-N (dotted line in Fig. 8) compared to other regions in New Zealand with  
852 0.2 mg/L NO<sub>3</sub>-N. The reason for elevated nitrate in water despite a high mean  
853 residence time is the high degree of mixing in the groundwater discharges from the  
854 highly porous unconfined Rotorua ignimbrite aquifers (see next section). In such  
855 aquifer conditions, groundwater from short and long flow paths converge at the  
856 groundwater discharges, causing a high degree of mixing of young and old water. For  
857 example, groundwater with a mean residence time (MRT) as high as 170 years, using  
858 an exponential piston flow model with 95% exponential flow volume within the total  
859 flow volume (see next section), contains over 20% of water recharged after 1955.  
860 This post-1955 water can contribute significant amounts of nitrate from high-intensity  
861 land use, raising the nitrate concentration of the water mix considerably, despite such  
862 a long mean residence time.

863 A significant increase in nitrate occurred only recently (Fig. 8). Apart from  
864 one data point, an increase up to 1.5 mg/L NO<sub>3</sub>-N was observed only in water with  
865 MRTs of less than 75 years, and a dramatic increase up to 14 mg/L NO<sub>3</sub>-N was  
866 observed in water with MRTs of less than 50 years. Note the dramatic increase of  
867 nitrate in water with MRT < 20 years, reflecting the increased conversions to dairy  
868 farming during the 1980s and 1990s (Rutherford et al., 2011). As the majority of the  
869 water discharges into Lake Rotorua are significantly older than a few decades, with  
870 MRTs of up to 120 years, the impact of the dairy conversions and their nitrate loads  
871 over the recent decades has to a large extent not yet reached the lake. Increased nitrate  
872 loads to the lake over the next decades must be expected as these nitrate loads work  
873 their way through the large groundwater system and eventually discharge into the  
874 streams and lake.

875 Sulphate and potassium are part of fertilizers and also show elevated  
876 concentrations in young groundwater (Fig. 9). Note that sulphate in groundwater in  
877 the eastern lake catchment has much elevated concentrations, up to 40 mg/L SO<sub>4</sub>, due  
878 to geothermal influence. Groundwater with indications of geothermal influence is not

879 discussed in this study. Also note that sulphate can be biased due to anoxic  $\text{SO}_4$   
880 reduction. The data shown are, however, not from anoxic groundwater environments.  
881 Sulphate and potassium show slightly elevated concentrations, up to a factor of three,  
882 only in water with MRT <50 y, corresponding to water recharged after approximately  
883 1950. Sulphate concentrations (in mg/L  $\text{SO}_4$ ) in the Rotorua volcanic aquifers are  
884 significantly lower compared to a national survey: pre-anthropogenic concentration of  
885 2 versus 12, and high intensity land-use concentrations of up to 6 versus 94 for the  
886 Rotorua volcanic aquifers and the national survey (Morgenstern and Daughney,  
887 2012), respectively.

888 Phosphate, in conjunction with nitrate the cause for lake eutrophication, is not  
889 elevated in young groundwater (Fig. 7a) despite its frequent application as super-  
890 phosphate fertilisers. Absence of elevated  $\text{PO}_4$  in young groundwater implies that  
891 fertiliser phosphate from non-point sources has not yet reached the saturated  
892 groundwater systems and is still retained in the soil. This finding is consistent with the  
893 usually high P-retention scores for ashfall soils and thick unsaturated zones across this  
894 region, which are very efficient at buffering P loss. P-retention in soils was also  
895 observed in the New Zealand National Groundwater Monitoring Programme across  
896 other soil types (Morgenstern and Daughney, 2012).

897 The presence of elevated  $\text{PO}_4$  only in old groundwater indicates that its source  
898 is purely due to geological factors, because these waters were recharged before land-  
899 use intensification.  $\text{PO}_4$  concentrations up to 0.1 mg/L  $\text{PO}_4$ -P are observed, due to  
900 phosphate leaching from the rhyolite ignimbrite and lava formations. With most  
901 groundwater discharging into Lake Rotorua being very old (MRT >50 y), the water  
902 has naturally high  $\text{PO}_4$  concentrations, well above the threshold for primary algae  
903 production of c. 0.03 mg/L total phosphate (Dodds, 2007).

904 The high phosphate load to the lake via groundwater is natural. As the turn-  
905 over time of the lake water is only 2.2 years via the high  $\text{PO}_4$ -bearing streams, there is  
906 a constantly high  $\text{PO}_4$  load reaching the lake via all streams. Therefore, the only  
907 effective way to limit algae blooms and improve lake water quality in such  
908 environments is by limiting the nitrate load.

909  
910

#### 911 **4.5 Prediction of future nitrate load**

912

913 The water quality of Lake Rotorua has declined continuously over the past 60  
914 years, responding very slowly to historical agricultural and urban development in the  
915 catchment, and large amounts of groundwater have insidiously become contaminated  
916 over the last 60 years because of the long travel times through the groundwater system  
917 of the Lake Rotorua catchment. The response time of the groundwater system to  
918 mitigation action will also be lengthy; it will take similar time frames until the  
919 contaminated water is flushed out of the aquifers. To improve lake water quality and  
920 define reduction targets for nutrients that affect lake water quality, prediction of future

921 contaminant loads from current and historic activities in the Lake Rotorua catchment  
922 are required.

923 In the previous section we have shown that of the two main contaminants that  
924 together cause lake eutrophication, phosphorus is naturally present in the volcanic  
925 lake environment, but nitrate from anthropogenic sources has been leaching into the  
926 groundwater since the onset of industrial agriculture, delivering increasing nitrate  
927 loads to the lake. Figure 8 shows significantly elevated nitrate concentrations in  
928 groundwater recharged after 1955.

929 Due to the large lag time in the groundwater system, these younger  
930 groundwaters, with their higher nitrate load, have not yet worked their way fully  
931 through the groundwater system. Significant fractions of the groundwaters  
932 discharging to the lake are older (Figs. 3 and 4), with MRT > 50 years, and were  
933 recharged before land-use intensification. Therefore the water discharges into the lake  
934 are currently still diluted by old pristine water. With the delayed arrival of nitrate  
935 from historic land use, which ultimately will discharge from the groundwater system  
936 via the springs and streams into the lake, nitrate loads to the lake from historic land-  
937 use activities must be expected to increase further in the future. No significant  
938 denitrification can be expected in the Rotorua groundwater system (Fig. 6a).

939 The age distributions functions derived from the tritium time series data in the  
940 stream discharges to Lake Rotorua (Table 2) can be used to project the future arrival  
941 to the lake of water that was recharged since land-use development in the catchment  
942 (Morgenstern and Gordon, 2006). The age distribution function for Hamurana Stream,  
943 the largest stream (Table 1), which discharges some of the oldest water to the lake  
944 (Fig. 2), is shown in Fig. 10.

945

946 Fig. 10

947

948 Figure 10 shows the two superimposed age distributions of exponential piston  
949 flow models: EPM2 with younger water of MRT = 12 y and EPM1 with significantly  
950 older water of MRT = 185, together with the average MRT = 125 y between the two  
951 models (blue). Only the water younger than 55 years has been recharged after land-  
952 use intensification (red shaded) and contains elevated nitrate. The cumulative fraction  
953 of land-use impacted water is about 45%, implying that more than half of the water is  
954 still pristine old water. After this old water is completely displaced by land-use  
955 impacted water, the nitrogen load of Hamurana Stream will approximately double.  
956 The projected increase in nitrogen load over time, as derived from the age  
957 distribution, is shown in Fig. 11.

958

959 Fig. 11

960

961 Nitrate, **as opposed to other nitrogen fractions,** is clearly the major component  
962 of nitrogen in the Rotorua groundwater system (Morgenstern et al., 2004).  
963 Concentrations of nitrate in the catchment were low (0.14 mg/L NO<sub>3</sub>-N) prior to  
964 catchment development, as determined from old groundwater (Morgenstern and

965 Gordon, 2006). The prediction of nitrogen load increase is calculated by scaling the  
966 nitrogen load currently measured in the stream (full symbol, Fig. 11) according to its  
967 fraction of land-use impacted water to the various years over time, using the age  
968 distribution (Fig. 10).

969 Good agreement with average historic monitoring results of nitrogen loads  
970 (Rutherford, 2003, hollow symbols in Fig. 11) confirms that the assumptions  
971 regarding the baseline concentration and timing of nitrogen input in the catchment are  
972 reasonable. Using the age distributions derived for all stream discharges to the lake,  
973 we also projected the total nitrogen load increase to Lake Rotorua (Morgenstern and  
974 Gordon, 2006). In regards to phosphorus, there are no elevated phosphorus  
975 concentrations in young groundwater (Fig. 7a) and the phosphorus load to Lake  
976 Rotorua is projected to stay constant, as long as fertilizer phosphate does not break  
977 through the soil into the groundwater.

978 The time scales necessary for the Hamurana Stream to adjust to changes in  
979 land-use activities in the catchment are long. Due to the long residence time of the  
980 water in the large aquifer system, it takes more than a hundred years for the  
981 groundwater discharge to the lake to adjust to changes in land-use activities. These  
982 time scales apply to activities that cause contamination, but also to remediation action.

983 This projection of nitrogen load via the stream is based on actual nitrogen  
984 concentrations in the stream (combined with the age of the water) and accounts only  
985 for the nitrogen from land-use activities that leaches out of the root zone of  
986 agricultural land into the deeper part of the groundwater system. Any nitrogen uptake  
987 in the soil is already taken into account.

988 The above nitrogen prediction is based on constant nitrogen input since  
989 catchment development. This trend will, however, be exacerbated by any further  
990 intensification of land use within the catchment over recent decades, as this recently  
991 recharged water has largely not yet reached the streams.

992

993

994

## 995 **5. Conclusions**

996

997 This study shows how the isotopic and chemistry signature of groundwater can be  
998 used to help determine the sources and the dynamics of groundwater and  
999 contaminants that travel with it, in particular in complex groundwater systems that are  
1000 difficult to characterise using conventional hydrogeologic methods, such as that of the  
1001 Lake Rotorua catchment. The isotopic and chemistry signatures of the major  
1002 groundwater-dominated stream discharges to the lake, after passing through the large  
1003 aquifer system of the catchment, allow us to understand groundwater processes and  
1004 lag time on a catchment scale.

1005 Tritium time series data and complementary age tracers SF<sub>6</sub> and CFCs can be used  
1006 to establish age distribution parameters, allowing for understanding of groundwater  
1007 processes and dynamics, and the timing of groundwater contamination. This is



1008 particular useful in catchments where little information is available on historic land-  
1009 use activities.

1010 After long-standing controversies (e.g. White et al., 2004; Rutherford et al., 2011),  
1011 hierarchical cluster analysis of the water chemistry parameters has provided evidence  
1012 about the recharge areas and hydraulic connections of the large springs near the  
1013 northern shore of Lake Rotorua. Streams and shallow wells that gain most of their  
1014 flow and recharge within the lacustrine sediments display a characteristic  
1015 hydrochemical signature. Hydrochemistry of the water draining the Ngongotaha lava  
1016 dome also has a characteristic signature due to interaction with lava formations. Only  
1017 where the lava dome intercepts the paleo-lake sediments is the groundwater flow from  
1018 the lava formation forced to the surface due to the low permeability of the sediments.  
1019 The water from the ignimbrite also displays a characteristic hydrochemical signature.  
1020 Similarly to the discharges from the lava formation, the water from the ignimbrite  
1021 discharges near the intercept of the ignimbrite formation with the paleo-lake  
1022 sediments, indicating the groundwater flow from the ignimbrite is forced to the  
1023 surface due to the low permeability of the sediments. The largest springs, discharging  
1024 in the north-west of the lake, emerge close to the lake shore within the sediment area,  
1025 but the ignimbrite signature of these water discharges implies that these springs drain  
1026 the Mamaku ignimbrite plateau, which has negligible surface runoff, through the lake  
1027 sediment layers in slope areas where the sediments are thinner and weaker.

1028 Groundwater chemistry and age data show clearly the source of nutrients that  
1029 discharge with the groundwater into the lake and cause lake eutrophication. Low  
1030 nitrate concentration in old oxic groundwater and high nitrate concentration in young  
1031 groundwater recharged after catchment development in the 1950s implies an  
1032 anthropogenic source of nitrate from agricultural activities, while low phosphate  
1033 ( $\text{PO}_4$ ) concentrations in young groundwater but high  $\text{PO}_4$  concentrations in old  
1034 groundwater imply a geologic source. High  $\text{PO}_4$  is a natural constituent of the  
1035 groundwater that discharges via the streams into the lake, and with a turn-over time of  
1036 the lake water of only 2.2 years, there is a constantly high  $\text{PO}_4$  load reaching the lake  
1037 via all streams. Therefore, the only effective way to limit algae blooms and improve  
1038 lake water quality in such environments is by limiting the nitrate load.

1039 The groundwater in the Rotorua catchment, once it has passed through the soil  
1040 zone, shows no further decrease in dissolved oxygen over the full range of residence  
1041 time of the water in the aquifer, indicating absence of significant microbial reactions  
1042 due to limitation of electron donors in the aquifer (e.g. organic matter) that could  
1043 facilitate microbial denitrification reactions (Kendall and McDonnell, 1998, Tesoriero  
1044 et al., 2007). Nitrate from land-use activities that leaches out of the root zone of  
1045 agricultural land into the deeper part of the groundwater system is unlikely to undergo  
1046 any significant degree of reduction through denitrification and must be expected to  
1047 travel with the groundwater to the lake.

1048 The old age of the water, with mean residence time of >50 years for most water  
1049 discharges to the lake, implies that there is a large lag time for transmission of the  
1050 nitrate through the groundwater system. Younger groundwaters, with their higher  
1051 nitrate load, have not yet worked their way fully through the groundwater system.

1052 With increasing arrival of this nitrate from historic land uses, a further increase of the  
1053 nitrate load to the lake must be expected in the future.

1054 The old age and the highly mixed nature of the water discharges imply a very  
1055 slow and lagged response of the streams and the lake to anthropogenic contaminants  
1056 in the catchment, such as nitrate. Using the age distribution as deduced from tritium  
1057 time series data measured in the stream discharges to the lake allows extrapolation of  
1058 the nutrient load from historic land-use activities into the future. For Hamurana  
1059 Stream, the largest stream to Lake Rotorua, it takes more than a hundred years for the  
1060 groundwater-dominated stream discharge to adjust to changes in land-use activities.  
1061 These time scales apply to activities that cause contamination, but also to remediation  
1062 action.

1063 Without age information on the groundwater-dominated streams, it would be  
1064 difficult to obtain such an understanding of groundwater process, groundwater  
1065 dynamics, and contaminant loads that travel with the groundwater.

1066

1067

1068

## 1069 **Acknowledgements**

1070 We thank personnel from Bay of Plenty Regional Council for assistance with  
1071 sample collection and for provision of some information used in this study, and Eileen  
1072 McSaveney for editing the manuscript. This research was supported by funding from  
1073 Bay of Plenty Regional Council and the Ministry of Science and Innovation (Contract  
1074 C05X1002).

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1289 **Tables**

1290

1291 **Table 1.** Estimates of baseflow from White et al. (2007). Abbreviations refer to Fig. 1

Stream	Abbreviation	Baseflow (l/s)
Hamurana	Ham	2750
Awahou	Awh	1700
Waiteti	Wtt	1300
Ngongotaha	Ngo	1700
Waiowhiro	Wwh	370
Utuhina	Utu	1600
Puarenga	Pua	1700
Waingaehe	Wgh	250
Waiohewa	Whe	390
Minor streams	n/a	350

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1296 **Table 2.** Age distribution parameters for the two binary exponential piston flow models (EPM) for the  
 1297 major stream discharges to Lake Rotorua. Average MRT is the mean residence time between the two  
 1298 EPMs.

Stream	EPM1		Fraction of EPM1	EPM2		Average MRT [y]
	MRT1	f1		MRT2	f2	
Hamurana	185	0.82	0.65	12	0.77	125
Awahou	80	1.00	0.92	6	0.91	75
Waiteti	60	1.00	0.78	3	0.90	45
Ngongotaha	35	1.00	0.82	1	0.91	30
Waiowhiro	40	0.63	1.00	na	na	40
Utuhina	85	0.60	0.70	1	1.00	60
Puarenga	44	1.00	0.95	2	1.00	40
Waingaehe	160	0.94	0.90	3	1.00	145
Waiohewa	55	1.00	0.75	1	1.00	40

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1303 **Figure Captions**

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1305 **Fig. 1.** Location and geology of the Lake Rotorua catchment, with sampling sites. The assumed  
1306 groundwater catchment is from White and Rutherford (2009). Surficial geology is based on the  
1307 1:250,000 map of Leonard et al. (2010). For abbreviations of the names of the major streams refer to  
1308 Table 1. The approximate trace of the caldera is shown. Cross-section A'-A is shown on the cut-away  
1309 face of Fig. 2.

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1311 **Fig. 2.** Three-dimensional geological model of the Lake Rotorua catchment area (from White et al.,  
1312 2007). The location of the vertical cut-away face A'-A is shown in Fig. 1. The vertical exaggeration is  
1313 5×, with a 1-km vertical scale bar shown in the centre. The colour scheme is similar to that in Fig. 1  
1314 (bottom to top): Pre-Mamaku formations (grey), Mamaku Plateau Formation (light blue), sediment  
1315 formation (dark yellow), lava domes (red) and Holocene alluvial deposits (light yellow).

1316

1317 **Fig. 3.** Tritium rain input for the Rotorua catchment, and measured tritium output at Hangarua Spring  
1318 and at Hamurana water intake spring. The input curve is based on monthly measurements in Kaitoke  
1319 near Wellington, New Zealand, scaled to the latitude of Rotorua with a factor 0.87, and smoothed by an  
1320 exponential piston flow model with 0.3 years mean residence time and 50% exponential flow within  
1321 the total flow volume. One TU = one tritium atom per 10<sup>18</sup> hydrogen atoms. For the spring samples  
1322 one-sigma measurement errors are shown. Note the logarithmic scale of the TU axis.

1323

1324 **Fig. 4.** Tritium time series data, together with their matching lumped parameter model outputs, for six  
1325 major streams. Grey line is tritium input via rain from Fig. 3. The locations of the streams are shown in  
1326 Fig. 1.

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1328 **Fig. 5.** Distribution of Hierarchical Cluster Analysis (HCA) clusters in Lake Rotorua catchment,  
1329 together with the geological units (Leonard et al., 2010) and stream reaches. Stream reaches shown as  
1330 dotted lines are usually dry. HCA clusters relate to origin of the groundwater from one of the three  
1331 main geologic formations: Mamaku ignimbrite (light blue), lava (red), and lacustrine sediment  
1332 (yellow).

1333

1334 **Fig. 6.** a) Dissolved Oxygen (DO) and b) pH versus mean residence time (MRT). The colour codes of  
1335 the samples indicate water from the relevant geologic formation, as indicated by Hierarchical Cluster  
1336 Analysis (HCA).

1337

1338 **Fig. 7.** a) Dissolved reactive phosphate (PO<sub>4</sub>-P), b) silica (SiO<sub>2</sub>), c) bicarbonate (HCO<sub>3</sub>), d) sodium  
1339 (Na), and e) fluoride (F) versus mean residence time (MRT). The sample colour code for all graphs is  
1340 shown in graph a), and indicates water origin from the relevant geologic formation, as indicated by  
1341 Hierarchical Cluster Analysis (HCA).

1342

1343 **Fig. 8.** Nitrate (NO<sub>3</sub>) versus mean residence time (MRT). The sample colour code indicates water  
1344 origin from the relevant geologic formation, as indicated by Hierarchical Cluster Analysis (HCA).

1345

1346 **Fig. 9.** a) Sulphate (SO<sub>4</sub>) and b) potassium (K) versus mean residence time (MRT). The sample colour  
1347 code indicates water origin from the relevant geologic formation, as indicated by Hierarchical Cluster  
1348 Analysis (HCA). The upper axis indicates calendar year.

1349

1350 **Fig. 10.** Age distribution for Hamurana Stream (at inflow into Lake Rotorua). The red shaded area  
1351 indicates the fraction of water that was recharged after land-use intensification. EPM1 and EPM2 are  
1352 exponential piston flow models.

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1354 **Fig. 11.** Projected increase over time of nitrogen load to Lake Rotorua from Hamurana Stream.

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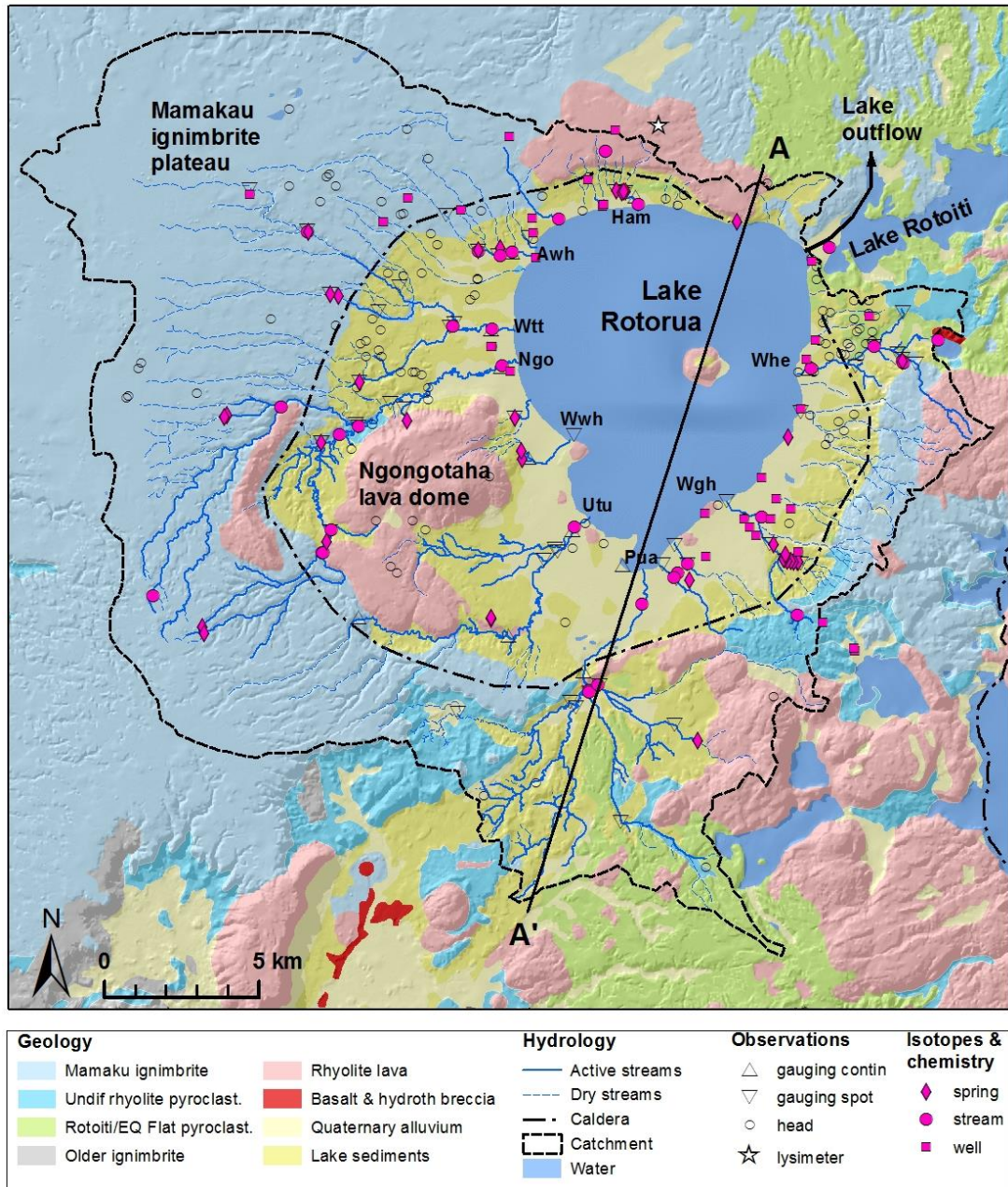
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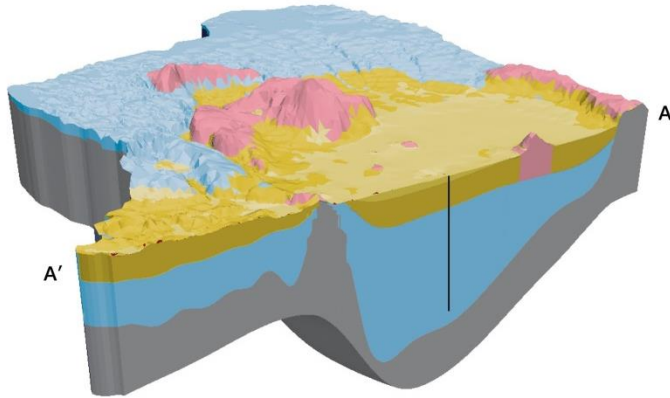
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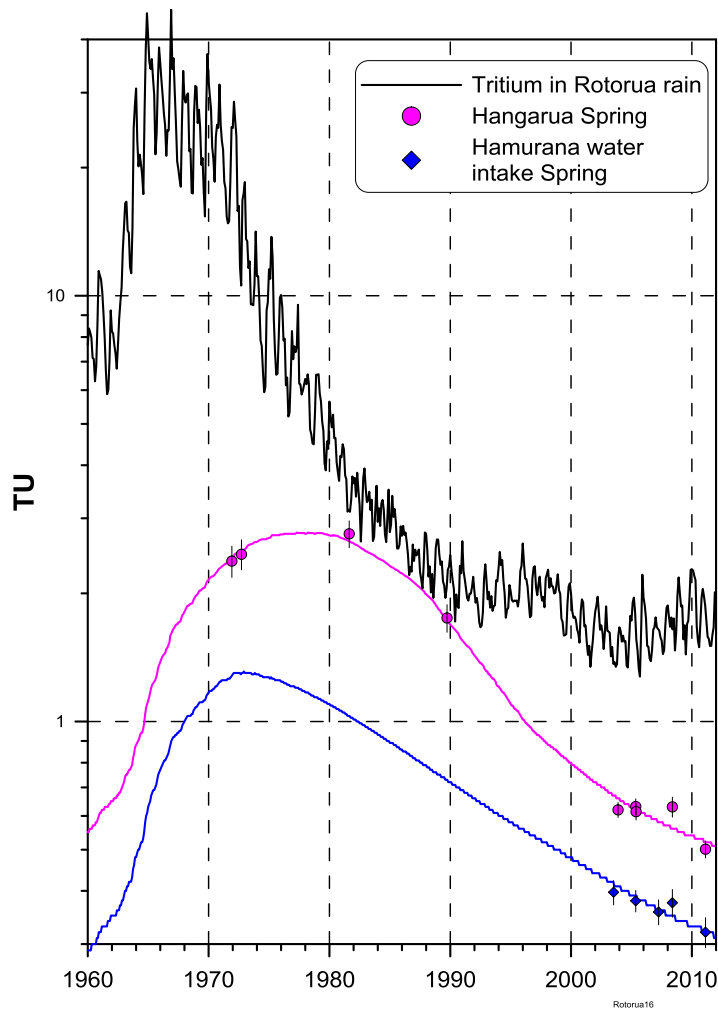
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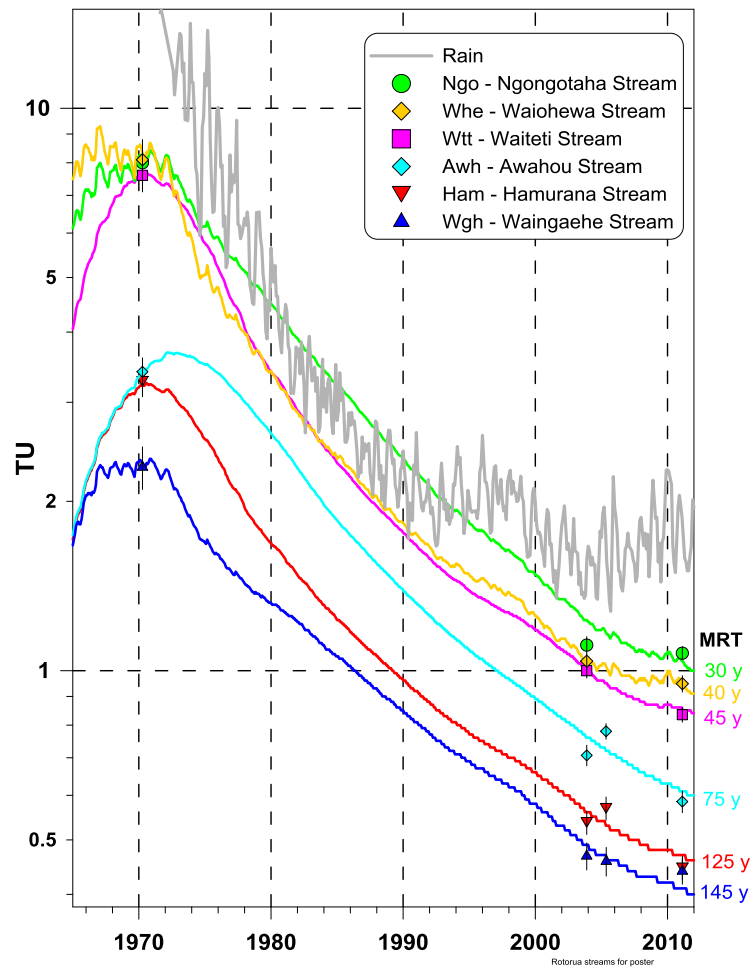
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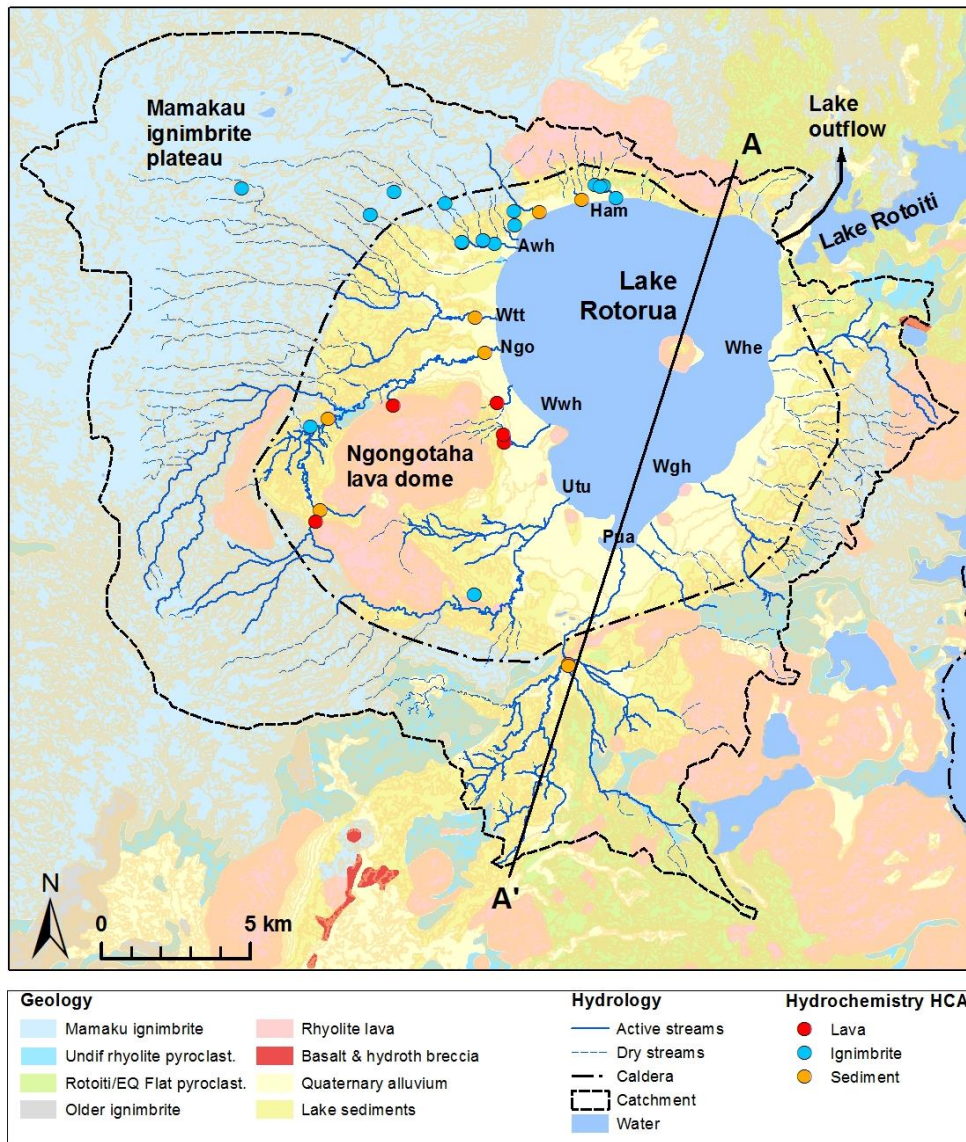
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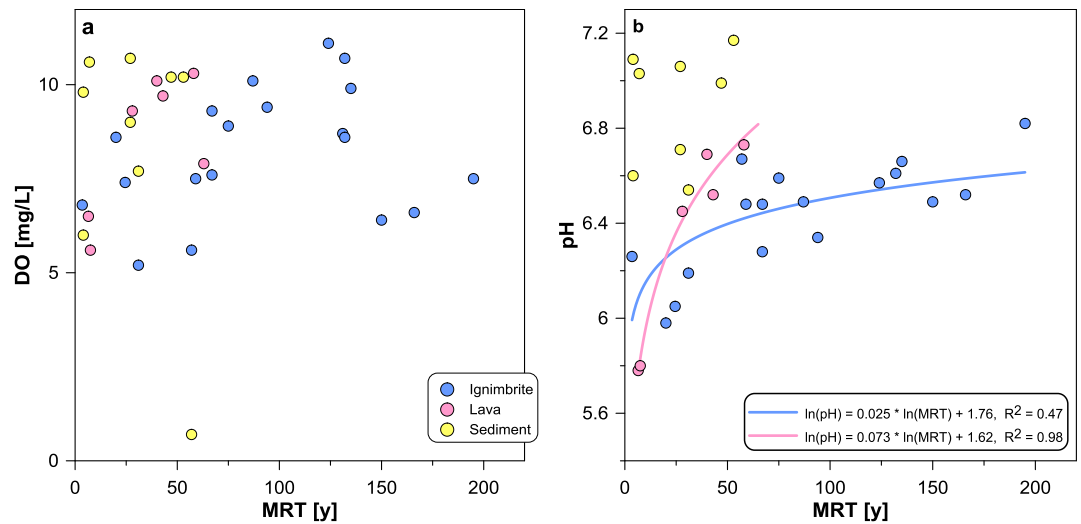
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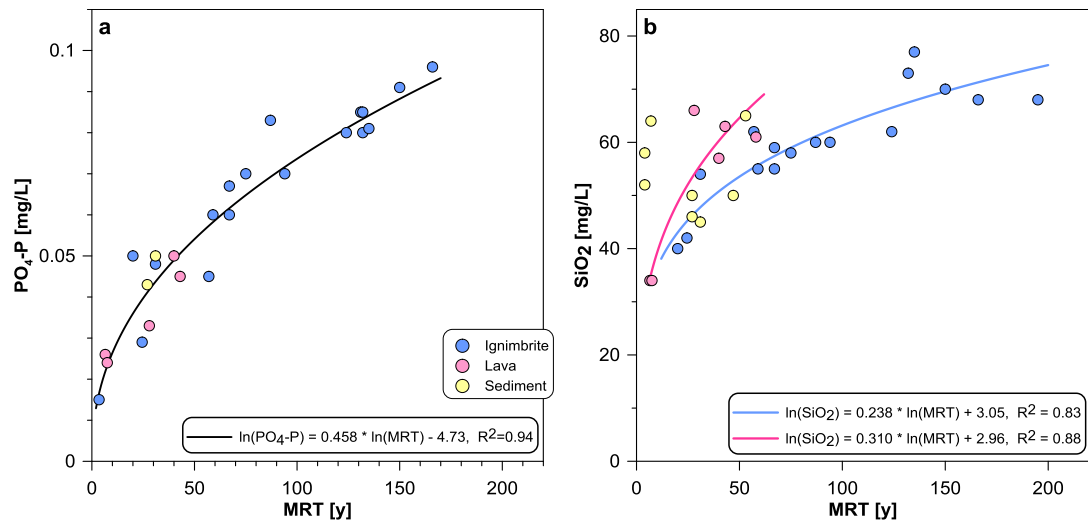
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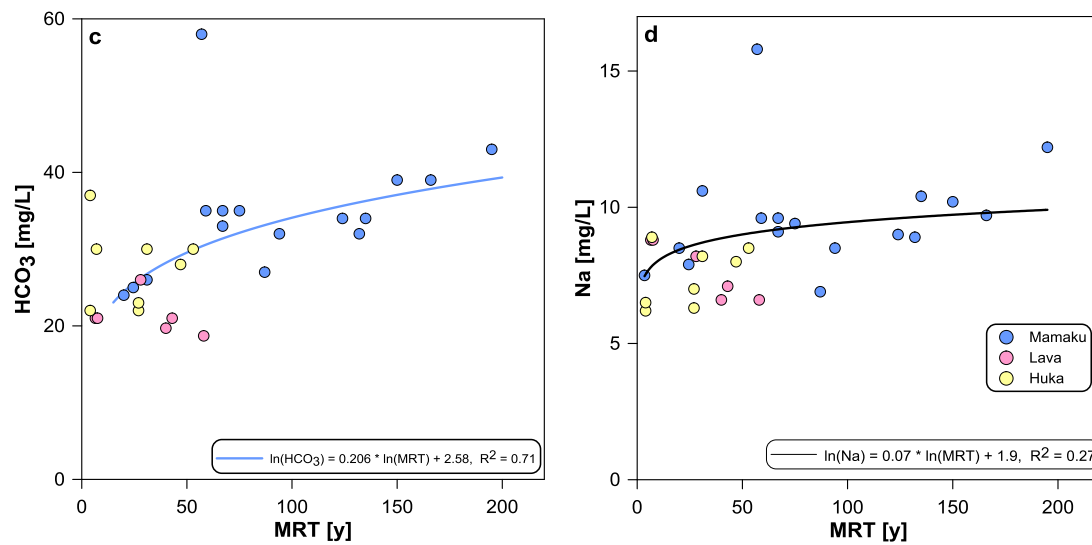


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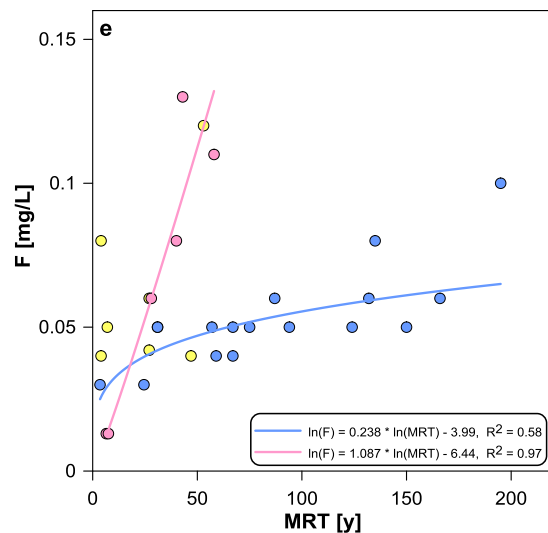
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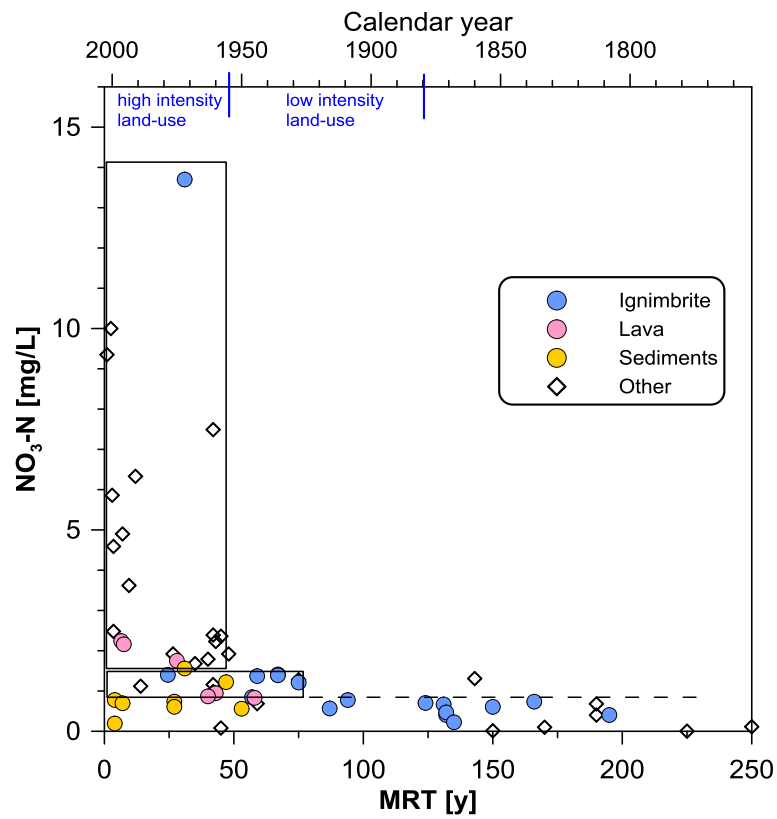


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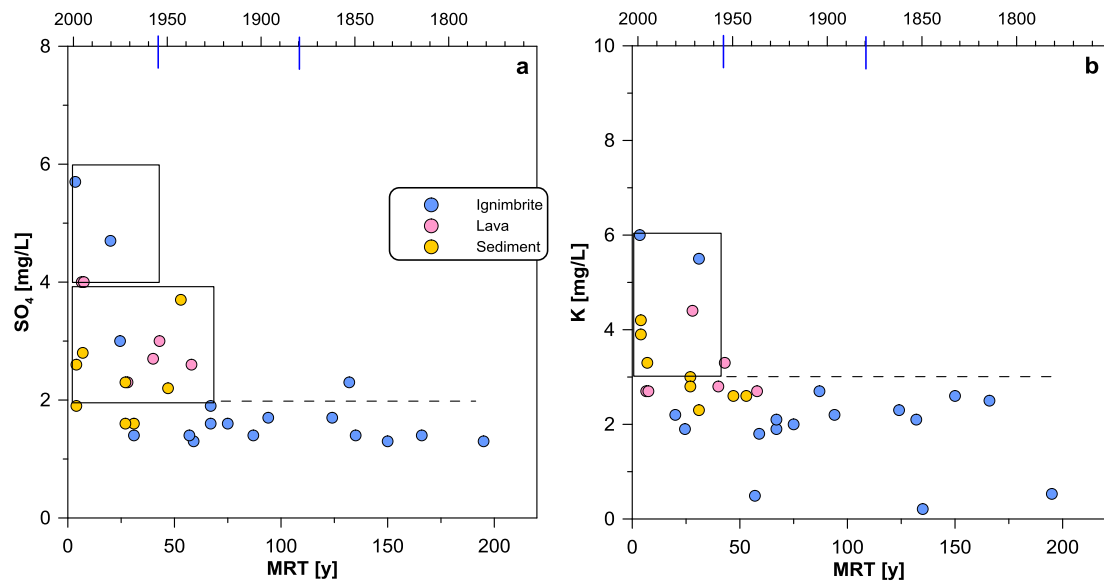
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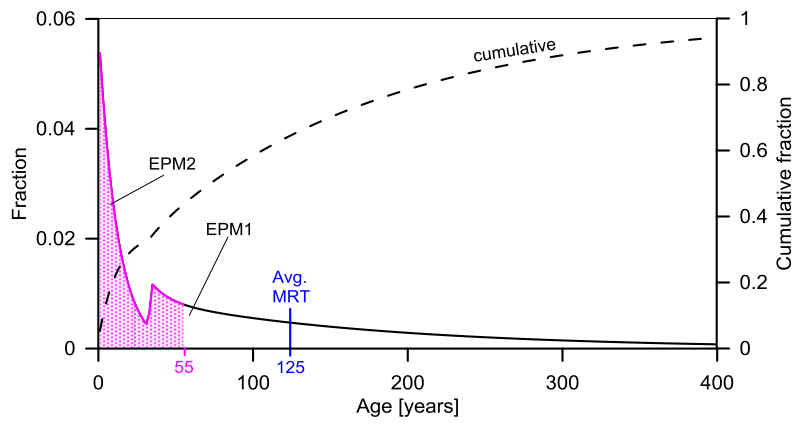
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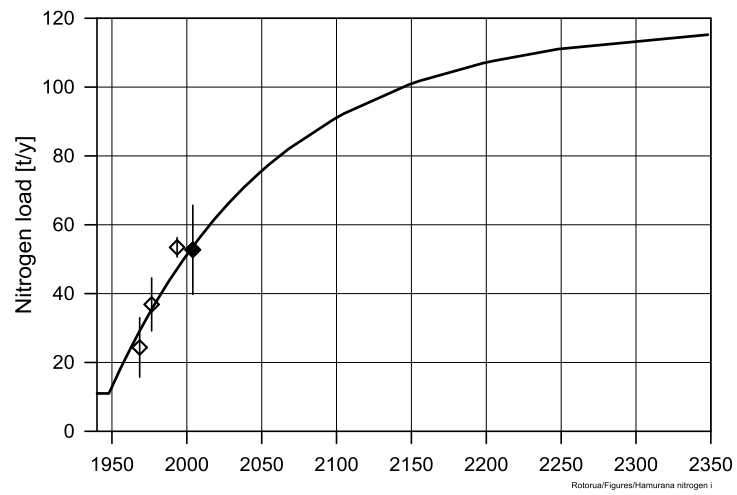
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