

1 Mobilisation or dilution? Nitrate response of karst springs to high rainfall  
2 events

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4 **Manuela Huebsch**<sup>a,b\*</sup>, **Owen Fenton**<sup>b</sup>, **Brendan Horan**<sup>c</sup>, **Deirdre Hennessy**<sup>c</sup>,  
5 **Karl G. Richards**<sup>b</sup>, **Philip Jordan**<sup>d</sup>, **Nico Goldscheider**<sup>a</sup>, **Christoph Butscher**<sup>a</sup>,  
6 **Philipp Blum**<sup>a</sup>

7 <sup>a</sup> Karlsruhe Institute of Technology (KIT), Institute of Applied Geosciences (AGW), Kaiserstr. 12,  
8 76131 Karlsruhe, Germany

9 <sup>b</sup> Teagasc, Environmental Research Centre, Johnstown Castle, Co Wexford, Ireland

10 <sup>c</sup> Teagasc, Animal and Grassland Research and Innovation Centre, Moorepark, Fermoy, Co. Cork,  
11 Ireland

12 <sup>d</sup> University of Ulster, School of Environmental Sciences, Cromore Road, Coleraine, BT52 1SA,  
13 Northern Ireland

14 \*Corresponding author: manuela.huebsch@partner.kit.edu (+49 721 608 45017)

15

## 16 **Abstract**

17 Nitrate (NO<sub>3</sub><sup>-</sup>) contamination of groundwater associated with agronomic activity is of  
18 major concern in many countries. Where agriculture, thin free draining soils and karst  
19 aquifers coincide, groundwater is highly vulnerable to nitrate contamination. As  
20 residence times and denitrification potential in such systems are typically low, nitrate  
21 can discharge to surface waters unabated. However, such systems also react quickest  
22 to agricultural management changes that aim to improve water quality. In response to  
23 storm events, nitrate concentrations can alter significantly, i.e., rapidly decreasing or  
24 increasing concentrations. The current study examines the response of a specific karst  
25 spring situated on a grassland farm in south Ireland to rainfall events utilising high-  
26 resolution nitrate and discharge data together with on-farm borehole groundwater  
27 fluctuation data. Specifically, the objectives of the study are to formulate a scientific  
28 hypothesis of possible scenarios relating to nitrate responses during storm events, and

29 to verify this hypothesis using additional case studies from the literature. This  
30 elucidates the controlling key factors that lead to mobilisation and/or dilution of  
31 nitrate concentrations during storm events. These were land use, hydrological  
32 condition and karstification, which in combination can lead to differential responses  
33 of mobilised and/or diluted nitrate concentrations. Furthermore, the results indicate  
34 that nitrate response in karst is strongly dependent on nutrient source, whether  
35 mobilisation and/or dilution occur and the pathway taken. This will have  
36 consequences for the delivery of nitrate to a surface water receptor. The current study  
37 improves our understanding of nitrate responses in karst systems and therefore can  
38 guide environmental modellers, policy makers and drinking water managers with  
39 respect to the regulations of the European Union (EU) Water Framework Directive  
40 (WFD). In future, more research should focus on high resolution monitoring of karst  
41 aquifers to capture the high variability of hydrochemical processes, which occur at  
42 time intervals of hours to days.

43

44 **Keywords:** nitrate; karst; mobilisation; dilution; groundwater quality

45

## 46 **1 Introduction**

47 The consequences of groundwater contamination by reactive nitrogen ( $N_r$ , e.g. nitrate  
48  $NO_3^-$ ), derived from agricultural sources, is of major concern in many countries  
49 (Galloway and Cowling, 2002; Spalding and Exner, 1993; L'hirondel, 2002). As  
50 groundwater response times affect the physical and economic viability of different  
51 mitigation measures, there is a realisation that such responses must be incorporated  
52 into environmental policy. However, such processes are poorly understood  
53 (Sophocleous, 2012), particularly where nitrate discharges unabated from high N  
54 input agricultural systems underlain by thin free draining soils and karst aquifers  
55 (Huebsch et al., 2013). Denitrification potential and response times in such systems  
56 are low (Jahangir et al., 2012) and at karst springs processes such as mobilisation  
57 and/or dilution during rainfall events inevitably control nitrate concentrations. In the  
58 European Union (EU) the Water Framework Directive (WFD; OJEC, 2000) aims to  
59 achieve at least good water quality status in all water bodies by 2015 and for

60 groundwater a maximum admissible concentration (MAC) of  $50 \text{ mg NO}_3^- \text{ L}^{-1}$  is in  
61 place. In karst regions, characterising nitrate dynamics in aquifers can help to predict  
62 when concentrations are likely to breach this MAC or not. No such standard exists for  
63 surface water but instead, in countries such as the Republic of Ireland, a much lower  
64 MAC of  $11.5 \text{ mg NO}_3^- \text{ L}^{-1}$  exists for estuaries (Statutory Instruments S.I. No. 272 of  
65 2009). Recent assessments have found that 16% of Irish groundwater bodies were ‘at  
66 risk’ of poor status due to the potential deterioration of associated estuarine and  
67 coastal water quality by nitrate from groundwater (Tedd et al., 2014). Improving our  
68 conceptual model of nitrate mobilisation and/or dilution in karst systems will  
69 therefore allow us to better manage agricultural systems in the future.

70

71 Karst areas exhibit a challenge for the protection of groundwater resources, because  
72 high heterogeneity, high vulnerability and fast groundwater flow result in low natural  
73 attenuation of contamination (Bakalowicz, 2005). Karst systems can vary significantly  
74 in the vadose zone from direct to slow infiltration and in the phreatic zone due to the  
75 complexity of conduit systems, fracture development and matrix porosity  
76 (Bakalowicz and Mangion, 2003). Episodic rainfall events can lead to rapid recharge,  
77 which has strong impact on discharge at and contaminant transport to karst springs,  
78 particularly if the conduit system is well developed (Butscher et al., 2011;  
79 Goldscheider et al., 2010). In addition, karst specific surface features (e.g. swallow  
80 holes) can contribute to a rapid contamination of the underlying aquifer (Ryan and  
81 Meiman, 1996). As a result of all these specific characteristics, karst aquifers overlain  
82 by thin free draining soils respond quickest to changes in N loading on the surface  
83 (Huebsch et al., 2013).

84

85 Leaching of organic and inorganic N can vary significantly. Organic N that has been  
86 applied on the surface provides mineral N to the plant on a longer basis due to  
87 mineralisation processes, whereas inorganic N is immediately available for the plant  
88 and hence, highly susceptible to leaching, especially in the first hours to days after  
89 application (Di et al., 1998). Due to its high solubility and mobility, nitrate responds  
90 much quicker and stronger to changes in hydrologic conditions and land use than less

91 mobile ions such as phosphorus (Hem, 1992). Because of this, in karst aquifers, low-  
92 resolution monitoring of nitrate (e.g., time intervals on a weekly basis) is unlikely to  
93 adequately characterise the system. This is especially true during rainfall events (Pu et  
94 al., 2011). As the dynamics of the system can change not only within, but also across  
95 events, it is important to have high resolution monitoring over long time periods.  
96 Long-term high-resolution monitoring can reveal rapid dilution of nitrate  
97 concentrations (Mahler et al., 2008), rapid mobilisation of nitrate concentrations  
98 (Baran et al., 2008; Plagnes and Bakalowicz, 2002; Pu et al., 2011; Yang et al., 2013)  
99 or a combination of mobilisation and dilution of nitrate concentrations during one or  
100 several rainfall events (Stueber and Criss, 2005; Rowden et al., 2001; Peterson et al.,  
101 2002).

102

103 In recent years, high-resolution monitoring in karst catchments over extended periods  
104 of time received greater attention (Mellander et al., 2013; Schwientek et al., 2013).  
105 Also, spectrophotometrical ultraviolet/visible (UV/VIS) light monitoring, which has  
106 originally been developed for monitoring waste water treatment plants (Drovc and  
107 Vrtovšek, 2010), has been applied to karst springs in recent years to continuously  
108 monitor nitrate concentrations (Grimmeisen et al., 2012; Pu et al., 2011). Such  
109 techniques offer the opportunity to observe both long-term trends, sudden changes of  
110 nitrate concentrations (Storey et al., 2011) and to increase the understanding of nitrate  
111 transport dynamics.

112

113 In this study, high-resolution UV monitoring, discharge and groundwater level  
114 fluctuation measurements were performed to observe nitrate concentration patterns  
115 and their relation to karst spring discharge and groundwater level fluctuations in  
116 response to storm events. The study site in Southern Ireland represents an ideal test  
117 site for nitrate responses in karst springs to storm events because of the combination  
118 of intensive agronomic N loading on the surface, an underlying karst aquifer and  
119 hydrometeorological conditions that ensure storm events throughout the year.

120

121 By looking at different nitrate characteristics during storm events, we aim to answer  
122 the following questions: What are the key factors controlling increased (i.e.  
123 mobilised) or decreased (i.e. diluted) nitrate concentrations in karst springs as  
124 response to storm events? Does it depend on the karst system alone, the hydrological  
125 situation or land use and/or of a combination of all these components together?  
126 Specifically, the objectives of the present study are to formulate a conceptual model  
127 of possible scenarios of nitrate responses during storm events, and to verify this  
128 hypothesis using other examples from the literature together with data from our study  
129 site. The results of this study can contribute to an improved understanding of when  
130 and under what conditions nitrate is released to fresh surface waters and, therefore,  
131 can guide environmental modellers, drinking water suppliers and environmental  
132 policy makers with respect to the regulations of the EU Water Framework Directive.

133

## 134 **2 Materials and methods**

### 135 **2.1 Site description**

136 The study site of 1.1 km<sup>2</sup> is located approximately 35 km north of Cork city in the  
137 Republic of Ireland and adjacent to the Teagasc, Animal and Grassland Research and  
138 Innovation Centre, Moorepark, in Fermoy (8°15'W, 52°10'N). About 0.97 km<sup>2</sup> (~ 90  
139 %) of the area is farmed. To the east, the study site is bounded by the River Funshion  
140 (Fig. 1). A public water supply well is located approximately 50 m up-gradient from  
141 the most westerly part of the study site at the River Funshion. Due to the topography,  
142 the study site can be sectioned into three parts. The upper part is intensively used as  
143 grassland for dairy farming, whereas the lower part is only periodically utilized as  
144 grassland, as it can be flooded for large periods of the year due to the proximity to the  
145 River Funshion and a shallow groundwater table. A steep slope between these two  
146 parts, which is the third part of the study site, has been forested to prevent erosion.  
147 The farm yard is located centrally on the study site. It includes the housing for the  
148 dairy herd and an intensively operated piggery.

149

150 The study site has been a research farm (dairy) with a commercially farmed, intensive  
151 pig farm in the farm yard since 2006. Prior to 2006, the farm was an intensive  
152 commercial dairy and pig farm with high fertiliser and feed inputs. All nutrients  
153 (slurry, cattle and pig manures) generated on the farm were applied to the farm land.  
154 No historic nutrient records are available. Since 2006, the dairy farm has been  
155 operating as a research farm and nitrogen fertiliser application rates are maintained  
156 within the Nitrates Directive (EC, 1991) which was implemented in Ireland in 2007.  
157 Jahangir et al. (2012a) calculated the annual N surplus for the research farm between  
158 2009 and 2010 at 263 kg N ha<sup>-1</sup> by subtracting the annual N output (35 kg N ha<sup>-1</sup>)  
159 from input (298 kg N ha<sup>-1</sup>). Furthermore, they estimated the possible amount of N  
160 leached at 148 kg N ha<sup>-1</sup> for the same years by taking N losses via volatilization and  
161 denitrification in soil surface into account. All slurry and manure generated from the  
162 dairy enterprise is applied to the grassland on the farm. The piggery is privately  
163 operated and all associated nutrients (slurry and manure) are exported off the farm.  
164 The present study site is comparable with a dairy farm approx. 2 km apart in terms of  
165 agronomic N-loading, local weather conditions, hydrogeological and geological site  
166 characteristics. The neighboring dairy farm has been described in detail by Huebsch et  
167 al. (2013). In this study agricultural practices were analyzed and the applied nitrogen  
168 input on the surface was related to recorded nitrate occurrence in groundwater over an  
169 11-year period whilst also considering a time lag from source to groundwater. N-  
170 inputs at this study site were 335 to 274 kg ha<sup>-1</sup> between 2001 and 2011 whereas the  
171 calculated N surplus (N inputs – N exports) at farm level was 260 to 174 kg ha<sup>-1</sup>.  
172 Those findings can also be compared to the study of Landig et al. (2011) who  
173 calculated N-inputs at the present study site for 2008. N inputs were 337 kg ha<sup>-1</sup> while  
174 209 kg ha<sup>-1</sup> were derived from organic N sources and 128 from inorganic N sources  
175 (Landig et al., 2009). In addition, on the present study site the availability of N on the  
176 land surface during autumn has increased as the farm has extended grazing during that  
177 period.

178 • **FIG. 1: Site map for the study area (comments on figure)**

179

180 The top soil (0 – 0.5 m) of the study site consists of sandy loam, whereas the subsoil  
181 (0.5 – 10.0 m) is composed of sand and gravel (Jahangir et al., 2012b). Two different

182 types of Carboniferous limestone occur at the study site: the Waulsortian Limestone  
183 and the Ballysteen Formation (Fig. 1) (GSI, 2000). The Waulsortian Limestone is in  
184 general less bedded and more karstified than the Ballysteen Formation due to the  
185 occurrence of massive calcareous mud-mounds and a lower content of shale  
186 components (GSI, 2000). In Fig. 1 the boundary of the two limestone types is adapted  
187 from mapping by the Geological Survey of Ireland (GSI), which was conducted at a  
188 larger scale. Therefore, and because of the lack of bedrock cores of the wells that have  
189 been drilled, the exact boundary on the local scale is uncertain.

190

191 Six boreholes (BH1 to BH6) with diameters of 150 mm were drilled in 2005 (Fig. 1).  
192 Five wells (BH1 and BH3 to BH6) consist of a 50 mm diameter piezometer casing. A  
193 multilevel piezometer was installed in BH1 with 6 m screen sections beginning at  
194 25.18 m AOD and 43.18 m AOD. BH3 to BH6 each consist of a single piezometer  
195 with a 6 m screen section beginning at 19.85, 24.68, 20.38 and 17.57 m AOD,  
196 respectively. BH2 is an open borehole with 150 mm diameter. It was found to be dry  
197 to a drilling depth of 62.9 m and subsequently filled with water already the day after  
198 drilling. The average drilling depth on site is 45.9 m with a minimum depth of 31.2 m  
199 at BH6 and a maximum depth of 62.9 m at BH2.

200

201 A perennial spring is located at the foot of the slope area (Fig. 1). The spring  
202 discharge is captured in a reservoir of about 23 m<sup>2</sup> and used as water supply for the  
203 dairy farm and the piggery. Water that is not needed for the farm flows over a weir via  
204 a channel towards the river.

205

## 206 **2.2 Spring, water level and meteorological data**

207 High-resolution monitoring of nitrate-nitrogen (NO<sub>3</sub>-N) in spring water was  
208 performed photometrically between the 11<sup>th</sup> of July 2011 and the 20<sup>th</sup> of April 2013 at  
209 15 min intervals with a two-beam UV sensor (NITRATAX plus sc, Hach Lange  
210 GmbH, Germany) using a 5 mm measuring path. The sensor reports NO<sub>3</sub>-N by

211 measuring total oxidised N (TON), and assuming negligible nitrite (NO<sub>2</sub>-N). To verify  
212 the UV sensor measurements, 12 water samples (50 ml) were taken at the sensor  
213 location in July 2011, 4 water samples in October 2012 and 12 water samples in May  
214 2013. Half of the samples were filtered immediately using a 0.45-µm micropore  
215 membrane, the other half were kept unfiltered to determine the influence of organic  
216 substances, as the accuracy of the sensor can be affected by those. All samples were  
217 transferred to 50 ml polyethylene screw top bottles, which were kept frozen prior to  
218 chemical analysis. TON and NO<sub>2</sub>-N content were determined in the laboratory  
219 (Aquakem 600A, Thermo Scientific, Finland), from which the nitrate concentration  
220 was calculated. For TON and NO<sub>2</sub>-N determination the hydrazine reduction method  
221 was used (Kamphake et al., 1967). The analysis of the unfiltered and filtered samples  
222 showed that UV sensor measurements were reliable and not affected by organic  
223 substances. NO<sub>2</sub>-N was negligible and the measured TON was reported as NO<sub>3</sub>-N.

224

225 To determine spring discharge, a trapezoidal weir was installed at the outlet of the  
226 spring capture reservoir (e.g. Walkowiak, 2006). The water level in the reservoir was  
227 measured with an electronic pressure transducer (Mini-Diver, Eijelkamp,  
228 Netherlands) in a stilling well at 15 min intervals. As the reservoir is used to provide  
229 water to the farm, a flow metre with data logger was also installed in the water supply  
230 pipe to measure pumped outflow. Changes in groundwater levels were continuously  
231 monitored at 15 min intervals in BH1, BH3, BH4 and BH6 using electronic pressure  
232 transducers (Mini-Diver, Eijelkamp, Netherlands).

233

234 Rainfall was recorded every hour at a Met Èireann weather station of approximately  
235 500 m from the study site. Effective Drainage (ED) was calculated as precipitation  
236 minus actual evapotranspiration, which was calculated from daily recordings of  
237 maximum and minimum temperature, precipitation, wind speed and solar radiation at  
238 the Met Èireann weather station after Schulte et al. (2005). In 2011 the annual rainfall  
239 was 855 mm and ED 364 mm, whereas in 2012 the annual rainfall was 1097 and ED  
240 578 mm.

241



## 242 3 Results

### 243 3.1 Observations at the study site

244 Two periods were evaluated: (1) from 13<sup>th</sup> November 2011 to 20<sup>th</sup> January 2012  
245 including high-resolution observations of NO<sub>3</sub>-N concentrations in spring water,  
246 precipitation and discharge (Fig. 2) and (2) from 1<sup>st</sup> February to 1<sup>st</sup> October 2012  
247 including high-resolution observations of NO<sub>3</sub>-N concentrations in spring water,  
248 precipitation and groundwater level fluctuations in BH1, BH3, BH4 and BH6 (Fig. 3).

249

250 Fig. 2 illustrates the impact of four storm events on discharge and nitrate patterns at  
251 the spring for period (1). Storm events were separated from each other if precipitation  
252 was less than 0.2 mm h<sup>-1</sup> for at least 24 hours in accordance to Kurz et al. (2005).  
253 Only storm events with a total amount of minimum 10 mm precipitation were taken  
254 into account.

255 • **FIG. 2: Observations at study site in period (1) between the 13<sup>th</sup> of**  
256 **November 2011 and the 20<sup>th</sup> of January 2012.**

257

258 The first storm event started on the 16<sup>th</sup> of November 2011 at 4 pm and ended on the  
259 19<sup>th</sup> of November at 10 am. A total of 60.3 mm precipitation was recorded during this  
260 time. Discharge started to rise on the 16<sup>th</sup> at 11.30 pm at 0.2 L s<sup>-1</sup> and reached its  
261 maximum of 1.7 L s<sup>-1</sup> on the 19<sup>th</sup> of November at 8:30 pm. After the maximum was  
262 reached, discharge decreased at first, and then showed a second increase, probably  
263 due a recurrence of intensified rainfall. NO<sub>3</sub>-N concentrations increased around 18.5  
264 hours later than discharge on the 17<sup>th</sup> of November at 5 pm and rose to 13.8 mg L<sup>-1</sup>  
265 until the 19<sup>th</sup> of November at 10:45 am. Hence, the NO<sub>3</sub>-N increase started later than  
266 the discharge increase but reached its maximum 9.75 hours earlier. After the  
267 maximum was reached, NO<sub>3</sub>-N exponentially decreased to 11.0 mg L<sup>-1</sup> until the 29<sup>th</sup> of  
268 November at 9 am.

269

270 The second storm event started on the 28<sup>th</sup> of November 2011 at 5 pm. Rainfall  
271 intensified and reached a total of 33.5 mm by the 30<sup>th</sup> of November at 10 pm.  
272 Discharge started to increase at 0.5 L s<sup>-1</sup> on the 28<sup>th</sup> of November at 10:45 pm, and the  
273 first maximum discharge of 1.2 L s<sup>-1</sup> was measured on the 29<sup>th</sup> of November at 7:30  
274 pm. However, the maximum discharge could have been higher and earlier. Intensive  
275 pumping at the reservoir between 12:15 and 7 pm led to a lack of stationary discharge  
276 values during that time. The increased discharge value of 1.0 L s<sup>-1</sup> or more was  
277 maintained until the 30<sup>th</sup> of November 2:30 am and decreased afterwards. The NO<sub>3</sub>-N  
278 concentrations started to increase at the 29<sup>th</sup> of November at 9 am at 11.0 mg L<sup>-1</sup> and  
279 reached its maximum of 12.1 mg L<sup>-1</sup> on the 29<sup>th</sup> of November at 5:45 pm. The NO<sub>3</sub>-N  
280 peak was observed about 1.45 hours earlier than the discharge peak.

281

282 During the third and fourth storm event, the same characteristics as described in the  
283 aforementioned storm events were observed at the spring. The total amount of  
284 precipitation was 28.8 mm for the third event and 18.7 mm for the fourth event. After  
285 rainfall intensified, discharge rose followed by increased NO<sub>3</sub>-N concentrations a few  
286 hours later. Again, the maximum NO<sub>3</sub>-N concentrations were reached earlier than the  
287 discharge peak. Specifically, during the third storm event discharge started to rise at  
288 0.4 L s<sup>-1</sup> on the 12<sup>th</sup> of December 2011 at 11:45 am, while NO<sub>3</sub>-N started to increase at  
289 10.6 mg L<sup>-1</sup> on the 12<sup>th</sup> of December 2011 at 3:15 pm. Highest discharge values were  
290 observed at 1.1 L s<sup>-1</sup> on the 13<sup>th</sup> of December 2011 at 12:30 pm. The NO<sub>3</sub>-N peak was  
291 reached at 11.0 mg L<sup>-1</sup> at 11:15 am on the same day and was therefore 1.15 hours  
292 earlier than the discharge peak. During the fourth storm event discharge started to  
293 increase at 0.3 L s<sup>-1</sup> on the 3<sup>rd</sup> of January 2012 at 4:30 am and NO<sub>3</sub>-N started to rise at  
294 10.6 mg L<sup>-1</sup> on the same day at 5:00 am. The maximum discharge was reached at 1.5  
295 L s<sup>-1</sup> on the 4<sup>th</sup> of January 2012 at 00:15 am and the maximum NO<sub>3</sub>-N concentration at  
296 11.0 mg L<sup>-1</sup> on the 3<sup>rd</sup> of January 2012 at 7 pm. Thus, the discharge maximum was  
297 reached 5.25 hours later than the NO<sub>3</sub>-N maximum.

298

299 In addition, groundwater level fluctuations at BH1 and BH3 to BH6 were observed  
300 and can be related to precipitation and NO<sub>3</sub>-N concentrations at the spring (Fig. 3).

301 During the 1<sup>st</sup> of February 2012 and the 1<sup>st</sup> of October 2012 groundwater level  
302 fluctuations in the boreholes accounted for up to 7.60 m. BH1 and BH3 had maximum  
303 water level fluctuations of 5.98 m on the 15<sup>th</sup> of August 2012 and 7.60 m on the 17<sup>th</sup>  
304 of August 2012, respectively. In the eastern part of the study site (Fig. 1), maximum  
305 water level fluctuations were lower. At BH4 and BH6 maximum values of 3.06 m on  
306 the 20<sup>th</sup> of August 2012 and 1.62 m on the 17<sup>th</sup> of August 2012, respectively, were  
307 observed. In all wells, the lowest groundwater level was observed at the beginning of  
308 June 2012 after a longer period of sparse precipitation. BH1 and BH3 in particular  
309 showed similar groundwater level fluctuation patterns as the response of NO<sub>3</sub>-N  
310 concentrations at the spring. Groundwater level fluctuations are reflecting ED.  
311 Between 11<sup>th</sup> of February 2012 and the 25<sup>th</sup> of April 2012 no ED occurred. Little ED  
312 was observed between 26<sup>th</sup> of April 2012 and 10<sup>th</sup> of June 2012 with a maximum peak  
313 of 13.3 mm and 27.3 mm in total. Between 11<sup>th</sup> of June 2012 and the 2<sup>nd</sup> of July 2012  
314 no ED occurred. During those periods groundwater levels dropped and no significant  
315 change in nitrate concentrations was observed at the spring. In the following period  
316 ED increased and three higher ED events > 20 mm were observed on the 7<sup>th</sup> of June  
317 2012 (23.7 mm), the 15<sup>th</sup> of June 2012 (21.4 mm) and the 28<sup>th</sup> of June 2012 (27.4  
318 mm). In August 2012 on the 12<sup>th</sup> and on the 15<sup>th</sup> high ED > 20 mm of 25.4 mm and  
319 25.1 mm, respectively, was observed. In Fig. 3 the high amounts of ED match with  
320 significantly increased nitrate concentrations at the spring. The maximum nitrate  
321 concentrations during the 5 events were 13.2 mg L<sup>-1</sup> on the 7<sup>th</sup> of June 2012 at 5.30  
322 pm, 13.7 mg L<sup>-1</sup> on the 15<sup>th</sup> of June 2012 at 6.30 pm, on the 28<sup>th</sup> of June 2012 13.6 mg  
323 L<sup>-1</sup> at 9.00 am, 13.6 mg L<sup>-1</sup> on the 12<sup>th</sup> of August 2012 at 7 pm and 14.1 mg L<sup>-1</sup> on the  
324 15<sup>th</sup> of August 2012 at 6 pm.

325 • **FIG. 3: Observations at study site in period (2) between the 1<sup>st</sup> of**  
326 **February and the 1<sup>st</sup> of October 2012.**

327

### 328 3.2 Conceptual model of nitrate responses in karst systems

329 A conceptual model of nitrate responses in karst groundwater systems was developed  
330 to elucidate the relationship between nitrate responses in karst springs and proposed

331 driving factors such as hydrological conditions, N availability through land use and  
332 karst features (Fig. 4).

333 • **FIG. 4: Conceptual model of nitrate response in karst systems**

334

335 Agriculture is known to be a main contributor of nitrate in groundwater, mainly  
336 because of inorganic and organic N fertilisation (Stigter et al., 2011). Current and past  
337 N applications, storage capacity and hydrological conditions can result in nitrate  
338 accumulation in the soil and epikarst (Fig. 4), while rainwater itself is typically low in  
339 nitrate concentration (about  $0.3 \text{ mg L}^{-1}$ , (Gächter et al., 2004)).

340

341 Groundwater flow in karst aquifer aquifers can be conceptualized by a dual flow  
342 system: water flows in pipe-like conduits and open cave stream channels (conduit  
343 flow system) as well as flow through fractures and pores (diffuse flow system). This  
344 dual flow concept is described in the literature and widely used in karst studies (e.g.,  
345 Shuster and White, 1971; Atkinson, 1977; White, 1988; Kiraly, 1998; Ford and  
346 Williams, 2007). Other researchers use a triple porosity concept for the description of  
347 karst aquifers, where groundwater flow is attributed to conduits, pores of the rock  
348 matrix and an intermediate flow system representing fissures and joints (e.g.,  
349 Worthington et al., 2000; Baedke and Krothe, 2001). In the conceptual model of the  
350 present study, the simpler dual porosity concept is used, which is well suited to  
351 describe the nitrate characteristics of the observed karst springs. Nitrate that recharges  
352 into the diffuse flow system during a storm event can hardly change nitrate  
353 concentrations within this large groundwater storage (Peterson et al., 2002). Hence,  
354 groundwater in the diffuse flow system is characterised by relatively stable nitrate  
355 concentrations that reflect average nitrate values of groundwater recharge and long-  
356 term trends. At the spring, stable nitrate concentrations representing water from the  
357 diffuse flow systems can be observed during base flow conditions.

358

359 During a storm event, water recharges also into the conduit flow system and bypasses  
360 the diffuse flow system. Nitrate concentrations of this recharge water strongly depend  
361 on hydrological conditions and land use. If nitrate concentrations in the soil and  
362 epikarst are high prior to a storm event, for example after N fertilisation, nitrate  
363 becomes mobilised and water with high nitrate concentration enters the conduit flow  
364 system. At the spring, a fast increase of nitrate concentrations can be observed as a  
365 storm response, which reflects nitrate mobilisation in the soil and epikarst by storm  
366 water. If nitrate concentrations in the soil and epikarst are low prior to a storm event,  
367 rainwater with low nitrate concentration enters the conduit flow system without a  
368 marked increase in nitrate concentration. At the spring, a fast decrease of nitrate  
369 concentrations can be observed as a storm response, which reflects the dilution of  
370 spring water by storm water.

371

372 Our conceptual model of karst spring responses to storm events can be summarized in  
373 four possible scenarios (Fig. 5). Scenario 1 (Fig. 5a) shows mobilisation of nitrate in  
374 the soil/epikarst during storm events and fast increasing nitrate concentrations as  
375 response at the spring, corresponding to observations of period (1) and (2) in the  
376 present study. Scenario 2 (Fig. 5b) shows dilution of spring water after storm events  
377 with fast decreasing nitrate concentrations. In Scenario 3 (Fig. 5c), nitrate in the  
378 soil/epikarst becomes mobilized during storm events, resulting in an initial increase in  
379 nitrate concentrations in spring water, followed by dilution of spring water with low  
380 nitrate storm water when groundwater recharge continues after mobilised nitrate has  
381 been flushed through the system. Scenario 4 (Fig. 5d) shows different responses to  
382 storm events depending on the availability of nitrate in the soil/epikarst. During the  
383 first event, little nitrate was available and dilution can be observed at the spring.  
384 Before the second event, high nitrate concentrations accumulated in the soil/epikarst.  
385 Nitrate then becomes mobilised during the second storm event and a sharp nitrate  
386 peak can be observed as response at the spring.

387 • **FIG. 5: Hypothesis of nitrate response scenarios**

388

389 The fast increase in nitrate concentrations after storm events indicates that  
390 mobilisation is the main process influencing nitrate patterns at the spring (Figs. 2 and  
391 3). At the site, intensive agriculture is the dominant land use including application of  
392 inorganic and organic N fertiliser. During dry weather, soil moisture deficit leads to  
393 an accumulation of nitrate and minor to zero leaching in the soil. This can be  
394 recognised at the spring during base flow conditions when nitrate concentrations  
395 remain fairly constant (for example between March and May 2012, Fig. 3). During  
396 storm events (for example in June 2012), residual nitrate that was not consumed by  
397 plants gets mobilised in the soil (Fig. 5a). At the spring, the rapid increase of nitrate  
398 concentrations, only a few hours after the start of a storm event, indicates that  
399 recharging water rapidly bypasses the diffuse flow systems in the rock matrix in  
400 activated conduit systems.

401

### 402 **3.3 Comparison with other studies**

403 To further test our conceptual model, documented nitrate responses to storm events  
404 were reanalysed with respect to the proposed processes (Fig. 4) and related to the  
405 various possible scenarios (Fig. 5). Four representative studies were selected that  
406 correspond to Scenarios 1 – 4 (Fig. 6).

#### 407 **• FIG. 6: Four illustrating case studies.**

408

409 Study 1 – Yverdon karst aquifer system, Switzerland (Pronk et al., 2009)

410 In this study, a similar response of discharge and nitrate concentrations after a storm  
411 event as in the present study was observed (Fig. 6a). During the whole study period, a  
412 nitrate range of 1.0 to 7.0 mg NO<sub>3</sub>-N L<sup>-1</sup> and a discharge range of 21 to 539 L s<sup>-1</sup> was  
413 monitored. After the storm event, discharge increased at the spring, followed by a  
414 steep nitrate increase with a slower drop down after the maximum was reached.  
415 According to our conceptual model, this pattern corresponds to mobilisation (Scenario  
416 1, Fig. 5a). Pronk et al. (2007) observed that a stream draining into a swallow hole in  
417 an agricultural dominated area contributes significantly to nitrate variations at the

418 spring during storm events. Their interpretation is in line with the conceptual model of  
419 the present study, where mobilisation in the soil/epikarst and subsequent transport of  
420 nitrate via the conduit flow system occur, i.e. rapidly by-passing the diffuse flow  
421 system of the rock matrix.

422

423 Study 2 – Chalk aquifer in Normandy, France, and Edwards aquifer, Texas, U.S.A.  
424 (Mahler et al., 2008)

425 In the second study, the observed predominant process after storm events (Fig. 6b)  
426 corresponds to dilution according to our conceptual model (Scenario 2, Fig. 5b). The  
427 observed NO<sub>3</sub>-N concentrations in the aquifer range between 2.2 and 9.0 mg L<sup>-1</sup>.  
428 Three days after the storm event, nitrate concentration decreased rapidly and rose  
429 gradually afterwards. The authors state that (recharging) surface runoff was rapidly  
430 transported through the conduit system, leading to dilution effects during the storm  
431 event. When the event water became increasingly replaced after the event by  
432 groundwater stored in the rock matrix, nitrate concentrations started to rise again.

433

434 Study 3 – Big Spring basin, Iowa, U.S.A. (Rowden et al., 2001)

435 In the third study, a storm event of 20 mm in total caused first predominance of  
436 mobilisation, directly followed by dilution during one event (Fig. 6c). This nitrate  
437 pattern corresponds well to Scenario 3 in our conceptual model (Fig. 5c). Rising  
438 nitrate concentrations during the event can be explained by first mobilisation of nitrate  
439 by infiltrating recharge, followed by dilution after mobilised nitrate is already flushed  
440 through the system and storm water continues to recharge into the conduit flow  
441 system. During the study period, discharge ranged from 300 to 7300 L s<sup>-1</sup> and NO<sub>3</sub>-N  
442 from 1.3 to 6.0 mg L<sup>-1</sup>.

443

444 Study 4 – Karst watershed, Illinois, U.S.A. (Stueber and Criss, 2005)

445 In this study, predominance of mobilisation during one and dilution during other  
446 events were observed (Fig. 6d), corresponding to Scenario 4 (Fig. 5d) of our  
447 conceptual model. Between May 2000 and December 2002, the authors frequently  
448 observed dilution during storm events. However, during one storm event, nitrate  
449 concentrations showed a different response – the concentrations increased rapidly  
450 (Fig. 5d, grey bar). The cause of the sharp nitrate increase was detected as heavy N  
451 fertilisation in the catchment during this time. A relatively constant NO<sub>3</sub>-N trend was  
452 monitored at 3.5 mg L<sup>-1</sup>, whereas during storm events concentrations decreased to 0.2  
453 mg L<sup>-1</sup> and increased up to 5.6 mg L<sup>-1</sup>.

454

#### 455 **4 Discussion**

456 In this chapter, the role of different key drivers in resulting nitrate responses at karst  
457 springs is discussed, including the hydrogeological setting of the karst system, mixing  
458 of water from different sources, hydrological conditions and land use practises. In  
459 addition, adequate sampling strategies for studying nitrate characteristics of karst  
460 systems are briefly discussed.

461

462 Transport of nitrate can occur quickly within conduits and fissures or be strongly  
463 retarded in less mobile water within the rock matrix (Baran et al., 2008). Hence, the  
464 development of the karst system itself plays an important role. But what karst features  
465 are most relevant for dilution and mobilisation processes?

466

467 In the study of Pronk et al. (2009), a sinking stream strongly impacts nitrate  
468 concentrations (and faecal bacteria) in spring water after storm events. The sinking  
469 stream points at the presence of a well-developed conduit system in the karst aquifer.  
470 The spring investigated in their study shows the same nitrate characteristics as the  
471 spring investigated in the present study. Also at the present study site, the existence of  
472 a well-developed conduit network is likely. For example, a cave exists at the study  
473 site (Fig. 1). However, the exact hydraulic properties of the karst system are uncertain.



474

475 In the study by Mahler et al. (2008) two karst systems that differ significantly in  
476 matrix porosity, thickness of soil and epikarst and land use were compared. In both  
477 karst systems, dilution was the observed predominant process after storm events. One  
478 karst system of this study is illustrated as an example in Fig. 6b. In contrast, the study  
479 of Baran et al. (2008), which focuses on a chalk aquifer in northern France  
480 comparable to one of the karst systems described in the aforementioned study of  
481 Mahler et al. (2008), shows predominance of nitrate mobilisation and not dilution, just  
482 as in the present study. Both chalk aquifers are characterised by a total matrix porosity  
483 of 30 to 40 %, low hydraulic conductivity of about  $10^{-9} - 10^{-8} \text{ m s}^{-1}$  and the presence  
484 of a conduit system with an observed hydraulic conductivity of  $10^{-3} \text{ m s}^{-1}$  (Mahler et  
485 al., 2008) and  $10^{-5}$  to  $10^{-3} \text{ m s}^{-1}$  (Baran et al., 2008). Nevertheless, a dual flow system  
486 will react differently to an isolated conduit system. A lower magnitude of the varying  
487 concentration is expected and the time lag between rise in spring discharge and  
488 response in concentration should be higher (Birk et al., 2006).

489

490 Similarly, Rowden et al. (2001) observed that the combination of infiltration and  
491 runoff recharge can have a significant influence on nitrate patterns at springs. The  
492 proportion of runoff recharge can vary significantly and changed in the study by  
493 Ribolzi et al. (2000) between 12 % for low intensity rain fall events and 82 % for high  
494 intensity rainfall events. In the study by Peterson et al. (2002) a step multiple  
495 regression analysis technique was used. The authors state that base flow conditions  
496 had an influence of 74 % of the nitrate concentrations at the karst spring and storm  
497 events made up to 26 %. Even if higher nitrate concentrations in soil cores can be  
498 directly related to fertilisation, during storm events surface runoff is dominating in  
499 well-developed karst systems. Thus, recharging water contains mainly surface derived  
500 nitrate and the impact of soil nitrate is only minor (Peterson et al., 2002). Zhijun et al.  
501 (2010) related a higher increase in nitrate concentrations in groundwater to rapid  
502 transportation after storm events combined with previous intensive N fertilisation in  
503 the catchment.

504

505 Ribolzi et al. (2000) monitored nitrate concentrations in a spring in a Mediterranean  
506 catchment and observed the predominance of either dilution or mobilisation during  
507 different rainfall events. Their results are similar to the results of the study by Stueber  
508 and Criss (2005) which were reanalysed in this study (Fig. 6d). They observed that  
509 mobilisation of nitrate concentrations occurred only after heavy N fertilisation  
510 coinciding with increased rainfall intensity of 107 mm during a four-week period.  
511 From this it follows that the different nitrate behaviour at the spring depends on source  
512 combination of land use and hydrological conditions. Similarly, Ribolzi et al. (2000)  
513 stated that dilution during one event was to the result of mixing of rainwater  
514 containing low nitrate concentrations and groundwater, whereas mobilisation during  
515 another event occurred due to mixing of two different groundwater types while water  
516 levels increased. This is similar to the interpretations of Toran and White (2005), who  
517 suggest that nitrate changes can depend on changing recharge pathways in karst  
518 environments.

519

520 Denitrification potential can vary in space and time in karst aquifers (Heffernan et al.,  
521 2011). Musgrove et al. (2014), for example, studied two hydrogeologically differing  
522 karst aquifers regarding their denitrification potential: the oxic Edward aquifer and the  
523 anoxic Upper Floridan aquifer in Florida (US). They concluded that, despite the  
524 differences in hydrogeology and in oxic/anoxic conditions, nitrate concentrations of  
525 spring water were strongly influenced by fast conduit-driven flow. These observations  
526 are in line with the conceptual model of the present study, where nitrate responses to  
527 storm events at karst springs are mainly influenced by rapid flow in the conduit  
528 system, and denitrification in the diffuse flow system (rock matrix) may influence  
529 nitrate characteristics of the spring (only) during base flow conditions significantly.  
530 Also Panno et al. (2001) observed a significant degree of denitrification in karst  
531 springs on the western margin of the Illinois Basin (Illinois, US). These authors  
532 reported a high density of sinkholes which caused rapid influx of agrichemicals to the  
533 springs, accounting for highest nitrate concentrations (Panno, 1996). These  
534 observations also justify the conceptual model of the present study, which is based on  
535 the assumption that the diffuse flow system transfers average nitrate concentrations  
536 and may account for long-term trends, while rapid bypass of lower or higher nitrate

537 concentrations after storm events via karst conduits accounts for (mobilized or  
538 diluted) peak concentrations at the spring. Nevertheless, water that flows through the  
539 karst matrix with longer travel time is likely to be affected by denitrification and  
540 redox processes (Einsiedl et al., 2005; Liao et al., 2012; White, 2002). One should  
541 therefore bear in mind that such processes can also contribute to variable nitrate  
542 concentrations at karst springs.

543

544 In the conceptual model (Fig. 4), precipitation is conceptualized as a low N source.  
545 However, precipitation can also be enriched with atmospheric derived nitrate  
546 (Einsiedl and Mayer, 2006). Sebestyen et al. (2008) showed for a catchment in an  
547 upland forest in northeast Vermont, USA, that atmospheric derived nitrate can  
548 account for more than 50% of nitrate concentrations in groundwater, especially during  
549 snowmelt. In the same catchment, Campbell et al. (2004) estimated the average total  
550 N input from atmospheric derived nitrate to be  $13.2 \text{ kg ha}^{-1} \text{ a}^{-1}$ , which can be  
551 significant in such a catchment where atmospheric nitrogen is the most influencing  
552 nitrate source. However, this N-input is relatively low compared to an intensively  
553 operated agricultural area. In Ireland, for example, the Nitrates Directive (EC, 1991)  
554 allows cattle stocking rates with a nitrate input of  $170 \text{ kg ha}^{-1} \text{ a}^{-1}$  or  $250 \text{ kg ha}^{-1} \text{ a}^{-1}$  on  
555 derogation farms.

556

557 Several authors discussed the link between land use practices, hydrological conditions  
558 and N availability (Andrade and Stigter, 2009; Badruzzaman et al., 2012; Kaçaroğlu,  
559 1999). Although nitrate is often not the major form of N application to agricultural  
560 land, it is usually the major form observed in recharge (Böhlke, 2002). In addition, in  
561 agricultural dominated areas not only the total amount of N application is relevant.  
562 Also different agronomic practices of N application have a consequence on the  
563 likelihood and amount of N leaching (Liu et al., 2013; Oenema et al., 2012). For  
564 example, the type of N applied has an influence on the leaching behaviour throughout  
565 the year. Inorganic N fertilisers are on the one hand immediately available for the  
566 plant, but on the other hand highly susceptible to leaching, whereas organic N  
567 fertiliser provide a more constant source of nitrate for the plant on a long term basis

568 due to mineralisation processes (Whitehead, 1995). Best nutrient management  
569 practices are contributing to an increased N use efficiency which directly implies  
570 reduced nitrate loss from surface to groundwater (Rahman et al., 2011; Buckley and  
571 Carney, 2013; Oenema et al., 2005). Huebsch et al. (2013) used multiple linear  
572 regression to explore the impact of agronomic practices on nitrate concentrations in  
573 karst groundwater on a similar site and concluded that improvements in management,  
574 such as timing of slurry application, reductions in inorganic fertiliser usage or the  
575 change from ploughing to minimum cultivation reseeded, contributed to reduced  
576 nitrate concentrations in groundwater.

577

578 **In addition to mobilisation and dilution processes, seasonal variations need to be**  
579 **addressed.** Mineralisation of organic N can also lead to a different leaching behaviour  
580 throughout the year. For example, Mudarra et al. (2012) linked increased mobilisation  
581 of nitrate at the Sierra del Rey-Los Tajos carbonate aquifer in autumn with increased  
582 soil microbial activities, which are directly related to decreased evaporation and  
583 increased soil moisture. In contrast, Panno and Kelly (2004) recorded a seasonal trend  
584 with greatest nitrate concentrations during late spring and summer and lowest during  
585 late fall and winter. Interestingly, Arheimer and Lidén (2000) monitored riverine  
586 inorganic and organic N concentrations from agricultural catchments and showed that  
587 inorganic N concentrations were lower during summer and higher during autumn,  
588 whereas organic N was higher in summer than during the rest of the year.

589

590 Similarly, Bende-Michel et al. (2013) linked riverine nitrate response with agricultural  
591 source availability throughout the year (e.g. time of inorganic and organic N  
592 fertilisation; nitrate build-up from organic matter in summer after organic N fertiliser  
593 application) and with hydrologic mobilisation due to a change from low to high flow  
594 conditions. They assumed that higher peaks of nutrient concentration response should  
595 occur (1) during spring after inorganic fertiliser application, (2) during autumn  
596 because of increased mineralisation and nitrification processes of organic matter in  
597 summer and eventually (3) during winter due to possible expansion of the source area  
598 during high flow conditions. In addition, Rowden (2001) showed that larger losses of

599 applied N occurred during wetter years (concentrations and loads). Rainfall intensity  
600 and duration is influencing soil moisture. Wet conditions coupled with high nitrate  
601 availability in soil due to accumulation intensify leaching from the soil and in the  
602 unsaturated zone (Di and Cameron, 2002; Stark and Richards, 2008). In the present  
603 study site, the highest peaks of mobilised nitrate concentrations occurred in November  
604 2011 and between June and September of 2012. Seasonal variations are driven by  
605 recharge and N availability at the surface. During the summer period, on the one hand,  
606 intensive recharge may transport lower nitrate concentrations if there is a lot of plant  
607 growth but on the other hand, it also may increase transport if there is inorganic N in  
608 the soil after fertilisation application. During autumn reduced crop uptake and  
609 increased recharge due to longer and more intensified rainfall events typically  
610 increases leaching of residual N in soil (Patil et al., 2010).

611

612 Because of rapidly changing concentrations of nitrate and other chemical or microbial  
613 contaminants in karst systems, traditional sampling strategies with sampling intervals  
614 of weeks to months are inadequate to assess water quality in such systems. This is  
615 especially of interest in context of the EU Water Framework Directive, which requires  
616 improving the quality of critical water bodies affected by high nitrate from  
617 groundwater, such as estuaries and coastal waters. In addition, high-resolution  
618 monitoring offers the possibility to detect predominance of mobilisation that can lead  
619 to sudden nitrate peaks above the MAC. Hence, if karst groundwater is used as  
620 drinking water this technique can help to prevent serious threat to humans and animals  
621 such as toxicity in livestock (Di and Cameron, 2002) or methemoglobinemia in  
622 infants also known as the 'blue baby syndrome' which can progress rapidly to cause  
623 coma and death (Knobeloch et al., 2000). An intensification of high-resolution  
624 monitoring in the future is therefore essential to assure good water quality of karst  
625 groundwater and water bodies highly affected by karst groundwater.

626

## 627 **5 Conclusions**

628 The proposed conceptual model of nitrate response in karst systems is able to explain  
629 various nitrate response scenarios, the nitrate patterns at the spring of the current study

630 and the findings from other studies. In the current study, four possible nitrate response  
631 scenarios in karst aquifers to storm events were hypothesized. Scenario 1 relates to  
632 mobilised nitrate concentrations, Scenario 2 diluted nitrate concentrations, Scenario 3  
633 a combination of mobilised and diluted nitrate concentrations during one event and  
634 Scenario 4 mobilised and diluted nitrate concentrations during multiple events. The  
635 proposed conceptual model of nitrate in karst systems elucidates the relation of nitrate  
636 responses at karst springs with driving factors such as hydrological conditions, N  
637 availability through land use and karst features. Predominance of mobilisation or  
638 dilution and therefore rapid rise or decline of nitrate concentrations during storm  
639 events depend highly on the availability of nitrate accumulated in soil and unsaturated  
640 zone. A well-developed karst system as well as wet conditions are crucial for rapid  
641 transport and have an influence on the intensity and time lag of nitrate concentration  
642 changes. Differences regarding predominance of dilution or mobilisation processes  
643 during different storm events on the same study site occur if 1) the source of N at the  
644 surface changes over time and/or 2) the activation of different flow paths causes  
645 mixing of water sources containing more or less nitrate than the average nitrate  
646 concentration in groundwater at the study site. The presented conceptual model of  
647 nitrate responses in karst systems contributes to a more comprehensive understanding  
648 of nitrate occurrences in the environment and therefore also facilitates an improved  
649 implementation of the EU Water Framework Directive in environmental activities,  
650 planning and policy. Finally, the study also highlighted the important role of  
651 continuous and long-term nitrate monitoring in karst systems.

652

### 653 **Acknowledgements**

654 The authors would like to acknowledge the Teagasc Walsh Fellowship scheme for  
655 funding this study. Furthermore, we gratefully acknowledge Tristan G. Ibrahim from  
656 Teagasc, the farm staff at Dairygold Farm for supporting the field work and in  
657 particular the farm manager Steven Fitzgerald as well as the staff of Johnstown Castle  
658 water laboratories for analysing the groundwater samples. In addition, we would like  
659 to thank the Irish Meteorological Service Met Éireann for providing meteorological  
660 data.

661

## 662 **References**

663

## 664 **Captions for Figures**

665 **Fig. 1:** Site map for the study area. The smaller arrows indicate the water flow  
666 direction of the continuous spring in a ditch to the river.

667 **Fig. 2:** Observations at the study site in period (1) between the 13<sup>th</sup> of November 2011  
668 and the 20<sup>th</sup> of January 2012. The symbols 1 to 4 indicate different storm events,  
669 which had a visible influence on the discharge and nitrate pattern at the spring.

670 **Fig. 3:** Observations at the study site in period (2) between the 1<sup>st</sup> of February and the  
671 1<sup>st</sup> of October 2012: a) precipitation; b) to e) groundwater fluctuation at BH1, BH3,  
672 BH4 and BH6 in [m] above minimum; f) NO<sub>3</sub>-N pattern at the spring.

673 **Fig. 4:** Conceptual model of nitrate response in karst systems.

674 **Fig. 5:** Hypothesis of nitrate response scenarios: Predominance of a) mobilised  
675 nitrate; b) diluted nitrate; c) mobilisation and dilution during one event; d)  
676 mobilisation and dilution during multiple rainfall events.

677 **Fig. 6:** Illustrating 4 case studies: Predominance of a) mobilised nitrate; b) diluted  
678 nitrate; c) mobilisation and dilution during one single event; d) mobilisation and  
679 dilution during multiple rainfall events. The grey bar in the upper diagram shows the  
680 only event in the dataset where mobilisation occurred instead of dilution during storm  
681 events.

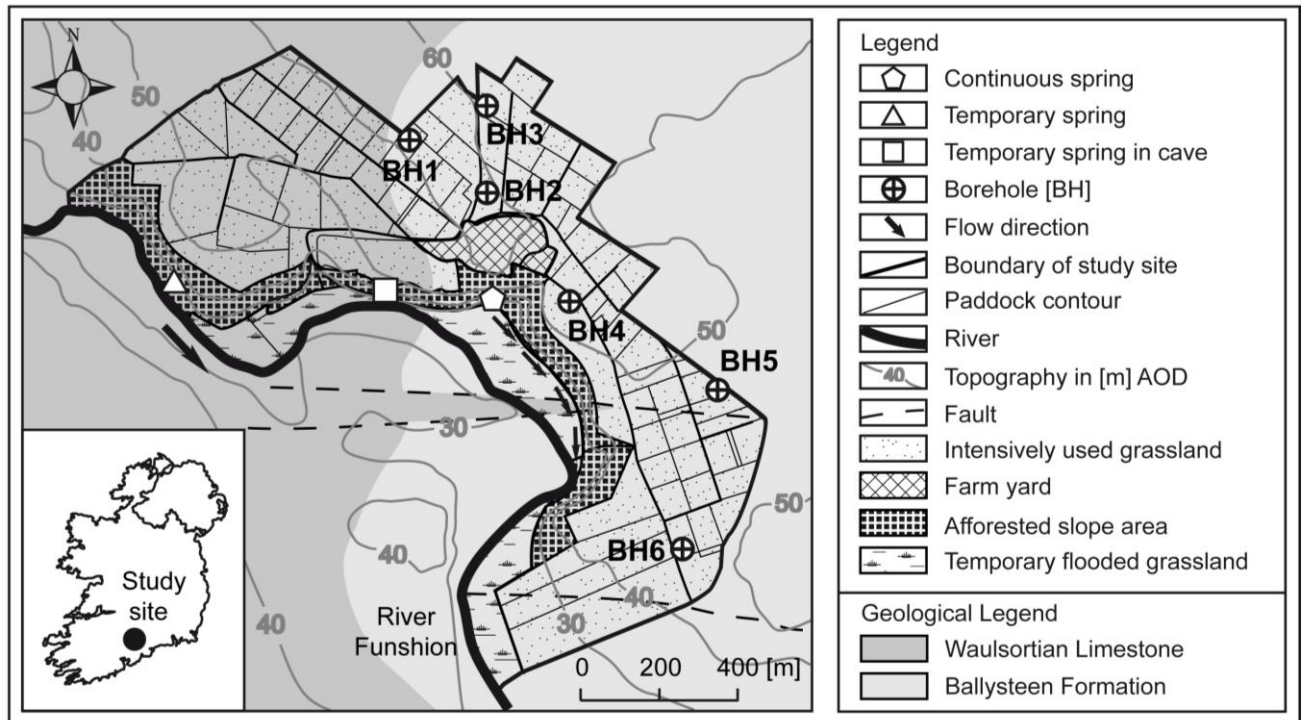
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685 **Figures**

686



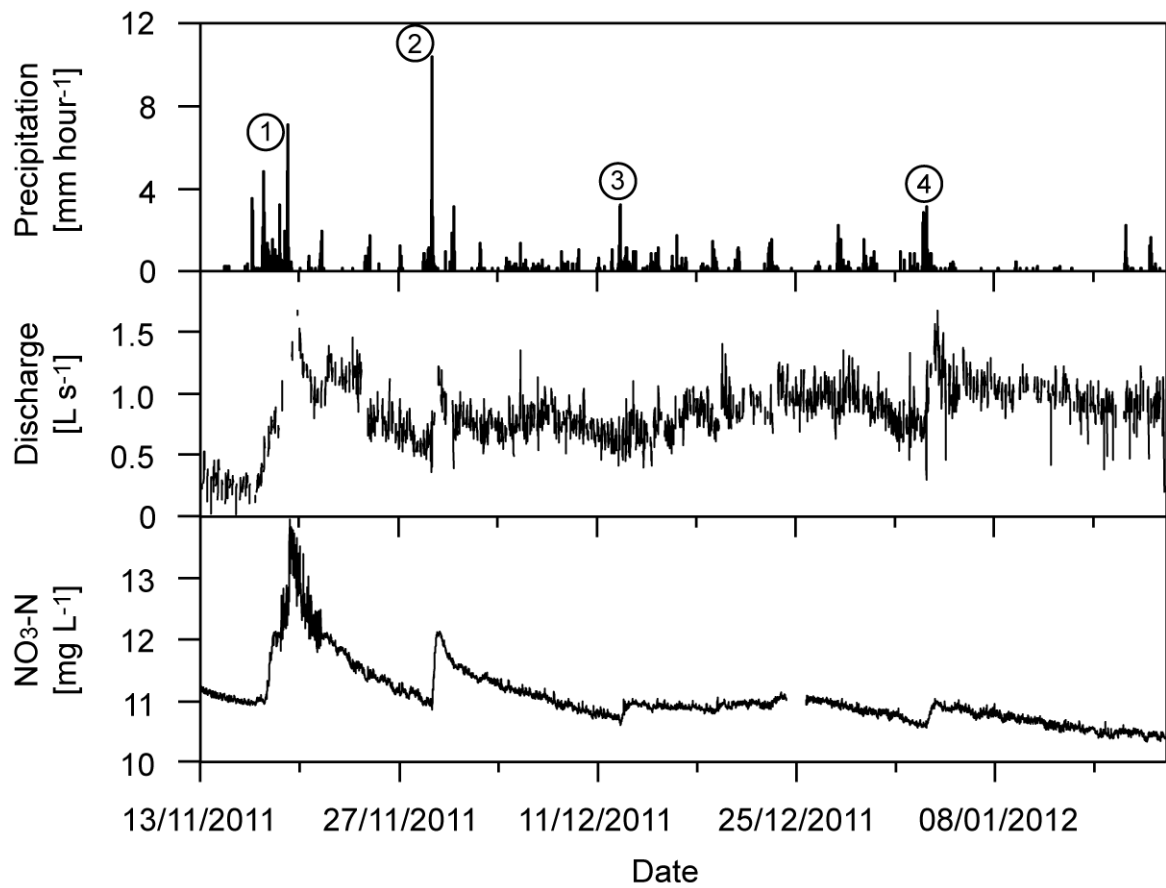
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689 **Fig. 1:** Site map for the study area in the Republic of Ireland. The smaller arrows  
690 indicate the water flow direction of the continuous spring in a ditch to the river.

691



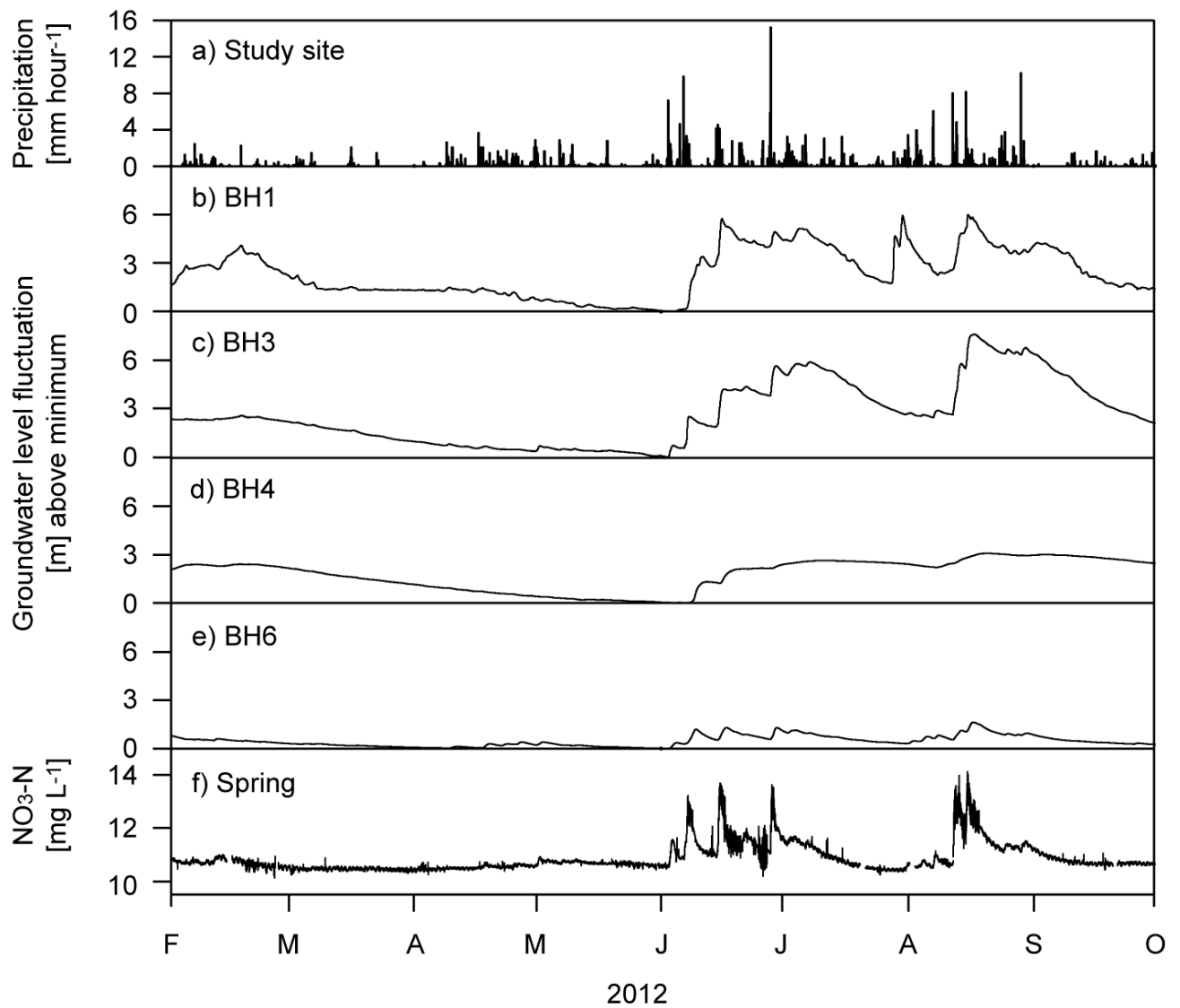


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693

694 **Fig. 2:** Observations at the study site in period (1) between the 13<sup>th</sup> of November 2011  
 695 and the 20<sup>th</sup> of January 2012. The symbols 1 to 4 indicate different storm events,  
 696 which had a visible influence on the discharge and nitrate pattern at the spring.

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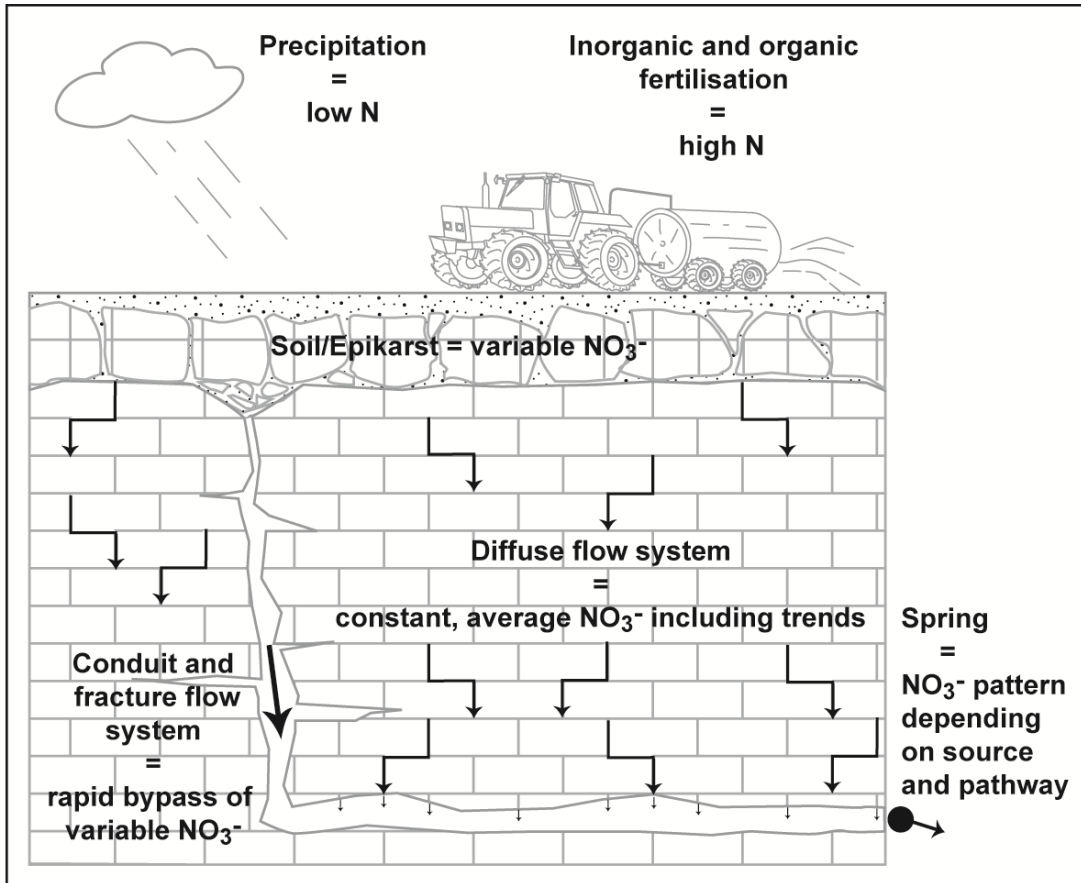


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700 **Fig. 3:** Observations at the study site in period (2) between the 1<sup>st</sup> of February and the  
 701 1<sup>st</sup> of October 2012: a) precipitation; b) to e) groundwater fluctuation at BH1, BH3,  
 702 BH4 and BH6 in [m] above minimum; f) NO<sub>3</sub>-N pattern at the spring.

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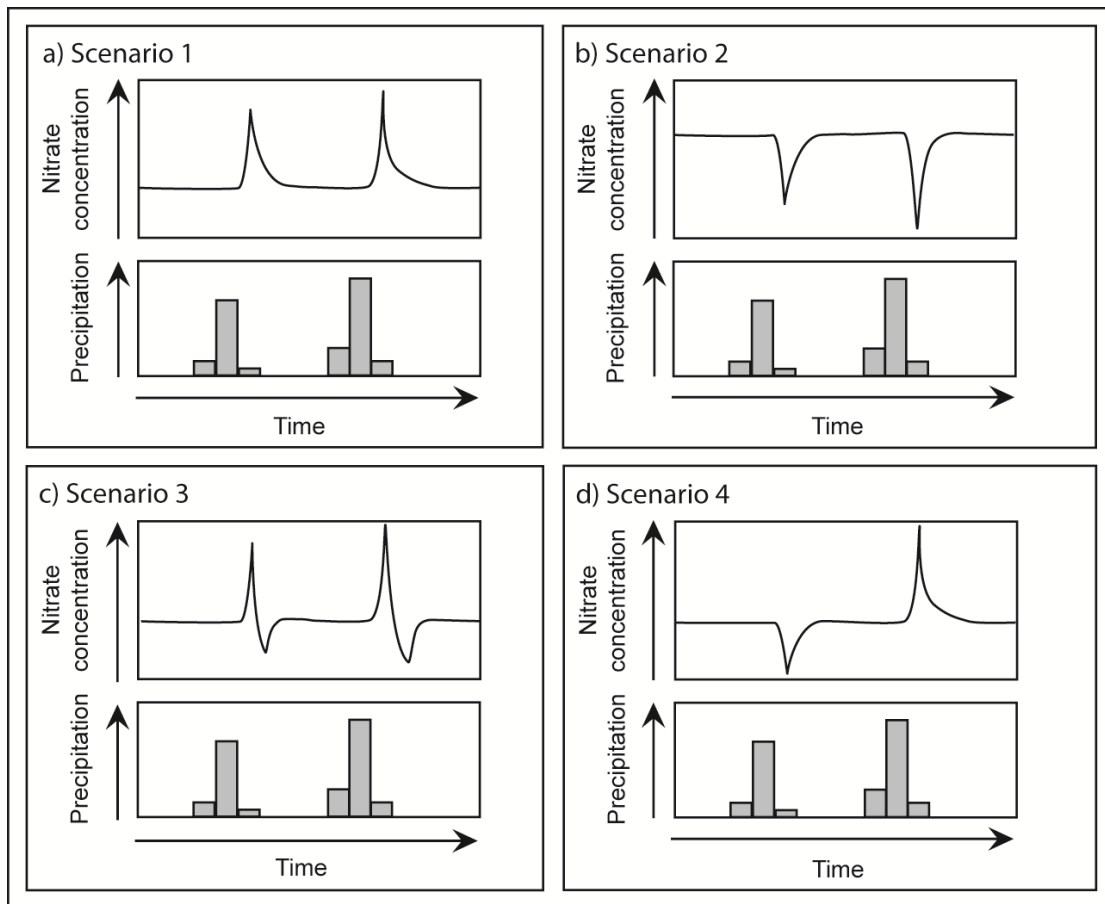


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706 **Fig. 4:** Conceptual model of nitrate response in karst systems.

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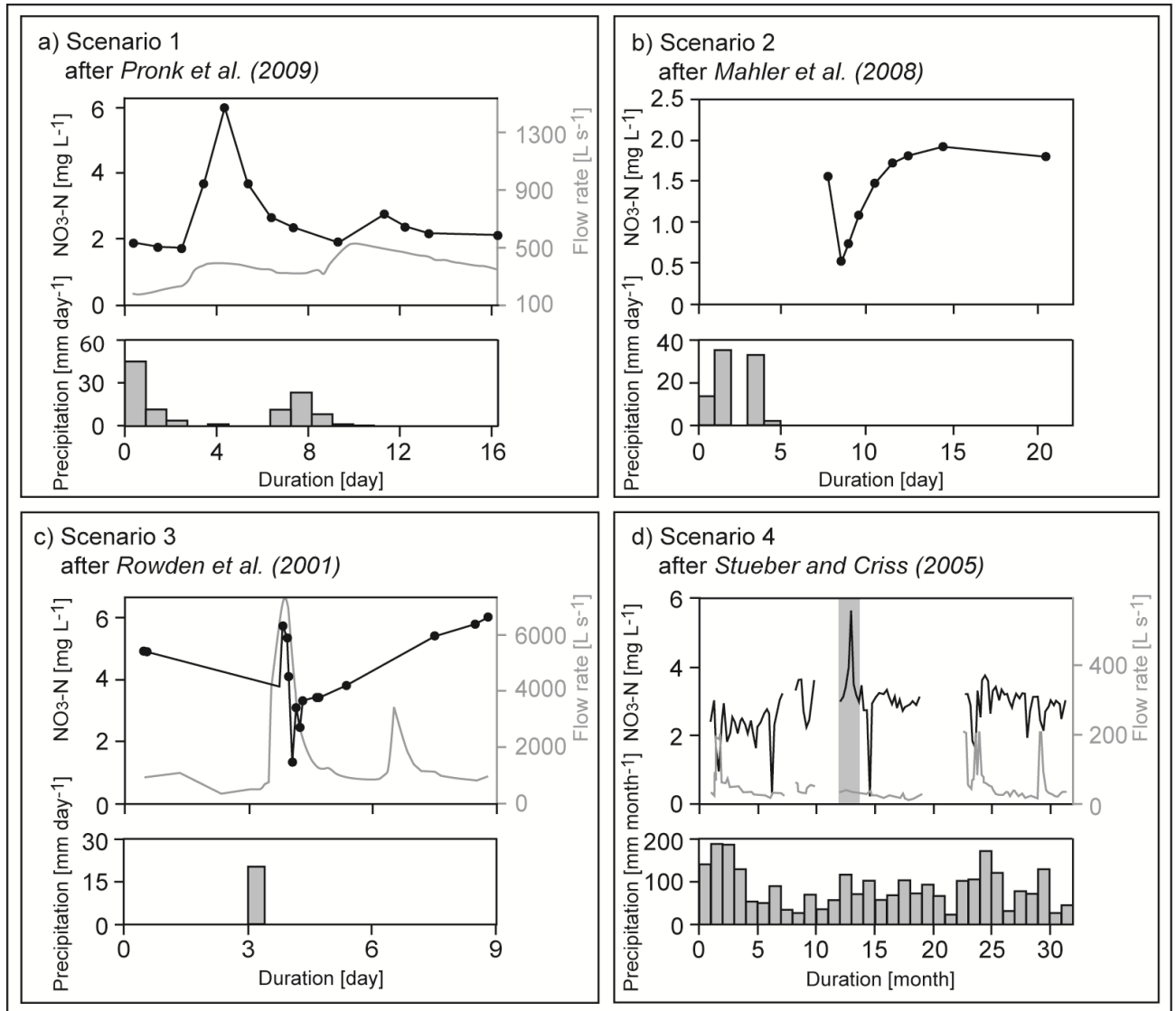


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709

710 **Fig. 5:** Hypothesis of nitrate response scenarios: Predominance of a) nitrate  
 711 mobilisation; b) nitrate dilution; c) mobilisation and dilution during one event; d)  
 712 mobilisation and dilution during multiple rainfall events.

713



714

715 **Fig. 6:** Four illustrating case studies: Predominance of a) nitrate mobilisation; b)  
 716 nitrate dilution; c) mobilisation and dilution during one single event; d) mobilisation  
 717 and dilution during multiple rainfall events (the grey bar in the upper diagram  
 718 indicates the event with nitrate mobilisation).

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